Noname manuscript No. (will be inserted by the editor)

- Mechanical Properties of Cellulose Nanofibril Films:
- ² Effects of Crystallinity and its Modification by
- ³ Treatment with Liquid Anhydrous Ammonia
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8 Received: date / Accepted: date

Abstract The influence of cellulose crystallinity on mechanical properties of 9 cellulose nano-fibrils (CNF) was investigated. Degree of crystallinity (DoC) 10 was modified using liquid anhydrous ammonia. Such treatment changes crys-11 tal allomorph from cellulose I to cellulose III, a change which was reversed by 12 subsequent boiling in water. DoC was measured using solid state nuclear mag-13 netic resonance (NMR). Crystalline index (CI) was also measured using wide 14 angle x-ray scattering (WAXS). Cotton linters were used as the raw material. 15 The cotton linter was ammonia treated prior to fibrillation. Reduced DoC is 16 seen to associate with an increased yield point and decreased Young modu-17 lus. Young modulus is here defined as the maximal slope of the stress-strain 18 curves. The association between DoC and Young modulus or DoC and yield 19 point are both statistically significant. We cannot conclude there has been an 20 effect on strainability. While mechanical properties were affected, we found no 21 indication that ammonia treatment affected degree of fibrillation. CNF was 22 also studied in air and liquid using atomic force microscopy (AFM). Swelling 23

 $_{24}$ of the nanofibers was observed, with a mean diameter increase of 48.9%.

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25 1 Introduction

²⁶ Society is increasingly valuing "green" materials and products. Non-renewable

²⁷ materials like non-biodegradable plastics and plastics from fossil feedstocks

 $_{\rm 28}$ like oil are under increasing pressure. Biodegradable, renewable alternatives

²⁹ are actively sought by consumers and policy makers alike - especially for short

³⁰ service life applications such as packaging and disposables used for food prepa-

³¹ ration and consumption (Brodin et al. 2017).

For packaging applications, cellulose nano-fibrils (CNF) may present one 32 such green alternative. Films from CNF may be transparent (Nogi et al. 2009), 33 have excellent gas barrier properties (Syverud and Stenius 2008; Aulin et al. 34 2010; Fukuzumi et al. 2009), and high tensile strength (Henriksson et al. 2008). 35 One of the properties not conducive to the adoption of this "green" material 36 is the limited ductility of CNF based films as compared to currently employed 37 plastic materials. CNF films are typically brittle - they are stiff and exhibit 38 low extensibility as compared to many plastic alternatives (Rodionova et al. 39 2012). This brittleness prevents folding and shaping of the CNF based product, 40 limiting applicability and appeal. 41

Controlling the stiffness, extensibility of CNF could open new markets for 42 the material and allow for greater flexibility in terms of applications. Various 43 strategies have previously been employed. Examples include addition of plasti-44 cizers (Myllytie et al. 2010; Minelli et al. 2010; Kumar et al. 2016), embedding 45 nanofibrils in a more ductile matrix material (Nishino et al. 2004), or chemical 46 or structural modification of the nanofibril surface (P. A. Larsson, Berglund, 47 et al. 2014; Codou et al. 2015; P. A. Larsson and Wågberg 2016). Both the 48 use of all-cellulose composites (Nishino et al. 2004), and amorphization of fib-49 ril surfaces (P. A. Larsson, Berglund, et al. 2014; P. A. Larsson and Wågberg 50 2016) make use of the link between flexibility and crystallinity of a material. 51 In the current work, we used submersion in anhydrous ammonia to reduce 52 the general crystallinity of the cellulose, while minimally affecting the degree 53 of polymerization or the surface chemistry. Ammonia treatment is associated 54 with strong reduction in crystallinity and a change in crystal allomorph. In 55 order to preserve the crystal allomorph, the ammonia submersion was followed 56 by a hydrothermal treatment, reverting it to cellulose I. Cotton linters were 57

⁵⁸ used for their high purity and high degree of crystallinity.

⁵⁹ 2 Materials and Methods

The current work describes CNF 60 prepared by homogenization of 61 cotton linters. Samples are re-62 ferred to as CNF-n where n is a 63 number indicating the number of 64 ammonia treatments the sample 65 has undergone, either 0, 1 or 3. 66 Most numerical and statistical 67 analysis was performed using the 68 statistical software "R" v. 3.5.0 69 (R Core Team 2015). Software de-70 veloped at Innventia AB, Sweden 71 was used for NMR data, as indi-72 cated. Plots were generated using 73

 $_{74}$ ggplot2 v. 3.0.0 (Ginestet 2011) or

 $_{75}$ TikZ (Figures 12, 5, 8, 1).

76 2.1 Ammonia Treatment

Liquid anhydrous ammo-77 nia at atmospheric pressure 78 was used in a closed sys-79 tem (Figure 1) to treat the 80 cotton linters. The ammo-81 nia (R717, AGA) was kept 82 liquid by use of a bath 83 of dry-ice in acetone. Af-84 ter 8 hours submerged in 85 ammonia, the reaction ves-86 sel was opened and removed 87 from the coolant bath. The 88 opened reaction vessel was 89 allowed to reach room tem-90 perature, and the ammo-91 nia was allowed to evapo-92 rate overnight. The follow-93 ing day, the treated lin-94 ters were submerged in de-95 ionized water and boiled for 96 6 hours. This boiling was 97 performed to revert the al-98 lomorph from cellulose III 99 to cellulose I (Figure 2). 100



Fig. 1 Ammonia treatment. Material to be treated was put in a closed reaction vessel, which was placed in a coolant bath containing acetone and dry ice (\bigcirc) . A cold finger with same was used to condense gaseous ammonia. Once set up anhydrous ammonia (\bigcirc) was introduced through a hose (a) until the linters were submerged. The setup is sealed, allowing one exit for escaping gas (b).

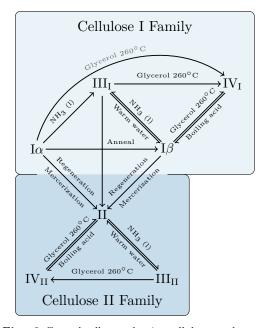


Fig. 2 Crystal allomorphs in cellulose and common pathways for conversion between these. (Kroon-Batenburg et al. 1996; Park et al. 2003; Perez and Mazeau 2005; Zugenmaier 2008; Dufresne 2012; Lavoine et al. 2012).

Once boiled, the linters were dried overnight at 100°C before further treatment. In the case where multiple treatments were conducted (CNF-3), these steps were performed once per treatment. Control samples with no ammonia treatment (CNF-0) were still boiled in DI-water and dried overnight as described.

106 2.2 Fibrillation

Once any pre-treatment was conducted, fibrillation was performed. Fibrillation 107 was a three-step process; first the linters were beaten in a PFI-mill for 10,000 108 revolutions at 10% solids. After beating, the linters were ground by running 109 them through a Masuko brand supermasscolloider 10 times at approximately 110 1% solids. After grinding, they were run through a Rannie 15 type 12.56x 111 homogenizer 5 times at slightly below 1% solids. The first pass through the 112 homogenizer was run at 600 bar pressure drop. All subsequent passes were run 113 at 1 kbar pressure drop. 114

¹¹⁵ 2.3 Characterization of CNF

Fibrillated material was run through a L&W FiberTester Plus+, assessing the 116 degree of fibrillation by optical means. 5 beakers containing 1 g dry matter sus-117 pended in 300 mL DI-water were analyzed for each of the three assessed treat-118 ments (CNF-0, CNF-1 and CNF-3). FiberTester Plus+ results were compared 119 with laser profilometry investigation of finished films. Fibrils were further in-120 vestigated by atomic force microscopy (AFM). AFM samples were prepared 121 by drop casting of 0.02 wt% CNF suspension, on freshly cleaved mica discs 122 (F7013 Agar Scientific) and dried by heating on a 65°C hotplate. Fibril di-123 ameter was ascertained by measuring the fibril height above the mica surface, 124 avoiding overestimation due to tip-sample convolution, expected to be large 125 due to the tip radius of the used tips, Bruker ScanAsyst Fluid, which have 126 a nominal tip radius of 20 nm and a maximal radius 60 nm. Areas chosen 127 for diameter estimation were chosen if judged to be a sufficient distance from 128 points where fibrils meet, cross or split/merge. This was done to maximize the 129 likelihood that the measured fibrils are in close contact with the underlying 130 mica, as opposed to suspended above it. A Veeco Multimode V AFM was 131 used in Veeco's software assisted tapping mode named ScanAsyst. Oscillation 132 parameters were controlled by the software. Tips used were ScanAsyst Fluid, 133 whether the image was taken in air or liquid. Samples were first imaged in 134 air, the tip was withdrawn and deionized water was added. After 30 minutes 135 of submersion, the tip was engaged again, and a new image was recorded. 136 AFM micrograph analysis was performed using Gwyddion v. 2.42 (Nečas and 137 Klapetek 2012). AFM micrographs were flattened/leveled using median of dif-138 ferences row alignment, and profiles used for fibril diameter measurements were 139 created using linear interpolation and 10 pixel wide measurement lines span-140 ning one fibril each. Images used for measurements were 1024 by 1024 pixel 141

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large, with 11.7 nm/pixel within the plane of the micrograph. The instrument manufacturer (Veeco) reports a z-noise below 30 pm RMS for the diMultimode V AFM used in the current work. The instrument is located in a cleanroom environment (ISO 7), where it is placed on a vibration dampened table positioned on a reinforced concrete plinth, which is separate from the main floor of the lab to reduce vibrations. Scales in z-direction are indicated by look-uptable (LUT) for each recorded image. For shown visual representations, a 256 value (0-255) gray-scale LUT is applied.

150 2.3.1 NMR and WAXS - Structural Changes and Crystallinity Measurements

¹⁵¹ Crystallinity was assessed using two techniques which will be described in-¹⁵² dividually below. The used techniques were Cross-Polarization Magic Angle ¹⁵³ Spinning Carbon-13 Nuclear Magnetic Resonance (CP/MAS ¹³C-NMR) - here ¹⁵⁴ referred to as NMR - and Wide Angle X-Ray Scattering (WAXS). NMR re-

¹⁵⁵ sults were also used to estimate lateral fibril dimension (LFD), lateral fibril ¹⁵⁶ aggregate dimension (LFAD) and LFAD specific surface area (LFAD SSA).

CP/MAS ¹³C-NMR spectra were recorded in a Bruker Avance III AQS 400 157 SB instrument operating at 9.4 T. All samples were packed uniformly in a 158 zirconium oxide rotor. All measurements were carried out at 295 (± 1) K with 159 a magic angle spinning (MAS) rate of 10 kHz. A 4 mm double air-bearing 160 probe was used. Data acquisition was performed using a cross-polarization 161 (CP) pulse sequence, i.e., a 2.95 microseconds proton 90-degree pulse and an 162 800 microseconds ramped (100-50%) falling contact pulse, with a 2.5 s delay 163 between repetitions. A SPINAL64 pulse sequence was used for 1H decoupling. 164 The Hartmann-Hahn matching procedure was based on glycine. The chemical 165 shift scale was calibrated to the TMS-scale (tetramethylsilane, $(CH_3)_4Si$) by 166 assigning the data point of maximum intensity in the alpha-glycine carbonyl 167 signal to a shift of 176.03 ppm. 4096 transients were recorded on each sample 168 leading to an acquisition time of about 3 h. The software for spectral fitting was 169 developed at Innventia AB and is based on a Levenberg-Marquardt algorithm 170 (P. T. Larsson et al. 1997). All computations were based on integrated signal 171 intensities obtained from spectral fitting (Wickholm et al. 1998). The errors 172 given for parameters obtained from the fitting procedure are the standard 173 error of the mean with respect to the quality of the fit. 174

When used to estimate LFD, LFAD and LFAD SSA, the calculations were performed as described in Nocanda et al. 2007. Briefly, Nocanda et al. uses known peak intensities from the signal as a basis for the estimate. The estimate, which relies on further processing, can be expressed thus;

$$q_{LFD} = \frac{I_A + I_I}{\sum I_x} \tag{1}$$

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$$q_{LFAD} = \frac{I_A}{\sum I_x},\tag{2}$$

subscripts for q denote which metric it can be used to estimate. I_x is a given 181 peak intensity for the allomorphs present in the sample, which is assumed to 182 be cellulose I_{α} and I_{β} as well as a peak for paracrystalline signal and sur-183 face signals for accessible (I_A) and inaccessible (I_I) surfaces. The resulting 184 qs can be used to estimate the dimensions as distances given a known fibril 185 cross-sectional geometry and known glucan dimensions (Nocanda et al. 2007; 186 Wickholm et al. 1998). Nocanda et al. assumed a square cross-section, which 187 means $q = \frac{4n-4}{n^2}$ where n is the number of glucan chains along one side of said 188 cross-section. Knowing the dimensions of a glucan chain cross-section LFD 189 and LFAD can be estimated by solving for n. Equations from Nocanda et al. 190 2007, where the topic is further discussed. 191

WAXS measurements were performed on an Anton Paar SAXSpoint 2.0 sys-192 tem (Anton Paar, Graz, Austria) equipped with a Microsource x-ray source 193 (Cu Ka radiation, wavelength 0.15418 nm) and a Dectris 2D CMOS Eiger R 194 1M detector with 75 µm by 75 µm pixel size. All measurements were performed 195 with a beam size of approximately 500 µm diameter, at a sample stage tem-196 perature between 25° C to 29° C (no temperature control was employed) with 197 a beam path pressure at about 1-2 mBar. The sample to detector distance 198 (SDD) was 111 mm. All samples were mounted on a Solid Sampler (Anton 199 Paar, Graz, Austria) mounted on a VarioStage (Anton Paar, Graz, Austria). 200 The samples were exposed to vacuum during measurement. For each sample 5 201 frames each of 17 minutes duration were read from the detector, giving a total 202 measurement time of 1.4 hour per sample. For all samples the relative trans-203 mission was determined and used for scaling of the scattering intensities. The 204 software used for instrument control was SAXSdrive version 2.01.224 (Anton 205 Paar, Graz, Austria), and post-acquisition data processing was performed us-206 ing the software SAXSanalysis version 3.00.042 (Anton Paar, Graz, Austria). 207 The crystalline index (CI) was estimated from WAXS, in transmission mode, 208 using the Segal method (Segal et al. 1959), where 209

$$CI = \frac{I_{200} - I_{am}}{I_{200}} \times 100.$$
(3)

Here I_{200} is the intensity of the (200) peak and I_{am} is the amorphous signal at $2\theta = 18^{\circ}$. The (200) peak was identified using data from untreated samples (CNF-0).

213 2.4 Production of Films

²¹⁴ CNF films were made by solvent casting in plastic petri dishes. CNF suspension
²¹⁵ with a concentration of 0.85 wt% was poured into the petri dishes and allowed
²¹⁶ to dry in room temperature. Target grammage was 22 g/m². Once dried, the
²¹⁷ films were conditioned according to ISO 187:1900 prior to characterization.

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218 2.5 Characterization of Films

Films were tensile tested using a Zwick/Roell at 23°C, 50% RH. Specimens 219 were 15 mm wide, the clamp-to-clamp gap was 20 mm. 40 tensile tests were 220 run per sample group (0, 1 or 3 treatments). Tensile indexes of the CNF films 221 were calculated from the average basis weight and average thicknesses of the 222 films. Film thickness was measured by micrometer. The thickness of a given 223 film is assumed to be the average of five measurements on said film. Young 224 modulus was estimated using the point of maximum slope. Yield point was 225 found using an offset of 0.2%. Films were imaged using a Nikon D80 with 226 an HB-47 lens. Electron microscopy (SEM) was employed for high resolution 227 imaging. The electron microscope used was a Hitachi SU4400. Micrographs 228 were recorded using an Everhart-Thornley detector set to collect secondary 229 electrons. Acceleration voltage was set to 5 kV, beam current at 138 µA. All 230 investigated samples were sputter coated with a layer of approximately 12 nm 231 gold prior to SEM analysis. Images were analyzed using FIJI (Schindelin et al. 232 233 2012).

234 2.5.1 Laser Profilometry

From each series, samples of 20 by 10 mm were prepared for laser profilometry (LP). The samples were placed on double-sided tape on microscopy slides and were coated with a layer of gold. 20 laser profilometry (Lehmann, Lehman

²³⁸ Mess-Systeme AG, Baden-Dättwil, Germany) topography images were ac-²³⁹ quired from each sample. The size of the local areas was 1 mm by 1 mm,

²⁴⁰ having a lateral and z-resolution of 1 µm and 10 nm, respectively.

241 3 Results and Discussion

The current paper discusses the effects of reduced CNF DoC on mechanical properties of CNF films. Treatment by liquid anhydrous ammonia was chosen as the means to reduce DoC because ammonia treatment is known to reduce the DoC of the treated cellulose while maintaining the chemistry and minimally affecting the degree of polymerization (Hess and Trogus 1935; Barry et al. 1936; Rousselle et al. 1976; Saapan et al. 1984; Mittal et al. 2011; Sawada et al. 2014).

The reduced DoC following ammonia treatment is a consequence of the 249 swelling of cellulose by ammonia. Ammonia swells cellulose very well, even 250 penetrating into crystalline areas. When cellulose is swollen with ammonia, 251 hydrogen bonds between cellulose chains are broken, and a nitrogen bridge is 252 established in their place (Wada et al. 2006). This results in a conformational 253 change on a molecular scale - a complex is formed where ammonia molecules 254 are placed between individual cellulose chains. Ammonia can be removed from 255 this complex by evaporation or washing. Inter-chain hydrogen bonds will re-256 form once ammonia is gone, but the unit cell is altered from the original 257

cellulose I (or II) to a less dense, metastable allomorph named cellulose III_x where x is either I or II, indicating the allomorph before conversion to cellulose III (Figure 2). The transition from cellulose I to cellulose III_I is, when conducted at atmospheric pressures, also accompanied by a powerful reduction in DoC. How much DoC is affected depends on treatment time (Menachem and Roldan 1971). Reversion to the original allomorph, here cellulose I, is possible by hydrothermal treatment, viz. submersion of cellulose III_I in hot water.

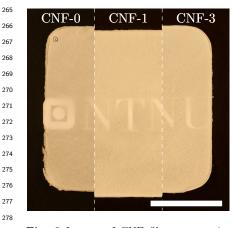


Fig. 3 Images of CNF films, ammoniatreated and controls. Linters received no ammonia treatments (left), one treatment (middle) or three treatments (right). Scale
bar is 5 cm long.

This hydrothermal treatment has been reported to increase the DoC somewhat as a function of temperature and treatment time (Menachem and Roldan 1971; Park et al. 2003). Given these aspects of treatment of cellulose with liquid anhydrous ammonia, such a treatment followed by hydrothermal treatment was chosen to investigate the effects of DoC on CNF film properties, fibril properties and degree of fibrillation after homogenization.

Ammonia treatment swells fibrils and films both on a molecular scale and on a macro-scale, resulting in pronounced shrinkage of cellulose based materials once the ammonia is evaporated (Hermann 1997; Thao Ho et al.

2013). In contrast, no immediately apparent difference between untreated and
treated CNF can be seen (Figure 3). This is to be expected as the cotton linters
were ammonia-treated before fibrillation and film formation. Any macro-scale

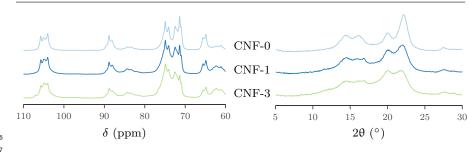
 $_{\tt 287}$ $\,$ shrinkage of the produced CNF films can therefore obviously not occur.

288 3.1 Effects on Crystallinity and Crystal Structure

To assess the change in DoC a wide range of techniques can be employed (Foster et al. 2018; Zugenmaier 2008), two of the experimental techniques that can be used are CP/MAS ¹³C-NMR, in brief referred to as NMR, and WAXS respectively. For both NMR and WAXS analysis of the materials we see a peak broadening and shift in relative peak intensities correlating with ammonia treatments (Figure 4). This change in the recorded NMR spectra and WAXS profiles is consistent with a reduction in crystalline index.

Analysis of data from the two experimental techniques (NMR, WAXS) supports the initial observations and quantitatively describes the reduction in DoC (Figure 5).

While a clear effect of ammonia treatment on DoC is seen, a stronger response may conceivably be possible if shorter - or no - hydrothermal treatment

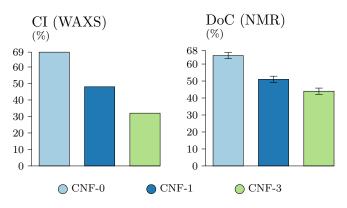


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Fig. 4 ¹³C NMR (left) spectra and WAXS (right) profiles from samples, as annotated. The plots are not normalized. DoC values are shown in Figure 5.

is used. Park et al. 2003 reported higher temperatures or longer submersions
in warm water reverses some of the loss of DoC caused by treatment with
anhydrous ammonia. We also note that successive treatments with ammonia
do have a compounding effect - successive treatments can lower DoC further

than a single one. While a compounding effect is seen, there are diminishing returns - the first treatment has the strongest effect.



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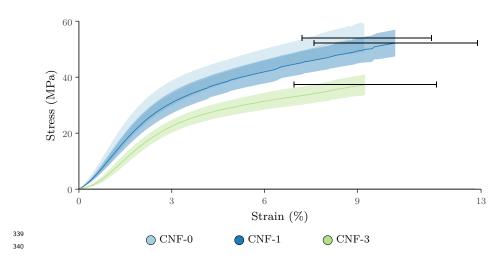
Fig. 5 Crystalline index (CI) from WAXS and degree of crystallinity (DoC) from NMR data. CI was estimated using the Segal method. DoC was calculated using spectral fitting based on Levenberg-Marquardt algorithm. Where error bars are supplied, these indicate standard error of the mean (SE). Cotton linter is nearly pure cellulose, and is assumed to be pure for the purposes of this work.

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317 3.1.1 The presence of cellulose II

From NMR data (Figure 4), in particular a peak at 107 ppm, a small quantity of cellulose II appears present in CNF-1 and CNF-3. The same peak is absent in CNF-0, revealing no cellulose II. The 107 peak, associated with C1 in cellulose II, is fairly well separated from peaks in cellulose I spectra, making

it a good identifier for the allomorph. Most other peaks identifying the allo-322 morph, such as at 105 ppm (C1) 76.5 ppm (C3) and 62.6 and 63 ppm (C6) 323 are present but due to more signal overlap these are less easily detected (Kono 324 and Numata 2004; Zugenmaier 2008). The cellulose II signal is nevertheless 325 small and only resolvable on NMR. In the event that any cellulose II is present 326 in the current experiment, its presence may have resulted from the ammonia 327 swelling step. Cellulose II is commonly a result of dissolution and redispersion 328 of cellulose. When cellulose re-organizes from a dissolved state, the polymeric 329 strands change from the parallel organization in cellulose I to a more thermo-330 dynamically stable, anti-parallel configuration - cellulose II. Cellulose II can 331 also be generated by swelling in concentrated alkali, here due to some cellu-332 lose chains moving from one fibril to another, adjacent fibril during swelling 333 (Okano and Sarko 1985; Dinand et al. 2002). Cellulose II may be formed by 334 a similar process during swelling with ammonia; polymer chains moving from 335 one fibril to another during swelling may be the cause of the observed cellulose 336 II. Further work is necessary to explore this hypothesis. 337



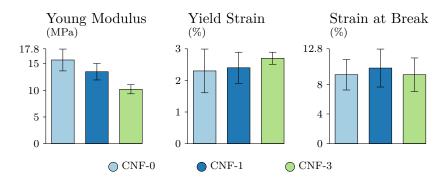
338 3.2 Mechanical Effects of Crystallinity Reduction

Fig. 6 Stress-Strain curves for tested samples. Colors denote number of ammonia treatments, 0, 1 or 3 as indicated. Ribbons show standard deviation, lines show mean values. For each curves n = 40. Lines and ribbons terminate at the mean strain at break, error bars denote standard deviation for strain at break. Strain at break, Young modulus and yield strain are shown in Figure 7. Statistical analysis is shown in Table 1.

As DoC falls, flexibility should increase. In simulations, reduced DoC is shown to weaken fibrils (Youssefian and Rahbar 2015). Increased free volume, which can result from reduction of DoC, has been shown to decrease the Young modulus of the fibrils themselves (Youssefian, Jakes, et al. 2017). As the fibrils

become more flexible, this should be reflected in the properties of produced 350 films. Stress/strain curves (Figure 6) show that there are differing mechanical 351 properties which correlate with DoC. We can note that successive treatments 352 appear to have cumulative effects on the maximum slope of the curves, indi-353 cating a reduced Young modulus. Note that Young modulus is estimated as 354 the point of maximum slope. There is a strong correlation between several 355 mechanical properties for the tested films and DoC. This correlation is con-356 sistent with the hypothesis that nanofibril DoC has an effect on mechanical 357 properties of films made from them. 358

While Young modulus and yield point are significantly affected, we cannot conclude strain at break is affected. There is only a barely statistically significant (p=0.047) effect on strain at break between samples CNF-0 and CNF-1 (Table 1). The effect is slight, and not consistent (Figure 7). Hence samples appear equally stretchable, and we cannot conclude there has been an effect.



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Fig. 7 Plotted values for Young modulus, yield strain and strain at break for tested samples. For each run n = 40. Yield strain is defined at 0.2% strain. Young modulus is approximated using the point of maximum slope. Scope and error bars defined by standard deviation. FiberTester Plus+ recorded between 17 059 and 20 027 fibers per run, five runs were completed for each CNF class. Comparisons between group and statistical analysis in Table 1.

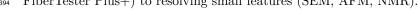
As shown in Figure 7 and accompanying statistical analysis in Table 1, 370 we see there is significant differences between treated and untreated samples. 371 Treated and untreated samples differ on both Young modulus (as approxi-372 mated by the maximal slope of the stress-strain curves) and yield stress (de-373 fined at 0.2%). From graph (Figure 7) and accompanying statistical analysis 374 (Table 1), we can conclude that successive ammonia treatments do increase the 375 strain a film can experience before deformation becomes plastic (yield point), 376 and decrease its stiffness (Young modulus) significantly. These observations 377 correlate with change in DoC. 378

Table 1 Statistical analysis of mechanical properties of ammonia-treated cellulose. Mechan-379 ical differences between CNF from cotton linters treated 0, 1 and 3 times were analyzed. 380 Similarly, untreated CNF films were compared with ammonia-treated CNF films. The statis-381 tical test used was a Mann-Whitney test, given that upon inspection of the data distribution, 382 normality can not be assumed. For each run n = 40. Leftmost column indicates which groups 383 384 were compared.

		Young modulus		Yield point		Strain at break	
		p-value	\mathbf{W}	p-value	\mathbf{W}	p-value	\mathbf{W}
385	CNF-0-CNF-1		1347	0.017	551	0.047	593
	CNF-0-CNF-3		1558	4.61×10^{-8}	212.5	0.95	807
	CNF-1-CNF-3	$<2.2\times10^{-16}$	1544	2.56×10^{-7}	264	0.060	996

3.3 Structural Differences - for Films and Fibrils 386

DoC can be seen to correlate with, likely causally, certain mechanical proper-387 ties of resulting films. It is prudent to ask if the same process might also af-388 fect fibrillation and microtopography of the films. Fibril strength is expected 389 to decline as DoC declines (Youssefian and Rahbar 2015), which might in-390 crease fibrillation. Several techniques have been employed to ascertain what, 391 if any, effects the ammonia treatment has had on degree of fibrillation. The 392 approaches used have spanned from resolving large features (Profilometry, 393 FiberTester Plus+) to resolving small features (SEM, AFM, NMR). 394



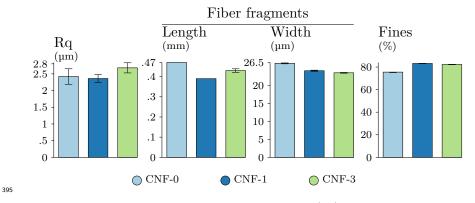


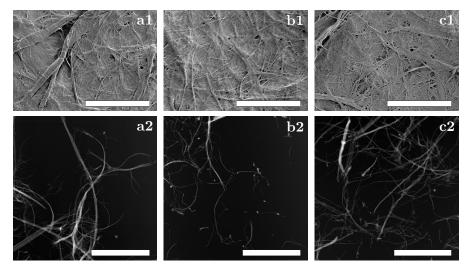
Fig. 8 Metrics pertaining to surface roughness parameter (Rq) obtained by laser profilom-396 etry and FiberTester Plus+ results. Scope and error bars defined by standard deviation. 397

Larger features in pulp or CNF suspensions were quantitatively assessed in 398 suspension (using FiberTester Plus+) as well as using dried films (using laser 399 profilometry). Profilometry can be used as an indirect measure of degree of 400 fibrillation, as smaller fibrils will yield smoother surfaces than larger, less fibril-401 lated samples (Chinga-Carrasco et al. 2008). FiberTester Plus+ uses a camera 402 to conduct measurements of fiber suspensions to measure fiber fragment and 403 fine size distributions among other metrics. Results from FiberTester+ and 404

⁴⁰⁵ profilometry (Figure 8) do not show consistent correlations with DoC, sug⁴⁰⁶ gesting fibrillation - and fibril morphology - was not affected by ammonia
⁴⁰⁷ treatments.

⁴¹⁷ SEM (Figures 9) and AFM (figures 9 and 11) also show few tangible dif-⁴¹⁸ ferences between samples, further supporting the conclusion that ammonia

⁴¹⁹ treatment has not affected fibrillation.



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Fig. 9 SEM micrographs of films and AFM micrographs of fibrils. SEM micrographs show
CNF-0 (a1, a2), CNF-1 (b1, b2) and CNF-3 (c1, c2). SEM micrographs are numbered 1
and AFM micrographs are numbered 2. SEM scale bars are 40 µm long. AFM micrographs
show scale bars 5 µm long. AFM micrographs show height data. AFM LUT goes from black
to white, where black represents 0 nm and white represents 159 (a2), 118 (b2) or 252 (c2).
AFM micrographs are recorded in air.

NMR is another technique which can assess fibril dimensions. LFAD and 420 LFAD SSA (Wickholm et al. 1998; Hult et al. 2001; Šturcova et al. 2004) (See 421 equation 2 and Figure 10) are measures of available surface area on the fibrils 422 as they would be seen in a microscope. This metric, which shows no large 423 differences between samples and largely overlapping standard errors, further 424 suggests strong morphological similarities between the samples. In other words, 425 NMR results further support the hypothesis that ammonia treatment does not 426 affect fibrillation. 427

The third metric provided, LFD, includes contribution from *inaccessible* surface area, viz. fibril surfaces within aggregates. As we can see from Figure 10 LFD is strongly affected by ammonia treatment, even though the fibril aggregates appear not to be. This reduction in lateral fibril dimensions is consistent with the appearance of cellulose II in ammonia treated samples; if cellulose II is present and formed by polymer chains leaving one fibril and attaching to another, existing fibril - or joining to form new structures with the cellulose

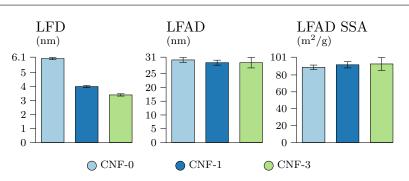


Fig. 10 Lateral fibril dimension (LFD), lateral fibril aggregate dimension (LFAD) and LFAD specific surface area (LFAD SSA) as calculated from ¹³C NMR (Segal et al. 1959; P. T. Larsson et al. 1997; Wickholm et al. 1998; Hult et al. 2001; Peciulyte et al. 2015). A detailed account for the calculations used is previously reported by Nocanda et al. (Nocanda et al. 2007). Where error bars are supplied, these indicate standard error of the mean (SE).

II allomorph - fibril, if not fibril aggregate, cross-sectional dimensions can be
 expected to be reduced.

⁴³⁷ 3.4 Fibril Properties- Swelling

AFM as a microscopy technique allows for very high precision measurements 438 of micro to nano-scopic materials. The technique has the added benefit of 439 being usable in atmosphere as well as on samples submerged in liquid. Using 440 this technique, acquisition of the micrographs of the same area while dry and 441 submerged (Figure 11) allows for measurement of fibril diameters in dry and 442 wet state, allowing for measurement of swelling of individual fibrils. It has long 443 been held that swelling can and does occur on a fibril level (Stone and Scallan 444 1968) while Wang et al. (2012) have shown similar results for fibers of bacterial 445 cellulose during enzymatic degradation, direct measurements for nano-fibrils 446 in dry state as compared with submerged in water is currently absent from 447 the literature. 448

By measuring fibril diameters on 38 separate locations in dry and swelled 461 state, we were able to measure the swelling of dried CNF, as shown in Fig-462 ure 12. The mean and median swelling are 48.9 and 48.6% respectively, with 463 a standard deviation of 27.8%. As these fibrils have been dried, it is possible 464 some loss of swell-ability has occurred and that the results do not reflect the 465 state of the elementary fibrils in the cell wall. In the event that swell-ability 466 has been lost, we note that the swelling is still measurable, which probably has 467 consequences for the properties of films produced from the material upon intro-468 duction of moisture or change in relative humidity. Given that the increase in 469 diameter measured is not constant, but changes from fibril to fibril, we believe 470 we are measuring swelling and not merely a layer of water molecules tightly 471 associated with the fibril surface. Likely areas for swelling may include regions 472 of the fibril where the degree of crystallinity is low, or less ordered / more 473

452

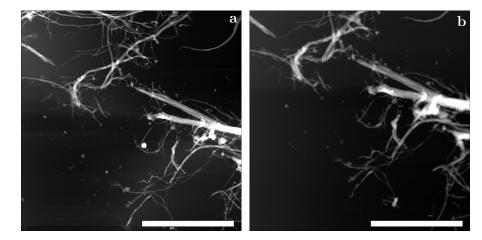


Fig. 11 AFM micrographs of ammonia-treated CNF in air (a) and water (b). Scale bars
are 5 µm long. The LUT goes from zero (black) to white, which is at 186.4 (a) and 195.5
nm (b) respectively. Both images show fibrillated linters from CNF-3.

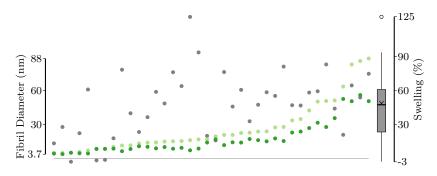


Fig. 12 Fibril diameters from Figure 11. Measurements are from 38 individual points in 453 air (\bigcirc) and water (\bigcirc) , respectively. Swelling, in %, is also shown in the same plot (\bigcirc) and 454 as a boxplot by the rightmost axis. The horizontal line across the plot shows a swelling 455 of 0% and a fibril diameter of 0 nm respectively. y-scales are limited to the range of the 456 presented data. Measurements sorted according to fibril diameter measured in water. The 457 points aligned vertically are from the same point on the same fibril in both air and water. 458 The box plot shows mean (horizontal line) and median (\times). An outlier is shown above the 459 box plot as a hollow circle. 460

amorphous. In disordered regions it seems likely water molecules are more 474 likely to enter a given fibril and swell it. For bundles of elementary fibrils, the 475 area between these may also serve as a likely candidate for swelling. Given that 476 cotton linters are nearly pure cellulose, the swelled material is unlikely to be 477 other components of the source plant, e.g. hemicellulose. As crystallinity is re-478 duced, e.g. by ammonia treatment, it seems possible swell-ability of nanofibrils 479 may be increased. As CNF from CNF-3 was used for the above measurements, 480 the swelling may be higher than for most CNF qualities due to the sample's 481 reduced degree of crystallinity. Further work can shed light on the role of DoC 482 in swelling. 483

484 4 Conclusions

The degree of crystallinity (DoC) of cotton linters was reduced while retaining a predominantly cellulose I allomorph by submersion in liquid anhydrous ammonia and subsequent boiling in de-ionized (DI)-water.

⁴⁸⁸ Our main conclusion based on the work presented herein is that the de⁴⁸⁹ gree of crystallinity does affect mechanical properties; reduced degree of crystallinity (DoC) correlates with increased elastic domain (increased yield point)
⁴⁹¹ as well as reduced Young-modulus. In summary:

The crystallinity of the sample was reduced by the used ammonia treatment scheme. The first ammonia treatment had the most effect; subsequent
 treatments had less effect on DoC and CI. Crystallinity was primarily reported as DoC measured using ¹³C NMR, though crystalline index (CI),
 measured using WAXS, was also supplied.

- 497
 A small amount of cellulose II is seen in the ammonia treated samples. This
 498 is slight, below WAXS detection limit. We hypothesize this is due to some
 499 few cellulose polymer chains being removed from their host fibrils during
 500 ammonia treatment, reforming in an anti-parallel fashion, giving rise to
 501 cellulose II.
- ⁵⁰² 3. DoC was seen to correlate with Young modulus. The effect was notable,
 ⁵⁰³ and highly significant.
- 4. DoC was seen to have an inverse correlation with yield point. The effect was slight, but statistically significant.
- 506 5. DoC is not shown to affect strain at break; no correlation was found for 507 strain at break.
- 6. Ammonia treatment was seen to alter the internal structure of the fibril
 aggregates, increasing inaccessible surface areas, reducing LFD. A similar
 effect on fibril aggregates (LFAD) was not seen.
- Ammonia treatment was not seen to affect degree of fibrillation, which was
 assessed using profilometry, FiberTester +, AFM, SEM and NMR.

Using AFM to examine CNF-3 in air and water we were also able to demonstrate swelling of individual CNFs (fibrils and fibril aggregates) upon submersion in DI water. From the range of fibril diameters measured (3.7 to 88 nm), swelling appears to affect nano-fibers with a wide range of diameters. This result demonstrates that swelling of dry cellulose materials is not relegated to the fiber or network size range, but also occurs on a nano-fibril level.

Acknowledgements This work is performed as a part of the NORCEL Project: The NOR-519 wegian NanoCELlulose Technology Platform, initiated and led by The Paper and Fiber 520 Research Institute (PFI) in Trondheim and funded by the Research Council of Norway 521 through the NANO2021 Program (grant 228147 Research Council of Norway). The Re-522 search Council of Norway is further acknowledged for the support to the Norwegian Micro-523 and Nano-Fabrication Facility, NorFab. Thanks are extended to CELSUR for providing 524 cotton linters. Thanks are further extended to Jasna Stevanic Srndovic for assistance with 525 NMR and WAXS measurements, Kelly McCammon-Ottesen for proof-reading. 526

527 References

Aulin, Christian, Mikael Gällstedt, and Tom Lindström (Jan. 2010). "Oxygen and oil barrier properties of microfibrillated cellulose films and coatings".
In: Cellulose 17.3, pp. 559–574. ISSN: 0969-0239. DOI: 10.1007/s10570-

⁵³¹ 009-9393-y.
⁵³² Barry, A. J., F. C. Peterson, and A. J. King (Feb. 1936). "x-Ray Studies of ⁵³³ Reactions of Cellulose in Non-Aqueous Systems. I. Interaction of Cellulose ⁵³⁴ and Liquid Ammonia 1". In: J. Am. Chem. Soc. 58.2, pp. 333-337. ISSN:

⁵³⁵ 0002-7863. DOI: 10.1021/ja01293a043.

Brodin, Malin et al. (Sept. 2017). "Lignocellulosics as sustainable resources for
production of bioplastics – A review". In: J. Clean. Prod. 162, pp. 646–664.
ISSN: 09596526. DOI: 10.1016/j.jclepro.2017.05.209.

⁵³⁹ Chinga-Carrasco, G. et al. (2008). "New advances in the 3D characterization
 ⁵⁴⁰ of mineral coating layers on paper". In: J. Microsc. 232.2, pp. 212–224.
 ⁵⁴¹ ISSN: 00222720. DOI: 10.1111/j.1365-2818.2008.02092.x.

⁵⁴² Codou, Amadine et al. (2015). "Partial periodate oxidation and thermal cross ⁵⁴³ linking for the processing of thermosetall-cellulose composites". In: Compos.

Sci. Technol. 117, pp. 54–61. DOI: 10.1016/j.compscitech.2015.05.022. Dinand, Elizabeth et al. (2002). "Mercerization of primary wall cellulose and

its implication for the conversion of cellulose I \rightarrow cellulose II". In: *Cellulose* 9.1, pp. 7–18. ISSN: 09690239. DOI: 10.1023/A:1015877021688.

- ⁵⁴⁸ Dufresne, Alain (2012). Nanocellulose: From Nature to High Performance Tai ⁵⁴⁹ lored Materials. Walter de Gruyter. ISBN: 3110254603.
- Foster, E Johan et al. (2018). Current characterization methods for cellulose
 nanomaterials. DOI: 10.1039/c6cs00895j.
- ⁵⁵² Fukuzumi, Hayaka et al. (Jan. 2009). "Transparent and High Gas Barrier
 ⁵⁵³ Films of Cellulose Nanofibers Prepared by TEMPO-Mediated Oxidation".
 ⁵⁵⁴ In: *Biomacromolecules* 10.1, pp. 162–165. ISSN: 1525-7797. DOI: 10.1021/
 ⁵⁵⁵ bm801065u.
- Ginestet, Cedric (2011). "ggplot2: Elegant Graphics for Data Analysis". In: J.
 R. Stat. Soc. Ser. A (Statistics Soc. 174.1, pp. 245–246. ISSN: 0006341X.
 DOI: 10.1111/j.1541-0420.2011.01616.x.

Henriksson, Marielle et al. (June 2008). "Cellulose Nanopaper Structures of
High Toughness". In: *Biomacromolecules* 9.6, pp. 1579–1585. ISSN: 15257797. DOI: 10.1021/bm800038n.

- Hermann, Christine K. F. (Nov. 1997). "The shrinking dollar bill". In: J. Chem.
 Educ. 74.11, p. 1357. ISSN: 0021-9584. DOI: 10.1021/ed074p1357.2.
- Hess, K. and C. Trogus (1935). "Über Ammoniak-Cellulose (Vorläuf. Mitteil.)"
- In: Berichte der Dtsch. Chem. Gesellschaft (A B Ser. 68.10, pp. 1986–1988.
 ISSN: 03659488. DOI: 10.1002/cber.19350681016.

Hult, Eva L., Per T. Larsson, and Tommy Iversen (2001). "Cellulose fibril ag gregation - An inherent property of kraft pulps". In: *Polymer (Guildf)*. 42.8,
 pp. 3309–3314. ISSN: 00323861. DOI: 10.1016/S0032-3861(00)00774-6.

- Kono, Hiroyuki and Yukari Numata (June 2004). "Two-dimensional spin-
- exchange solid-state NMR study of the crystal structure of cellulose II". In:

572	Polymer (Guildf). 45.13, pp. 4541–4547. ISSN: 00323861. DOI: 10.1016/j.
573	polymer.2004.04.025.
574	Kroon-Batenburg, L M J, B Bouma, and J Kroon (1996). "Stability of Cel-
575	lulose Structures Studied by MD Simulations. Could Mercerized Cellulose
576	II Be Parallel?" In: <i>Macromolecules</i> 29.17, pp. 5695–5699. ISSN: 0024-9297.
577	DOI: 10.1021/ma9518058.
578	Kumar, Vinay et al. (June 2016). "Influence of nanolatex addition on cellulose
579	nanofiber film properties". In: Nord. Pulp Pap. Res. J. 31.02, pp. 333–340.
580	ISSN: 0283-2631. DOI: 10.3183/NPPRJ-2016-31-02-p333-340.
581	Larsson, Per A, Lars A Berglund, and Lars Wågberg (June 2014). "Ductile all-
582	cellulose nanocomposite films fabricated from core-shell structured cellu-
583	lose nanofibrils." In: Biomacromolecules 15.6, pp. 2218–23. ISSN: 1526-4602.
584	DOI: 10.1021/bm500360c.
585	Larsson, Per A and Lars Wågberg (2016). "Towards natural-fibre-based ther-
586	moplastic films produced by conventional papermaking". In: Green Chem.
587	18.11, pp. 3324-3333. DOI: 10.1039/c5gc03068d.
588	Larsson, Per Tomas, Kristina Wickholm, and Tommy Iversen (July 1997). "A
589	CP/MAS13C NMR investigation of molecular ordering in celluloses". In:
590	Carbohydr. Res. 302.1-2, pp. 19–25. ISSN: 00086215. DOI: 10.1016/S0008-
591	6215(97)00130-4.
592	Lavoine, Nathalie et al. (Oct. 2012). "Microfibrillated cellulose - its barrier
593	properties and applications in cellulosic materials: a review." In: Carbohydr.
594	<i>Polym.</i> 90.2, pp. 735–64. ISSN: 1879-1344. DOI: 10.1016/j.carbpol.2012.
595	05.026.
596	Menachem, Lewin and Luis G. Roldan (1971). "the Effect of Liquid Anhy-
597	drous Ammonia in the structure and morphology of cotton cellulose". In:
598	J. Polym. Sci. Part C Polym. Symp. 229.36, pp. 213–229.
599	Minelli, Matteo et al. (Aug. 2010). "Investigation of mass transport properties

- of microfibrillated cellulose (MFC) films". In: J. Memb. Sci. 358.1-2, pp. 67–
 75. ISSN: 03767388. DOI: 10.1016/j.memsci.2010.04.030.
- Mittal, Ashutosh et al. (Jan. 2011). "Effects of alkaline or liquid-ammonia treatment on crystalline cellulose: changes in crystalline structure and effects on enzymatic digestibility". In: *Biotechnol. Biofuels* 4.41, pp. 1–16.
 ISSN: 1754-6834. DOI: 10.1186/1754-6834-4-41.
- Myllytie, Petri et al. (2010). "Viscoelasticity and water plasticization of polymer cellulose composite films and paper sheets". In: *Cellulose* 17.2, pp. 375–385.
 ISSN: 09690239. DOI: 10.1007/s10570-009-9376-z.
- Nečas, David and Petr Klapetek (Jan. 2012). "Gwyddion: an open-source software for SPM data analysis". In: *Open Phys.* 10.1. ISSN: 2391-5471. DOI: 10.2478/s11534-011-0096-2.
- Nishino, Takashi, Ikuyo Matsuda, and Koichi Hirao (2004). "All-Cellulose
 Composite". In: *Macromolecules* 37.20, pp. 7683–7687. ISSN: 00249297. DOI:
 10.1021/ma049300h.
- ⁶¹⁵ Nocanda, Xolani et al. (Jan. 2007). "Cross polarisation/magic angle spinning
- ⁶¹⁶ 13C-NMR spectroscopic studies of cellulose structural changes in hard-

- wood dissolving pulp process". In: *Holzforschung* 61.6, pp. 675–679. ISSN:
 1437434X. DOI: 10.1515/HF.2007.095.
- Nogi, Masaya et al. (Apr. 2009). "Optically Transparent Nanofiber Paper". In:
 Adv. Mater. 21.16, pp. 1595–1598. ISSN: 09359648. DOI: 10.1002/adma.

200803174.
Okano, Takeshi and Anatole Sarko (Jan. 1985). "Mercerization of cellulose. II.
Alkali-cellulose intermediates and a possible mercerization mechanism".
In: J. Appl. Polym. Sci. 30.1, pp. 325–332. ISSN: 00218995. DOI: 10.1002/

⁶²⁵ app.1985.070300128.

- Park, Sun-Ji et al. (2003). "Effect of Dry Heat and Hot Water Processings on
 Cellulose III Crystallite of Cotton and Lyocell Fibers Treated with Liquid
- Ammonia." In: Sen'i Gakkaishi 58.8, pp. 299–303. ISSN: 0037-9875. DOI:
 10.2115/fiber.58.299.
- Peciulyte, Ausra et al. (Dec. 2015). "Impact of the supramolecular structure of
 cellulose on the efficiency of enzymatic hydrolysis". In: *Biotechnol. Biofuels*8.56, pp. 1–13. ISSN: 1754-6834. DOI: 10.1186/s13068-015-0236-9.
- Perez, Serge and and Karim Mazeau (2005). "Conformations, Structures, and
 Morphologies of Celluloses". In: *Polysaccharides Struct. Divers. Funct. Versatility*, pp. 41–68. DOI: 10.1201/9781420030822.ch2.
- ⁶³⁶ R Core Team (2015). *R: A Language and Environment for Statistical Com-*⁶³⁷ *puting.* DOI: 10.1017/CB09781107415324.004.
- Rodionova, Galina et al. (Feb. 2012). "Mechanical and oxygen barrier properties of films prepared from fibrillated dispersions of TEMPO-oxidized Norway spruce and Eucalyptus pulps". In: *Cellulose* 19.3, pp. 705–711. ISSN: 0969-0239. DOI: 10.1007/s10570-012-9664-x.
- Rousselle, Marie A. et al. (Apr. 1976). "Liquid-ammonia And Caustic Mercerization Of Cotton Fibers: Changes In Fine Structure And Mechanical Properties". In: *Text. Res. J.* 46.4, pp. 304–310. ISSN: 00405175. DOI:
 10.1177/004051757604600412.
- Saapan, A.A., Sherif H. Kandil, and Abdel M. Habib (Dec. 1984). "Liquid Ammonia and Caustic Mercerization of Cotton Fibers Using X-Ray, Infrared, and Sorption Measurements". In: *Text. Res. J.* 54.12, pp. 863–867.
 ISSN: 0040-5175. DOI: 10.1177/004051758405401212.
- Sawada, Daisuke et al. (Mar. 2014). "The initial structure of cellulose during
 ammonia pretreatment". In: *Cellulose* 21.3, pp. 1117–1126. ISSN: 0969-0239.
 DOI: 10.1007/s10570-014-0218-2.
- Schindelin, Johannes et al. (July 2012). "Fiji: an open-source platform for
 biological-image analysis." In: *Nat. Methods* 9.7, pp. 676–82. ISSN: 15487105. DOI: 10.1038/nmeth.2019.
- Segal, Leon et al. (1959). "An Empirical Method for Estimating the Degree of Crystallinity of Native Cellulose Using the X-Ray Diffractometer". In: *Text. Res. J.* 29.10, pp. 786–794. ISSN: 0040-5175. DOI: 10.1177/
 004051755902901003.
- Stone, J. E. and A. M. Scallan (1968). "A Structural Model for the Cell Wall
 of Water Swollen Wood Fibres Based on Their Accessibility to Macro molecules". In: Cellul. Chem. Technol. 2, pp. 343–358.

63	Sturcova, Adriana et a	l. (2004). '	"Structural	details	of crystall	ine cellul	lose from
54	higher plants". In:	Biomacr	omolecules.	ISSN:	15257797.	DOI: 10	. 1021/

bm034517p. 665

666	Syverud, Kristin and Per Stenius (Aug. 2008). "Strength and barrier properties
667	of MFC films". In: Cellulose 16.1, pp. 75–85. ISSN: 0969-0239. DOI: 10.
668	1007/s10570-008-9244-2.

- Thao Ho, Thi Thu et al. (Apr. 2013). "Liquid ammonia treatment of (cationic) 669 nanofibrillated cellulose/vermiculite composites". In: J. Polym. Sci. Part 670 B Polym. Phys. 51.8, pp. 638-648. ISSN: 08876266. DOI: 10.1002/polb. 671
- 23241. 672

Wada, Masahisa, Yoshiharu Nishiyama, and Paul Langan (Apr. 2006). "X-673

- ray Structure of Ammonia-Cellulose I: New Insights into the Conversion 674 of Cellulose I to Cellulose III I". In: Macromolecules 39.8, pp. 2947-2952. 675
- ISSN: 0024-9297. DOI: 10.1021/ma060228s. 676
- Wang, Jingpeng et al. (2012). "Real-time observation of the swelling and hy-677 drolysis of a single crystalline cellulose fiber catalyzed by cellulase 7B from 678 Trichoderma reesei". In: Langmuir 28.25, pp. 9664–9672. ISSN: 07437463. 679 DOI: 10.1021/la301030f. 680
- Wickholm, Kristina, Per Tomas Larsson, and Tommy Iversen (1998). "Assign-681
- ment of non-crystalline forms in cellulose I by CP/MAS 13C NMR spec-682 troscopy". In: Carbohydr. Res. 312.3, pp. 123-129. ISSN: 00086215. DOI: 683
- 10.1016/S0008-6215(98)00236-5. 684

Youssefian, Sina, Joseph E. Jakes, and Nima Rahbar (2017). "Variation of 685

- Nanostructures, Molecular Interactions, and Anisotropic Elastic Moduli of 686 Lignocellulosic Cell Walls with Moisture". In: Sci. Rep. 7.1, pp. 1–10. ISSN: 687 20452322. DOI: 10.1038/s41598-017-02288-w. 688
- Youssefian, Sina and Nima Rahbar (2015). "Molecular origin of strength and 689
- stiffness in bamboo fibrils". In: Sci. Rep. 5, pp. 1–13. ISSN: 20452322. DOI: 690 10.1038/srep11116. 691
- Zugenmaier, Peter (2008). Crystalline cellulose and cellulose derivatives. Springer. 692
- ISBN: 9783540739333. 693

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