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# A Study of Metabolites in Breast Cancer Xenografts 

Optimisation of Extraction Method for MS-based Analysis and Investigation of Subtypes using Metabolite Profiling and Isotope Labelling

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## Preface

This Master's thesis was based on collaboration between the Department of Biotechnology and the Department of Circulation and Medical Imaging at the Norwegian University of Science and Technology (NTNU). The thesis was carried out for the Master of Technology degree in Chemical Engineering and Biotechnology in the spring of 2012. I would like to express my sincere gratitude to my supervisor, Associate Professor Per Bruheim at the Department of Biotechnology for his expert guidance and full support. Grateful acknowledgements are also due to PhD Candidate Siver Moestue and Research Scientist Tone Bathen at the Department of Circulation and Medical Imaging for their valuable advice and for providing xenograft tissue. I wish to express my sincere appreciation to PhD Candidate Hans Fredrik N. Kvitvang for his assistance and patience. Special thanks go to Senior Engineer Kåre Kristiansen and PhD Candidate Stina K. Lien for their help in the laboratory.

This report was written by me alone but the laboratory work was carried out together with Master student T. Marita Madsen. I would like to express gratitude to her for encouragement, support and for making the laboratory work an enjoyable experience.

## Contents

1 Introduction ..... 1
1.1 Breast cancers and prospects for improvement of treatment ..... 1
1.2 Metabolite profiling ..... 2
1.3 Extraction of metabolites ..... 2
1.4 Analytical techniques ..... 3
1.5 Aim ..... 4
2 Experimental methods ..... 5
2.1 Xenograft mouse models ..... 5
2.2 Extraction of polar and non-polar metabolites ..... 6
2.2.1 Extraction of polar metabolites ..... 6
2.2.2 Combining extraction of polar and non-polar metabolites ..... 6
2.2.3 Optimised extraction method ..... 8
2.3 Derivatisation ..... 9
2.3.1 MCF derivatisation of polar extracts ..... 9
2.3.2 TMS derivatisation of polar extracts ..... 10
2.3.3 HCl in methanol derivatisation of non-polar extracts ..... 11
2.4 GC-MS analysis ..... 12
2.4.1 Triple Quadrupole GC-MS ..... 12
2.4.2 Single Quadrupole GC-MS ..... 12
2.5 Q-TOF LC-MS analysis ..... 13
2.6 Software for identification and quantitation ..... 14
2.6.1 Chemstation for GC-Q-MS ..... 14
2.6.2 Masshunter for GC-QQQ-MS and Q-TOF LC-MS ..... 14
3 Results and discussion ..... 15
3.1 Optimisation of a method for metabolite extraction ..... 15
3.1.1 Extraction of polar metabolites ..... 15
3.1.2 Extraction of non-polar metabolites ..... 19
3.1.3 Combining extraction of polar and non-polar metabolites ..... 21
3.1.4 Method reproducibility and variation ..... 23
3.1.5 Sample preparation for polar extracts ..... 23
3.1.6 Optimised extraction method ..... 25
3.1.7 Protocol ..... 26
3.2 Metabolite profiling of polar extracts ..... 27
3.2.1 Metabolite profiles of luminal-like and basal-like subtypes ..... 27
3.2.2 Comparison of metabolite profiles ..... 29
3.3 Investigation of metabolic pathways using isotope labelling ..... 34
3.4 Q-TOF LC-MS analysis of non-polar extracts ..... 37
3.5 Recommendations for future work ..... 38
3.5.1 Loss of lipids in polar extracts ..... 38
3.5.2 Analysis of non-polar extracts ..... 38
3.5.3 Metabolite profiles ..... 38
4 Conclusion ..... 39
A List of abbreviations ..... 42
B Tumour and sample names ..... 44
C MCF derivatisation of polar extracts ..... 45
C. 1 Raw data ..... 45
C. 2 Normalised data ..... 51
D TMS derivatisation of polar extracts ..... 57
D. 1 Raw data ..... 57
D. 2 Normalised data ..... 58
E HCl in methanol derivatisation of non-polar extracts ..... 59
E. 1 Raw data ..... 59
E. 2 Normalised data ..... 60
E. 3 Unidentified compound ..... 60
F Lipid analysis of non-polar extracts ..... 61
G Isotope labelling experiment ..... 63
H Data analysis ..... 65
H. 1 Standard deviation ..... 65
H. 2 Student's t-test ..... 65
H. 3 Principal component analysis (PCA) ..... 67


#### Abstract

The most common cancer among women in Europe and the United States is breast cancer. However, treatment remains a major challenge. If prediction of prognosis and treatment response become more reliable, treatment can be improved. An approach to overcome this challenge is to establish a technique for identification of subtypes. Metabolite profiling of breast cancer has shown potential to identify subtype. The aim was divided into three parts, including metabolite extraction and profiling and isotope labelling of metabolites.

The first aim was to optimise a method for extraction of polar and non-polar metabolites for mass spectrometry-based analysis. To achieve the aim, beads-based homogenisation of xenograft tissue in $60 \%$ methanol solution and in chloroform was performed in a series of experiments. Requirements were fulfilled, including complete extraction, simple performance and high reproducibility. Method was therefore stated successfully optimised.

The second aim was to obtain and compare metabolite profiles of luminal-like and basal-like subtypes of breast cancers. To achieve the aim, polar metabolites were extracted from xenograft tissue using the optimised method prior to gas chromatography mass spectrometry (GC-MS) analysis. Profiles comprising more than 30 metabolites and their concentration were obtained. For comparison of subtypes, data analysis including $\log _{2}$ ratios, principal component analysis (PCA) and Student's t-test were used. All data analyses indicated differences in metabolite concentrations between subtypes. 15 metabolites were found by Student's t-test to significantly differ in concentration between subtypes, including lactate, glycine, citrate, lysine and aspartate. Therefore, metabolite profiling is a potential tool for identification of subtype. Furthermore, metabolites shown to significantly differ may provide insight into metabolic changes in breast cancers that remain poorly understood.


The third aim was to investigate metabolic pathways in luminal-like subtype, possessing reduced tumour growth rates due to treatment. To achieve the aim, ${ }^{13} \mathrm{C}$-labelled glucose was injected into xenograft models prior to tumour excision, extraction, GC-MS analysis and calculation of summed fractional labelling (SFL). $8.3 \%$ lactate, $2.2 \%$ citrate and $1.6 \%$ fumarate were found labelled using SFL. ${ }^{13} \mathrm{C}$ labelling was therefore shown retained throughout glycolysis, to enter the tricarboxylic acid cycle (TCA) cycle and to give rise to TCA intermediates.

## Sammendrag

Brystkreft er den vanligste kreftformen blant kvinner i Europa og i USA. Likevel er optimal behandling en stor utfordring. Bedre behandling kan oppnås med mer nøyaktige metoder for å stille prognose og forutsi respons på behandling. Utfordringen kan angripes ved å etablere en metode for identifisering av brystkreftsubtype. Metabolittprofiler har vist potensiale til å identifisere subtype. Formålet med oppgaven ble delt inn i tre deler; ekstraksjon av metabolitter, metabolittprofiler og isotopmerking av metabolitter.

Det første formålet var å optimalisere en ekstraksjonsmetode for polare og upolare metabolitter for massespektroskopi-basert analyse. En rekke eksperimenter ble utført med perle-basert homogenisering av xenograft vev i en $60 \%$ metanol-løsning og i kloroform. Metodekravene; fullstendig ekstraksjon, enkel gjennomførelse og høy reproduserbarhet ble innfridd og metoden erklært optimal.

Det andre formålet var å lage og sammenligne metabolittprofiler av luminal-like og basal-like subtyper av brystkreft. Polare metabolitter ble ekstrahert fra xenograft vev ved hjelp av den optimaliserte metoden i forkant av GC-MS analyse. Profilene inneholdt mer enn 30 metabolitter. Dataanalyser som $\log _{2}$ forhold, prinsipal komponent analyse (PCA) og Students t-test ble brukt til å sammenligne subtyper. Dataanalysene indikerte forskjeller i metabolittkonsentrasjoner mellom subtyper. 15 metabolitter ble funnet signifikant forskjellige i subtyper ved hjelp av Students t-test blant annet laktat, glysin, citrat, lysin og aspartat. Dette viser at metabolittprofiler har potensiale til å kunne identifisere subtype. Dessuten kan de signifikant forskjellige metabolittene bidra til $\varnothing$ kt kunnskap om metabolske forandringer i brystkreft som man per dags dato vet lite om.

Det tredje formålet var å unders $\varnothing$ ke metabolske spor i luminal-like subtype med redusert tumor vekst rate som følge av behandling. ${ }^{13} \mathrm{C}$-merket glukose ble injisert inn i xenograft mus før fjerning av tumour, ekstraksjon av metabolitter, GC-MS analyse og beregning av summed fractional labelling (SFL). $8.3 \%$ laktat, $2.2 \%$ citrat og $1.6 \%$ fumarat ble funnet merket. Det ble dermed vist at ${ }^{13} \mathrm{C}$-merkinga ble beholdt gjennom hele glykolysen, gikk inn i sitronsyresyklusen (TCA) og resulterte i TCA-mellomprodukter.

## 1 Introduction

### 1.1 Breast cancers and prospects for improvement of treatment

The most common cancer among women in Europe and the United States is breast cancer and as many as one in every ten Norwegian women are diagnosed with the disease at some time during their lives [1, 2]. However, optimal treatment remains a major challenge. An approach to overcome this challenge is to establish better techniques for the prediction of prognosis and treatment response. If these techniques become more reliable, treatment can be tailored specifically to individual patients. However, the establishment of better techniques are dependent on a deeper understanding of the biology of breast cancers [1, 3].

Cancers arise from abnormalities in cell function. Consequently, cells proliferate in an uncontrolled manner causing a mass of growing tissue, a tumour. Tumours are classified as either benign (non-cancerous) or malignant (cancerous). A benign tumour is rarely dangerous because it grows in a confined local area. In contrast, a malignant tumour is capable of spreading and may therefore spread to and invade neighbouring tissues [4].

Nearly all breast cancers are carcinomas. Breast carcinoma arises from epithelial cells constituting the lining layer of the breast. The two main groups are ductal and lobular carcinoma. Ductal carcinoma starts in the ducts, tubes that transport milk to the nipple. Lobular carcinoma has its origin in the lobules, glands where milk production occurs [4]. The anatomy of the female breast, including lobules (arrow 3) and ducts (arrow 6), is illustrated in Figure 1.1.


Figure 1.1: Cross-section scheme of the female breast illustrating chest wall (arrow 1), chest muscles (2), lobules (3), nipple (4), areola (5), duct (6), fatty tissue (7) and skin (8) [5].

Five subtypes have been defined, including luminal-like and basal-like subtypes, named after the epithelial cells in which the cancer arises. If techniques for prediction of prognosis and treatment response include identification of subtype, they become more reliable. However, the usage of subtypes remains limited [6, 3]. More insight into subtypes may be provided by investigating metabolites contained in subtypes [3].

### 1.2 Metabolite profiling

Metabolites are low-molecular-weight compounds used by or formed by cells in association with their metabolism. A cell's metabolism involves bioprocesses required for proliferation and survival [7]. Due to the abnormal cell function of cancers, metabolic changes occur [8]. Breast cancers have shown increased concentrations of lactate, glycine and choline metabolites [9, 3]. Metabolite profiles are able to show metabolic changes. Xenograft mouse models were exploited because they are considered unique tools for investigation of human breast cancers [1]. To obtain such a profile, metabolites have to be extracted from tissue and quantitated using an analytical technique $[8,10,3]$.

### 1.3 Extraction of metabolites

To extract metabolites from tissue, cells and tissue are broken up to enable metabolites to liberate to an extraction solvent. Most commonly, this is achieved by grinding tissue with mortar and pestle [11, 10]. Beads-based and rotating blade homogenisers can also be employed to break up tissue. Several extraction methods including mortar and pestle, beads-based and rotating blade homogenisation, freezing-thawing, boiling ethanol and ultrasonication were previously evaluated for their suitability for xenograft tissue. Beads-based homogenisation equipped with cooling was shown superior to the other methods [12].


Figure 1.2: The beads-based homogeniser used in this project, a Precellys 24 tissue homogeniser. Cryo tubes containing beads are seen loaded into homogeniser.[13].

At the time tissue is broken up, metabolites can liberate to extraction solvents in which they are soluble. Methanol mixed with water has shown to be a suitable solvent for polar metabolites and chloroform for non-polar metabolites [11, 14]. Often, these solvents are in contact with tissue simultaneously and a biphasic system is obtained $[15][14,10]$.

### 1.4 Analytical techniques

Metabolites are commonly analysed using nuclear magnetic resonance (NMR) spectroscopy and mass spectroscopy (MS). MS is coupled with gas chromatography (GC) and liquid chromatography (LC) in order to separate metabolites in extracts. GC requires chemical derivatisation of metabolites. In contrast, extracts can be injected directly into LC. MS-based approaches, however, are more sensitive than NMR and provides therefore access to metabolites with lower concentrations [16, 10, 7, 11]. In addition to obtain metabolite profiles, the aforementioned analytical techniques can be used to investigate for instance the passage of an isotope through pathways [9].


Figure 1.3: Flowchart of experimental procedures performed in order to obtain metabolite profile of tissue.

### 1.5 Aim

This project includes three aims, all investigating the metabolites in breast cancer xenograft tissue.

The first aim is to optimise a method for extraction of polar and non-polar metabolites for MS-based analysis. An optimised method requires complete extraction, simple performance and high reproducibility. For the reason that beads-based homogenisation has shown potential to fulfill these requirements, beads-based homogenisation of tissue directly in extraction solvent is attempted optimised [12]. Extraction solvents of polar and non-polar metabolites include a $60 \%$ methanol solution and chloroform, respectively.

The second aim is to obtain and compare metabolite profiles of luminal-like and basal-like subtypes of breast cancer. To achieve the aim, the optimised method can be used to extract metabolites from xenograft tissue prior to MS-based analysis. Data analysis of profiles may show differences between subtypes.

The third aim is to investigate metabolic pathways in treated luminal-like subtype using isotope labelling. Carbon 13-labelled glucose had been injected into xenograft mice and the passage of ${ }^{13} \mathrm{C}$ is tracked obtaining summed fractional labelling (SFL) from GC-MS data.

## 2 Experimental methods

### 2.1 Xenograft mouse models

The MR Cancer Group at the Department of Circulation and Medical Imaging at NTNU provided the tumours after approval by the National Animal Research Authority in Norway. Xenograft mouse models were established by serial transplantation of human breast cancer isolates in severe combined immunodeficient (SCID) female mice. Prior to transplantation, isolates were classified as invasive grade III ductal carcinoma. Tumours were excised from xenograft models at the time tumour diameter reached 15 mm . Immediate storage on liquid nitrogen was performed to halt biological activity [17].

Xenograft models used in this study, MAS98.06 and MAS98.12, represent luminal-like and basal-like subtypes of ductal carcinoma, respectively. Luminal-like subtype has been charcterised as estrogen receptor positive (ER + ) and basal-like subtype as estrogen receptor negative (ER-). Basal-like subtype has demonstrated a worse prognosis than luminal-like subtype $[17,3,1]$.

Drinking water was supplied with estrogen and given to mice bearing luminal-like tumours to promote growth of tumours [3]. However, some of the mice with luminal-like tumours were not given estrogen. In consequence, their tumours possessed reduced growth rates. The low abundance of estrogen in breast tissue was meant to resemble treatment with tamoxifen. Moreover, $29 \mathrm{mg}\left[1-{ }^{13} \mathrm{C}\right]$ glucose was injected into mice ten minutes prior to tumour excision. In this project, tissue from these mice are therefore referred to as treated [18, 19]. Full tumour names are shown in Table B.1.


Figure 2.1: Xenograft mouse model bearing two tumours of ductal carcinoma.

### 2.2 Extraction of polar and non-polar metabolites

To optimise extraction method for xenograft tissue (2.1) for MS-based analysis, a beads-based homogeniser, more specifically a Precellys 24 tissue homogeniser (Bertin Technologies) was employed. Homogeniser was coupled with a Cryolys cooling system (Bertin Technologies) to ensure low temperatures. Method was developed for cancerous tissue within the range of $30 \pm 10 \mathrm{mg}$. 0.50 g of ceramic (zirconium oxide) beads with a diameter of 1.4 mm were used [14]. Optimisation experiments extracted metabolites exclusively from luminal-like subtype. At last, optimised method was performed for different purposes, including metabolite profiling and investigation of isotope labelling.

### 2.2.1 Extraction of polar metabolites

Tumours were cut and weighed to obtain tissue samples of $30 \pm 10 \mathrm{mg}$. Tissue samples were placed in precooled $\left(-20^{\circ} \mathrm{C}\right) 2 \mathrm{ml}$ cryo tubes containing beads. $400 \mu \mathrm{l} 60 \%$ methanol and 10 $\mu \mathrm{l}$ internal standard ( 10 mM d3-alanine) were added to tubes. Tissue samples were homogenised for three intervals of 20 sec at 5500 rpm using the beads-based homogeniser. Homogeniser was set to pause for 30 sec in between intervals. Samples were centrifuged for 5 min at 5000 rpm and $-9{ }^{\circ} \mathrm{C}$. Extracts were removed and three technical replicates of 100 $\mu \mathrm{l}$ were obtained in 5 ml polypropylene ( PP ) tubes.

Experimental procedure described in previous paragraph was repeated twice to obtain a second and a third extraction round. d5-glutamate was used as internal standard for second round. Third round lacked internal standard. Polar extracts were derivatised using MCF prior to GC-QqQ-MS analysis.

### 2.2.2 Combining extraction of polar and non-polar metabolites

Attempts to combine extraction of polar and non-polar metabolites were made. To achieve complete extraction of non-polar metabolites, number of extraction rounds and volume of chloroform were investigated.

## Three extraction rounds in $400 \mu \mathrm{l}$ chloroform

To extract polar metabolites, experimental procedure previously described was performed (2.2.1 ). Some changes were made; internal standards for first and second extraction round were 10 mM d 3 -alanine and 1 mM d4-succinate, respectively. Extracts were pooled and technical replicates obtained from pool of extracts.

To extract non-polar metabolites, $400 \mu \mathrm{l}$ precooled $\left(-20^{\circ} \mathrm{C}\right)$ chloroform and $10 \mu \mathrm{l} 10 \mathrm{mM}$ internal standard (d5-glutamate) were added to cryo tubes containing beads and homogenised tissue. Tissue samples were homogenised for three intervals of 20 sec at 5500 rpm using tissue homogeniser. Homogeniser was set to pause for 30 sec in between intervals. Samples were centrifuged for 5 min at 5000 rpm and $-9{ }^{\circ} \mathrm{C}$. Extracts were removed and three technical replicates of $100 \mu \mathrm{l}$ were obtained in vials.

Experimental procedure described in previous paragraph was repeated twice to obtain a second and a third extraction round. No internal standard were added.

To investigate whether polar metabolites remained in tissue, a new extraction round in methanol solution was performed. $400 \mu \mathrm{l}$ precooled $\left(-20^{\circ} \mathrm{C}\right) 60 \%$ methanol and $10 \mu \mathrm{l}$ internal standard ( 1 mM d4-succinate) were added to tubes containing beads and homogenised tissue. Tissue samples were homogenised for three intervals of 20 sec at 5500 rpm using tissue homogeniser. Homogeniser was set to pause for 30 sec in between intervals. Samples were centrifuged for 5 min at 5000 rpm and $-9{ }^{\circ} \mathrm{C}$. Extracts were removed and three technical replicates of $100 \mu \mathrm{l}$ were obtained in 5 ml PP tubes. Polar extracts were derivatised using MCF prior to GC-Q-MS analysis.

## Five extraction rounds in $400 \mu \mathrm{l}$ chloroform

Experimental procedure as described in 2.2.2 was performed with some changes. 1 mM d 4 succinate, 10 mM d3-alanine and 10 mM d8-valine were added prior to first, second and third extraction round, respectively. Polar extracts were derivatised using TMS prior to GC-Q-MS analysis.

To extract non-polar metabolites, five extraction rounds in chloroform was performed with settings as previously. Three technical replicates of $100 \mu \mathrm{l}$ were obtained in vials each round. Non-polar extracts were derivatised using HCl in methanol prior to GC-Q-MS analysis. No last extraction round in methanol solution as previously described was performed.

## Biphasic system prior to five extraction rounds in $800 \mu l$ chloroform

Experimental procedure described in 2.2.2 was performed as previously up until extraction of non-polar metabolites. Then, 2 ml chloroform was added to pool of polar extracts (of approximately 1 ml ) to obtain a biphasic system of approximately $2: 1$ volume-to-volume (v/v) chloroform/60\% methanol solution. After 5 min of separation, three technical replicates of $100 \mu \mathrm{l}$ of both phases were obtained. Methanol phase was derivatised using MCF prior to GC-QqQ-MS analysis and chloroform phase was derivatised using HCl in methanol prior to GC-Q-MS analysis.

Cell pellet was homogenised in $800 \mu \mathrm{l}$ chloroform for three intervals of 20 sec at 6500 rpm . Homogeniser was set to pause for 30 sec in between intervals. Samples were centrifuged for 5 min at 5000 rpm and $-9{ }^{\circ} \mathrm{C}$. Extracts were removed and three technical replicates of 100 $\mu \mathrm{l}$ obtained in vials.

Experimental procedure described in previous paragraph was repeated twice. Non-polar extracts were derivatised using HCl in methanol prior to GC-Q-MS analysis.

### 2.2.3 Optimised extraction method

A few adjustments were made to experimental procedure described in 2.2.2 to obtain the optimised method that is described in detail below.

Tumours were cut and weighed to obtain tissue samples of $30 \pm 10 \mathrm{mg}$. Tissue samples were placed in precooled $\left(-20^{\circ} \mathrm{C}\right) 2 \mathrm{ml}$ cryo tubes containing 0.50 g of ceramic (zirconium oxide) beads with a diameter of 1.4 mm . To maintain low temperatures, tubes were kept on ice.

To extract polar metabolites, homogenisation of tissue in a $60 \%$ methanol solution was performed. $400 \mu \mathrm{l} 60 \%$ methanol and $10 \mu \mathrm{l} 10 \mathrm{mM}$ internal standard (d3-alanine) were added to tubes. Tissue samples were homogenised for three intervals of 20 sec at 5500 rpm using the beads-based homogeniser. Homogeniser was set to pause for 30 sec in between intervals. Samples were centrifuged for 5 min at 5000 rpm and $-9^{\circ} \mathrm{C}$. Extracts were removed and stored in tubes until later.

Experimental procedure described in previous paragraph was repeated twice with the only change being addition of other internal standards ( d 4 -succinate and d8-valine for second and third round, respectively). Extracts were pooled and vortexed. Three technical replicates of $100 \mu \mathrm{l}$ obtained in 5 ml polypropylene (PP) tubes. Pool of extracts were vortexed in between obtaining replicates.

To extract non-polar metabolites, homogenisation of tissue in chloroform was performed. $800 \mu \mathrm{l}$ chloroform and $10 \mu \mathrm{l} 10 \mathrm{mM}$ internal standard (d31-palmitic acid) were added to tubes containing beads and cell pellet. Tissue samples were homogenised for three intervals of 20 sec at 6500 rpm . Homogeniser was set to pause for 30 sec in between intervals. Samples were centrifuged for 5 min at 5000 rpm and $-9^{\circ} \mathrm{C}$. Extracts were removed and stored until later.

Experimental procedure described in previous paragraph was repeated twice with the only change being addition of another internal standards (d35-stearic acid). Extracts were pooled. Three technical replicates of $100 \mu \mathrm{l}$ were obtained in GC-MS vials.

Polar extracts obtained from luminal-like and basal-like subtypes were derivatised using MCF prior to GC-QqQ-MS analysis. Polar extracts obtained from treated luminal-like subtypes were also derivatised using MCF but analysed using GC-Q-MS.

### 2.3 Derivatisation

Prior to derivatisation, extracts were vacuum-dried by a Savant SPD2010 SpeedVac concentrator (Thermo Electron Corporation).

### 2.3.1 MCF derivatisation of polar extracts

Samples were derivatised using an established protocol for methyl chloroformate (MCF) derivatisation, frequently described in literature [20]. However for an adaption to tissue, some changes were made. $390 \mu \mathrm{l} 1 \mathrm{M}$ sodium hydroxide ( NaOH ), $10 \mu \mathrm{l} 10 \mathrm{mM}$ internal standard (d5-glutamate), $333 \mu \mathrm{l}$ methanol and $67 \mu \mathrm{l}$ pyridine were added to vaccum-dried samples in PP tubes. The mixture was voretexed for 5 sec . To start the derivatisation
reaction, $80 \mu \mathrm{l}$ MCF was added followed by 1 min of vortexing. $400 \mu \mathrm{l}$ chloroform was added and mixture vortexed for 10 sec . Then, $400 \mu \mathrm{l} 50 \mathrm{mM}$ sodium bicarbonate $\left(\mathrm{NaHCO}_{3}\right)$ solution was added followed by 10 sec of vortexing. Lower chloroform phase was transferred to another PP tube and dried by adding a small portion of anhydrous sodium sulfate $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. After vortexing for 5 sec , water-free chloroform phase was transferred to a GC vial with insert and analysed using triple quadrupole GC-MS. Standard curve samples were obtained by making a dilution serie of a 1.408 mM MCF standard mix (std mix) in $1 \mathrm{M} \mathrm{NaOH} .100 \mu \mathrm{l}$ std mix dilutions of $1: 1,1: 3,1: 10$ and $1: 100$ were made to obtain a four point standard curve. Derivatisation was performed in PP tubes using the same procedure as samples.

## Absolute quantitation

Absolute quantitation was enabled by spiking samples with the 1.408 mM MCF std mix, separately derivatised with deuterated derivatisation reagents. To obtain std mix for spiking, often referred to as deuterised internal std mix (d3-istd), $100 \mu \mathrm{l}$ 1:3 diluted MCF std mix was derivatised following the same procedure as samples but using $300 \mu \mathrm{NaOH}$ and deuterated derivatisation reagents, respectively MCF and MeOH. The derivatisation of $100 \mu \mathrm{l}$ 1:3 diluted MCF std mix was repeated three times. All the derivatised std mixes were mixed before $170 \mu \mathrm{l}$ derivatised sample solution was spiked with $30 \mu \mathrm{l}$ std mix in an insert. Sample solution and std mix were carefully mixed with a pipette. Likewise, $170 \mu \mathrm{l}$ derivatised standard curve samples were spiked with $30 \mu \mathrm{l}$ of internal std mix. MCF derivatised samples were analysed using GC-QqQ-MS.

### 2.3.2 TMS derivatisation of polar extracts

$25 \mu \mathrm{l}$ internal standard ( $3 \mathrm{mg} / \mathrm{ml}$ d27-myristic acid in a $2: 5: 2(\mathrm{v} / \mathrm{v} / \mathrm{v}$ ) water:methanol:isopropanol) was added to samples in vials and empty vials, labelled d27 and FAME. $75 \mu \mathrm{l} 2 \mathrm{mM}$ std mix was added to empty vials, labelled std1 and std2. Vials were vacuum-dried. $10 \mu \mathrm{l}$ 1:100 FAME R1 solution ( 10 mM and equal volumes of fructose, glucose, lactose, maltose and raffinose) was added to FAME vial. $20 \mu \mathrm{l} 4 \%$ methoxyamine HCl in pyridine was added to samples, $50 \mu \mathrm{l}$ to d27, FAME and to an empty vial, labeled blank, and $100 \mu \mathrm{l}$ to std1 and std2. Vials were shaken at $30{ }^{\circ} \mathrm{C}$ for $90 \mathrm{~min} .20 \mu \mathrm{l}$ of samples,
d27, FAME and blank were transferred to new vials, $50 \mu \mathrm{l}$ of std1 and std2. $180 \mu \mathrm{l}$ MSTFA $+1 \%$ TMCS was added to samples, d27, std1, std2 and blank. $150 \mu \mathrm{l}$ to FAME. Vials were shaken at $37{ }^{\circ} \mathrm{C}$ for 30 min . Then cooled down to room temperature. Solutions were transferred to inserts and placed back into vials. Sequence order: 2xblank, d27, FAME, std1, samples, blank, std2. TMS derivatised samples were analysed using GC-Q-MS.

### 2.3.3 HCl in methanol derivatisation of non-polar extracts

Preparation of FAMEs from lipids and free fatty acids was performed with some changes to an experimental procedure described in literature[21]. $100 \mu \mathrm{l} 1.25 \mathrm{M}$ hydrogen chloride $(\mathrm{HCl})$ in methanol and $10 \mu \mathrm{l} 10 \mathrm{mM}$ internal standard (d27-myristic acid) were added to vacuum-dried samples in vial. The mixture was mixed and set to react overnight on a heating block of $50{ }^{\circ} \mathrm{C}$. In addition, derivatisation reagent was also added to a vial containing $10 \mu \mathrm{l}$ super standard mix.

The next day, samples were transferred to glass tubes. $100 \mu \mathrm{l}$ MQ-grade water and $300 \mu \mathrm{l}$ hexane were added to tubes followed by 10 sec vortexing. Upper layer hexane phase was transferred to a PP tube and a small portion of anhydrous $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ added. After vortexing for 5 sec , water-free hexane phase was transferred to vial with insert and analysed using a GC-Q-MS.

### 2.4 GC-MS analysis

### 2.4.1 Triple Quadrupole GC-MS

A Triple Quadrupole GC-MS (GC-QqQ-MS) system was employed. The GC-MS system included an Agilent 7890A GC coupled with an Agilent 7000B QqQ MS. A Triple-Axis HED-EM Detector was used. The GC was operated under a constant pressure of 1 bar and with helium as the carrier gas. d5-glutamate was used as a standard for retention time (RT) locking of method. Injection of $1 \mu \mathrm{l}$ sample was performed in the pulsed splitless mode into an inlet of $290{ }^{\circ} \mathrm{C}$. The GC compartment was equipped with a $30 \mathrm{~m} \times 250 \mu \mathrm{~m} \times$ $0.25 \mu \mathrm{~m}$ Agilent J\&W 122-5532G DB-5MS+DG capillary column. Its temperature was set at $325{ }^{\circ} \mathrm{C}$. Run time was 20 min including 3 min hold time at $40^{\circ} \mathrm{C}$, temperature gradient of $20{ }^{\circ} \mathrm{C} / \mathrm{min}$ to $320{ }^{\circ} \mathrm{C}$ and 3 min hold time. Post run was set to 5 min . Methane reagent gas was used for positive chemical ionisation (PCI). Multiple reaction monitoring (MRM) was applied. 19 cycles/sec and 78 cycles/sec were obtained by performing absolute quantitation and semi-quantitation, respectively. Std mix dilutions and d3-istd were analysed twice, prior and subsequent to samples. The most concentrated dilution was the first to be analysed.

### 2.4.2 Single Quadrupole GC-MS

A Single Quadrupole GC-MS (GC-Q-MS) was employed. The GC-MS system was an Agilent 7890A GC coupled with an Agilent 5975 inert Mass selective detector (MSD)/MS system. The GC compartment was equipped with a $30 \mathrm{~m} \times 250 \mathrm{~mm} \times 0.25 \mathrm{~mm}$ Agilent J\&W 122-5532G DB-5MS+DG capillary column. Electron impact (EI) source operated at 70 eV . Method settings were chosen according to derivatisation method.

## Subsequent to TMS derivatisation

d27-myristic acid was used as a standard for RT locking of method. Injection of $1 \mu \mathrm{l}$ sample was performed in the split mode using $1: 5$ split. Inlet temperature was set to $250{ }^{\circ} \mathrm{C}$. Run time was 37.5 min including 1 min hold time at $60^{\circ} \mathrm{C}$, temperature gradient of $10{ }^{\circ} \mathrm{C} / \mathrm{min}$ to $325{ }^{\circ} \mathrm{C}$ and 10 min hold time. Post run was set to 1 min . The scan mode started after 5.9 min with a mass range of $50-600$ a.m.u. at $2.66 \mathrm{scans} / \mathrm{sec}$.

## Subsequent to FAME analyis of lipids

d27-myristic acid was used as a standard for RT locking of method. Injection of $2 \mu \mathrm{l}$ sample was performed in the split mode using 1:5 split. Inlet temperature was set to $290{ }^{\circ} \mathrm{C}$. Run time was 35 min including 2 min hold time at $45^{\circ} \mathrm{C}$, temperature gradient of $10{ }^{\circ} \mathrm{C} / \mathrm{min}$ to $325{ }^{\circ} \mathrm{C}$ and 5 min hold time. Post run was set to 1 min . The scan mode started after 6 min with a mass range of $50-550 \mathrm{a} . \mathrm{m} . \mathrm{u}$. at $1.5 \mathrm{scans} / \mathrm{sec}$.

## Subsequent to MCF derivatisation of isotopically labelled metabolites

d5-glutamate was used as a standard for retention time (RT) locking of method. Injection of $2 \mu \mathrm{l}$ sample was performed in the splitless mode. Inlet temperature was set to $290{ }^{\circ} \mathrm{C}$. Run time was 35 min including 7.5 min hold time at $45^{\circ} \mathrm{C}$, temperature gradient of 10 ${ }^{\circ} \mathrm{C} / \mathrm{min}$ to $300{ }^{\circ} \mathrm{C}$ and 2 min hold time. Post run was set to 1 min . The scan mode started after 6 min with a mass range of 50-550 a.m.u. at $2.5 \mathrm{scans} / \mathrm{sec}$.

### 2.5 Q-TOF LC-MS analysis

A Quadrupole Time-of-Flight (Q-TOF) LC-MS was employed. The LC-MS system was an Agilent 6510 Q-TOF LC-MS including an Agilent 1200 Series LC. $500 \mu \mathrm{l}$ non-polar extracts were vaccum-dried and dissolved in $100 \mu \mathrm{l}$ dichloromethane (DCM). $10 \mu \mathrm{~L}$ of the DCM solution was directly infused into the MS compartment of the Q-TOF LC-MS. Flow was 0.2 $\mathrm{mL} / \mathrm{min}$. Electrospray ionisation (ESI) was carried out in positive ion mode with a gas temperature of $325{ }^{\circ} \mathrm{C}$. Fragmentor was set at 100 V . Mass-to-charge range was chosen as $100-1500 \mathrm{~m} / \mathrm{z}$. Reference masses were $121.050873 \mathrm{~m} / \mathrm{z}$ and $922.009798 \mathrm{~m} / \mathrm{z} .1 .03$ spectra was obtained per sec. Run time was 2 min .

### 2.6 Software for identification and quantitation

### 2.6.1 Chemstation for GC-Q-MS

Identification and semi-quantitation subsequent to GC-Q-MS analysis were enabled using Chemstation software, version E. 02.00.493 (Agilent Technologies) in combination with Automated Mass Spectral Deconvolution and Identification Software (AMDIS) (National Institute of Standards and Technology (NIST)).

## Construction of a semi-quantitative DRS method

For the semi-quantitation of FAMEs obtained from fatty acids and lipids, a Deconvolution Reporting Software (DRS) method was constructed. AMDIS and Chemstation libraries were made separately and then linked together. Compounds and their corresponding RT obtained from a previous run were added to an empty AMDIS library. Among added compounds were the corresponding methyl esters of palmitate, stearate, d27-myristic acid and "unknown RT 21.107". A tab file was made in order to make the Chemstation part. Target and qualifier ions were chosen for each compound based on the mass spectrum from the previous run. Target and qualifier ions, and also the calculated ratio of these ions were added to tab file. The constructed DRS method was performed on chloroform extracts subsequent to GC-Q-MS analysis. Peak area was obtained.

### 2.6.2 Masshunter for GC-QQQ-MS and Q-TOF LC-MS

Identification and absolute quantitation subsequent to GC-QqQ-MS analysis were enabled using Masshunter Workstation software - Qualitative Analysis, version B.04.00 (Agilent Technologies). Data obtained from Q-TOF LC-MS analysis were also investigated using Masshunter.

## 3 Results and discussion

A method for extraction of polar and non-polar metabolites from breast cancer xenograft tissue for GC-MS analysis was optimised (3.1). Metabolite profiles of luminal-like and basal-like subtypes of ductal carcinoma were obtained and compared (3.2). The passage of carbon 13 of glucose through pathways in luminal-like subtype was investigated (3.3). Non-polar extracts were analysed using LC-MS for investigation of lipids (3.4). At last, recommendations for future work were proposed (3.5).

### 3.1 Optimisation of a method for metabolite extraction

Beads-based homogenisation of tissue in polar and non-polar solvents was chosen as the basis for extraction. In combination with cooling, beads-beating directly in solvents has previously shown the potential to completely extract metabolites [12]. A methanol solution and chloroform were used as extraction solvents for polar and non-polar metabolites, respectively. Method was developed for xenograft tissue within the range of $30 \pm 10 \mathrm{mg}$. Method requirements included complete extraction of metabolites, simple performance and reproducible method. In order to fulfill the requirements several aspects of method were investigated; extraction of polar metabolites (3.1.1), extraction of non-polar metabolites (3.1.2), combining extraction of polar and non-polar metabolites (3.1.3), sample preparation of polar extracts (3.1.5) and reproducibility of method (3.1.4). Conclusions throughout method development were summarised (3.1.6) and at last, protocol for the optimised extraction method was obtained (3.1.7).

### 3.1.1 Extraction of polar metabolites

To extract polar metabolites, beads-based homogenisation of tissue in $60 \%$ methanol solution was performed. Metabolite concentrations were obtained by performing MCF derivatisation and GC-QqQ-MS analysis. Extraction round was defined as the experimental procedure including three 20 sec intervals of homogenisation, centrifugation and collection of extract. The number of extraction rounds needed in order to completely extract polar metabolites was investigated.

## Complete extraction with three extraction rounds in $60 \%$ methanol solution

Three extraction rounds in $400 \mu \mathrm{l} 60 \%$ methanol solution were performed. Metabolite concentrations were obtained each round. To investigate whether metabolites were completely extracted, percentage concentrations were calculated. The calculation of percentage concentration was based on the assumption that metabolites were completely extracted with three extraction rounds. Even if assumption is incorrect, percentage concentrations still serve as a good indication of completeness. Polar metabolites that achieved complete extraction with three extraction rounds are shown in Figure 3.1. Raw data, including incompletely extracted metabolites, are shown in Table C.8.

In this project, extraction was stated complete for the round obtaining $5 \%$ concentration. 16 polar metabolites obtained a percentage concentration of $5 \%$ or less in third extraction round and were therefore stated completely extracted. These metabolites were all high-abundance metabolites, contained in high concentrations in xenograft tissue. Several metabolites were not completely extracted. However, incomplete extraction was seen for low-abundance metabolites. Optimisation experiments were chosen to focus on some metabolites and for this reason, complete extraction was stated for polar metabolites. Therefore, three extraction rounds were shown to achieve complete extraction of polar metabolites.

## Reassurance of complete extraction

Another experiment was conducted to reassure that three extraction rounds achieve complete extraction of polar metabolites. It was investigated whether chloroform has the ability to break up more cells and thus enable more polar metabolites to leak out. Tissue was therefore homogenised in chloroform subsequent to the apparent complete extraction. To check for remaining polar metabolites, an extraction round in $60 \%$ methanol solution was performed. Percentage concentration was calculated for the last extraction round, the check round, and shown in Table 3.1 for metabolites previously stated as completely extracted. Corresponding data are shown in Table C.9.
Figure 3.1: Percentage concentration of polar metabolites obtained from each of the three extraction rounds of beads-based homogenisation in $60 \%$ methanol solution. Only metabolites completely extracted with three extraction rounds were included in the graph.

Table 3.1: Percentage concentration [\% $\mathrm{pM} / \mathrm{mg}$ ] to check for remaining polar metabolites obtained in extraction round in $60 \%$ methanol solution subsequent to the apparent complete extraction and homogenisation in chloroform.

| Metabolite | Percentage concentration |
| :--- | :---: |
| Lactate/Methylglyoxal | 0.9 |
| Alanine | - |
| Glycine | 1.8 |
| O-acetyl-L-serine | 4.0 |
| 2-aminobutyrate | - |
| Beta-alanine | - |
| Valine | 1.5 |
| Leucine | - |
| Malate | 1.0 |
| Threonine | - |
| Proline | - |
| Citrate | - |
| Glutamate | - |
| Cysteine | - |
| Histidine | - |
| Tyrosine | - |

All metabolites previously stated as completely extracted were obtained in so small concentrations or not obtained at all, that complete extraction with three extraction rounds was reassured. Cells were therefore adequately broken up during three extraction rounds in $60 \%$ methanol solution. Further breakage of cells using chloroform was not necessary as complete extraction of polar metabolites was already achieved. The $60 \%$ methanol solution was shown suitable as a polar extraction solvent.

## Conclusion

Complete extraction of polar metabolites is achieved with three extraction rounds in $400 \mu \mathrm{l}$ $60 \%$ methanol solution. Therefore three extraction rounds, each round including three intervals of 20 sec , are implemented in experimental procedure. Extracts were pooled in order to simplify method.

### 3.1.2 Extraction of non-polar metabolites

Beads-based homogenisation of tissue in chloroform was performed in order to extract non-polar metabolites. The number of extraction rounds needed to completely extract non-polar metabolites was investigated. Metabolites were quantitated by performing hydrogen chloride in methanol derivatisation and GC-Q-MS analysis. Extraction round was defined as previously (3.1.1).

## Incomplete extraction with five extraction rounds in $400 \mu$ l chloroform

Five extraction rounds, set at 6500 rpm , of tissue in $400 \mu \mathrm{l}$ chloroform was performed subsequent to polar metabolite extraction. Approximate values of metabolite abundance from chromatogram are shown in Figure 3.2. Corresponding data are shown in Table E.1.


Figure 3.2: Approximate abundance of non-polar metabolites obtained in each of the five extraction rounds in $400 \mu \mathrm{l}$ chloroform.

The extraction of non-polar metabolites was clearly incomplete as about the same abundances were obtained each round. This is likely due to saturation of chloroform. In order to extract more non-polar metabolites into chloroform, an increase in volume is necessary. Alternatively, number of extraction rounds can be increased.

Only three compounds were obtained with GC-Q-MS analysis. More compounds are likely to be obtained with a more sensitive analysis. However, optimisation experiments made use of GC-Q-MS analysis of non-polar extracts for investigation of complete extraction.

Complete extraction with five extraction rounds in $800 \mu \mathrm{l}$ chloroform

To address the issue of saturation, the volume of chloroform was increased to $800 \mu \mathrm{l}$. The alternative of increasing number of extraction rounds leads to a more labour-intensive method and was therefore not chosen. Five extraction rounds in $800 \mu \mathrm{l}$ chloroform was performed subsequent to polar metabolite extraction. Response of metabolites, given in peak area, was obtained and normalised against tissue weight as shown in Figure 3.3. Percentage response was obtained using the same assumption as for polar metabolites (3.1.1). Data are shown in Table E. 3 and mass spectra of unidentified compound in Figure E.1.


Figure 3.3: Percentage response, given in percentage peak area per tissue, obtained from each of the five extraction rounds in chloroform subsequent to polar metabolite extraction.

All non-polar metabolites obtained percentage response of about $5 \%$ in the third extraction round. Round four obtained responses even lower and round five no responses at all. Therefore, extraction was stated complete with three extraction rounds. Chloroform was shown suitable as a non-polar extraction solvent. Although complete extraction was shown for fatty acids, complete extraction was assumed for other non-polar metabolites.

## Conclusion

Complete extraction of non-polar metabolites is achieved with three extraction rounds in $800 \mu \mathrm{l}$ chloroform. Therefore three extraction rounds, each round including three intervals of 20 sec , are implemented in experimental procedure. Extracts were pooled in order to simplify method.

### 3.1.3 Combining extraction of polar and non-polar metabolites

Complete extraction of polar and non-polar metabolites was previously achieved (3.1.1 and 3.1.2). For this reason, approaches to combine and implement extractions of polar and non-polar metabolites into a single method were investigated. An approach to separately extract polar and non-polar metabolites was investigated. Separate extraction was achieved by performing polar metabolite extraction prior to and separately from non-polar metabolite extraction. Furthermore, an approach to simultaneously extract metabolites was investigated. Simultaneous extraction involved a biphasic system of methanol solution and chloroform.

## Separate extraction

Extraction of polar metabolites was performed prior to and separately from extraction of non-polar metabolites for a separate extraction. For the reason that polar metabolites were the first to be extracted, performance of extraction was identical to previous (3.1.1). Polar metabolites are therefore completely extracted using the separate approach. The subsequent extraction of non-polar metabolites was also stated complete because previous investigation (3.1.2) was also performed subsequent to polar metabolite extraction. Some
non-polar metabolites may be extracted into polar extracts thus complete extraction into chloroform is not achieved. However, this applies to a few out of many non-polar metabolites. Therefore the approach to separately extract metabolites is stated complete for polar and non-polar metabolites.

## Simultaneous extraction

A biphasic system of methanol solution and chloroform was obtained to enable simultaneous extraction of polar and non-polar metabolites. Chloroform was added to polar extract of methanol/water. The biphasic system was obtained subsequent to vortexing and centrifugation. Both phases were collected and analysed using GC-QqQ-MS and GC-Q-MS for methanol/water phase and chloroform phase, respectively. Polar metabolites obtained by analysis of methanol/water phase are shown in Table C.7. Low standard deviations of technical replicates indicate an homogenous methanol/water phase thus reproducible quantitation of polar metabolites.

Non-polar metabolites, however, were not obtained by analysing chloroform phase. An insensitive analysis could have been the explanation. However, a more likely explanation is that non-polar metabolites remain in tissue. Tissue was not present in the biphasic system because chloroform was exclusively added to extracts. The biphasic system described in literature, however, includes tissue [15, 14, 10].

An advantage of the simultaneous approach is that non-polar metabolites, with fairly polar charcteristics, cannot be lost in polar extracts.

## Conclusion

The approach to separately extract polar and non-polar metabolites was shown to achieve complete extraction. Separate extraction was therefore implemented in experimental procedure. In contrast, simultaneous extraction obtaining a biphasic system did not achieve complete extraction.

### 3.1.4 Method reproducibility and variation

Variation in metabolite concentrations were investigated for the reason that method is required reproducible. As a measure of variation, standard deviation based on concentrations of all technical replicates was calculated. Standard deviations obtained for basal-like and luminal-like subtypes based on three technical replicates are listed in Table C. 10 and Table C.11, respectively.

Standard deviations obtained from basal-like subtype were generally around $20 \%$. In contrast, tissue samples of luminal-like subtype resulted in great differences in standard deviation. Two tissue samples of luminal-like subtype were associated with high standard deviations of about $40 \%$, another with much lower. These high standard deviations represent great heterogeneity of extracts likely to be due to inadequate mixing. A reduction of standard deviations are required in order to state a reproducible and optimised method. To address this problem, extracts were vortexed in between the making of replicates. With this change in experimental procedure, adequate mixing was assumed. Therefore, method was stated reproducible.

In addition to variation created by extraction method, some variation is expected created by analysis. Day-to-day variation of analytical instruments are likely to affect the sensitivity of analysis which in turn affects concentration. Moreover, variation due to biological diversity cannot be avoided. However, it is assumed smaller than the variation created by experimental procedures. If not, experimental procedures would be highly inadequate. Similarily, there are always going to be some heterogeneity associated with tissue samples (3.2.2).

### 3.1.5 Sample preparation for polar extracts

Derivatisation methods for GC-QqQ-MS analysis of polar metabolites were evaluated in terms of their suitability to derivatise metabolites in breast cancer xenografts.

Class and number of metabolites obtained with MCF and TMS derivatisation of polar extracts were compared as shown in Table 3.2. The basis of the comparison is optimisation experiments, given in Table C. 7 to Table C.11. Metabolites only occasionally obtained are excluded from comparison.

Table 3.2: Metabolites obtained performing MCF and TMS derivatisation on polar extracts. Class of metabolites are included.

| Metabolites | MCF | TMS |
| :---: | :---: | :---: |
| Amino acids |  |  |
| Lactate/Methylglyoxal | x | x |
| Glutamate | x | x |
| O-acetyl-L-serine | x |  |
| Alanine | x |  |
| Glycine | x | x |
| Proline | x | x |
| Threonine | x |  |
| Aspartate | x |  |
| Serine | x | x |
| Lysine | x |  |
| Valine | x |  |
| Malate | x | x |
| Fumarate | x | x |
| Phenylalanine | x |  |
| Methionine | x |  |
| Leucine | x |  |
| Cysteine | x |  |
| Non-amino organic acids |  |  |
| Salicylate | x |  |
| Citrate | x | x |
| Glyoxylate | x |  |
| Succinate | x |  |
| Pyruvate | x |  |
| Phenylacetate | x |  |
| Beta-hydroxypyruvate | x |  |
| Benzoate | x |  |
| 3-methyl-oxovalerate | x |  |
| Citraconate/Itaconate | x |  |
| 4-methylvalerate | x |  |
| Malonate | x |  |
| 2-aminobutyrate | x |  |
| Urea |  | x |
| Phosphoric acid |  | x |
| alpha-ketoglutarate |  | x |
| Sugars |  |  |
| D-glucose |  | x |
| D-allose 2 |  | x |
| Maltose 1 |  | x |
| Sugar alcohols |  |  |
| Glycerol 1-phosphate |  | x |
| Sterols |  |  |
| Cholesterol |  | x |
| Other |  |  |
| Allantoin | x |  |
| Creatinine |  | x |
| O-phosphocolamine |  | x |
| Methyl-beta-D-galactopyranoside |  | x |

MCF and TMS derivatisation obtained 31 and 19 metabolites, respectively. TMS obtained sugar alcohol and sugars not obtained with MCF. The number of metabolites obtained with MCF was considerably greater than the number obtained with TMS. Therefore, MCF is associated with a more sensitive analysis of breast cancer xenografts. MCF was considered suitable for this purpose and was therefore implemented in experimental procedure of extraction method.

### 3.1.6 Optimised extraction method

Method requirements were shown fulfilled. Consequently, method was stated optimised. Complete extraction was achieved for polar and non-polar metabolites using an approach to separately extract metabolites. The first aim of this project was achieved.

## Extraction of polar and non-polar metabolites from breast cancer xenograft tissue for MS-based analysis

1. Cut and weigh tumours to obtain tissue samples of $30 \pm 10 \mathrm{mg}$.
2. Place tissue samples in precooled $\left(-20^{\circ} \mathrm{C}\right) 2 \mathrm{ml}$ cryo tubes containing 0.50 g of ceramic (zirconium oxide) beads with a diameter of 1.4 mm .
3. Keep tubes on ice to maintain low temperatures.

## Polar metabolite extraction

4. Add $400 \mu \mathrm{l} 60 \%$ methanol and $10 \mu \mathrm{l}$ internal standard to tubes.
5. Homogenise tissue samples for three intervals of 20 sec at 5500 rpm using Precellys 24 tissue homogeniser coupled with Cryolys cooling system (Bertin Technologies).
6. Centrifuge tissue samples at 5000 rpm and $-9^{\circ} \mathrm{C}$ for 5 min .
7. Remove extracts and store in tubes until later.
8. Repeat 4-7 twice to obtain a second and a third extraction round but add other internal standards.
9. Pool extracts obtained from same tissue sample. Vortex pooled extracts.
10. Make three technical replicates of $100 \mu \mathrm{l}$ in 5 ml polypropylene (PP) tubes. Vortex pooled extracts in between the making of replicates.

## Non-polar metabolite extraction

11. Add $800 \mu \mathrm{l}$ chloroform and $10 \mu \mathrm{l}$ internal standard to tubes containing beads and cell pellet.
12. Homogenise tissue samples for three intervals of 20 sec at 6500 rpm .
13. Centrifuge tissue samples at 5000 rpm and $-9^{\circ} \mathrm{C}$ for 5 min .
14. Remove extracts and store in tubes until later.
15. Repeat 11-14 twice to obtain a second and a third extraction round but add other internal standards.
16. Pool extracts obtained from same tissue sample. Vortex pooled extracts.
17. Make three technical replicates of $100 \mu \mathrm{l}$ in vials. Vortex pooled extracts in between the making of replicates.

## Sample preparation and analysis

18. Vacuum-dry polar and non-polar extracts.
19. Derivatise vacuum-dried polar extracts using MCF.
20. Analyse MCF-derivatised samples using GC-QQQ-MS.
21. Derivatise vacuum-dried non-polar extracts using HCl in methanol.
22. Analyse non-polar extracts using GC-Q-MS.

### 3.2 Metabolite profiling of polar extracts

Metabolite profiles of luminal-like and basal-like subtypes comprising metabolite concentrations were obtained (3.2.1). To obtain profiles, xenograft tissue from SCID mice was extracted following the optimised method and polar extracts analysed using GC-QqQ-MS. Furthermore, profiles were compared using $\log _{2}$ ratios, principal component analysis (PCA) and Student's t-test (3.2.2).

### 3.2.1 Metabolite profiles of luminal-like and basal-like subtypes

Six biological replicates, each producing one technical replicate, were the basis of metabolite profiles shown in Table 3.3. To represent uncertainty in concentration, standard deviation between biological replicates, in picomolar/mg tissue ( $\mathrm{pM} / \mathrm{mg}$ ) and in percentage, was included. Raw data are shown in Table C.12.

Metabolite profiles of luminal-like and basal-like subtypes comprise of 34 and 35 polar metabolites, respectively. The polar metabolites include amino and non-amino organic acids. Citraconate/itaconate was only found in basal-like subtype. Lactate and glycine, metabolites that have been noted to increase in breast cancers, were obtained in profiles [3].

Standard deviations range widely and are generally higher than 30\%. This indicates high heterogeneity of tissue samples. Moreover, heterogeneous tissue samples indicate heterogeneous tissue. Heterogeneity of samples was attempted reduced by cutting tumours in a certain way. The average of standard deviations obtained for luminal-like and basal-like subtypes was $36 \%$ and $30 \%$, respectively. The higher value obtained for luminal-like subtype indicates that tissue samples of luminal-like subtype are more heterogeneous than tissue samples of basal-like subtype. Therefore, it was suggested that luminal-like subtype possesses a more heterogeneous tissue than basal-like subtype. The suggestion is in accordance with earlier studies [17].

Table 3.3: Metabolite profiles of luminal-like and basal-like subtypes of breast cancer comprising polar metabolites, their concentration and corresponding standard deviation, given in $p M / m g$ tissue. Percentage standard deviation is given in paranthesis.

| Metabolite | Concentration $[$ pM $/ \mathbf{m g} \mid$ |  |  |  |  |  |
| :--- | ---: | :--- | :--- | ---: | :--- | :--- |
|  | Luminal-like |  |  | Basal-like |  |  |
| Lactate/Methylglyoxal | 59,886 | $\pm 23,501$ | $(39 \%)$ | 101,489 | $\pm 29,027$ | $(29 \%)$ |
| Glutamate | 36,735 | $\pm 11,160$ | $(30 \%)$ | 106,700 | $\pm 27,821$ | $(26 \%)$ |
| O-acetyl-L-serine | 33,629 | $\pm 6,752$ | $(20 \%)$ | 16,943 | $\pm 6,573$ | $(39 \%)$ |
| Allantoin | 32,652 | $\pm 10,904$ | $(33 \%)$ | 72,088 | $\pm 31,164$ | $(43 \%)$ |
| Alanine | 23,503 | $\pm 5,227$ | $(22 \%)$ | 19,397 | $\pm 6,351$ | $(33 \%)$ |
| Glycine | 17,326 | $\pm 4,416$ | $(25 \%)$ | 60,867 | $\pm 16,809$ | $(28 \%)$ |
| Salicylate | 12,742 | $\pm 5,295$ | $(42 \%)$ | 11,574 | $\pm 3,522$ | $(30 \%)$ |
| Proline | 15,265 | $\pm 4,651$ | $(30 \%)$ | 17,547 | $\pm 3,326$ | $(19 \%)$ |
| Threonine | 10,473 | $\pm 2,561$ | $(24 \%)$ | 9,844 | $\pm 3,023$ | $(31 \%)$ |
| Glyoxylate | 3,388 | $\pm 1,438$ | $(42 \%)$ | 8,031 | $\pm 5,105$ | $(64 \%)$ |
| Aspartate | 8,233 | $\pm 3,270$ | $(40 \%)$ | 1,569 | $\pm 571$ | $(36 \%)$ |
| Citrate | 6,628 | $\pm 2,633$ | $(40 \%)$ | 15,829 | $\pm 3,779$ | $(24 \%)$ |
| Serine | 4,347 | $\pm 2,726$ | $(63 \%)$ | 3,633 | $\pm 1,252$ | $(34 \%)$ |
| Lysine | 4,277 | $\pm 1,581$ | $(37 \%)$ | 10,331 | $\pm 2,906$ | $(28 \%)$ |
| Valine | 3,608 | $\pm 1,341$ | $(37 \%)$ | 3,414 | $\pm 870$ | $(25 \%)$ |
| Tyrosine | 2,293 | $\pm 875$ | $(38 \%)$ | 2,673 | $\pm 646$ | $(24 \%)$ |
| Malate | 1,819 | $\pm 1,204$ | $(66 \%)$ | 3,298 | $\pm 1,117$ | $(34 \%)$ |
| Succinate | 3,111 | $\pm 774$ | $(25 \%)$ | 3,001 | $\pm 492$ | $(16 \%)$ |
| Pyruvate | 1,457 | $\pm 857$ | $(59 \%)$ | 2,503 | $\pm 1,096$ | $(44 \%)$ |
| Fumarate | 1,197 | $\pm 404$ | $(34 \%)$ | 2,154 | $\pm 587$ | $(27 \%)$ |
| Phenylalanine | 1,630 | $\pm 513$ | $(31 \%)$ | 1,584 | $\pm 513$ | $(32 \%)$ |
| Citramalate | 895 | $\pm 272$ | $(30 \%)$ | 1,447 | $\pm 386$ | $(27 \%)$ |
| Methionine | 1,453 | $\pm 433$ | $(30 \%)$ | 1,296 | $\pm 281$ | $(22 \%)$ |
| Hippurate | 1,256 | $\pm 352$ | $(28 \%)$ | 1,302 | $\pm 285$ | $(22 \%)$ |
| Ornithine | 1,125 | $\pm 686$ | $(61 \%)$ | 647 | $\pm 306$ | $(47 \%)$ |
| Phenylacetate | 977 | $\pm 407$ | $(42 \%)$ | 1,042 | $\pm 241$ | $(23 \%)$ |
| Leucine | 975 | $\pm 344$ | $(35 \%)$ | 617 | $\pm 292$ | $(47 \%)$ |
| m-Toluate | 569 | $\pm 207$ | $(36 \%)$ | 550 | $\pm 107$ | $(19 \%)$ |
| Beta-hydroxypyruvate | 563 | $\pm 172$ | $(31 \%)$ | 1,059 | $\pm 370$ | $(35 \%)$ |
| Benzoate | 484 | $\pm 143$ | $(30 \%)$ | 753 | $\pm 185$ | $(25 \%)$ |
| 2-isopropylmalate | 451 | $\pm 118$ | $(26 \%)$ | 694 | $\pm 126$ | $(18 \%)$ |
| 3-methyl-oxovalerate | 260 | $\pm 94$ | $(36 \%)$ | 643 | $\pm 123$ | $(19 \%)$ |
| Citraconate/Itaconate |  | -179 | $\pm 53$ | $(30 \%)$ | 540 | $\pm 156$ |
| 4-methylvalerate | 179 | 275 | $\pm 69$ | $(25 \%)$ |  |  |
| 2-aminobutyrate | 245 | $\pm 80$ | $(33 \%)$ | 269 | $\pm 67$ | $(25 \%)$ |

### 3.2.2 Comparison of metabolite profiles

Metabolite profiles were compared in order to find differences and similarities in metabolite content between subtypes. Results from $\log _{2}$ ratios, Student's t-test and PCA are shown in Table 3.4.

## $\log _{2}$ ratios

$\log _{2}$ ratios of concentrations in metabolite profiles were obtained and presented in Figure 3.4. Ratios were calculated with concentration in luminal-like subtype as the nominator and with concentration in basal-like subtype as the denominator. As a consequence, positive ratios indicated a higher concentration in luminal-like than in basal-like subtypes. Negative ratios indicated the opposite. It was assumed that ratios with absolute values of 0.5 indicates a difference between subtypes. Therefore, the coloured area of the plot illustrates $\log _{2}$ ratios not associated with a difference in concentration.


Figure 3.4: $\log _{2}$ ratio of concentration in luminal-like subtype to basal-like subtype. Difference in concentration was assumed for ratios greater than absolute value of 0.5.

20 metabolites were found to differ in concentration between subtypes. Most of these metabolites were contained in higher concentrations in basal-like than in luminal-like subtypes, for instance glycine. This finding has previously been described in literature [3]. Metabolites that were suggested by $\log _{2}$ plot to be contained in different concentrations between subtypes are shown in Table 3.4.

## Principal component analysis (PCA)

PCA analysis was performed on concentrations used to obtain metabolite profiles, shown in Table C.12. A score plot of principal component 1 ( $\mathrm{PC}-1$ ) versus principal component 2 (PC-2) is shown in Figure 3.5. Each data point represents a biological replicate. Lines were drawn to link biological replicates in order to investigate for grouping of subtypes.

Score plot showed a grouping of subtypes into two distinct groups. The two distinct groups that appears in score plot indicate considerable differences between subtypes. Otherwise, these groups would have overlapped. Biological replicates of luminal-like subtype were seen spread in a greater extent than biological replicates of basal-like subtype. Luminal-like subtype is therefore associated with greater standard deviations and higher heterogeneity of tissue samples and tissue compared to basal-like subtype. This is in accordance with previous discussion of standard deviation and heterogeneity (3.2.1).

A loading plot of PC-1 versus PC-2 is shown in Figure 3.6. Each data point represents the concentration of a specific metabolite. Loading plot was compared with score plot in order to suggest correlations between metabolites and subtypes. Correlation with subtype was shown in Table 3.4 as it highlights difference between subtypes. Position of metabolite in loading plot coinciding with position of subtype in score plot suggests positive correlation between metabolite and subtype. Several metabolites were shown to positively correlate with basal-like subtype and a few with luminal-like subtype. Metabolites found contained in higher concentrations in luminal-like subtype than in basal-like subtype were compared with results of $\log _{2}$ ratios in Figure 3.4. Metabolites found contained in higher concentrations in basal-like subtype than in luminal-like subtype were also compared. All correlations found by PCA coincided with conclusions drawn from $\log _{2}$ ratios.


Figure 3.5: Score plot based on metabolite concentrations of six biological replicates of basallike and luminal-like subtypes obtained by performing PCA analysis.


Figure 3.6: Loading plot based on metabolite concentrations of six biological replicates of basal-like and luminal-like subtypes obtained by performing PCA analysis.

## Student's t-test

Metabolites that were shown to significantly differ in concentration between subtypes using Student's t-test are shown in Table 3.4. Student's t-test was performed on concentrations used to obtain metabolite profiles, shown in Table C.12. Probability describing likeliness of concentrations to obtain the same average was obtained. Student's t-test is therefore a great indication of whether a metabolite is contained in different concentrations in subtypes. Because two subtypes giving rise to two data sets were investigated, Stundent's t-test rather than analysis of variance (ANOVA) was chosen. Significance level was set to $5 \%$ meaning that a significant difference in concentration is stated for metabolites obtaining this probability. All metabolites associated with a significant difference between subtypes were also suggested contained in different concentrations by $\log _{2}$ ratios and PCA analysis.

## Differences and similarities in metabolite content between subtypes

Results of $\log _{2}$ ratios, PCA analysis and Student's t-test are shown in Table 3.4. $\log _{2}$ ratios and PCA analysis suggested that 20 metabolites differed in concentration between subtypes. However, significant difference were shown for 14 metabolites using Student's t-test. These metabolites can possibly be exploited in order to identify subtypes. Metabolite profiles were therefore shown a potential tool for identification of subtype.

Whether a difference was obtained due to a higher or a lower concentration in luminal-like subtype compared to basal-like subtype was readily seen in Figure 3.4. $\log _{2}$ ratios were shown in accordance with results from PCA analysis. Lactate, glycine, citrate and lysine were some of the metabolites associated with a significant difference and a higher concentration in basal-like subtype. Generally, basal-like subtype contained higher concentrations than luminal-like subtype. O-acetyl-L-serine and aspartate were some of the metabolites associated with a significant difference and a higher concentration in luminal-like than basal-like subtypes. In addition to possibly identify subtype, these metabolites may provide insight into metabolic changes in breast cancers. Furthermore, some of these metabolites can potentially serve as metabolic markers for prognosis and treatment response.

Table 3.4: Results of $\log _{2}$ ratios, PCA and Student's $t$-test highlighting differences and similarities in metabolite concentration between subtypes. x represents difference in concentration, blank similar concentration and $n / a$ not applicable. p-values associated with the $t$-test are shown in parantheses.

| Metabolite | Data analysis |  |  |
| :---: | :---: | :---: | :---: |
|  | $\log _{2}$ ratio | PCA | Student's t-test |
| Lactate/Methylglyoxal | x | x | $\mathrm{x}(2.2 \%)$ |
| Glutamate | x | x |  |
| O-acetyl-L-serine | x | x | x (0.1\%) |
| Allantoin | x | x | $\mathrm{x}(2.5 \%)$ |
| Alanine |  |  |  |
| Glycine | x | x | $\mathrm{x}(0.1 \%)$ |
| Salicylate |  |  |  |
| Proline |  |  |  |
| Threonine |  |  |  |
| Glyoxylate | x | x |  |
| Aspartate | x | x | x (0.4\%) |
| Citrate | x | x | x (0.1\%) |
| Serine |  |  |  |
| Lysine | x | x | $\mathrm{x}(0.2 \%)$ |
| Valine |  |  |  |
| Tyrosine |  |  |  |
| Malate | x | x |  |
| Succinate |  |  |  |
| Pyruvate | x | x |  |
| Fumarate | x | x | x (1.0\%) |
| Phenylalanine |  |  |  |
| Citramalate | x | x | $\mathrm{x}(1.9 \%)$ |
| Methionine |  |  |  |
| Hippurate |  |  |  |
| Ornithine | x | x |  |
| Phenylacetate |  |  |  |
| Leucine | x | x |  |
| m-Toluate |  |  |  |
| Beta-hydroxypyruvate | x | x | $\mathrm{x}(2.0 \%)$ |
| Benzoate | x | x | $\mathrm{x}(1.9 \%)$ |
| 2-isopropylmalate | x | x | $\mathrm{x}(0.6 \%)$ |
| 3-methyl-oxovalerate | x | x | $\mathrm{x}(0.0 \%)$ |
| Citraconate/Itaconate | $\mathrm{n} / \mathrm{a}$ | $\mathrm{n} / \mathrm{a}$ | $\mathrm{x}(0.0 \%)$ |
| 4-methylvalerate <br> 2-aminobutyrate | x | x | $\mathrm{x}(2.4 \%)$ |

### 3.3 Investigation of metabolic pathways using isotope labelling

Metabolic pathways of treated luminal-like subtype, shown to possess reduced tumour growth rates, was investigated using carbon-13 ( $\left.{ }^{13} \mathrm{C}\right)$-labelled glucose. Carbon 1 of glucose was replaced by ${ }^{13} \mathrm{C}$, giving rise to $\left[1-{ }^{13} \mathrm{C}\right]$ glucose. Injection of $\left[1-{ }^{13} \mathrm{C}\right]$ glucose into xenograft models was performed 10 min prior to tumour excision and halting of biological activity. Polar metabolites were extracted and analysis using GC-Q-MS.

To track the passage of ${ }^{13} \mathrm{C}$ through metabolic pathways, summed fractional labelling (SFL) was calculated directly from mass spectrum. See Appendix G, Equation G.1. SFL given in percentage and corrected for naturally occuring ${ }^{13} \mathrm{C}$ is shown in Table 3.5 for lacate, citrate, proline and fumarate.

Table 3.5: Summed fractional labelling (SFL) given in percentage for metabolites obtained in treated luminal-like subtype. Possible positions of ${ }^{13} \mathrm{C}$ in metabolite are included.

| Metabolite | Possible position of ${ }^{13} \mathbf{C}$ | SFL [\%] |
| :--- | :---: | :---: |
| Lactate | C-3 | 8.3 |
| Citrate | C-4 | 2.2 |
| Proline | C-4 | 0 |
| Fumarate | C-4 | 1.6 |

Metabolic pathways of glycolysis and tricarboxylic acid cycle (TCA) are illustrated in Figure 3.7. Formation of lactate and proline were also illustrated. Red coloured carbons represent ${ }^{13} \mathrm{C}$. The passage of ${ }^{13} \mathrm{C}$ shown in Figure 3.7 is only a suggestion based on literature [8].

SFL indicated ${ }^{13} \mathrm{C}$ labelling of $8.3 \%$ lactate, $2.2 \%$ citrate and $1.6 \%$ fumarate. Given this, it can be concluded that ${ }^{13} \mathrm{C}$ is retained throughout glycolysis. Moreover, ${ }^{13} \mathrm{C}$ enters the TCA cycle and gives rise to ${ }^{13} \mathrm{C}$-labelled TCA intermediates such as citrate and fumarate. This is in accordance with previous studies [9]. Figure 3.7 suggests that carbon 3 of lactate and carbon 4 of citrate and fumarate were labelled as a result of $\left[1-{ }^{13} \mathrm{C}\right]$ glucose.

No labelling was obtained for proline. It took 10 minutes from injection to excision, and this period may be inadequate to give rise to labelled proline. Furthermore, a very small fraction of alpha-ketoglutarate is likely to be labelled as only $2.2 \%$ citrate was obtained. Alpha-ketoglutarate can form other TCA inermediates in addition to proline. This reduces the possibility of labelled proline. SFL of glutamate, alanine and succinate were not calculated because their corresponding deuterised internal standards were previously added to extracts.

[1-13C]glucose


[3-13C]pyruvate


[2-13C]acetyl-CoA


Figure 3.7: Metabolic pathways in glycolysis, TCA cycle and in formation of lacate and proline drawn with ChemBioDraw Ultra software. Number of successive reactions (rx) is shown beside arrows, for arrows that represent more than one reaction. Possible passage of ${ }^{13} C$ from glucose labelled in position 1 is shown using red coloured carbons to represent ${ }^{13} C$.

SFL of metabolites in untreated luminal-like subtype may differ from the ones obtained above for the treated luminal-like subtype. Moreover, differences in metabolic pathways between subtypes may be noted using isotope labelling.

### 3.4 Q-TOF LC-MS analysis of non-polar extracts

Non-polar extracts were vaccum-dried, redissolved in dichloromethane (DCM) and infused directly into MS compartment of Q-TOF LC-MS. Total ion current (TIC) plot, showing mass scan data collected over time, were obtained using electrospray ionisation (ESI) in positive ion mode. Figure 3.8 shows TIC plots of blank (DCM) and sample obtained from non-polar extract.


Figure 3.8: TIC plots of blank at the top and sample at the bottom showing mass scan data collected over time.

The peaks obtained after 0.3 min in blank and sample are probably due to DCM. However, peak in sample is greater than the one in blank and may therefore contain non-polar metabolites as lipids. Sample obtained a slight increase of abundance from 0.4 min and on. This increase is not shown in blank and is therefore likely to represent non-polar metabolites. Mass fragments of peak and of the slight increase are shown in F. 1 and F.2, respectively. Injection of more concentrated samples and adjustment of LC-MS method settings are assumed to lead to more peaks in TIC plot of non-polar extracts.

### 3.5 Recommendations for future work

### 3.5.1 Loss of lipids in polar extracts

The approach to separately extract polar and non-polar metabolites was included in the optimised method (3.4). However, it is recommended to investigate a possible loss of lipids in polar extracts. Alternatively, if complete extraction is achieved by simultaneous extraction using a biphasic system, optimised method can include simultaneous extraction instead. Loss of lipids is not of any great concern for extractions using biphasic system.

### 3.5.2 Analysis of non-polar extracts

Optimisation was based on fatty acid analysis of non-polar extracts using GC-Q-MS. However, low sensitivity leading to detection of only three non-polar metabolites indicates inadequacy of method. An approach to overcome this problem, is to make samples more concentrated and to adjust method settings of GC-Q-MS.

Investigation of lipids in non-polar extracts can be performed using LC-MS analysis.

### 3.5.3 Metabolite profiles

Metabolite profiles were obtained using six biological replicates. To obtain more accurate profiles, extraction of more biological replicates are recommended. Prior to future extraction experiments, expiring dates of MCF and deuterated-MCF should be checked. It is suggested that an expired deuterated-MCF was the reason GC-QqQ-MS analysis obtained unexpected shapes of peaks. These peaks made data handling more difficult. For this reason, cysteine was excluded. Expiring date was assumed longer for MCF than deuterated-MCF, thus it was concluded that deuterated-MCF was expired [22].

To increase the number of metabolites presented in metabolite profiles, non-polar metabolites is suggested quantitated and included. Metabolite profiles representing more metabolite classes are desired.

In order to start using metabolite profiles in clinical settings, comprehensive investigation remains.

## 4 Conclusion

Optimal treatment of breast cancer remains a major challenge. To improve treatment, prediction of prognosis and treatment response have to become more reliable. Metabolite profiling of breast cancer tissue can potentially contribute to better techniques for prediction. The aim of this project was divided into three parts: Metabolite extraction and profiling, and isotope labelling of metabolites.

The first aim was to optimise a method for extraction of polar and non-polar metabolites for MS-based analysis. Beads-based homogenisation of tissue in $60 \%$ methanol solution and in chloroform was shown to fulfill method requirements, including complete extraction, simple performance and high reproducibility. Consequently, extraction method was successfully optimised.

The second aim was to obtain and compare metabolite profiles of breast cancer subtypes. Profiles comprised of polar metabolites and their concentration, given in $\mathrm{pM} / \mathrm{mg}$ tissue, were obtained for luminal-like and basal-like subtypes using the optimised extraction method in combination with GC-MS analysis. $\mathrm{Log}_{2}$ ratios, PCA analysis and Student's t-test were used to compare metabolite profiles, hence subtypes. $\log _{2}$ ratios and PCA pointed out that basal-like subtype is generally associated with higher concentrations than luminal-like subtype. PCA obtained a grouping of subtypes, indicating considerable differences in metabolite content between subtypes. 15 metabolites were found by Student's t-test to significantly differ in concentration between subtypes, including lactate, glycine, citrate, lysine, o-acetyl-L-serine and aspartate. Metabolite profiling is therefore a potential tool for identification of subtype. Established metabolite profiles may contribute to a more reliable prediction of prognosis and treatment response. Lactate and glycine have previously been noted to increase in breast cancers. The 13 other metabolites found to differ significantly, may increase the knowledge of metabolic changes in breast cancers and possibly serve as potential metabolic markers of breast cancer.

The third aim was to investigate metabolic pathways of treated luminal-like subtype using isotope labelling. Injection of $\left[1-{ }^{13} \mathrm{C}\right]$ glucose into xenograft models prior to tumour excision, extraction and GC-MS analysis was performed. Summed fractional labelling (SFL) indicated that ${ }^{13} \mathrm{C}$ was retained throughout glycolysis and that it entered the TCA cycle and gave rise to ${ }^{13} \mathrm{C}$-labelled TCA intermediates.

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## A List of abbreviations

Abbreviations used in report are listed in table A. 1 and A.2.

Table A.1: Abbreviations used in report

| Word | Abbreviation |
| :--- | :--- |
| Automated Mass Spectral Deconvolution and Identification Software | AMDIS |
| Average | avg |
| Basal-like | bas |
| Carbon 13 | ${ }^{13} \mathrm{C}$ |
| Carbon 13 at position 1 | $[1-13 \mathrm{C}]$ |
| Deconvolution Reporting Software | DRS |
| Dichloromethane | DCM |
| Deuterised internal standard | d3-istd |
| Deuterised methyl chloroformate | d3-MCF |
| Deviation | dev |
| Electron impact | EI |
| Electrospray ionisation | ESI |
| Estrogen receptor negative | ER- |
| Estrogen receptor positive | ER+ |
| Fatty acid methyl esters | FAME |
| Gas chromatography | GC |
| Gas chromatography mass spectrometry | GC-MS |
| Glucose | Glc |
| High Energy Dynode Detector - Electron Multiplier | HED-EM |
| Hydrogen chloride | HCl |
| Id est | i.e. |
| Liquid chromatography | LC |
| Liquid chromatography mass spectrometry | LC-MS |
| Luminal-like | lum |
| Mass selective detector | MSD |
| Mass spectrometry | MS |
| Mass-to-charge | m/z |
| Methyl chloroformate | MCF |
| Methyl tert-betyl ether | MTBE |
| Millimetre | mm |
| Micro | M |
| Minute | min |
| Molar | Milli-Q-grade |
| National Institute of Standards and Technology | MQ |
| Norwegian University of Science and Technology | Not applicable |
| Nuclear magnetic resonance | NTNT |
| Principal component analysis | n/a |
| Principal component | NMR |
| Pico | PCA |
|  | PC |
| P | p |

Table A.2: Abbreviations used in report continued

| Word | Abbreviation |
| :--- | :--- |
| Polypropylene | PP |
| Positive chemical ionisation | PCI |
| Quadrupole Time-of-Flight | Q-TOF |
| Reaction | rx |
| Retention time | RT |
| Second | sec |
| Severe combined immunodeficient | SCID |
| Single Quadrupole | Q |
| Sodium sulfate | $\mathrm{Na} \mathrm{SO}_{4}$ |
| Standard | std |
| Summed fractional labelling | SFL |
| Total ion current | TIC |
| Tricarboxylic acid cycle | TCA |
| Trimethylsilyl/-ation | TMS |
| Triple Quadrupole | QqQ |
| Volume | V |
| Volume-to-volume | v/v |

## B Tumour and sample names

Table B. 1 relates sample names to full tumour names, specific to each individual tumour. Section in report where experimental procedure is described is also given.

Sample name includes date of extraction and a number representing each individual sample. Samples extracted following the optimised method include subtype. Bas and lum denote basal-like and luminal-like subtypes, respectively. T represents treatment. Tumour name includes date of excision, subtype and location of mouse in cage. Treated xenograft models are not associated with any date but type of treatment. No supply of estrogen is denoted as -estrogen and injection of ${ }^{13} \mathrm{C}$-labelled glucose into xenograft models as 13 C Glc.

Table B.1: Sample name, full tumour name and experimental procedure of extraction.

| Sample name | Full tumour name | Experimental |
| :---: | :---: | :---: |
| 180112 1-3 | 20.11.08 MAS98.12 1-1V | 2.2.1 |
| 130212 1-3, 1 check-3 check | 21.11.08 MAS98.12 2-4V | 2.2.2 |
| 210212 1-3 | 21.11.08 MAS98.12 2-4V | 2.2.2 |
| 280212 1-2 | 21.11.08 MAS98.12 2-1V | 2.2.2 |
| 140312 bas 1-bas 3 | 21.11.08 MAS98.12 2-1V | 2.2 .3 |
| 140312 lum 1-lum 3 | 07.11.08 MAS98.06 6-2V | 2.2.3 |
| 210312 lumT 1 | MAS98.06 10-6 -estrogen 13C Glc | 2.2 .3 |
| 210312 lumT 2-3 | MAS98.06 10-3 -estrogen 13C Glc | 2.2.3 |
| 210312 lumT 4-5 | MAS98.06 10-1 -estrogen 13C Glc | 2.2 .3 |
| 210312 lumT 6-7 | MAS98.06 10-2 -estrogen 13C Glc | 2.2.3 |
| 210312 lumT 8-9 | MAS98.06 10-4 -estrogen 13C Glc | 2.2 .3 |
| 210312 lumT 10-12 | MAS98.06 10-5 -estrogen 13C Glc | 2.2 .3 |
| 270312 bas 1 | 18.11.08 MAS98.12 2-3H | 2.2.3 |
| 270312 bas 2 | 18.11.08 MAS98.12 2-2H | 2.2 .3 |
| 270312 bas 3 | 18.11.08 MAS98.12 2-2V | 2.2 .3 |
| 270312 bas 4 | 18.11.08 MAS98.12 2-4H | 2.2 .3 |
| 270312 bas 5 | 21.11.08 MAS98.12 2-5V | 2.2.3 |
| 270312 bas 6 | 17.11.08 MAS98.12 1-3V | 2.2 .3 |
| 270312 lum 1 | 04.11.08 MAS98.06 4-1V | 2.2.3 |
| 270312 lum 2 | 07.11.08 MAS98.06 6-2H | 2.2 .3 |
| 270312 lum 3 | 07.11.08 MAS98.06 5-4H | 2.2 .3 |
| 270312 lum 4 | 07.11.08 MAS98.06 6-1V | 2.2.3 |
| 270312 lum 5 | 07.11.08 MAS98.06 6-2V | 2.2.3 |
| 270312 lum 6 | LAI1u H1 MRI MAS98.06 03.02.10 EMH | 2.2.3 |

## C MCF derivatisation of polar extracts

## C. 1 Raw data

Raw data were obtained in picomol per microliter injected sample. However, Table C. 3 is an exception with raw data in peak area per microliter injected sample.

Table C.1: Metabolite concentration of injected sample [pmol/ $\mu \mathrm{l}$ ] of methanol/water phase in the biphasic system obtained subsequent to three extraction rounds in $60 \%$ methanol. Each tissue sample obtained three technical replicates.

| Metabolite | Sanple name 280212 |  |  |  |  |  |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: |
|  | 11 | 12 | 13 | 21 | 22 | 2.3 |
|  | 3.94 | 3.05 | 3.68 | 3.40 | 3.22 | 3.27 |
|  | 11.73 | 11.13 | 11.34 | 17.01 | 15.46 | 12.75 |
|  | 8.53 | 8.44 | 8.11 | 9.11 | 9.49 | 8.13 |
|  | 14.08 | 14.66 | 13.71 | 15.80 | 16.17 | 14.93 |
|  | 130.05 | 142.16 | 138.25 | 194.66 | 192.09 | 175.06 |
| Succinate | 9.42 | 9.89 | 9.56 | 11.54 | 12.86 | 11.55 |
| Citraconate/Itaconate | 7.34 | 7.79 | 7.19 | 8.79 | 8.34 | 9.07 |
| Benzoate | 4.01 | 3.91 | 3.57 | 3.50 | 3.57 | 3.26 |
| Glyoxylate | 33.37 | 36.14 | 32.92 | 34.32 | 35.80 | 34.95 |
| Beta-3-hydroxybutyrate | - | 5.03 | 3.52 | - | - | - |
| Glycine | 98.71 | 113.99 | 107.35 | 127.89 | 134.35 | 127.89 |
| O-acetyl-L-serine | 61.64 | 70.20 | 65.36 | 80.59 | 85.98 | 78.91 |
| Alanine | 5.88 | 7.46 | 6.72 | 10.17 | 11.48 | 10.00 |
| Phenylacetate | 0.63 | 0.52 | 0.39 | 0.39 | 0.33 | 0.30 |
| Salicylate | 30.15 | 23.46 | 23.95 | 23.94 | 19.33 | 19.62 |
| 2-aminobutyrate | 3.33 | 3.32 | 3.23 | 3.74 | 3.80 | 3.54 |
| m-Toluate | 0.17 | 0.12 | 0.08 | 0.07 | 0.07 | 0.05 |
| Beta-alanine | 2.75 | 30.17 | 28.47 | 37.25 | 38.31 | 36.43 |
| Adipate | 7.31 | 8.80 | 9.46 | 8.79 | 9.35 | 8.58 |
| Valine | 6.56 | 7.75 | 7.57 | 10.33 | 10.88 | 9.85 |
| Alpha-ketoglutarate | 43.74 | 46.66 | 43.82 | 68.60 | 64.07 | 56.80 |
| Beta-hydroxypyruvate | 25.23 | 26.92 | 24.83 | 26.98 | 27.42 | 24.69 |
| Leucine | 18.12 | 19.53 | 19.31 | 23.54 | 24.39 | 23.29 |
| Isoleucine | 19.29 | 21.24 | 20.76 | 25.79 | 25.98 | 24.99 |
| Threonine | 51.18 | 59.75 | 53.50 | 66.63 | 69.09 | 67.22 |
| Malate | 50.91 | 64.64 | 58.64 | 71.11 | 71.42 | 63.32 |
| Proline | 62.26 | 68.23 | 67.02 | 80.90 | 8.34 | 77.81 |
| Citrate | 33.60 | 36.69 | 37.25 | 43.69 | 46.65 | 44.71 |
| Serine | 37.64 | 40.58 | 39.51 | 43.43 | 43.49 | 40.51 |
| Allantoin | 76.10 | 82.83 | 80.12 | 85.00 | 98.96 | 92.56 |
| Glutamate | 9.65 | 110.06 | 108.32 | 126.32 | 144.84 | 134.74 |
| Methionine | 11.81 | 20.66 | 13.92 | 16.11 | 26.25 | 17.93 |
| Phenylalanine | 26.37 | 27.53 | 27.49 | 31.51 | 32.59 | 31.15 |
| Cysteine | - | 99.58 | - | 99.58 | 99.58 | - |
| d5-Glutamate | 711.35 | 801.99 | 779.16 | 839.02 | $1,052.22$ | 958.72 |
| d8-Valine | 59.43 | 65.77 | 65.04 | 63.95 | 65.07 | 61.56 |
| d4-Succinate | 54.65 | 58.00 | 56.55 | 56.49 | 62.73 | 56.20 |
| d3-Alanine | 77.22 | 85.22 | 81.47 | 84.79 | 88.00 | 83.72 |


in 60\% methanol.





Mincor oro

Table C.3: Metabolite response per injected sample [peak area/ $\mu$ l] of polar extracts obtained with three extraction rounds and obtained in a check round subsequent to extraction of non-polar metabolites. Response is an average of three technical replicates.

Table C.6: Metabolite concentration of injected sample [pmol/ $\mu$ l] of polar extracts obtained by performing optimised extraction


## C. 2 Normalised data

Normalisation against internal standards, dilutions, volume analysed and tissue weight was performed on raw data. As a result, metabolite concentrations are given in picomole per mg ( $\mathrm{pM} / \mathrm{mg}$ ) tissue.

Table C.7: Metabolite concentration per tissue mass [pM/mg] of methanol/water phase in the biphasic system obtained subsequent to three extraction rounds in $60 \%$ methanol. Each of the three technical replicates are shown.

| Sanple name 280212 | 11 | 12 | 13 | 21 | 22 | 23 | Std dev 1 | Std dev 2 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Tissue weight [mg] | 28.05 |  |  | 39.82 |  |  |  |  |
| Malonate | 8,204 | 4,341 | 5,851 | 3,463 | 2,230 | 3,086 | 26 | 18 |
| Pyruvate | 24,430 | 15,860 | 18,050 | 17,327 | 10,707 | 12,020 | 19 | 21 |
| 3-methyl-oxovalerate | 17,767 | 12,029 | 12,904 | 9,278 | 6,573 | 7,661 | 18 | 14 |
| Fumarate | 29,328 | 20,892 | 21,816 | 16,094 | 11,196 | 14,075 | 16 | 15 |
| Lactate/Methylglyoxal | 270,907 | 202,620 | 220,063 | 198,266 | 133,034 | 165,022 | 13 | 16 |
| Succinate | 19,616 | 14,096 | 15,221 | 11,752 | 8,904 | 10,884 | 15 | 11 |
| Citraconate/Itaconate | 15,293 | 11,107 | 11,444 | 8,955 | 5,773 | 8,547 | 15 | 18 |
| Benzoate | 8,355 | 5,573 | 5,679 | 3,560 | 2,472 | 3,072 | 20 | 15 |
| Glyoxylate | 69,517 | 51,505 | 52,408 | 34,958 | 24,791 | 32,947 | 14 | 14 |
| Beta-3-hydroxybutyrate | - | 7,174 | 5,599 | - | - | 120,50 | 12 | - |
| Glycine | 205,628 | 162,464 | 170,871 | 130,264 | 93,044 | 120,561 | 10 | 14 |
| O-acetyl-L-serine | 128,402 | 100,050 | 104,043 | 82,084 | 59,545 | 74,391 | 11 | 13 |
| Alanine | 12,239 | 10,627 | 10,697 | 10,362 | 7,950 | 9,429 | 7 | 11 |
| Phenylacetate | 1,310 | 739 | 627 | 393 | 226 | 286 | 34 | 23 |
| Salicylate | 62,803 | 33,442 | 38,127 | 24,387 | 13,384 | 18,498 | 29 | 24 |
| 2-aminobutyrate | 6,935 | 4,726 | 5,139 | 3,809 | 2,634 | 3,341 | 17 | 15 |
| m-Toluate | 348 | 176 | 129 | 69 | 46 | 48 | 43 | 19 |
| Beta-alanine | 55,715 | 42,999 | 45,316 | 37,935 | 26,535 | 34,345 | 12 | 14 |
| Adipate | 15,232 | 12,545 | 15,060 | 8,949 | 6,474 | 8,086 | 9 | 13 |
| Valine | 13,661 | 11,046 | 12,043 | 10,524 | 7,532 | 9,288 | 9 | 13 |
| Alpha-ketoglutarate | 91,111 | 66,506 | 69,758 | 69,867 | 44,373 | 53,548 | 14 | 19 |
| Beta-hydroxypyruvate | 52,557 | 38,374 | 39,520 | 27,477 | 18,988 | 23,272 | 15 | 15 |
| Leucine | 37,743 | 27,839 | 30,736 | 23,974 | 16,888 | 21,952 | 13 | 14 |
| Isoleucine | 40,189 | 30,276 | 33,041 | 26,265 | 17,992 | 23,557 | 12 | 15 |
| Threonine | 106,623 | 85,165 | 85,162 | 67,861 | 47,849 | 63,369 | 11 | 14 |
| Malate | 106,055 | 92,127 | 93,344 | 72,429 | 49,459 | 59,695 | 6 | 16 |
| Proline | 129,706 | 97,244 | 106,675 | 82,401 | 57,028 | 73,350 | 12 | 15 |
| Citrate | 69,995 | 52,294 | 59,297 | 44,495 | 32,306 | 42,147 | 12 | 13 |
| Serine | 78,415 | 57,839 | 62,894 | 44,231 | 30,121 | 38,187 | 13 | 15 |
| Allantoin | 158,537 | 118,049 | 127,527 | 86,578 | 68,533 | 87,252 | 13 | 11 |
| Glutamate | 207,588 | 156,858 | 172,426 | 128,664 | 100,310 | 127,019 | 12 | 11 |
| Methionine | 24,609 | 29,442 | 22,157 | 16,404 | 18,177 | 16,901 | 12 | 4 |
| Phenylalanine | 54,932 | 39,244 | 43,765 | 32,095 | 22,569 | 29,366 | 14 | 14 |
| Cysteine | - | 141,927 | - | 101,424 | 68,965 | - | - | 19 |

Table C.8: Metabolite concentration per tissue mass [pM/mg] of polar extracts obtained in each of the three extraction rounds in

| Sample name 180112 | 1 r 1 | 1 r 2 | 1 r 3 | 2 r 1 | 2 r 2 | 2 r 3 | 3 rl | 3 r 2 | 3 r 3 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Tissue weight [mg] | 38.9 |  |  | 28.7 |  |  | 32.8 |  |  |
| 4-methylvalerate | 112 | 98 | 83 | 91 | 91 | 82 | 76 | 75 | 71 |
| Malonate | 167 | 181 | 200 | 358 | 357 | 258 | 234 | 221 | 219 |
| Pyruvate | 2,055 | 1,041 | 582 | 2,679 | 1,022 | 587 | 1,424 | 643 | 406 |
| 3-methyl-oxovalerate | 31 | 9 | 9 | 22 | 12 | 8 | 36 | 5 | 5 |
| Fumarate | 1,950 | 532 | 150 | 1,869 | 683 | 172 | 1,600 | 311 | 96 |
| Lactate/Methylglyoxal | 430,795 | 99,480 | 19,791 | 996,586 | 263,163 | 46,407 | 547,093 | 66,431 | 14,715 |
| Succinate | 3,943 | 974 | 512 | 8,982 | 2,986 | 981 | 4,506 | 835 | 484 |
| Benzoate | 223 | 202 | 166 | 247 | 271 | 205 | 155 | 164 | 160 |
| Beta-3-hydroxybutyrate | 6,076 | 1,587 | 564 | 7,025 | 2,358 | 685 | 5,831 | 5,695 | 656 |
| Alanine | 48,236 | 9,479 | 1,826 | 72,053 | 16,023 | 2,856 | 52,057 | 6,210 | 1,263 |
| Glycine | 59,023 | 11,641 | 2,466 | 67,372 | 16,404 | 3,256 | 63,964 | 8,579 | 1,971 |
| O-acetyl-L-serine | 19,521 | 5,408 | 805 | 24,906 | 8,445 | 1,077 | 17,046 | 4,178 | 1,455 |
| Phenylacetate | 96 | 86 | 49 | 0 |  |  | 23 |  | 20 |
| 2 -aminobutyrate | 190 | 76 | 41 | 197 | 82 | 15 | 182 | 51 | 11 |
| m-Toluate | 223 | 191 | 143 | 159 | 175 | 153 | 121 | 126 | 126 |
| Beta-alanine | 4,363 | 1,134 | 299 | 4,918 | 1,388 | 357 | 4,666 | 783 | 229 |
| Adipate | 260 | 282 | 259 | 523 | 513 | 319 | 198 | 316 | 396 |
| Valine | 5,211 | 1,341 | 403 | 7,815 | 2,124 | 577 | 6,119 | 1,032 | 343 |
| 2-isopropylmalate | 139 | 136 | 98 | 110 | 116 | 108 | 93 | 87 | 88 |
| Beta-hydroxypyruvate | 897 | 376 | 164 | 1,100 | 452 | 182 | 1,276 | 252 | 205 |
| Alpha-ketoadipate | 951 | 821 | 856 | 15,675 | 5,180 | 1,562 | 541 | 713 | 559 |
| Leucine | 4,487 | 1,113 | 299 | 8,089 | 2,009 | 475 | 5,002 | 840 | 231 |
| Isoleucine | 2,481 | 665 | 288 | 3,819 | 1,136 | 440 | 2,932 | 544 | 272 |
| Malate | 4,076 | 635 | 88 | 3,000 | 600 | 87 | 2,447 | 231 | 41 |
| Oxaloacetate | 2,481 | 370 | 581 | 2,358 | 740 | 427 | 3,346 | 610 | - |
| Threonine | 9,160 | 1,946 | 534 | 8,919 | 3,247 | 681 | 10,679 | 1,550 | 450 |
| Proline | 13,255 | 2,766 | 575 | 18,288 | 4,417 | 828 | 16,683 | 2,118 | 444 |
| L-homoserine | 352 | 193 | 177 | 415 | 301 | 223 | 274 | 219 | 213 |
| Aspartate | 791 | 215 | 142 | 1,270 | 434 | 140 | 687 | 134 | 99 |
| Citrate | 7,198 | 1,273 | 301 | 5,568 | 1,709 | 305 | 5,218 | 677 | 150 |
| 5 -aminovalerate | 239 | 116 | 159 | 277 | 243 | 224 | 214 | 163 | 171 |
| Anthralinate | 106 | 122 | 159 | 103 | 148 | 175 | 79 | 89 | 139 |
| Serine | 4,555 | 1,372 | 520 | 7,320 | 2,690 | 834 | 5,218 | 1,119 | 422 |
| Allantoin | 20,569 | 8,551 | 5,817 | 57,871 | 13,193 | 8,954 | 47,652 | 8,468 | 16,208 |
| Glutamate | 62,266 | 9,138 | 2,157 | 74,816 | 18,724 | 3,237 | 58,825 | 6,469 | 1,530 |
| N-Acetyl-L-glutamate | 1,783 | 671 | 461 | 1,949 | 1,155 | 628 | 1,402 | 638 | 537 |
| Methionine | 1,119 | 295 | 102 | 1,713 | 515 | 143 | 1,429 | 260 | 93 |
| Hydroxyproline | 189,998 | 74,878 | 39,117 | 188,479 | 51,663 | 52,150 | 259,130 | 96,682 | 58,506 |
| Cysteine | 31,785 | 6,417 | 1,411 | 48,212 | 12,709 | 2,851 | 36,349 | 4,976 | 1,272 |
| Phenylalanine | 2,545 | 728 | 237 | 3,985 | 1,207 | 341 | 3,108 | 613 | 209 |
| Hippurate | 2,304 | 2,234 | 2,404 | 2,492 | 3,116 | 2,893 | 2,441 | 2,064 | 2,226 |
| Putrescine | 845 | 444 | 489 | 1,166 | 863 | 800 | 748 | 549 | 554 |
| 4-imidazoleacrylate | 112 | 74 | 73 | 94 | 88 | 47 | 56 | 38 | 37 |
| Histamine | 2,154 | 1,309 | 2,953 | 1,728 | 2,088 | - | 1,741 | 371 | 538 |
| Ornithine | 629 | 259 | 172 | 1,217 | 516 | 239 | 798 | 271 | 171 |
| Lysine | 10,690 | 3,225 | 1,121 | 16,014 | 5,080 | 1,474 | 12,653 | 2,901 | 1,253 |
| Histidine | 503 | 153 | - | 759 | 255 | - | 653 | 116 | - |
| Tyrosine | 1,814 | 473 | 129 | 2,925 | 830 | 172 | 2,223 | 380 | 107 |
| Thryptophane | 791 | 316 | 154 | 1,192 | 378 | - | 904 | 240 | 44 | 60\% methanol.

Table C.9: Metabolite response per tissue mass [peak area/mg] of polar extracts obtained with three extraction rounds and obtained in a check round subsequent to non-polar metabolite extraction.

| Sample name | 1302121 | 1302122 | 1302123 | 1302121 check | 1302122 check | 1302123 check | \% in check |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Tissue weight [mg] | 28.67 | 26.79 | 33.93 | 28.67 | 26.79 | 33.93 |  |
| 4-methylvalerate | 1,259 | 577 | 319 | - | - | - | - |
| Pyruvate | 13,790 | 21,842 | 41,400 | 219 | 238 | 243 | 0.90 |
| 3-methyl-oxovalerate | 5,013 | 3,638 | 6,245 | - | - | - | - |
| Fumarate | 83,164 | 62,980 | 110,651 | - | - | - | - |
| Lactate/Methylglyoxal | 500,413 | 397,915 | 665,661 | 5,048 | 4,841 | 4,703 | 0.92 |
| Succinate | 87,041 | 77,922 | 108,624 | 7,210 | 9,906 | - | - |
| Citraconate/itaconate | 7,302 | 7,652 | 8,303 | - | - | - | - |
| Benzoate | 24,584 | 13,270 | 12,283 | - | - | - | - |
| Alanine | 195,603 | 165,563 | 207,265 | - | - | - | - |
| Glyoxylate | 6,611 | 4,846 | 5,868 | - |  | - | - |
| Glycine | 934,291 | 934,073 | 1,348,812 | 19,286 | 20,809 | 18,192 | 1.78 |
| O-acetyl-L-serine | 28,512 | 31,026 | 40,784 | 1,285 | 1,545 | 1,353 | 4.00 |
| 2 -aminobutyrate | 6,839 | 5,815 | 6,167 | - | - |  | - |
| Salicylate | 14,907 | 8,131 | 6,635 | - | - | - | - |
| beta-alanine | 37,243 | 35,065 | 53,374 | - | - | - | - |
| Adipate | 28,037 | 35,158 | 39,849 | - | - | - | - |
| Valine | 69,628 | 68,571 | 98,703 | 1,163 | 1,279 | 1,148 | 1.49 |
| beta-hydroxypyruvate | 4,642 | 7,265 | 11,595 | 203 | 201 | 199 | 2.50 |
| Leucine | 19,083 | 20,247 | 30,748 | - | - | - | - |
| Isoleucine | 3,339 | 2,820 | 4,300 | - | - | - | - |
| Malate | 57,617 | 73,003 | 67,361 | 728 | 789 | 541 | 1.03 |
| OAA | 681 | 984 | 1,178 | - | - | - | - |
| Threonine | 6,825 | 4,948 | 7,122 | - | - | - | - |
| Proline | 55,360 | 41,388 | 70,220 | - | - | - | - |
| Aspartate | 590 | 633 | 840 | - | - | 223 | - |
| Citrate | 19,220 | 17,055 | 22,952 | - | - | - | - |
| Allantoin | 38,123 | 48,035 | 47,717 | 14,389 | 14,231 | 12,548 | 23.52 |
| Serine | 10,993 | 14,201 | 16,555 | 156 | 211 | 176 | 1.28 |
| Glutamate | 358,980 | 384,203 | 434,474 | 1,625 | 1,251 | 1,427 | - |
| N -Acetyl-L-glutamate | 5,644 | 7,427 | 7,055 | 2,388 | 2,422 | 2,153 | 25.70 |
| Methionine | 60,133 | 61,918 | 91,791 | 247 | 183 | 236 | - |
| Cysteine | 36,901 | 40,632 | 86,629 | - | - | - | - |
| Phenylalanine | 37,287 | 38,221 | 57,229 | 239 | 144 | 253 | - |
| Histamine | 392 | 287 | 415 | - | - | - | - |
| P-coumarate | 2,011 | 1,207 | 1,092 | - | - | - | - |
| Lysine | 16,429 | 16,500 | 20,894 | - | - | - | - |

Table C.10: Metabolite concentration per tissue mass $[\mathrm{pM} / \mathrm{mg}$ / of polar extracts obtained by performing optimised extraction

| Sample name 140312 | bas 11 | bas 12 | bas 13 | bas 21 | bas 22 | bas 23 | bas 31 | bas 32 | bas 33 | Std dev [\%] | Std dev [\%] | Std dev [\%] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Tissue weight [mg] |  | 22.8 |  |  | 28.13 |  |  | 25.7 |  | bas 1 | bas 2 | bas 3 |
| 4-methylvalerate | 409 | 369 | 447 | 347 | 154 | 274 | 289 | 243 | 275 | 8 | 31 | 7 |
| Pyruvate | 3,129 | 4,245 | 11,502 | 1,858 | 1,590 | 2,378 | 3,874 | 3,673 | 3,393 | 59 | 17 | 5 |
| 3-methyl-oxovalerate | 837 | 672 | 1,068 | 472 | 276 | 451 | 482 | 442 | 429 | 19 | 22 | 5 |
| Fumarate | 4,084 | 3,757 | 5,686 | 1,640 | 1,182 | 1,817 | 3,072 | 2,974 | 2,816 | 19 | 17 | 4 |
| Lactate/Methylglyoxal | 224,529 | 169,832 | 329,893 | 112,703 | 68,354 | 106,929 | 154,345 | 138,446 | 111,517 | 28 | 21 | 13 |
| Succinate | 4,411 | 4,075 | 5,093 | 1,789 | 1,613 | 2,567 | 2,139 | 2,017 | 1,894 | 9 | 21 | 5 |
| Citraconate/Itaconate | 1,409 | 988 | 1,603 | 711 | 339 | 560 | 644 | 521 | 471 | 19 | 29 | 13 |
| Benzoate | 2,610 | 2,281 | 2,150 | 1,440 | 896 | 1,464 | 1,238 | 1,220 | 1,121 | 8 | 21 | 4 |
| Citramalate | 570 | 559 | 492 | 321 | 187 | 322 | 300 | 270 | 238 | 6 | 23 | 9 |
| Glyoxylate | 5,871 | 5,484 | 4,975 | 5,817 | 2,332 | 3,778 | 3,222 | 2,261 | 1,713 | 7 | 36 | 26 |
| Beta-3-hydroxybutyrate | 1,267 | 1,364 | 1,959 | 1,064 | 620 | 938 | 1,165 | 990 | 1,102 | 20 | 21 | 7 |
| Glycine | 187,563 | 173,249 | 320,437 | 81,366 | 48,654 | 68,713 | 102,382 | 93,253 | 68,477 | 29 | 20 | 16 |
| Alanine | 41,966 | 38,991 | 67,768 | 18,392 | 11,699 | 17,674 | 23,697 | 21,814 | 19,968 | 26 | 19 | 7 |
| O-acetyl-L-serine | 19,527 | 18,280 | 38,631 | 5,887 | 5,275 | 7,010 | 11,860 | 11,643 | 10,966 | 37 | 12 | 3 |
| Nicotinate | 1,676 | 1,605 | 2,101 | 1,206 | 581 | 1,090 | 1,296 | 1,408 | 1,794 | 12 | 28 | 14 |
| Phenylacetate | 477 | 766 | 439 | 449 | 198 | 313 | 319 | 348 | 304 | 26 | 32 | 6 |
| Salicylate | 6,943 | 3,176 | 6,438 | 2,268 | 3,313 | 4,700 | 5,909 | 3,722 | 4,262 | 30 | 29 | 20 |
| 2-aminobutyrate | 819 | 795 | 991 | 472 | 257 | 461 | 477 | 456 | 438 | 10 | 25 | 3 |
| m -Toluate | 580 | 530 | 510 | 389 | 228 | 361 | 367 | 321 | 302 | 5 | 22 | 8 |
| Beta-alanine | 11,152 | 9,518 | 15,362 | 5,969 | 3,321 | 5,148 | 6,570 | 6,491 | 6,094 | 21 | 23 | 3 |
| OH-Glutarate | 1,264 | 1,205 | 1,406 | 639 | 324 | 713 | 677 | 687 | 379 | 7 | 30 | 25 |
| Adipate | 791 | 734 | 692 | 492 | 303 | 467 | 496 | 399 | 412 | 6 | 20 | 10 |
| Valine | 13,870 | 6,514 | 10,357 | 3,301 | 2,089 | 3,225 | 4,164 | 3,887 | 3,255 | 29 | 19 | 10 |
| Alpha-ketoglutarate | 11,940 | 12,234 | 23,060 | 7,236 | 3,576 | 6,149 | 6,560 | 5,410 | 4,742 | 33 | 27 | 13 |
| Beta-hydroxypyruvate | 2,981 | 2,625 | 6,694 | 1,045 | 601 | 1,001 | 1,448 | 1,279 | 1,048 | 45 | 23 | 13 |
| Alpha-ketoadipate | 1,218 | 528 | 570 | 486 | 269 | 459 | 444 | 303 | 309 | 41 | 24 | 19 |
| Leucine | 2,259 | 1,962 | 3,260 | 852 | 630 | 903 | 1,324 | 1,315 | 1,044 | 22 | 15 | 11 |
| Gamma-aminobutyrate | 3,002 | 3,071 | 3,012 | 2,465 | 1,161 | 1,992 | 2,226 | 1,938 | 2,148 | 1 | 29 | 6 |
| Threonine | 9,991 | 9,754 | 15,706 | 4,236 | 2,734 | 4,549 | 6,161 | 5,683 | 5,265 | 23 | 21 | 6 |
| Malate | 12,119 | 10,101 | 20,013 | 3,774 | 2,283 | 3,379 | 6,042 | 5,464 | 4,453 | 30 | 20 | 12 |
| Proline | 37,866 | 34,315 | 58,104 | 17,409 | 11,298 | 17,184 | 22,818 | 21,615 | 18,967 | 24 | 18 | 8 |
| Aspartate | 2,187 | 1,712 | 2,172 | 637 | 446 | 814 | 968 | 742 | 683 | 11 | 24 | 15 |
| Citrate | 12,128 | 13,811 | 17,172 | 4,788 | 3,675 | 6,718 | 7,207 | 7,284 | 6,851 | 15 | 25 | 3 |
| Serine | 5,671 | 5,422 | 9,684 | 2,128 | 1,294 | 2,302 | 3,704 | 3,776 | 3,106 | 28 | 23 | 9 |
| Anthralinate | 465 | 301 | 326 | 182 | 128 | 196 | 203 | 142 | 185 | 20 | 18 | 15 |
| Allantoin | 206,958 | 202,512 | 145,182 | 98,706 | 56,550 | 111,034 | 71,640 | 76,926 | 60,075 | 15 | 26 | 10 |
| N-Acetyl-L-glutamate | 16,633 | 12,908 | 11,952 | 8,146 | 6,072 | 10,392 | 7,131 | 7,072 | 6,109 | 15 | 22 | 7 |
| Glutamate | 210,411 | 199,599 | 290,175 | 71,749 | 53,090 | 91,190 | 114,826 | 103,243 | 92,525 | 17 | 22 | 9 |
| Methionine | 1,923 | 1,677 | 2,578 | 877 | 584 | 894 | 1,146 | 1,102 | 988 | 18 | 18 | 6 |
| Hippurate | 451 | 453 | 469 | 262 | 206 | 396 | 513 | 463 | 610 | 2 | 28 | 12 |
| Phenylalanine | 3,192 | 2,954 | 4,592 | 1,489 | 1,007 | 1,508 | 1,930 | 1,919 | 1,689 | 20 | 17 | 6 |
| P-coumarate | 219 | 202 | 126 | 64 | 37 | 46 | 42 | 35 | 32 | 22 | 22 | 12 |
| Ornithine | 5,182 | 4,322 | 5,978 | 2,817 | 1,427 | 2,541 | 3,037 | 2,872 | 2,423 | 13 | 27 | 9 |
| Lysine | 18,575 | 17,434 | 25,191 | 9,402 | 5,098 | 8,452 | 11,034 | 10,353 | 9,274 | 17 | 24 | 7 |
| Tyrosine | 5,383 | 3,546 | 6,058 | 2,320 | 1,368 | 2,570 | 2,877 | 2,353 | 2,297 | 21 | 25 | 10 |

Table C.11: Metabolite concentration per tissue mass [pM/mg] of polar extracts obtained by performing optimised extraction method on tissue of luminal-like subtype.

| Sample name 140312 | lum 11 | lum 12 | lum 13 | lum 21 | lum 22 | lum 23 | lum 31 | lum 32 | lum 33 | Std dev [\%] | Std dev [\%] | Std dev [\%] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Tissue weight [mg] |  | 28.09 |  |  | 24.76 |  |  | 33.44 |  | lum 1 | lum 2 | lum 3 |
| 4-methylvalerate | 623 | 267 | 330 | 440 | 238 | 174 | 131 | 141 | 109 | 38 | 40 | 11 |
| Pyruvate | 408 | 2,039 | 4,310 | 4,054 | 2,733 | 1,751 | 2,164 | 1,810 | 2,427 | 71 | 33 | 12 |
| 3-methyl-oxovalerate | 808 | 464 | 557 | 1,239 | 419 | 387 | 272 | 303 | 349 | 24 | 58 | 10 |
| Fumarate | 6,418 | 1,161 | 1,855 | 2,627 | 1,352 | 898 | 782 | 831 | 714 | 74 | 45 | 6 |
| Lactate/Methylglyoxal | 235,997 | 78,577 | 124,981 | 167,454 | 77,802 | 71,592 | 54,974 | 57,486 | 48,190 | 45 | 41 | 7 |
| Succinate | 3,881 | 2,386 | 4,454 | 5,630 | 2,455 | 1,286 | 1,530 | 1,549 | 1,404 | 24 | 59 | 4 |
| Citraconate/Itaconate | 931 | 454 | 630 | 854 | 372 | 317 | 255 | 252 | 187 | 29 | 47 | 14 |
| Benzoate | 3,396 | 1,135 | 1,468 | 1,947 | 975 | 648 | 573 | 581 | 548 | 50 | 46 | 3 |
| Citramalate | 695 | 267 | 335 | 399 | 217 | 140 | 120 | 120 | 90 | 43 | 43 | 13 |
| Glyoxylate | - | 1,205 | 3,549 | 3,352 | 2,597 | 1,863 | 1,301 | 1,127 | 1,455 | 93 | 23 | 10 |
| Beta-3-hydroxybutyrate | 4,583 | 1,591 | 2,213 | 3,312 | 1,799 | 1,275 | 1,354 | 1,231 | 970 | 46 | 41 | 14 |
| Glycine | 61,151 | 20,383 | 43,210 | 42,758 | 22,479 | 13,755 | 13,783 | 10,625 | 10,602 | 40 | 46 | 13 |
| Alanine | 103,839 | 38,129 | 67,595 | 73,848 | 39,667 | 28,767 | 24,933 | 24,313 | 16,405 | 38 | 40 | 18 |
| O-acetyl-L-serine | 61,471 | 19,905 | 44,892 | 34,954 | 24,584 | 17,600 | 14,107 | 13,913 | 15,616 | 41 | 28 | 5 |
| Phenylacetate | 885 | 334 | 435 | 796 | 351 | 242 | 209 | 213 | 177 | 43 | 52 | 8 |
| Salicylate | 10,826 | 4,922 | 4,711 | 11,939 | 2,448 | 3,609 | 3,367 | 2,904 | 3,664 | 42 | 70 | 9 |
| 2-aminobutyrate | 1,686 | 587 | 832 | 1,083 | 580 | 461 | 345 | 332 | 287 | 46 | 38 | 8 |
| m -Toluate | 874 | 332 | 401 | 655 | 357 | 262 | 245 | 220 | 170 | 45 | 40 | 15 |
| Beta-alanine | 2,562 | 884 | 1,204 | 1,634 | 902 | 681 | 460 | 495 | 276 | 47 | 38 | 23 |
| Adipate | 1,067 | 370 | 531 | 655 | 389 | 314 | 192 | 205 | 152 | 45 | 32 | 12 |
| Valine | 16,068 | 5,391 | 9,534 | 10,834 | 6,165 | 4,747 | 3,447 | 3,397 | 3,119 | 43 | 36 | 4 |
| Beta-hydroxypyruvate | 484 | 921 | 1,403 | 1,718 | 785 | 815 | 496 | 663 | 385 | 40 | 39 | 22 |
| Alpha-ketoadipate | 1,018 | 307 | 394 | 460 | 207 | 206 | 130 | - | 144 | 55 | 41 | 71 |
| Leucine | 3,749 | 1,312 | 2,685 | 3,168 | 1,714 | 1,311 | 1,005 | 997 | 953 | 39 | 39 | 2 |
| Gamma-aminobutyrate | 6,172 | 1,927 | 2,492 | 3,157 | 2,251 | 1,484 | 1,262 | 1,414 | 868 | 53 | 30 | 19 |
| Threonine | 14,779 | 5,761 | 11,366 | 13,330 | 7,180 | 5,324 | 4,134 | 4,384 | 3,851 | 35 | 40 | 5 |
| Malate | 12,210 | 3,616 | 6,322 | 6,762 | 3,198 | 2,923 | 1,984 | 1,888 | 1,818 | 49 | 41 | 4 |
| Proline | 53,239 | 20,724 | 37,773 | 37,949 | 20,432 | 15,103 | 13,848 | 13,538 | 12,703 | 36 | 40 | 4 |
| Aspartate | 15,795 | 4,419 | 9,158 | 11,799 | 5,256 | 2,881 | 2,641 | 2,703 | 2,442 | 48 | 57 | 4 |
| Citrate | 4,203 | 2,815 | 7,526 | 8,325 | 3,337 | 1,939 | 2,125 | 2,164 | 1,854 | 41 | 60 | 7 |
| Serine | 54,916 | 4,199 | 6,683 | 9,465 | 3,960 | 3,090 | 2,447 | 2,616 | 2,039 | 106 | 51 | 10 |
| Anthralinate | 460 | 217 | 204 | 274 | 95 | 82 | 77 | 63 | 66 | 40 | 58 | 9 |
| Allantoin | 390,010 | 70,939 | 109,555 | 124,528 | 103,994 | 31,612 | 29,610 | 33,757 | 27,769 | 75 | 46 | 8 |
| N-Acetyl-L-glutamate | 41,875 | 7,875 | 11,351 | 15,649 | 4,981 | 3,085 | 4,016 | 3,030 | 3,519 | 75 | 70 | 11 |
| Glutamate | 135,447 | 37,869 | 70,602 | 90,033 | 39,432 | 22,517 | 22,383 | 25,364 | 18,086 | 50 | 57 | 14 |
| Methionine | 3,071 | 1,312 | 2,112 | 2,486 | 1,375 | 1,012 | 788 | 821 | 726 | 33 | 39 | 5 |
| Hippurate | 1,209 | 403 | 525 | 711 | 506 | 318 | 351 | 325 | 315 | 50 | 31 | 5 |
| Phenylalanine | 5,663 | 1,960 | 3,406 | 3,952 | 2,165 | 1,621 | 1,188 | 1,152 | 1,119 | 41 | 39 | 2 |
| Ornithine | 10,035 | 4,442 | 4,499 | 6,199 | 3,290 | 2,712 | 2,088 | 2,255 | 1,788 | 41 | 38 | 9 |
| Lysine | 20,991 | 8,329 | 10,668 | 14,148 | 7,321 | 5,864 | 4,692 | 4,711 | 4,248 | 41 | 40 | 5 |
| Tyrosine | 6,568 | 2,932 | 3,670 | 4,576 | 2,547 | 1,801 | 1,415 | 1,087 | 1,121 | 36 | 39 | 12 |

Table C.12: Metabolite concentration per tissue mass [pM/mg] of polar extracts obtained by performing optimised extraction

| Sample name 270312 | bas 1 | bas 2 | bas 3 | bas 4 | bas 5 | bas 6 | lum 1 | lum 2 | lum 3 | lum 4 | lum 5 | lum 6 | \% std dev | \% std dev |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Tissue weight [mg] | 23.12 | 35.25 | 32.73 | 38.38 | 33.77 | 28.12 | 32.15 | 32.41 | 32.35 | 24.19 | 30.52 | 28.95 | bas | lum |
| 4-methylvalerate | 391 | 258 | 246 | 182 | 276 | 295 | 245 | 189 | 197 | 214 | 111 | 120 | 25 | 30 |
| Pyruvate | 921 | 1,596 | 2,526 | 3,124 | 2,879 | 3,972 | 2,287 | 1,852 | 2,368 | 1,278 | 643 | 316 | 44 | 59 |
| 3-methyl-oxovalerate | 688 | 643 | 761 | 419 | 734 | 611 | 272 | 284 | 277 | 412 | 166 | 152 | 19 | 36 |
| Fumarate | 2,787 | 1,706 | 1,475 | 2,654 | 2,604 | 1,697 | 1,379 | 1,120 | 1,613 | 1,594 | 842 | 633 | 27 | 34 |
| Lactate/Methylglyoxal | 73,673 | 138,459 | 137,952 | 83,452 | 83,388 | 92,013 | 55,158 | 65,877 | 99,076 | 64,528 | 45,372 | 29,304 | 29 | 39 |
| Succinate | 3,784 | 2,687 | 2,891 | 2,674 | 3,420 | 2,551 | 3,967 | 3,027 | 3,881 | 3,358 | 2,216 | 2,218 | 16 | 25 |
| Citraconate/Itaconate | 651 | 581 | 426 | 323 | 757 | 503 |  |  |  |  |  |  | 29 | - |
| Benzoate | 1,071 | 729 | 630 | 536 | 832 | 719 | 616 | 457 | 526 | 658 | 338 | 310 | 25 | 29 |
| Citramalate | 2,147 | 1,249 | 1,236 | 1,043 | 1,514 | 1,495 | 943 | 702 | 1,188 | 1,239 | 606 | 691 | 27 | 30 |
| Glyoxylate | 17,599 | 7,354 | 5,430 | 4,056 | 9,431 | 4,318 | 4,179 | 2,333 | 3,466 | 5,841 | 2,181 | 2,328 | 64 | 42 |
| Glycine | 83,824 | 48,059 | 66,206 | 63,792 | 67,707 | 35,613 | 21,055 | 17,782 | 22,117 | 18,958 | 11,160 | 12,881 | 28 | 25 |
| Alanine | 18,947 | 11,403 | 24,551 | 22,900 | 26,317 | 12,265 | 28,466 | 23,132 | 27,505 | 27,002 | 19,738 | 15,173 | 33 | 22 |
| O-acetyl-L-serine | 15,079 | 8,199 | 21,593 | 24,874 | 20,938 | 10,976 | 35,122 | 29,898 | 33,003 | 44,335 | 35,456 | 23,962 | 39 | 20 |
| Phenylacetate | 1,411 | 1,184 | 1,058 | 823 | 1,027 | 752 | 1,098 | 965 | 1,028 | 1,650 | 496 | 625 | 23 | 42 |
| Salicylate | 15,591 | 15,914 | 11,740 | 10,058 | 8,363 | 7,778 | 17,828 | 10,070 | 14,261 | 19,418 | 6,021 | 8,853 | 30 | 42 |
| 2-aminobutyrate | 370 | 255 | 259 | 161 | 292 | 280 | 237 | 297 | 303 | 325 | 191 | 115 | 25 | 33 |
| m-Toluate | 506 | 609 | 511 | 412 | 533 | 726 | 676 | 537 | 592 | 886 | 274 | 449 | 20 | 36 |
| Valine | 4,327 | 2,579 | 4,068 | 3,294 | 4,012 | 2,207 | 4,103 | 3,637 | 5,640 | 3,990 | 2,430 | 1,851 | 25 | 37 |
| 2-isopropylmalate | 845 | 566 | 633 | 558 | 815 | 747 | 516 | 556 | 470 | 547 | 345 | 270 | 18 | 26 |
| Beta-hydroxypyruvate | 1,041 | 1,070 | 886 | 861 | 1,768 | 726 | 688 | 654 | 579 | 751 | 359 | 346 | 35 | 31 |
| Leucine | 1,088 | 321 | 512 | 405 | 847 | 531 | 983 | 1,125 | 1,569 | 750 | 835 | 591 | 47 | 35 |
| Threonine | 11,797 | 6,938 | 12,770 | 10,234 | 11,974 | 5,352 | 11,210 | 10,395 | 12,399 | 13,688 | 8,406 | 6,740 | 31 | 24 |
| Malate | 4,895 | 2,755 | 2,232 | 3,547 | 4,219 | 2,138 | 1,713 | 1,320 | 1,952 | 4,114 | 1,044 | 771 | 34 | 66 |
| Proline | 15,479 | 18,328 | 23,470 | 14,020 | 18,055 | 15,927 | 17,751 | 12,988 | 20,153 | 20,078 | 10,869 | 9,749 | 19 | 30 |
| Aspartate | 2,668 | 1,303 | 1,299 | 1,129 | 1,706 | 1,306 | 10,262 | 6,059 | 13,287 | 9,217 | 4,686 | 5,884 | 36 | 40 |
| Citrate | 19,654 | 13,973 | 14,196 | 12,878 | 21,525 | 12,747 | 8,786 | 7,171 | 6,067 | 10,208 | 4,108 | 3,430 | 24 | 40 |
| Serine | 4,461 | 2,099 | 3,915 | 4,131 | 5,090 | 2,100 | 4,215 | 3,626 | 9,590 | 4,222 | 2,131 | 2,299 | 34 | 63 |
| Allantoin | 84,740 | 72,436 | 61,809 | 41,103 | 126,411 | 46,028 | 35,671 | 19,943 | 25,954 | 35,963 | 51,076 | 27,303 | 43 | 33 |
| Glutamate | 142,415 | 87,578 | 92,463 | 96,724 | 141,314 | 79,707 | 48,720 | 34,008 | 37,916 | 50,026 | 22,417 | 27,323 | 26 | 30 |
| Methionine | 1,552 | 942 | 1,510 | 1,247 | 1,539 | 986 | 1,847 | 1,359 | 2,033 | 1,500 | 931 | 1,048 | 22 | 30 |
| Hippurate | 1,676 | 1,205 | 968 | 1,051 | 1,326 | 1,587 | 1,320 | 1,318 | 1,354 | 1,797 | 957 | 788 | 22 | 28 |
| Phenylalanine | 2,004 | 1,346 | 2,398 | 1,332 | 1,409 | 1,013 | 2,127 | 1,565 | 2,252 | 1,725 | 1,002 | 1,108 | 32 | 31 |
| Ornithine | 555 | 691 | 487 | 376 | 1,236 | 536 | 877 | 1,338 | 2,429 | 782 | 705 | 620 | 47 | 61 |
| Lysine | 14,210 | 10,900 | 12,772 | 7,239 | 9,875 | 6,993 | 5,884 | 4,017 | 6,212 | 4,479 | 2,544 | 2,528 | 28 | 37 |
| Tyrosine | 3,504 | 2,214 | 3,351 | 2,414 | 2,692 | 1,862 | 2,772 | 2,325 | 3,531 | 2,548 | 1,321 | 1,264 | 24 | 38 |

## D TMS derivatisation of polar extracts

## D. 1 Raw data

Raw data were obtained in peak area per microliter injected sample.

Table D.1: Metabolite response [peak area] of polar extracts obtained by performing optimised extraction method prior to TMS derivatisation and $G C-Q-M S$ analysis.

| Sample name | Blank 1 | Blank 2 | Blank 3 | 21021223 | 21021231 | 21021232 | 21021233 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Lactate/Methylglyoxal | - | - | - | 2,213,730 | 2,477,960 | 2,627,620 | 2,504,960 |
| Urea | - | - | - | 613,172 | 635,031 | 710,652 | 687,498 |
| L-Serine | - | - | - | - | 41,643 | 48,492 | 46,947 |
| Glycerol | 834,986 | 759,963 | 674,495 | 1,409,160 | 1,360,390 | 1,441,060 | 1,419,290 |
| Phosphoric acid | - | - | - | 6,668,050 | 7,764,480 | 8,694,750 | 8,535,380 |
| L-Proline | - | - | - |  | 154,663 | 187,594 | 156,845 |
| Glycine | - ${ }^{-}$ | - - | - | 1,980,930 | 2,260,610 | 2,718,750 | 2,366,180 |
| Porphine | 420,799 | 376,832 | 342,033 | 352,304 | 363,060 | 316,833 | 336,729 |
| Fumarate | - | - | - | 30,174 | 33,287 | 33,690 | 30,593 |
| D-Malate | - | - | - | 99,821 | 117,536 | 126,737 | 108,039 |
| L-Glutamate 3 | - | - | - | 340,927 | 400,570 | 431,866 | 437,915 |
| L-Glutamate 1 | - | - | - | 602,858 | 853,561 | 989,755 | 966,027 |
| Creatinine | - | - | - | 24,293 | 37,132 | 38,082 | 41,748 |
| alpha-ketoglutarate | - | - | - | 49,984 | 61,144 | 60,239 | 62,167 |
| L-Glutamate 2 | - | - | - | 109,748 | 136,913 | 141,230 | 134,235 |
| Glycerol 1-phosphate | - | - | - | 145,192 | 176,206 | 206,267 | 183,342 |
| O-phosphocolamine | - | - | - | 103,291 | 142,467 | 175,087 | 150,956 |
| Citrate | - | - | - | 21,870 | 27,773 | 35,979 | 36,507 |
| Methyl-beta-D-galactopyranoside | - | - | - | 173,827 | 152,321 | 172,399 | 221,242 |
| D-glucose | - | - | - | 390,654 | 253,770 | 295,122 | 306,242 |
| D-allose 2 | - | - | - | 68,127 | 43,481 | 51,489 | 51,023 |
| Palmitic acid | 203,514 | 51,472 | 155,705 | 282,263 | 387,472 | 458,024 | 468,329 |
| Oleic acid | - | - | 44,963 | 30,460 | 83,452 | 100,629 | 102,135 |
| Stearic acid | 91,861 | 28,746 | 104,709 | 139,116 | 208,260 | 273,442 | 283,309 |
| Maltose 1 | - | - | - | 235,024 | 166,163 | 197,849 | 278,249 |
| Cholesterol | - ${ }^{-}$ | - | - | 637,707 | 730,234 | 842,769 | 749,360 |
| Myristic acid | 55,973 | - | - | 43,47,800 | 12,082,30- | 48,098, $60{ }^{-}$ | 48,631,300 |
| d27-Myristic acid | 50,435 | - | - | 43,437,800 | 42,082,300 | 48,998,600 | 48,631,300 |

## D. 2 Normalised data

Normalisation against internal standards and tissue weight was performed and response per mg tissue obtained.

Table D.2: Metabolite response per tissue mass [peak area/mg] of polar extracts obtained by performing optimised extraction method prior to TMS derivatisation and GC-Q-MS analysis.

| Sample name | 21021223 | 21021231 | 21021232 | 21021233 |
| :--- | ---: | ---: | ---: | ---: |
| Tissue weight [mg | 27.35 | 25.94 | 25.94 | 25.94 |
| Lactate/Methylglyoxal | 85,319 | 103,937 | 94,658 | 90,921 |
| Urea | 23,632 | 26,636 | 25,601 | 24,954 |
| L-Serine | - | 1,747 | 1,747 | 1,704 |
| Phosphoric acid | 256,993 | 325,679 | 313,221 | 309,802 |
| L-Proline | - | 6,487 | 6,758 | 5,693 |
| Glycine | 76,347 | 94,821 | 97,941 | 85,883 |
| Fumarate | 1,163 | 1,396 | 1,214 | 1,110 |
| D-Malate | 3,847 | 4,930 | 4,566 | 3,921 |
| L-Glutamate 1 | 23,235 | 35,802 | 35,655 | 35,063 |
| L-Glutamate 2 | 4,230 | 5,743 | 5,088 | 4,872 |
| L-Glutamate 3 | 13,140 | 16,802 | 15,558 | 15,895 |
| Creatinine | 936 | 1,557 | 1,372 | 1,515 |
| alpha-ketoglutarate | 1,926 | 2,565 | 2,170 | 2,256 |
| Glycerol 1-phosphate | 5,596 | 7,391 | 7,431 | 6,655 |
| O-phosphocolamine | 3,981 | 5,976 | 6,307 | 5,479 |
| Citrate | 843 | 1,165 | 1,296 | 1,325 |
| Methyl-beta-D-galactopyranoside | 6,699 | 6,389 | 6,211 | 8,030 |
| D-glucose | 15,056 | 10,644 | 10,632 | 11,115 |
| D-allose 2 | 2,626 | 1,824 | 1,855 | 1,852 |
| Maltose 1 | 9,058 | 6,970 | 7,127 | 10,099 |
| Cholesterol | 24,578 | 30,629 | 30,360 | 27,199 |

## E HCl in methanol derivatisation of non-polar extracts

## E. 1 Raw data

Raw data were obtained in peak area per microliter injected sample.

Table E.1: Approximate metabolite responses [abundance] of non-polar extracts obtained in each of the five extraction rounds in chloroform.

| Sample name | 2102121 r 1 | 2102121 r 2 | 2102121 r 3 | 2102121 r 4 | 2102121 r 5 |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: |
| Methyl nonanoate | 1,500 | - | - | - | - |
| d27-Myristic acid | $1,000,000$ | - | - | $2,000,000$ | $1,000,000$ |
| Methyl myristate | 10,000 | - | - | - | - |
| Unknown RT 20.774 | 6,000 | - | - | - | - |
| Methyl palmitate | 400,000 | 150,000 | 160,000 | 120,000 | 400,000 |
| Unknown RT 21.107 | 150,000 | 220,000 | 240,000 | 250,000 | 100,000 |
| Methyl stearate | 160,000 | 90,000 | 90,000 | 75,000 | 17,000 |
| Unknown RT 24.407 | 20,000 | - | - | - | - |
| Unknown RT 24.562 | 10,000 | - | - | - | - |

Table E.2: Metabolite response [peak areal of non-polar extracts obtained in each of the five extraction rounds in chloroform. Response is an average of two technical replicates.

| Sample name | Tissue weight [mg] | Methyl palmitate | Unknown RT 21.107 | Methyl stearate | d27-myristic acid |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2802121 r1 |  | 109,367 | 151,144 | 68,202 | 7,362,630 |
| 280212 1 r2 |  | 22,876 | 46,534 | 11,108 | 9,105,795 |
| 280212 1 r3 | 28.05 | 22,876 | 9,239 | 11,108 | 5,472,930 |
| 2802121 r4 |  | 4,676 | 12,488 | - | 9,400,875 |
| 2802121 r 5 |  | - | - | - | 10,779,500 |
| 2802122 r1 |  | 338,323 | 263,594 | 221,520 | 11,439,300 |
| 2802122 r2 |  | 13,572 | 34,888 | , | 7,652,700 |
| 2802122 r3 | 39.82 | 12,632 | 26,363 | - | 11,341,100 |
| 2802122 r4 |  |  | , | - | 6,513,320 |
| 2802122 r5 |  | - | - | - | 8,553,955 |

## E. 2 Normalised data

Raw data were normalised against internal standards and tissue weight to obtain peak area per mg tissue.

Table E.3: Metabolite response per tissue mass [peak area/mg] of non-polar extracts obtained in each of the five extraction rounds in chloroform.

| Sample name | 280212 r 1 | 280212 r 2 | 280212 r 3 | 280212 r 4 | 280212 r 5 |
| :--- | ---: | ---: | ---: | ---: | ---: |
| Methyl palmitate | 5,419 | 571 | 238 | 151 | - |
| Unknown RT 21.107 | 5,582 | 1,264 | 505 | 403 | - |
| Methyl stearate | 3,478 | 370 | - | - | - |

## E. 3 Unidentified compound

Scan of unknown compound RT 21.107 is shown in figure E.1.


Figure E.1: $M S$ scan for unknown compound with retention time 21.107 obtained with $G C$ -$Q-M S$ analysis of non-polar extracts.

## F Lipid analysis of non-polar extracts

Q-TOF LC-MS analysis with direct infusion was performed. Scan at 0.299 min and 0.300 min for DCM blank and sample, respectively, obtained with ESI in positive ion mode is shown in Figure F.1.


Figure F.1: Scan of mass fragments, showing abundance as a function of $m / z$, for $D C M$ blank at the top and sample at the bottom obtained at 0.3 min with ESI in positive ion mode.

Scan at 0.702 min for DCM blank and sample obtained with ESI in positive ion mode is shown in Figure F.2.


Figure F.2: Scan of mass fragments, showing abundance as a function of $m / z$, for DCM blank at the top and sample at the bottom obtained at 0.702 min with ESI in positive ion mode.

Mass fragments at $121.05 \mathrm{~m} / \mathrm{z}$ and $922.01 \mathrm{~m} / \mathrm{z}$ represent the reference masses. The comparison of blank and sample in Figure F. 1 and Figure F. 2 showed that several mass fragments were obtained in sample but not in blank. These may represent lipids contained in non-polar extracts.

## G Isotope labelling experiment

Isotope labelling with ${ }^{13} \mathrm{C}$ was investigated by calculating summed fractional labelling (SFL). SFL of metabolites can be obtained using Equation G. 1 [23].

$$
\begin{equation*}
S F L=\frac{\sum i \times I_{i}}{\sum I_{i}} \tag{G.1}
\end{equation*}
$$

for $\mathrm{i}=0$ and up until n C-atoms in the fragment. I is the intensity of isotopomer. An approach to correct for naturally occurring ${ }^{13} \mathrm{C}$, is to calculate SFL for unlabelled metabolites and subtract it from SFL for possibly labelled metabolites. Raw data and calculated SFL are shown in Table G.1.

## H Data analysis

## H. 1 Standard deviation

In this project, Excel's function STDEV.S, estimating standard deviation based on sample, was used. STDEV.S uses Equation H. 1 where $x$ is a quantity, $\bar{x}$ is the average value of quantity and $n$ is number of values obtained from quantity.

$$
\begin{equation*}
S T D E V \cdot S=\sum \frac{(x-\bar{x})^{2}}{(n-1)} \tag{H.1}
\end{equation*}
$$

## H. 2 Student's t-test

Student's t-test was performed on metabolite concentrations of six biological replicates of each subtype, the same data used to obtain metabolite profiles. The probabilities associated with Student's t-test are shown in Table H.1. The probability of metabolite concentration in luminal-like and basal-like subtypes to obtain the same average concentration was obtained. Excel's function TTEST was used. Tail was set to 2 for a two-tailed distribution. Type was set to 3 for the reason that variances were assumed to differ between subtypes. Significance level was set to $5 \%$ meaning that a significant difference in concentration between subtypes was stated for metabolites obtaining a probability of $5 \%$ or less.

Table H.1: Probability associated with Student's t-test obtained for polar metabolites in luminal-like and basal-like subtypes.

| Metabolite | Probability $[\%]$ |
| :--- | ---: |
| 4-methylvalerate | 2.4 |
| Pyruvate | 9.7 |
| 3-methyl-oxovalerate | 0.0 |
| Fumarate | 1.0 |
| Lactate/Methylglyoxal | 27.2 |
| Succinate | 7.5 |
| Citraconate/Itaconate | 0.0 |
| Benzoate | 1.9 |
| Citramalate | 1.9 |
| Glyoxylate | 7.7 |
| Glycine | 0.1 |
| Alanine | 25.1 |
| O-acetyl-L-serine | 0.1 |
| Phenylacetate | 74.2 |
| Salicylate | 66.4 |
| 2-aminobutyrate | 57.7 |
| m-Toluate | 84.5 |
| Valine | 77.3 |
| 2-isopropylmalate | 0.6 |
| Beta-hydroxypyruvate | 2.0 |
| Leucine | 8.1 |
| Threonine | 70.6 |
| Malate | 5.2 |
| Proline | 35.4 |
| Aspartate | 0.4 |
| Citrate | 0.1 |
| Serine | 57.8 |
| Allantoin | 2.5 |
| Glutamate | 0.1 |
| Methionine | 47.6 |
| Hippurate | 80.6 |
| Phenylalanine | 87.9 |
| Ornithine | 1.4 |
| Lysine | 0.2 |
| Tyrosine | 41.5 |

## H. 3 Principal component analysis (PCA)

In order to compare luminal-like and basal-like subtypes, PCA analysis was performed using Unscrambler X multivariate analysis software. PCA analysis was performed on concentrations used to obtain metabolite profiles, shown in Table C.12. Loading plot without weighting of concentrations is shown in Figure H.1.


Figure H.1: Loading plot of PC-1 versus PC-2 without weighting.

Five metabolites (o-acetyl-L-serine, lactate, glycine, allantonin, glutamate) were seen to explain all the variance between subtypes. O-acetyl-L-serine and glycine are only explained by principal component 1 (PC-1). For this reason, concentrations were weighted.

Subsequent to weighting of concentrations, PCA analysis was performed. Influence plot is shown in Figure H.2. Data points seen in influence plot represent biological replicates.


Figure H.2: Influence plot showing to what extent data points, i.e. biological replicates, spread.

Biological replicates were spread evenly out in influence plot. No outliers were seen thus no data needed to be excluded for PCA analysis.

Explained variance plot is shown in Figure H.3.


Figure H.3: Explained variance plot showing PCs up to PC-3.

Optimum number of principal components (PCs) was two as seen in explained variance plot. PC-1 and PC-2 explain around $70 \%$ of the variance in concentrations between subtypes. Morover, other PCs obtained a decrease in explained variance as shown by a decreasing red line. PC-1 and PC-2 were therefore used.

Loading plot in Figure 3.6 was enlarged in order to see the metabolites shown to positively correlate with luminal-like subtype. The enlargement is shown in Figure H.4.


Figure H.4: Enlargement of loading plot obtained using PCA analysis.

