

# Design and off-design analyses of a pre-combustion CO<sub>2</sub> capture process in a natural gas combined cycle power plant

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## Abstract

In this study, a cycle designed for capturing the greenhouse gas CO<sub>2</sub> in a natural gas combined cycle power plant has been analyzed. The process is a pre-combustion CO<sub>2</sub> capture cycle utilizing reforming of natural gas and removal of the carbon in the fuel prior to combustion in the gas turbine. The power cycle consists of a H<sub>2</sub>-fired gas turbine and a triple pressure steam cycle. Nitrogen is used as fuel diluent and steam is injected into the flame for additional NO<sub>x</sub> control. The heat recovery steam generator includes pre-heating for the various process streams. The pre-combustion cycle consists of an air-blown auto thermal reformer, water-gas shift reactors, an amine absorption system to separate out the CO<sub>2</sub>, as well as a CO<sub>2</sub> compression block. Included in the thermodynamic analysis are design calculations, as well as steady-state off-design calculations. Even though the aim is to operate a plant, as the one in this study, at full load there is also a need to be able to operate at part load, meaning off-design analysis is important. A reference case which excludes the pre-combustion cycle and only consists of the power cycle without CO<sub>2</sub> capture was analyzed at both design and off-design conditions for comparison. A high degree of

process integration is present in the cycle studied. This can be advantageous from an efficiency stand-point but the complexity of the plant increases. The part load calculations is one way of investigating how flexible the plant is to off-design conditions. In the analysis performed, part load behavior is rather good with efficiency reductions from base load operation comparable to the reference combined cycle plant.

*Key words:* Carbon capture and storage (CCS), CO<sub>2</sub> capture, Pre combustion capture, Off-design analysis, Process simulation

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## 1 Introduction

Levels of atmospheric carbon dioxide, methane, and other greenhouse gases are on the rise and are contributing to the warming of the atmosphere due to the greenhouse effect. Natural causes can only explain part of this global warming. Fossil fueled power generation, transportation, industrial processes, and other man-made greenhouse gas emission sources add to the picture, mainly because of CO<sub>2</sub> emissions. Out of the energy related carbon dioxide emission sources, the power generation sector is the largest emitter (International Energy Agency, 2006). Thus, if one tries to control and limit the emission of greenhouse gases and thereby attenuating the rise in atmospheric temperature, CO<sub>2</sub> capture from fossil fuel power plants can be a viable path. Among the fossil fuels, the capture of the carbon from coal is attracting the main attention because of the high carbon dioxide emissions per kilowatt hour of electricity and the abundance of coal-fired plants in the world. However, for

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15 Norway, with large natural gas reserves and the planned and already built  
16 gas-fired power plants in the country, CO<sub>2</sub> capture from such plants will be  
17 important.

18 The methods for capture of CO<sub>2</sub> from fossil fuel power generation sources can  
19 be divided into three main categories:

20 1) Post-combustion capture, where the CO<sub>2</sub> is captured at the tail end of the  
21 plant from the flue gases, i.e., after the combustion (Chapel and Mariz, 1999).  
22 Capture of CO<sub>2</sub> from the flue gases of a power plant may be the best option  
23 for capture retrofits of existing power plants. It is also a viable option for  
24 new plants. The currently preferred option is capture by absorption processes  
25 based on chemical solvents and have been implemented in a number of pilot  
26 projects world-wide for CO<sub>2</sub> capture purposes, for example, the Castor pilot  
27 project in Denmark (Le Thiez et al., 2004; Knudsen et al., 2006), and the  
28 Boundary Dam pilot plant in Canada (Wilson et al., 2004).

29 2) Pre-combustion capture, where the fossil fuel is used for producing a syngas  
30 and the carbon (as CO<sub>2</sub>) is separated out before the combustion takes place.  
31 The fuel for the combustion mainly consists of hydrogen mixed with a diluent,  
32 such as, nitrogen or steam. An existing technology for power plant applica-  
33 tions, the integrated gasification combined cycle (IGCC), could be attractive  
34 as part of a coal based pre-combustion CO<sub>2</sub> capture method (Bohm et al.,  
35 2007).

36 3) Oxy-fuel combustion, where the oxidizer for the combustion is oxygen in-  
37 stead of air. The combustion products are mainly carbon dioxide and steam,  
38 and the CO<sub>2</sub> can be separated out by condensing the steam. Many proposals  
39 for cycle configurations have been suggested in the oxy-fuel category. Exam-

40 ples include the Graz cycle (Jericha et al., 2004), the Matiant cycle (Mathieu  
41 and Nihart, 1999), the advanced zero emissions power plant (Griffin et al.,  
42 2005), and chemical looping combustion (Richter and Knoche, 1983; Ishida  
43 and Jin, 1994).

44 This study focuses on the pre-combustion approach. More specifically, pre-  
45 combustion capture utilizing an air-blown auto thermal reformer (ATR) in a  
46 natural gas fueled combined cycle (NGCC) plant. Similar process configura-  
47 tions have been studied by Andersen et al. (2000); Lozza and Chiesa (2002a,b);  
48 Corradetti and Desideri (2005); Ertesvåg et al. (2005). Their results from heat  
49 and mass balance analyses show lower heating value (LHV) net plant efficien-  
50 cies ranging from approximately 46% to 49%. Another possibility for this type  
51 of plant is to utilize it for co-production of hydrogen and electricity (Consonni  
52 and Viganò, 2005); however, the focus of this paper is on power production  
53 only. Kvamsdal et al. (2007) performs comparative heat and mass balance sim-  
54 ulations for a number of CO<sub>2</sub> capture cycles including pre-combustion cases.  
55 The cited studies focus on design case analysis. Little is found in the litera-  
56 ture in terms of off-design analysis of CO<sub>2</sub> capture cycles. Part load analyses  
57 of post-combustion systems are performed for coal cycles by Chalmers and  
58 Gibbins (2007) and for natural gas cycles by Möller et al. (2007). Haag et al.  
59 (2007) and Naqvi et al. (2007) analyze the part load behavior of some of  
60 the proposed oxy-fuel cycles. For NGCC pre-combustion plants no off-design  
61 publications have been found by the author.

62 The remainder of the paper is divided into the following sections: Section 2  
63 describes the process where the details of the cycle are explained. Section 3  
64 describes the methodology and lists the assumptions used in the study. The  
65 results are shown and analyzed in Section 4 and concluding remarks are given

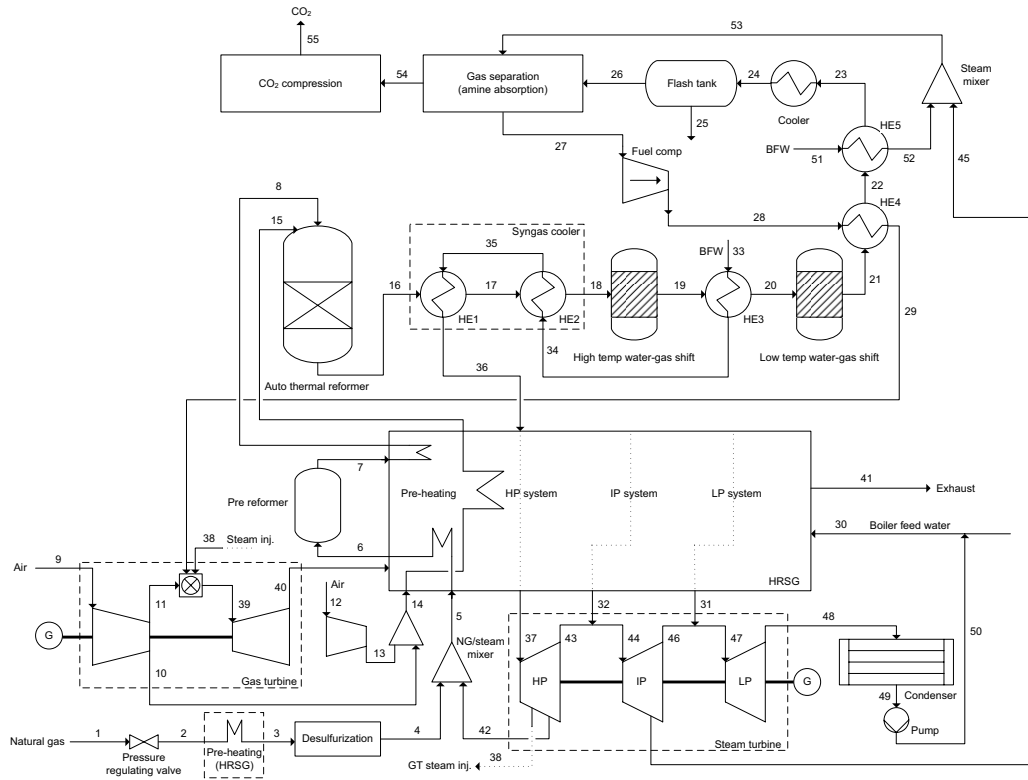


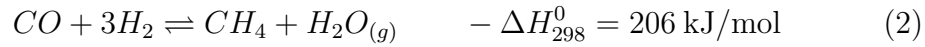
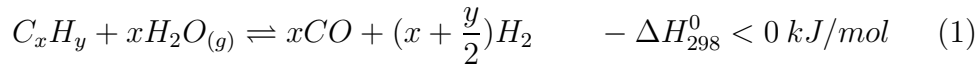
Fig. 1. Pre-combustion process flow sheet.

66 in Section 5.

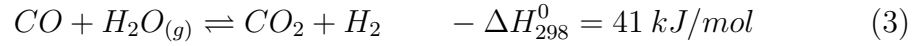
## 67 2 Process description

68 The selected process for the work is a pre-combustion CO<sub>2</sub> capture cycle in a  
 69 natural gas combined cycle power plant as shown in Fig. 1. The power cycle  
 70 consists of a General Electric (GE) 9FA H<sub>2</sub>-fired gas turbine (GT) and a triple  
 71 pressure steam cycle. The heat recovery steam generator (HRSG) includes pre-  
 72 heating for the various process streams. The pre-combustion cycle consists of a  
 73 pre-reformer, an air-blown auto thermal reformer, two water-gas shift reactors,  
 74 a gas separation stage in form of amine absorption to separate out the CO<sub>2</sub>,  
 75 as well as a CO<sub>2</sub> compression block.

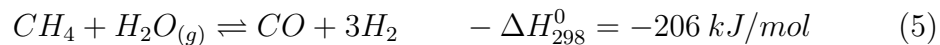
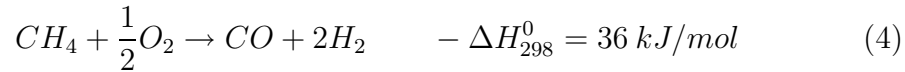
As mentioned, the fuel input to the process is natural gas (stream 1 in Fig. 1). After the natural gas has been regulated down to system pressure (stream 2), pre-heated to 400°C (3), and desulfurized (4), it is mixed with steam (5) before another pre-heating section (500°C) and introduced to the pre-reformer (6). The steam to carbon ratio (S/C) is set at 1.5. In the pre-reforming reactor the hydrocarbons higher than methane are converted to protect against coking in the primary reformer according to reactions (1) and (2).



Also, the exothermic water-gas shift reaction (3) converting the CO into CO<sub>2</sub> occurs to some degree in the pre-reforming reactor.



Before entering the ATR the stream from the pre-reformer (7) is again pre-heated to 500°C (8). Also, air extracted from the compressor discharge stream of the gas turbine (10) combined with an additional compressor air stream (13) is pre-heated and supplied to the ATR (15). The external compressor is introduced in order to better utilize the operation of the gas turbine. If too much air is removed prior to the combustion chamber in the gas turbine the effect on performance and temperature profile can be negative. With the additional compressor another degree of freedom is attained and the gas turbine can be utilized in a more efficient manner. In the ATR the exothermic reaction (4) provide heat to the endothermic reaction (5).



76 As in the pre-reformer the water-gas shift reaction (3) converts some of the CO  
77 into CO<sub>2</sub>. Further on, the syngas is cooled in the syngas cooler before entering  
78 the water-gas shift reactors where most of the remaining CO is converted into  
79 CO<sub>2</sub> according to reaction (3). The reasons behind dividing the water-gas shift  
80 reaction into a high temperature reactor and a low temperature one are due to  
81 conversion rate and catalysts. To get a higher degree of conversion of the CO to  
82 CO<sub>2</sub>, two reactors are favorable compared to a one-reactor setup. Also, there is  
83 a need for a more active catalyst at the lower region of the temperature range  
84 (Moulijn et al., 2007). It can therefore make sense to use a standard catalyst  
85 at the higher temperature range and then have a separate reactor with a more  
86 active catalyst for the low end temperature. Heat exchanger 3 (HE3) and  
87 the syngas cooler are utilized for producing high-pressure saturated steam to  
88 be added to the high-pressure superheater in the HRSG. The reason for not  
89 superheating the steam in the heat exchanger is because of the risk of metal  
90 dusting (Grabke and Spiegel, 2003). Heat exchanger 4 (HE4) is used to pre-  
91 heat the fuel to the gas turbine to 200°C (29). In this model the pre-combustion  
92 capture (Gas separation) is using the chemical absorbent activated MDEA  
93 (Zhang et al., 2003; van Loo et al., 2007) and is modeled as a 'black box'.  
94 Assumptions for the capture section include a CO<sub>2</sub> capture rate of 95% and  
95 the heat required for the stripper reboiler at 1.5 MJ/kg CO<sub>2</sub>. Heat exchanger 5  
96 (HE5) is producing some of the steam necessary for the reboiler in the amine  
97 absorption process. The CO<sub>2</sub> (54) is passed on to the compression section  
98 where the gas is compressed in the four compressor/intercooler stages and  
99 excess water is removed. To achieve the exit pressure of 110 bar a pump is  
100 utilized at the end of the compression train.

101 From the gas separation stage the fuel mix (27) is passed on to the gas turbine

102 via a fuel compressor. In principle, the fuel consists of an  $H_2/N_2$  mixture. The  
103  $N_2$  diluent is used to be able to operate with available IGCC-type combustors  
104 in the gas turbine. For further  $NO_x$  control, steam is injected into the flame. In  
105 addition to running the GT on a hydrogen based fuel, the idea is to be able to  
106 operate on natural gas if the pre-combustion process is shut-down and during  
107 plant start-up. This requires fuel flexibility for the combustor system (Tomczak  
108 et al., 2002; Shilling and Jones, 2003; Moliere, 2005). The gas turbine exhaust  
109 stream (40) passes through the HRSG for pre-heating of process streams and  
110 steam generation before emitted to the atmosphere through the stack (41).

111 The steam cycle is designed for pressure levels of approximately 83/10/3 bars  
112 for the high, intermediate, and low pressure (HP/IP/LP) systems respectively.  
113 The pre-heating makes the HRSG design more complex and a lot of heat is  
114 removed from the gas stream at the hot part of the HRSG due to the high  
115 temperature requirements of some of the process streams. Note that the pre-  
116 heating is not entirely in the hot end of the HRSG but instead inter-mixed  
117 with the low, intermediate, and high-pressure sections. The steam turbine  
118 (ST) has extractions for the GT steam injection (38), the reforming process  
119 steam (42), and for the reboiler in the amine absorption system (45). After  
120 exiting the last low pressure turbine stage (48) the steam is condensed in the  
121 direct seawater cooled condenser (49). The condenser pressure is assumed at  
122 0.04 bar.

123 There are certainly many configuration options for a plant like this. For ex-  
124 ample, one could operate the system at a higher pressure by boosting the air  
125 pressure from the gas turbine compressor discharge with an additional com-  
126 pressor. In this way a fuel compressor would not be necessary. The impact of  
127 this option was investigated by Andersen et al. (2000) where it was concluded



128 that operating at a lower system pressure and having a fuel compressor im-  
129 proves the overall efficiency for the cycle in their study. This effect was due  
130 to the need for extra process stream pre-heating in the elevated pressure case  
131 since the air was cooled before the compression to minimize compressor work.  
132 Other configuration options include utilizing an oxygen-blown ATR with an  
133 air separation unit (ASU) for the oxygen supply. Or using a steam reformer  
134 instead of the ATR. Configurations with less integration between the power  
135 cycle and syngas process could also be attractive. For the power cycle one  
136 could employ a more recent gas turbine model as for example the GE 9FB  
137 type with a higher turbine inlet temperature (TIT) and cycle efficiency. The  
138 steam cycle could include a reheat cycle leading to a higher efficiency but also  
139 more complexity. For the capture section one could use other absorbants, such  
140 as, hot potassium carbonate.

141 A reference case which excludes the pre-combustion cycle and only consist of  
142 the power cycle without CO<sub>2</sub> capture was analyzed at both design and off-  
143 design conditions for comparison. The reference case consists of the same type  
144 GE 9FA gas turbine but is instead of the IGCC combustor using a regular  
145 pre-mix natural gas combustor without steam injection. The steam cycle is  
146 again triple pressure without reheat.

### 147 **3 Methodology**

148 This section provides details into the process models simulated in the study.  
149 Assumptions for the design case analysis are described in Section 3.1. Included  
150 in the thermodynamic analysis are steady-state off-design calculations, that  
151 is, analysis when the plant is operating at part load. In a scenario where

152 CO<sub>2</sub> capture plants become common-place, part load operation will be an  
153 important part of the operation scheme. For a plant such as the one modeled in  
154 this work the goal is certainly to run it at base load operation for the majority  
155 of the time but as part of an overall grid strategy part load operation will come  
156 into play. Assumptions for the part load cases are described in Section 3.2.

157 The pre-combustion cycle, including the pre-heating section, was modeled with  
158 Aspen HYSYS. The property package was modeled with the Kabadi-Danner  
159 equation of state. The Kabadi-Danner is a modification of the Soave-Redlich-  
160 Kwong equation of state to take into account hydrocarbon solubility in the  
161 water phase. The power cycle was modeled with GT PRO for the design case  
162 and GT MASTER for the off-design cases. For the steam properties in GT  
163 PRO/GT MASTER the IAPSW-IF97 formulation was used (Wagner et al.,  
164 2000).

### 165 *3.1 Design model assumptions*

166 The selected gas turbine is a GE 9FA from the model library of GT PRO  
167 version 17. Steam is injected into the flame for NO<sub>x</sub> control at a rate of 20%  
168 of the fuel mass flow. The GT turbine inlet temperature has been reduced  
169 because of the high steam content in the turbine. The hydrogen fuel together  
170 with the injected steam lead to an H<sub>2</sub>O content entering the turbine of about  
171 18.2 vol%. This leads to a higher heat transfer rate to the blades compared to  
172 a natural gas fired turbine. As a result, the metal temperature of the turbine  
173 blades is higher for the same turbine inlet temperature as in a conventional gas  
174 turbine. To obtain similar life of the turbine parts, the turbine inlet tempera-  
175 ture reduction is necessary. Chiesa et al. (2005) report TIT decreases of 10-34

176 K for hydrogen combustion with nitrogen or steam diluent (VGV operation  
177 cases). As a model assumption, a TIT reduction of 30 K has been assumed  
178 for this work. The inlet filter pressure drop is set to 10 mbar and the total  
179 exhaust losses (GT exhaust and HRSG) to 25 mbar. The maximum allowable  
180 GT power output is increased from 260 to 286 MW (IGCC setup). Air from  
181 the compressor discharge is re-directed to the reforming section at a rate of  
182 75 kg/s. This is approximately 12% of the GT inlet air flow. Additional air  
183 required for the reforming is supplied by an external (to the GT) compres-  
184 sor with a polytropic efficiency of 85%. A polytropic efficiency of 85% is also  
185 assumed for the fuel compressor for the hydrogen-rich fuel.

186 The high-pressure steam is set to 83 bar at 568°C before the stop valve to  
187 the steam turbine. The intermediate-pressure level is 10.3 bar and the LP  
188 drum pressure is 2.8 bar. The pinch point temperature difference is assumed  
189 to be 10 K for all three pressure levels. The subcooling approach temperature  
190 difference at the exit of the economizers is assumed at 5 K.

191 The natural gas composition (stream 1) is listed in Table 2 with the exception  
192 of the H<sub>2</sub>S content which is set to be 5 ppmvd. The sulfur is removed in the  
193 desulfurizer unit, which is modeled as a separator. The air composition (9) is  
194 also listed in Table 2. The ambient pressure is assumed to be 1.013 bar with  
195 a temperature of 15°C and a relative humidity of 60%.

196 The pressure drops in the pre-reformer and ATR are set at 5% of the in-  
197 let pressure. The tube side pressure drop in the heat exchangers modeled in  
198 HYSYS is set to be 0.5 bar (approximately 3% of inlet pressure) with the ex-  
199 ception of HE3-HE5 which each has an assumed pressure drop of 0.85 bar due  
200 to two shell passes compared to one shell pass for the other heat exchangers.

201 The shift reactors are modeled with a 0.5 bar pressure drop. The pre-reformer  
202 and the water-gas shift reactors are modeled as equilibrium reactors. A Gibbs  
203 reactor model is used for the ATR.

204 A splitter is used for the amine absorption section model. The reboiler duty is  
205 set to 1.5 MJ/kg CO<sub>2</sub> and the total pump work is assumed to be 0.16 MJ/kg  
206 CO<sub>2</sub>. The reboiler temperature is set to 120°C. A 95% capture rate is assumed  
207 for the absorption system.

208 Polytropic efficiencies for the CO<sub>2</sub> compression train are assumed at 85%,  
209 80%, 80%, and 75% for the four compressor stages respectively (listed in flow  
210 direction). The pump that pressurizes the CO<sub>2</sub> stream to the end pressure of  
211 110 bar is assumed to have an adiabatic efficiency of 75%.

### 212 *3.2 Off-design model assumptions*

213 The selected part load points are 60% and 80% of the design case gas turbine  
214 load. The reason for selecting the relative part load points as a function of gas  
215 turbine load is because the GT dictates the overall plant load. By changing the  
216 GT load, the steam cycle, as well as the pre-combustion process, will follow.  
217 Gas turbine part load operation commonly employs variable inlet guide vanes  
218 (VIGV). This is the case for the GE 9FA which has one row of variable guide  
219 vanes where the flow angle entering the first stage of the compressor can be  
220 varied. The VIGV operation allows reduction of the air flow and the turbine  
221 exhaust temperature can remain high at part load operation. The high exhaust  
222 temperature means the part load combined cycle efficiency can be maintained  
223 at a high level. However, at the lower part load range the cycle efficiency drops

224 off quicker. The steam cycle part load operating concept involves sliding pres-  
225 sure operation with fully open steam valves down to approximately 50% steam  
226 turbine load (Kehlhofer et al., 1999). At lower loads the operating concept is  
227 based on fixed steam pressure operation by closing of the steam valves. This  
228 leads to throttling losses in the ST inlet valves. These factors combined may  
229 suggest that it does not make sense to operate a plant, such as the one in the  
230 study, at a much lower GT load than 60%. Certainly, the plant still has to be  
231 able to operate at lower part load points, not the least during transients such  
232 as start-ups and shut-downs; however, transient analysis is not covered in this  
233 study.

234 All the hardware in the off-design cases are identical to the design case. This  
235 also means that the extractions of the steam turbine are set. Since the part  
236 load operation is with sliding pressure operation of the steam cycle the steam  
237 pressures at the extraction points will decrease. In the case of the steam for the  
238 reboiler in the amine absorption system the design case was actually "over-  
239 designed" to allow for a sufficient steam pressure (and hence a sufficiently high  
240 condensation temperature) for the part load cases.

241 The turbine inlet temperature reduction was removed for the off-design simu-  
242 lations since the temperature was decreased anyway for part load operation at  
243 the 80% and 60% relative load levels. The air extraction from the compressor  
244 discharge was decreased to 60 kg/s (approximately 11% of GT inlet air flow)  
245 for the 80% case and 45 kg/s (approximately 10% of GT inlet air flow) for  
246 the 60% case. The fuel compressor exit pressure is assumed constant from the  
247 design case.

248 In the design case the inlet temperatures to the desulfurization unit, the re-

249 forming reactors, and the water-gas shift reactors were fixed. For the off-design  
 250 calculations these constraints were removed. Instead, for each part load case  
 251 a check was performed to see if the inlet temperatures were within the oper-  
 252 ational window of each reactor. Based on the resulting inlet temperatures it  
 253 was not necessary to use by-pass valves for the various heat exchangers at the  
 254 steady-state part load cases simulated (although likely needed during lower  
 255 part load and start-up and shut-down).

For the analysis of the various heat exchangers a correction of the heat transfer coefficient was done based on the gas massflow. The correction is based on course literature from Bolland (2006) as displayed in Equation (6).

$$\frac{U}{U_{design}} = \left( \frac{\dot{m}_{gas}}{\dot{m}_{gas,design}} \right)^m \quad (6)$$

$U$  is here the heat transfer coefficient,  $\dot{m}_{gas}$  the gas massflow, and  $m$  a constant. For a staggered tubes configuration with assumed tube pitches of 2.5 (Incropera and DeWitt, 1990):

$$\left. \begin{array}{l} \frac{S_T}{D} = 2.5 \\ \frac{S_L}{D} = 2.5 \end{array} \right\} \Rightarrow m \simeq 0.57$$

$S_T$  is the transverse pitch, that is, the distance  $90^\circ$  off from the flow direction between the centers of two adjacent tubes.  $S_L$  is the longitudinal pitch, that is, the distance in flow direction between the centers of two adjacent tubes.  $D$  is the tube diameter in the heat exchanger. In HYSYS there is the option to lock in the  $UA$  specification for a heat exchanger. Since the area  $A$  is constant one could re-write Equation (6) as:

$$UA = U_{design}A \left( \frac{\dot{m}_{gas}}{\dot{m}_{gas,design}} \right)^{0.57} \quad (7)$$

256 A similar expression, the exception being the  $m$ -factor which was set at 0.6,  
257 was used by Haag et al. (2007).

## 258 4 Results

259 The main results are summarized in Table 1. Included in the table is the power  
260 consumption for the air compressor (external to GT), the fuel compressor,  
261 the CO<sub>2</sub> compression, the pump work in the amine absorption system (gas  
262 separation pumps), as well as the additional boiler feed water pumps in the  
263 pre-combustion system, and the remaining plant auxiliaries. The auxiliaries  
264 post in Table 1 includes, among other items, the regular boiler feed water  
265 pumps and the cooling water pumps.

266 The design case LHV based cycle efficiency is 41.9% with a net power out-  
267 put of approximately 362 MW. The net power output is here defined as the  
268 gross power output at the generator terminals minus the power needed for  
269 air compression, fuel compression, CO<sub>2</sub> compression, pump work, and auxil-  
270 iaries, as displayed in Table 1. The cycle efficiency is the net power output  
271 divided by the natural gas lower heating value input. The design case results  
272 should be compared to the reference case net power output of approximately  
273 385 MW and efficiency of 55.9% leading to a capture efficiency penalty of  
274 approximately 14%-points. The calculated design case cycle efficiency is low  
275 and the capture efficiency penalty high compared to the literature (Ander-  
276 sen et al., 2000; Lozza and Chiesa, 2002a,b; Corradetti and Desideri, 2005;  
277 Ertesvåg et al., 2005). This can be explained to a large degree by the practical  
278 considerations included in this work. For one, steam is injected into the gas  
279 turbine for NO<sub>x</sub> control which lowers the overall efficiency. Also, the turbine

Table 1

Summary of results for design case (100%), off-design cases (80% and 60%), and reference cases (100% ref., 80% ref., and 60% ref).

	100%	100%	80%	80%	60%	60%
		ref.		ref.		ref.
Natural gas LHV input [MW]	865.2	689.1	729.8	599.0	588.2	501.0
Gross power output GT [MW]	277.0	253.5	221.6	204.0	166.2	153.8
Gross power output ST [MW]	137.6	137.2	122.5	127.6	103.8	113.7
Gross power output [MW]	414.6	390.7	344.1	331.6	270.0	267.5
Gross power output [% of LHV input]	47.9	56.7	47.1	55.4	45.9	53.4
Air compression [MW]	8.2	-	7.9	-	6.9	-
Air compression [% of LHV input]	0.9	-	1.1	-	1.2	-
Fuel compression [MW]	13.6	-	14.7	-	17.0	-
Fuel compression [% of LHV input]	1.6	-	2.0	-	2.9	-
CO <sub>2</sub> compression [MW]	17.7	-	15.0	-	12.2	-
CO <sub>2</sub> compression [% of LHV input]	2.0	-	2.1	-	2.1	-
Gas separation pumps [MW]	7.6	-	6.4	-	5.2	-
Gas separation pumps [% of LHV input]	0.9	-	0.9	-	0.9	-
BFW pumps in pre-comb process [MW]	1.0	-	0.8	-	0.5	-
BFW pumps in pre-comb process [% of LHV input]	0.1	-	0.1	-	0.1	-
Auxiliaries [MW]	4.5	5.4	4.4	5.3	4.3	5.2
Auxiliaries [% of LHV input]	0.5	1.4	0.6	1.6	0.7	1.9
<b>Net power output [MW]</b>	<b>362.2</b>	<b>385.3</b>	<b>294.9</b>	<b>326.3</b>	<b>223.8</b>	<b>262.3</b>
<b>Net plant efficiency [% of LHV input]</b>	<b>41.9</b>	<b>55.9</b>	<b>40.4</b>	<b>54.5</b>	<b>38.0</b>	<b>52.4</b>
Efficiency capture penalty [%-point loss to ref. case]	14.0	-	14.1	-	14.3	-
CO <sub>2</sub> emissions [g CO <sub>2</sub> /net kWh el.]	33.2	380.1	30.7	390.1	29.3	405.9
<b>CO<sub>2</sub> capture rate [%]</b>	<b>93.4</b>	<b>0</b>	<b>94.1</b>	<b>0</b>	<b>94.7</b>	<b>0</b>

280 inlet temperature is decreased by 30 K which further will bring the efficiency  
281 down. In addition, for the design case, considerations were taken of the part  
282 load scenarios. For example, a steam turbine extraction had to be taken at a  
283 higher than necessary pressure during design case analysis to have sufficient  
284 pressure also at the off-design cases. This also has a negative effect on the  
285 design case plant efficiency.



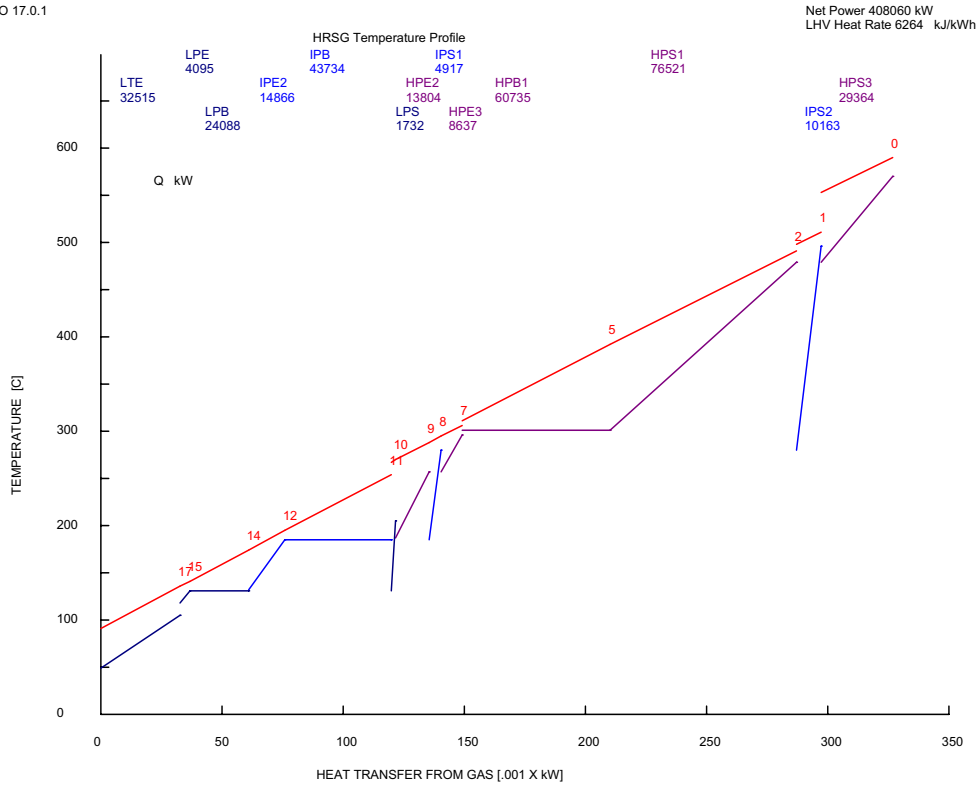


Fig. 2. GT PRO T-Q diagram for heat recovery steam generator.

286 The HRSG has a different design than would be present in a typical NGCC  
 287 plant. A large portion of the heat in the GT exhaust gases are utilized in the  
 288 pre-heating and in the HP superheaters, as displayed in Fig. 2. Because of  
 289 the saturated steam introduced from the syngas cooler the massflow to the  
 290 high-pressure superheaters are more than three times as high as the massflow  
 291 in the HP boiler. The vertical gas temperature jumps in the T-Q diagram  
 292 represent the pre-heating sections in the HRSG.

293 The off-design calculations resulted in net plant efficiencies of 40.4% and 38.0%  
 294 for the 80% and 60% load cases respectively. The capture penalties for the part  
 295 load cases are very similar to the design case, that is, around 14%-points.

296 The CO<sub>2</sub> capture rate varies between 93% and 95% for the different cases,  
 297 with CO<sub>2</sub> emissions of 29-33 g/net kWh electricity. The CO<sub>2</sub> capture rate is

298 defined as the fraction of formed  $\text{CO}_2$  that is captured.

299 Stream data for the design case is displayed in Table 2, for the 80% load case  
300 in Table 3, and for the 60% load case in Table 4.

## 301 **5 Conclusions**

302 The pre-combustion NGCC cycle is a system well worth studying. Advantages  
303 include the reduced size of the capture system and the increased  $\text{CO}_2$  partial  
304 pressure compared to post-combustion capture. A post-combustion capture  
305 system would have to deal with separating out  $\text{CO}_2$  from flue gases with very  
306 large flow rates at a low pressure. Disadvantages compared to post-combustion  
307 capture include conversion losses in the natural gas reforming process. Another  
308 advantage for a post-combustion capture system is that natural gas fired gas  
309 turbines are a more mature product than hydrogen fired ones. Pre-mix com-  
310 bustion with low  $\text{NO}_x$  emissions is one of the advantages of a standard GT  
311 fired with natural gas. In the case of the hydrogen diffusion combustion system,  
312 diluents such as steam and/or nitrogen are necessary. In this study, nitrogen  
313 was used as diluent and steam was injected directly into the flame in the  
314 combustor.

315 A high degree of process integration is present in the cycle studied. This can be  
316 advantageous from an efficiency stand-point but the complexity of the plant  
317 increases. This is exemplified in the HRSG where several of the process streams  
318 are pre-heated and high-pressure steam are introduced from the syngas cooler  
319 to the HP superheaters. The heat from the syngas is used for the economizing  
320 and boiling of the high-pressure water. This heat integration increases the cycle

Table 2

Stream results for the design case.

No.	$T$ (°C)	$p$ (bar)	$\dot{m}$ (kg/s)	$MW$ (kg/kmol)	CH <sub>4</sub> (vol%)	C <sub>2+</sub> (vol%)	H <sub>2</sub> (vol%)	CO (vol%)	CO <sub>2</sub> (vol%)	H <sub>2</sub> O (vol%)	O <sub>2</sub> (vol%)	N <sub>2</sub> (vol%)	Ar (vol%)
1	16.0	31.00	19.0	20.73	79.84	16.72	-	-	2.92	-	-	0.51	-
3	400.0	17.68	19.0	20.73	79.84	16.72	-	-	2.92	-	-	0.51	-
5	371.3	17.68	49.5	18.97	28.01	5.86	-	-	1.03	64.91	-	0.18	-
6	500.0	17.18	49.5	18.97	28.01	5.86	-	-	1.03	64.91	-	0.18	-
7	451.4	16.32	49.5	17.30	35.20	0.00	8.77	0.12	5.21	50.53	-	0.16	-
8	500.0	15.82	49.5	17.30	35.20	0.00	8.77	0.12	5.21	50.53	-	0.16	-
9	15.0	1.01	629.3	28.86	-	-	-	-	0.03	1.01	20.74	77.29	0.93
10	394.0	16.35	75.0	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
11	394.0	16.35	483.9	28.86	-	-	-	-	0.03	1.01	20.74	77.29	0.93
13	436.1	16.35	18.5	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
15	500.0	15.85	93.5	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
16	950.0	15.03	143.0	19.24	0.08	0.00	28.87	10.38	5.16	21.36	0.00	33.74	0.40
18	350.0	14.03	143.0	19.24	0.08	0.00	28.87	10.38	5.16	21.36	0.00	33.74	0.40
19	433.7	13.53	143.0	19.24	0.08	0.00	35.94	3.31	12.23	14.29	0.00	33.74	0.40
20	205.8	12.68	143.0	19.24	0.08	0.00	35.94	3.31	12.23	14.29	0.00	33.74	0.40
21	241.4	12.18	143.0	19.24	0.08	0.00	38.79	0.46	15.08	11.44	0.00	33.74	0.40
26	25.0	9.98	128.1	19.39	0.09	0.00	43.65	0.52	16.96	0.35	0.00	37.97	0.45
29	200.0	20.00	79.9	14.55	0.11	0.00	52.32	0.63	0.64	0.19	0.00	45.56	0.54
31	203.3	2.47	11.1	18.02	-	-	-	-	-	100.00	-	-	-
32	494.8	10.30	21.9	18.02	-	-	-	-	-	100.00	-	-	-
36	301.4	87.62	86.7	18.02	-	-	-	-	-	100.00	-	-	-
37	568.0	83.00	129.4	18.02	-	-	-	-	-	100.00	-	-	-
38	377.0	22.00	16.0	18.02	-	-	-	-	-	100.00	-	-	-
39	1295.0	15.70	579.8	26.73	-	-	-	-	0.37	18.21	9.28	71.28	0.86
40	591.0	1.04	650.1	26.94	-	-	-	-	0.34	16.48	10.43	71.89	0.87
41	90.6	1.01	650.1	26.94	-	-	-	-	0.34	16.48	10.43	71.89	0.87
42	346.0	17.68	30.5	18.02	-	-	-	-	-	100.00	-	-	-
45	227.0	4.00	24.0	18.02	-	-	-	-	-	100.00	-	-	-
53	209.6	4.00	30.5	18.02	-	-	-	-	-	100.00	-	-	-
55	40.9	110.00	47.4	43.89	0.00	0.00	0.12	0.00	99.58	0.25	0.00	0.05	0.00

Table 3

Stream results for the 80% load case.

No.	$T$ (°C)	$p$ (bar)	$\dot{m}$ (kg/s)	$MW$ (kg/kmol)	CH <sub>4</sub> (vol%)	C <sub>2+</sub> (vol%)	H <sub>2</sub> (vol%)	CO (vol%)	CO <sub>2</sub> (vol%)	H <sub>2</sub> O (vol%)	O <sub>2</sub> (vol%)	N <sub>2</sub> (vol%)	Ar (vol%)
1	16.0	31.00	16.0	20.73	79.84	16.72	-	-	2.92	-	-	0.51	-
3	397.5	17.68	16.0	20.73	79.84	16.72	-	-	2.92	-	-	0.51	-
6	494.9	15.56	41.8	18.97	28.01	5.86	-	-	1.03	64.91	-	0.18	-
8	494.2	14.28	41.8	17.30	35.21	0.00	8.76	0.11	5.22	50.53	-	0.16	-
9	15.0	1.01	535.6	28.86	-	-	-	-	0.03	1.01	20.74	77.29	0.93
10	377.0	13.91	60.0	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
15	489.3	13.41	79.5	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
18	334.9	11.70	121.3	19.27	0.06	0.00	28.74	10.34	5.18	21.38	0.00	33.90	0.40
20	195.4	10.35	121.3	19.27	0.06	0.00	36.03	3.06	12.46	14.10	0.00	33.90	0.40
21	229.2	9.85	121.3	19.27	0.06	0.00	38.71	0.37	15.15	11.41	0.00	33.90	0.40
29	200.4	20.00	67.8	14.57	0.08	0.00	52.24	0.50	0.65	0.20	0.00	45.78	0.55
31	207.5	2.21	8.0	18.02	-	-	-	-	-	100.00	-	-	-
32	486.2	9.26	18.9	18.02	-	-	-	-	-	100.00	-	-	-
36	297.5	82.88	74.1	18.02	-	-	-	-	-	100.00	-	-	-
37	568.0	73.69	114.4	18.02	-	-	-	-	-	100.00	-	-	-
39	1270.0	13.35	497.0	26.75	-	-	-	-	0.33	17.98	9.45	71.38	0.86
40	602.0	1.03	556.8	26.96	-	-	-	-	0.30	16.28	10.58	71.98	0.87
45	228.0	3.68	19.9	18.02	-	-	-	-	-	100.00	-	-	-
53	208.3	3.68	25.8	18.02	-	-	-	-	-	100.00	-	-	-
55	40.9	110.00	40.3	43.89	0.00	0.00	0.12	0.00	99.58	0.25	0.00	0.05	0.00

321 efficiency but the price is paid in the resulting increased plant complexity.

322 Part load calculations are one way of investigating how flexible the plant is  
323 to off-design conditions. In the analysis performed in the study, part load  
324 behavior is rather good with efficiency reductions from baseload operation  
325 comparable to the reference combined cycle plant. Based on the analysis per-  
326 formed in the paper, it is possible to operate a complex plant like this one  
327 at part loads down to 60% GT load and possibly lower. Not included in the  
328 part load study are compressor mapping for off-design calculations for the air

Table 4

Stream results for the 60% load case.

No.	$T$ (°C)	$p$ (bar)	$\dot{m}$ (kg/s)	$MW$ (kg/kmol)	CH <sub>4</sub> (vol%)	C <sub>2+</sub> (vol%)	H <sub>2</sub> (vol%)	CO (vol%)	CO <sub>2</sub> (vol%)	H <sub>2</sub> O (vol%)	O <sub>2</sub> (vol%)	N <sub>2</sub> (vol%)	Ar (vol%)
1	16.0	31.00	12.9	20.73	79.84	16.72	-	-	2.92	-	-	0.51	-
3	400.5	13.22	12.9	20.73	79.84	16.72	-	-	2.92	-	-	0.51	-
6	501.6	12.72	33.7	18.97	28.02	5.86	-	-	1.03	64.90	-	0.18	-
8	497.9	11.58	33.7	17.24	34.89	0.00	9.50	0.12	5.38	49.95	-	0.16	-
9	15.0	1.01	465.3	28.86	-	-	-	-	0.03	1.01	20.74	77.29	0.93
10	355.0	11.84	45.0	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
15	489.6	11.34	63.8	28.85	-	-	-	-	0.03	1.02	20.73	77.29	0.92
18	314.5	9.76	97.4	19.24	0.04	0.00	28.89	10.39	5.16	21.32	0.00	33.79	0.40
20	182.4	8.41	97.4	19.24	0.04	0.00	36.49	2.79	12.76	13.72	0.00	33.79	0.40
21	214.2	7.91	97.4	19.24	0.04	0.00	39.00	0.29	15.27	11.22	0.00	33.79	0.40
29	205.1	20.00	54.3	14.49	0.06	0.00	52.56	0.39	0.65	0.21	0.00	45.58	0.55
31	204.0	1.92	6.4	18.02	-	-	-	-	-	100.00	-	-	-
32	489.6	7.90	15.6	18.02	-	-	-	-	-	100.00	-	-	-
36	286.0	69.98	60.2	18.02	-	-	-	-	-	100.00	-	-	-
37	568.1	62.25	95.5	18.02	-	-	-	-	-	100.00	-	-	-
39	1200.0	11.36	433.5	26.89	-	-	-	-	0.28	16.82	10.25	71.79	0.86
40	590.0	1.03	485.5	27.09	-	-	-	-	0.26	15.23	11.30	72.35	0.87
45	233.0	3.27	14.8	18.02	-	-	-	-	-	100.00	-	-	-
53	207.0	3.27	20.8	18.02	-	-	-	-	-	100.00	-	-	-
55	40.9	110.00	32.7	43.89	0.00	0.00	0.12	0.00	99.58	0.25	0.00	0.05	0.00

329 compressor, fuel compressor, and CO<sub>2</sub> compression train. Energy requirement  
330 changes per kg of CO<sub>2</sub> for the reboiler in the amine absorption system at off-  
331 design points are not considered either. Including these details in the model  
332 could show a different part load behavior.

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## References

- Andersen, T., Kvamsdal, Hanne, M., Bolland, O., 2000. Gas turbine combined cycle with CO<sub>2</sub> capture using auto-thermal reforming of natural gas. In: ASME Turbo Expo 2000. Munich, Germany.
- Bohm, M. C., Herzog, H. J., Parsons, J. E., Sekar, R. C., 2007. Capture-ready coal plants - Options, technologies and economics. *International Journal of Greenhouse Gas Control* 1 (1), 113–120.
- Bolland, O., 2006. Thermal power generation. p. 126, part of the literature in the course "Thermal power cycles and co-generation" at NTNU.
- Chalmers, H., Gibbins, J., 2007. Initial evaluation of the impact of post-combustion capture of carbon dioxide on supercritical pulverised coal power plant part load performance. *Fuel* 86 (14), 2109–2123.
- Chapel, D. G., Mariz, C. L., October 4-6 1999. Recovery of CO<sub>2</sub> from flue gases: Commercial trends. In: Canadian Society of Chemical Engineers annual meeting. Saskatoon, Saskatchewan, Canada.
- Chiesa, P., Lozza, G., Mazzocchi, L., 2005. Using hydrogen as gas turbine fuel. *Journal of Engineering for Gas Turbines and Power* 127 (1), 73–80.
- Consonni, S., Viganò, F., 2005. Decarbonized hydrogen and electricity from natural gas. *International Journal of Hydrogen Energy* 30 (7), 701–718.
- Corradetti, A., Desideri, U., 2005. Analysis of gas-steam combined cycles with natural gas reforming and CO<sub>2</sub> capture. *Journal of Engineering for Gas Turbines and Power* 127 (3), 545–552.
- Ertesvåg, I. S., Kvamsdal, H. M., Bolland, O., 2005. Exergy analysis of a gas-turbine combined-cycle power plant with precombustion CO<sub>2</sub> capture. *Energy* 30 (1), 5–39.
- Grabke, H. J., Spiegel, M., 2003. Occurrence of metal dusting - Referring to

- failure cases. *Materials and Corrosion* 54 (10), 799–804.
- Griffin, T., Sundkvist, S. G., Åsen, K., Bruun, T., 2005. Advanced zero emissions gas turbine power plant. *Journal of Engineering for Gas Turbines and Power* 127 (1), 81–85.
- Haag, J. C., Hildebrandt, A., Hönen, H., Assadi, M., Kneer, R., 2007. Turbo-machinery simulation in design point and part-load operation for advanced CO<sub>2</sub> capture power plant cycles. Vol. 3 of *Proceedings of the ASME Turbo Expo*. American Society of Mechanical Engineers, New York, NY, United States, pp. 239–250.
- Incropera, F. P., DeWitt, D. P., 1990. *Fundamentals of heat and mass transfer*, 3rd Edition. John Wiley & Sons, Inc., p. 422.
- International Energy Agency, 2006. *World Energy Outlook*.
- Ishida, M., Jin, H., 1994. A new advanced power-generation system using chemical-looping combustion. *Energy* 19 (4), 415–422.
- Jericha, H., Göttlich, E., Sanz, W., Heitmeir, F., 2004. Design optimization of the graz cycle prototype plant. *Journal of Engineering for Gas Turbines and Power* 126 (4), 733–740.
- Kehlhofer, R. H., Warner, J., Nielsen, H., Bachmann, R., 1999. *Combined-Cycle Gas & Steam Turbine Power Plants*, 2nd Edition. PennWell Publishing Company.
- Knudsen, J. N., Vilhelmsen, P. J., Biede, O., Jensen, J. N., 19-22 June 2006. Castor 1 t/h CO<sub>2</sub> absorption pilot plant at the Elsam Kraft A/S Esbjerg power plant - First year operation experience. In: *8th International Conference on Greenhouse Gas Control Technologies*. Trondheim, Norway.
- Kvamsdal, H. M., Jordal, K., Bolland, O., 2007. A quantitative comparison of gas turbine cycles with CO<sub>2</sub> capture. *Energy* 32 (1), 10–24.
- Le Thiez, P., Mosditchian, G., Torp, T., Feron, P., Ritsema, I., Zweigel, P.,

- Lindeberg, E., 2004. CASTOR: from pie in the sky to commercial reality. *Modern Power Systems* 24 (7), 19–20.
- Lozza, G., Chiesa, P., 2002a. Natural gas decarbonization to reduce CO<sub>2</sub> emission from combined cycles - Part I: Partial oxidation. *Journal of Engineering for Gas Turbines and Power* 124 (1), 82–88.
- Lozza, G., Chiesa, P., 2002b. Natural gas decarbonization to reduce CO<sub>2</sub> emission from combined cycles - Part II: Steam-methane reforming. *Journal of Engineering for Gas Turbines and Power* 124 (1), 89–95.
- Mathieu, P., Nihart, R., 1999. Zero-emission MATIANT cycle. *Journal of Engineering for Gas Turbines and Power* 121 (1), 116–120.
- Moliere, M., 2005. Expanding fuel flexibility of gas turbines. Vol. 219 of *Proc. Inst. Mech. Eng. A, J. Power Energy (UK)*. Mech. Eng. Publications, Brussels, Belgium, pp. 109–119.
- Möller, B. F., Genrup, M., Assadi, M., 2007. On the off-design of a natural gas-fired combined cycle with CO<sub>2</sub> capture. *Energy* 32 (4), 353–359.
- Moulijn, J. A., Makkee, M., Diepen, A. v., 2007. *Chemical Process Technology*. John Wiley & Sons Ltd, pp. 158–159.
- Naqvi, R., Wolf, J., Bolland, O., 2007. Part-load analysis of a chemical looping combustion (CLC) combined cycle with CO<sub>2</sub> capture. *Energy* 32 (4), 360–370.
- Richter, H. J., Knoche, K. F., 1983. Reversibility of combustion processes. In: Gaggioli, R. A. (Ed.), *A.C.S. Symposium Series 235*. Washington D.C., USA, pp. 71–85.
- Shilling, N. Z., Jones, R. M., 2003. The impact of fuel flexible gas turbine control systems on integrated gasification combined cycle performance. Vol. 1 of *Proceedings of the ASME Turbo Expo*. American Society of Mechanical Engineers, New York, NY, United States, pp. 259–265.



- Tomczak, H.-J., Benelli, G., Carrai, L., Cecchini, D., 2002. Investigation of a gas turbine combustion system fired with mixtures of natural gas and hydrogen. *IFRF Combustion Journal*.
- van Loo, S., van Elk, E. P., Versteeg, G. F., 2007. The removal of carbon dioxide with activated solutions of methyl-diethanol-amine. *Journal of Petroleum Science and Engineering* 55 (1-2), 135–145.
- Wagner, W., Cooper, J. R., Dittmann, A., Kijima, J., Kretzschmar, H. J., Kruse, A., Mares, R., Oguchi, K., Sato, H., Stöcker, I., Šifner, O., Takaishi, Y., Tanishita, I., Trübenbach, J., Willkommen, T., 2000. The IAPWS industrial formulation 1997 for the thermodynamic properties of water and steam. *Journal of Engineering for Gas Turbines and Power* 122 (1), 150–180.
- Wilson, M., Tontiwachwuthikul, P., Chakma, A., Idem, R., Veawab, A., Aroonwilas, A., Gelowitz, D., Barrie, J., Mariz, C., 2004. Test results from a CO<sub>2</sub> extraction pilot plant at Boundary Dam coal-fired power station. *Energy* 29 (9-10), 1259–1267.
- Zhang, X., Wang, J., Zhang, C.-f., Yang, Y.-h., Xu, J.-j., 2003. Absorption rate into a MDEA aqueous solution blended with piperazine under a high CO<sub>2</sub> partial pressure. *Industrial and Engineering Chemistry Research* 42 (1), 118–122.