

Precipitation behavior of Al-Si-Cu-Mg(-Fe) alloys by a deformation-semisolid extrusion process

DaeHan Kim¹, JaeHwang Kim², * Sigurd Wenner^{3,4}, Elisabeth Thronsen⁴, Calin Daniel Marioara³, Randi Holmestad⁴, Equo Kobayashi¹

¹Department of Materials Science and Engineering, Tokyo Institute of Technology, 2-12-1-S8-18 Ookayama, Meguro-ku, Tokyo, Japan, 152-8552

²Carbon and Light Materials Application Group, Korea Institute of Industrial Technology, 222 Palbok-ro, Jeonju-si, South Korea, 54896

³SINTEF Industry, N-7465 Trondheim, Norway

⁴Department of Physics, Norwegian University of Science and Technology (NTNU), N-7491 Trondheim, Norway

*Corresponding author at: Korea Institute of Industrial Technology, Carbon and Light Materials Application Group, 222 Palbok-ro, Jeonju-si, South Korea, 54896

E-mail address: raykim@kitech.ac.kr

Abstract

Al-4.5Si-1Cu-0.3Mg(-1Fe) (wt.%) alloys fabricated by a deformation-semisolid extrusion (D-SSE) process have been investigated by transmission electron microscopy, down to the atomic level. T5 and T6 heat treatments were conducted to understand the age-hardening behavior of the alloys. Disordered Mg-Si(-Cu) precipitates with strong Cu enrichments at their interfaces with the Al matrix have been observed in the overaged conditions of both heat treatments and in the peak hardness of the T6 condition, but only Cu-containing atomic clusters were detected in the peak hardness of the T5 heat treatment. Despite having a lower bulk precipitate number density at comparable precipitate size and volume fraction, hardness in the T6 condition was higher in the alloy with highest Fe content due to the extra contribution from the precipitates nucleated on fragmented β -Al₅FeSi particles and grain boundaries. Many of these precipitates were Q'-phase, and two new coherent interfaces with the Al matrix are reported for this phase.

Keywords: hybrid precipitates, grain boundary precipitates, Fe-intermetallic compounds, Al-Si-Cu-Mg(-Fe) alloys.

1. Introduction

Aluminum alloys are often used as automotive and structural components due to light weight, specific strength and formability. Fe is introduced to aluminum alloy from scraps, secondary ingots, and during recycling and processing. Various Fe-IMCs (Fe-intermetallic compounds) i.e. α -, β - and π -AlFeSi particles appear when Fe is introduced to Al-Si alloys [1-2]. They are detrimental to the ductility acting as crack-initiating sites and paths [3].

1 Recently, the negative effect of Fe-IMCs on the mechanical properties has been reduced by deformation-
2 semisolid forming (D-SSF) and caliber rolling processes [4-6]. The D-SSF process is an effective method for
3 breaking Fe-IMCs and inducing large strain in the Al matrix. Since Fe is introduced during recycling, it is
4 important to control the irregular-shaped Fe-IMCs and to optimize the mechanical properties of high Fe-
5 containing alloys. Further enhancement of the mechanical properties in high Fe-containing alloys can be
6 obtained by the formation of precipitates through a post heat treatment.

7 Al-Si based alloys are used for casting and do not require heat treatments. However, precipitates can nucleate
8 during aging if solute elements such as Mg, Si, and Cu are added. For example, those elements are added to
9 enhance the strength by precipitate hardening in A319 (Al-Si-Cu-Mg) and A356 (Al-Si-Mg) cast alloys. The
10 alloys studied in the present work have composition Al-4.5Si-1Cu-0.3Mg(-1Fe) (wt%) and are considered heat
11 treatable. By adding elements such as Mg, Si and Cu to pure Al, metastable precipitates nucleate during heat
12 treatments at elevated temperature. The precipitates are favorable for strength and different types can nucleate
13 depending on the composition of the alloy and thermomechanical treatment [7]. The precipitation sequence of
14 Al-Cu alloys is usually given as follows [8]:

15 super saturated solid solution (SSSS) \rightarrow atomic clusters \rightarrow Guinier-Preston (GP) 1 zone \rightarrow GP2 zone (θ') \rightarrow
16 $\theta' \rightarrow \theta$. The precipitation sequence of Al-Mg-Si-Cu alloy is summarized as follows [9-10]:

17 SSSS \rightarrow atomic clusters \rightarrow GP zones \rightarrow β'' , L, QP, QC \rightarrow β' , Q' \rightarrow β , Q.

18 The GP1 zone in the Al-Cu system is actually an enriched {200}Al plane [8, 11] and will be referred to as either
19 'GP zone' or 'enriched {200}Al plane', or 'Cu wall' in this paper. The GP2 zone is also called θ' and consists of
20 two enriched {200}Al planes separated by $4d_{Al_{200}}$ [12].

21 The kinetics of precipitation is accelerated by increasing Cu content in Al-Mg-Si-Cu alloys [13]. Moreover, Cu
22 additions to the Al-Mg-Si system enable nucleation of complex and interesting precipitates. The number density
23 of the metastable β'' phase gradually decreases while the occurrence of Q', S, L and θ' phases increases with
24 increasing Cu content [14]. A high amount of Cu addition (~4.5 wt%) causes the hardness to decrease more
25 rapidly as compared to corresponding Cu-free alloys during over-aging [13].

26 Gazizov et al. [15] have investigated the precipitation behavior of an Al-4.9Cu-0.74Mg-0.51Si-0.48Mn-0.1Cr-
27 0.08Ti-0.02Fe alloy. The terms of 'Cu capsule' and 'hybrid precipitate' are introduced to clarify the crystal
28 structure of the precipitates. A 'Cu capsule' refers to a Mg-Si(-Cu) needle-type precipitate with Cu enrichment at
29 all interfaces as viewed along the needle-length, giving the appearance of a tube, encapsulating precipitate.
30 'Hybrid precipitates' are defined as needle or lath type disordered precipitates containing structural units of

1 known precipitates from both the Al-Mg-Si-Cu and Al-Cu systems. Wenner et al. [16] studied the precipitation
2 formed during the artificial aging of an aluminum alloy containing Zn, Cu, Mg and Si. Combinations of
3 disordered Al-Mg-Si-Cu phases as well as S and θ' were found. These studies have used the high-angle annular
4 dark-field scanning transmission electron microscopy (HAADF-STEM) technique, which provides Z-contrast
5 spatial atomic resolution that enables the investigation of precipitate atomic arrangement in great details [17-19].

6 In the present study, two Al-Si- cast alloys containing approximately the same amount of Si, Cu and Mg and
7 different amount of Fe are studied for developing recycled aluminum alloys using a deformation-semisolid
8 extrusion (D-SSE) process [4]. Low Fe content is required to improve the ductility of aluminum alloys, while a
9 high amount of Fe (~1 %) is purposefully added to form Fe-IMCs. A high Fe-containing aluminum alloys have
10 been neglected for studying structural analysis of precipitates since Fe-IMCs are detrimental to the mechanical
11 properties and lifetime of materials. A high strength and ductile aluminum alloys containing Fe contents would
12 be feasible using a D-SSE process and post heat treatments. Cost saving and recycled resources will be followed.
13 T5 and T6 heat treatments are considered in this study to control the precipitation behavior and the balance
14 between strength and ductility. The formation and kinetics of precipitates are also affected by the cooling rates
15 from solution heat treatment (SHT) before natural- and artificial aging (NA and AA). Meanwhile, the role of Fe-
16 IMCs on the formation of nano-sized precipitates has not been fully documented and the crystal structures of
17 hybrid precipitates containing Cu have not been fully understood. Therefore, the aim of the present study is to
18 investigate the precipitation behavior of Al-4.5Si-1Cu-0.3Mg(-1Fe) alloys produced by a D-SSE process. We
19 focus on precipitate statistics and the atomic structure of hybrid precipitates using atomic resolution HAADF-
20 STEM. As all the metastable precipitates in the Al-Mg-Si(-Cu) and Al-Cu system have plate/needle/lath
21 morphologies with the main growth direction along $\langle 100 \rangle_{Al}$, all the TEM investigations shown in this work
22 were performed with the Al matrix in this orientation.

24 **2. Experiment procedure**

25 Chemical compositions of the two Al-Si based alloys are given in Table 1. From now on the two alloy
26 compositions are termed '0.2 Fe and 1 Fe'. The alloys were cast and homogenized at 520 °C for 24 h. The D-
27 SSE process, which is a combination of deformation and a semi-solid extrusion process, was conducted to
28 uniformly fragment the Fe-IMCs. 50% of deformation at 350 °C was applied to samples with a dimension of 20
29 x 30 x 40 mm³. The samples were heated to a semi-solid temperature (~555 °C) and cylinder type samples with
30 a radius of 4 mm were produced by the semi-solid extrusion process. The extrusion ratio was 40:1 and the ram

1 speed was 0.5 mm / sec. The extruded profiles were stored at room temperature for approximately six months.
2 T5 (AA at 170 °C without SHT) and T6 (AA at 170 °C with SHT at 520 °C for 2 h followed by water
3 quenching) treatments were carried out. Vickers hardness testing was performed using a load of 200 g and a
4 dwell time of 15 s. At least five indentations were acquired on each sample to obtain an average value of
5 hardness.

6 TEM samples were made by polishing to less than 100 µm thickness and punching out 3 mm discs. A mixture
7 of 2/3 methanol and 1/3 nitric acid was used for electropolishing. The electrolyte was cooled down to a
8 temperature lower than -25 °C and a voltage of 20 V was used. A JEOL JEM-2100 operated at 200 kV, equipped
9 with a GIF-2000 Electron energy loss spectroscopy (EELS) spectrometer, was used for bright-field (BF) TEM
10 imaging of the precipitate microstructure. The area of precipitates is estimated using Image-J software based on
11 the cross section of precipitates pointing in the viewing direction. 6 TEM images corresponding to
12 approximately 1000 precipitate cross sections were acquired. TEM images were filtered by band pass and
13 precipitates with a circularity from 0.7 to 1.0 were selected to increase accuracy of the quantification. The
14 statistics were obtained from the Al matrix away from regions with grain boundaries (GBs) and β-Al₅FeSi
15 particles. The number, cross-section area and length of precipitates including the thickness of TEM samples
16 were used to estimate the volume fraction of the precipitates. The full statistical approach can be found from the
17 previous studies [20-21]. The thickness of the specimen is estimated from EELS measurements. EELS has
18 become a common technique for measuring the thickness of an electron-transparent specimen, which is required
19 for the estimation of precipitate number density by TEM imaging. The thickness, t is given by the ratio of the
20 total intensity in the EELS spectrum to the total intensity of the zero-loss spectrum $\ln(I_t/I_0)$ as [22],

$$21 \quad t = \lambda \ln(I_t/I_0) \quad (1)$$

22 where λ represents the inelastic mean free path (IMFP) in the aluminum alloy. The value of the IMFP (λ_{exp}) used
23 in this work is 143 nm [23].

24 An image and probe Cs-corrected JEOL ARM-200F cold FEG microscope operated at 200 kV was employed
25 for HAADF-STEM imaging. The samples were plasma-cleaned to remove contaminations using a Fishione
26 1020 plasma cleaner before they were inserted into the TEM. The probe size was 0.08 nm, the convergence
27 semi-angle was 27 mrad and the inner and outer collection angles were 42 and 178 mrad, respectively. Some
28 images are slightly distorted due to specimen drift during acquisition. HAADF-STEM images are filtered unless
29 otherwise specified. A circular band pass mask is applied on the FFTs and inverse FFTs (IFFTs) were performed
30 on the masked area, suppressing all features with separation shorter than 0.15 nm in real space. It is noted that

1 0.15 nm is close to the minimum projected atomic column separation for precipitates in the Al-Mg-Si(-Cu)
2 system viewed along their needle lengths.

3. Results and discussions

3.1 Age-hardening of T6 heat-treated alloys

6 Fig. 1 (a) shows the age-hardening curves of T5 and T6 heat-treated 0.2 Fe and 1 Fe alloys. The arrows
7 indicate which samples were selected for TEM observations. Hardness values of Al-Si alloys increase with
8 alloying elements [24]. Hardness of Fe-IMC is higher than the primary aluminum in Al-Si alloys [25]. Namely,
9 the formation of Fe-IMCs is contributed to an increase in hardness. Hardness of 1 Fe is slightly higher than 0.2
10 Fe right after solution heat treatment in Fig. 1 (a) (green and blue lines), indicating the difference of Fe contents.
11 The higher HV in 1 Fe is not only be attributed to the precipitation sequence and precipitates, but also related to
12 the fragmented Fe-IMCs. Micrographs of as-extruded 0.2 and 1 Fe alloys before aging are presented as shown in
13 Fig. 2. Si and Fe-IMCs are marked by yellow arrows, respectively. A high number density of Fe-IMCs is
14 identified in case of 1 Fe (Fig. 2 (b)).

15 The precipitation kinetics was accelerated for the conditions T6 by rapid cooling from SHT while relatively
16 broad aging curves were identified in the T5 treated alloys. The hardness difference of T5 and T6 treated 1 Fe
17 alloys is as shown in Fig. 1 (b). The hardness difference is calculated based on the difference between the
18 hardness values of the aged sample and the as-quenched (for T6) or as-extruded (for T5) ones. The slope of the
19 aging curves is connected to the kinetics of clustering (up to 2 ks) and precipitation. It is observed that the
20 hardness obtained from precipitates during AA increases with SHT. The hardness of T5 and T6 1 Fe alloys are
21 comparable at the peak aging stage.

22 Fig. 3 shows bright-field TEM images of the T6 treated alloys after aging for 6 and 72 h. Nano-sized needle
23 and lath precipitates in the bulk oriented along the $\langle 001 \rangle_{Al}$ zone axis are observed in both 0.2 Fe and 1 Fe
24 alloys. A slightly higher number density of precipitates is measured in the 0.2 Fe alloy aged for 6 h compared to
25 that of 1 Fe. The average needle lengths of precipitates in 0.2 Fe and 1 Fe alloys are very short, 16.2 and 13.3
26 nm, respectively. The average precipitate cross sections in 0.2 Fe and 1 Fe alloys are 2.9 and 3.3 nm²,
27 respectively. This gives a higher precipitate volume fraction in the 0.2 Fe alloy aged for 6 h. The full statistical
28 results of precipitates on the matrix are given in Table 2. It is observed that T6 treated 1 Fe alloy aged for 6 h has
29 lower precipitate number densities than the 0.2 Fe alloy since solute Si content consumed to form the Fe-IMCs
30 with Fe and Al in the 1 Fe. ~~at comparable precipitate sizes and volume fractions.~~ Thus, the strength contribution

1 from bulk hardening precipitates should be lower in this alloy. However, Fig. 1 (a) shows that 1 Fe alloy is
2 harder than 0.2 Fe, implying the existence of additional strength contributions. The Q' phase on grain boundaries
3 and β -Al₅FeSi (which are not included in the numbers) contribute to the higher hardness than that of 0.2 Fe.

4 Figures 4 and 5 show high-resolution HAADF-STEM images of precipitates in the T6 treated 0.2 Fe and 1 Fe
5 alloys aged for 6 h. The L phase, incorporating local symmetries of the C phase [9], GP like structure [8, 15] and
6 Guinier-Preston-Bagaryastsky (GPB) zone [26-27] is shown in Fig. 3 (a). Although the L phase is disordered,
7 the precipitates sometimes have mirror and/or rotation symmetries [28], as exemplified by the mirror plane
8 marked by a dotted line in Fig. 4 (a). Both GP like structures and incomplete GPB zone units from the Al-Cu
9 and Al-Cu-Mg systems were identified [8, 15, 26-27]. A hybrid β'' phase mixed with a GP like structure is
10 shown in Fig. 4 (b). The GP-like structure, or 'Cu wall' is located at the left side of the particle. The 'eye' is the
11 building block of β'' , which can be stacked in various ways [29-31]. Two terms β_2'' and β_3'' were found by
12 previous works [29-31]. Different β'' variants are overlaid as shown in Fig 4 (c). A half monoclinic β'' unit at the
13 upper side, β_2'' in the middle, β_3'' at the lowest, and another incomplete (only 3 eyes) β'' at the low-left side can
14 be identified in Fig. 4 (b). Some Cu columns, which have higher atomic number than Al, Mg and Si, and thus
15 appear bright, are found at certain sites in the β'' phase.

16 In the 1 Fe alloy, a Cu capsule (middle), Cu walls (upper right) and β'' (lower right) are shown in Fig 5. (a).
17 The Cu capsule type [15] refer to Mg-Si-Cu precipitates with high coherency with the Al matrix that are
18 encapsulated by Cu-enriched {200}Al planes (GP-like structures). The Cu walls precipitate refers to a new type
19 found by Marioara et al. [12] that contains a Mg-Si core delimited by two Cu walls (enriched {200}Al planes)
20 separated by 6d Al₂₀₀. In addition, a precipitate consisting of a fragment of β'' , segregated Cu atoms at the
21 interfaces and a GP like structure, was observed as shown in Fig. 5 (b). All these precipitates can be classified as
22 hybrids because they incorporate structural elements from the precipitates formed in the Al-Mg-Si(-Cu) and Al-
23 Cu systems.

24 In the following, we focus on details in the 1 Fe alloy to investigate the crystal structure of the hybrid
25 precipitates. HAADF-STEM images of the T6 treated 1 Fe alloy aged for 72 h are presented in Fig. 6. In (a-b),
26 two Q' phases surrounded by GP like structures are shown. A disordered precipitate containing local Cu sub-unit
27 clusters (the building blocks of Q' and C phases [28]) and units isostructural with the stacking fault reported in
28 [12] is observed in Fig. 5 (c). A β'' phase with Cu-enriched interfaces in (d), L phase with a disordered structure
29 in (e) and the L phase containing fragments of C and Q' phases in (f) are clearly identified. It seems like the Cu
30 columns are systematically ordered in all types of precipitates as they grow during aging. Apart from Cu sites in

1 the C and Q' structures, they tend to position at the vicinity of the interfaces between precipitates and matrix.

3 3.2 Q' phase on β -Al₅FeSi particles and grain boundaries

4 Fe-containing particles are known as impurities in aluminum alloys. The nucleation of α - and β -AlFeSi
5 particles is determined by chemical composition and heat treatment. Lervik et al. [32] have reported that β -
6 Mg₂Si and η -MgZn₂ were nucleated on AlFeSi particles in an AA 7003 alloy. Only β -Al₅FeSi particles were
7 detected in the 1 Fe alloy owing to the high amount of Si and Fe contents [33] and the ratio of Fe/Si was found
8 to be approximately 1:1 by EDS analysis. In this alloy, the Q' phase has been found to nucleate on β -Al₅FeSi
9 particles. One example is shown in Fig. 7 (a-c). It is noted that β -Al₅FeSi particles are thermally stable at
10 elevated temperature, and act as nucleation sites for precipitates. HAADF-STEM images in Fig 7 (d-f) zoom on
11 the Q' phases nucleated on the β -Al₅FeSi particle. Coherent (along $\langle 510 \rangle$ Al) and irregular interfaces between
12 the Q' and the Al matrix are shown in Fig. 7 (e) and (f), respectively. It is interesting to notice that Cu atoms
13 positioned at the coherent interface have a triangle-shaped feature, which has not been reported before. This
14 atomic arrangement is also sporadically observed at the interface of the particle in Fig. 7 (f).

15 In addition, the Q' phase nucleates at grain boundaries, either during the cooling after extrusion or during
16 artificial aging. A Q' phase with different interfaces is shown in Fig. 8. The interfaces with Al of the left [001]Al
17 oriented grain and the right grain with unknown orientation are different. An example of periodic arrangement
18 of Cu atomic columns at the interface between the precipitates and the matrix, is identified in Fig. 8 (b) [12, 31].
19 It consists of short Cu walls (enriched {200}Al planes viewed edge-on) with 1.04 nm periodicity along $\langle 510 \rangle$ Al.
20 It should be noted that this interface is very different from a regular Q' $\langle 510 \rangle$ Al interface [12] and from the
21 interface of the Q' phase nucleated on the β -Al₅FeSi particle and presented in Fig. 7 (e). The coherent interfaces
22 with atomic overlay along $\langle 510 \rangle$ Al of the Q' precipitates from [12], from Fig. 7 (e) and Fig. 8 (b) are presented
23 in Fig. 9. The overlay is based on the construction rules for precipitates in Al-Mg-Cu and Al-Mg-Si(-Cu) alloys
24 published in [30], and on the Z-contrast of atomic columns provided by the HAADF-STEM images. In the case
25 of the normal interface (Fig. 9 (a)), the connection between Q' and Al matrix along $\langle 510 \rangle$ Al is done directly, as
26 the Q' unit cells have direct contact with the Al matrix. However, in the case of the Q' nucleated at the β -Al₅FeSi
27 (Fig. 9 (b)) and of the Q' nucleated on the GB (Fig. 8 (c)), the connection between Q' unit cells and the Al matrix
28 is done **through** the insertion of Cu-rich buffer layers. One layer is in the form of GP-like units (Cu walls) in the
29 case of the Q' nucleated at the GB, and another as a more triangular atomic arrangement in the case of Q'
30 nucleated on the β -Al₅FeSi. The introduction of the Cu-rich buffer layers facilitates the connection of Q' with

1 the Al matrix in different orientations, giving the Q' phase more growth flexibility. In the case of the Q'
2 nucleated at the GB, the Al matrix is shifted one Al unit cell diagonal (along $\langle 110 \rangle_{\text{Al}}$) as compared to the
3 normal orientation. Interestingly, some Q' unit cell Si corners inside the precipitate, marked by arrows, are
4 unusually bright, indicating Cu enrichment. In the case of the Q' nucleated on the $\beta\text{-Al}_5\text{FeSi}$, the Al matrix has a
5 30° rotation with respect to the normal orientation as shown in the supplementary material. Fig. 9 (b) was
6 flipped and rotated to have the same precipitate orientation as the normal interface image.

7 The average grain sizes of as-extruded 0.2 Fe and 1 Fe are 134.8 and 33.8 μm , respectively, estimated by
8 optical microscope. The nucleation sites at grain boundaries increase with the refining grain size of the 1 Fe
9 alloy. Although the precipitation hardening in the matrix of the 1 Fe alloy is reduced compared to the 0.2 Fe
10 alloy as seen in Table 2, diverse nucleation sites for hardening precipitates i.e. grain boundaries and $\beta\text{-Al}_5\text{FeSi}$
11 particles during aging are higher in numbers. Solute elements and vacancies tend to diffuse toward grain
12 boundaries and interfaces of large particles such as $\beta\text{-Al}_5\text{FeSi}$, during cooling and aging. Therefore, higher
13 density of grain boundaries and fragmented $\beta\text{-Al}_5\text{FeSi}$ particles in this alloy provides high-diffusivity paths for
14 vacancies. We suggest that this is the reason for the improved mechanical properties of the 1 Fe alloy.

16 3.3 Age-hardening of T5 treated 1 Fe alloys

17 The kinetics of precipitate formation are decelerated by the lack of solution heat treatment in the T5 condition,
18 as shown in Fig. 1 (b). It is due to the low vacancy concentration and supersaturation of solute atoms in the
19 material from slow air cooling and long NA after extrusion. TEM micrographs of the T5 treated 1 Fe alloy aged
20 for 8 h are shown in Fig. 10. Fine dots are detected with BF (a) and dark-field (DF) (b) TEM. The selected area
21 diffraction pattern (c) shows diffuse dots, which are a signature of atomic clustering [34-35]. A HAADF-STEM
22 image is displayed in (d), filtered by selecting the corresponding cluster spots in the FFT of the image, and using
23 Gaussian blur in DigitalMicrograph on the IFFT image to improve the visibility of the atomic clusters. The Cu-
24 rich clusters are extremely small and with much less structure than, for example, the clusters in the Al-Mg-Zn
25 system [36].

26 Micrographs of the T5 treated 1 Fe alloy aged for 72 h are shown in Fig. 11. The microstructure contains large
27 plate-shaped particles in addition to the fine needle-shaped precipitates similar to those found in the 72 h aged
28 T6 condition (shown in Fig. 2 (d)). The nano-sized precipitates were investigated by HAADF-STEM to identify
29 their crystal structure and a representative selection is shown in Fig. 12. A new type of precipitate containing Mg
30 and Si delimited by Cu walls with $5d\text{Al}_{200}$ separation is shown in (a). Two examples of β'' phase with Cu

1 segregation at interfaces are presented in (b-c). Strong Cu enrichment of the Si_3/Al interior sites [37] is observed
2 in (c). Cu capsules with Mg-Si-Cu phase interiors are presented in (d) and (e). In (d) the interfaces consist of Cu
3 walls, while the precipitate in (e) has more irregular interfaces. The interface at the lower precipitate side in this
4 case incorporates the same triangular Cu structure also observed at the Q'-Al interface in Fig. 7 (e). Therefore,
5 two types of segregation were identified. The most common one is the Cu wall and the other one is the new
6 triangular configuration observed in connection to the Q' coherent interface. L phases with asymmetric and
7 symmetric structures are identified in (f-g), respectively. The larger plate-shaped particles are identified as θ' and
8 pure Si (h-i). The size of the hybrid precipitates remains quite smaller as compared to the θ' and Si precipitates
9 even during prolonged aging. The θ' phase and Si precipitates, which are nucleated by clusters formed from
10 room temperature storage and excess Si content [8, 12, 38], are relatively coarse after 72 h aging.

11 During the early aging, the T6 state has its lowest hardness as all solute elements are in solution, while T5
12 retains clusters and some Q' phases nucleated on GBs and $\beta\text{-Al}_3\text{FeSi}$. Extruded samples were cooled down from
13 semi solid temperature. Natural aging was conducted for approximately six months. Those offer the formation
14 of clusters. Fine Cu clusters are identified in T5 treated 1 Fe aged at 8 h (Fig. 10). It is deduced that rapid
15 coarsening may occurs due to the dissolution and remaining of atomic clusters.

16 In addition, dislocations generated from the semi solid extrusion affects the kinetic of precipitates. J. Hu et. al
17 [39] concluded that the formation of precipitates of Al-Mg-Si alloys was accelerated by a high number density
18 of dislocations using an extrusion forming process. More nucleation sites were provided with deformed
19 materials compared to undeformed one. The nucleation of β'' phase prefer to form in matrix while the nucleation
20 of β' favors to form at dislocations when a certain dislocation induced [40]. Thus, T5 treated alloys are forced to
21 offer numerous nucleation sites from dislocations by the D-SSE process though there was lower SSSS. The
22 precipitate microstructure of Al-Cu-Mg-Si alloys is affected by a combination of dislocations and natural aging
23 [41]. The dislocations formed from the D-SSE aid in the formation of Q' phase [36, 42]. The high density of
24 disordered and fine precipitates with good thermal stability observed in the present work is probably due to Cu
25 encapsulation of the Mg-Si phases, preventing their growth [43]. Based on the initial hardness drop in the T5
26 treatment it is deduced that part of the clusters formed during natural aging dissolve initially, while a second
27 wave of cluster precipitation takes place during longer aging time, until peak hardness. At longer aging times
28 (during overaging) no clusters are observed as the microstructure in this case consists of Cu-encapsulated
29 needles/laths. It is debatable whether clusters formed from natural aging and those from artificial aging are
30 similar. Clearly, the thermo mechanical process and natural aging in the T5 treatment give a favorable

1 combination of high strength and good thermal stability in this condition. The HAADF-STEM technique has
2 proven very valuable for investigating the crystal structure of hybrid phases i.e. hybrid β'' , Q', L, Cu capsule in
3 Al-Si-Cu-Mg(-Fe) alloys. This technique allows insights of precipitation behavior and provides information of
4 hybrid phases containing Cu.

6 **4. Conclusions**

7 Semisolid extruded profiles were artificially aged with T6 and T5 heat treatments. Solution heat treatment and
8 aging (T6) led to the formation of needle/lath metastable precipitates in the matrix at both peak hardness and
9 overaged conditions. Their number densities were slightly reduced in the 1 Fe alloy, as compared to the 0.2 Fe.
10 However, the 1 Fe alloy produced a higher hardness due to the additional contribution from precipitates
11 nucleated on grain boundaries and on fragmented β -Al₃FeSi phases. Interestingly, only atomic clusters were
12 observed up to peak hardness in the T5 treatment despite an initial hardness drop. ~~However,~~ Metastable
13 needle/lath precipitates were observed in the overaged conditions of this heat treatment, together with large Si
14 plates and θ' particles.

15 Most of the needle/lath metastable precipitates in the T6 treatment consist of disordered β'' , L/C type with Cu
16 walls at their interfaces, which can be described as 'Cu capsule' types. ~~A recently discovered precipitate type
17 with 6dAl₂₀₀ between the Cu walls was also found. With overaging Q' was observed in addition to the
18 aforementioned phases, but with the main difference that most of the precipitates are completely encapsulated
19 by Cu walls or Cu enriched interfaces. This was also found in the overaged conditions of the T5 treatment.~~ Two
20 new coherent Q'-phase interface along $\langle 510 \rangle$ Al have been found, as well as a new type of precipitate consisting
21 of a Mg-Si core delimited by Cu walls separated by 5dAl₂₀₀. The new interfaces offer the Q' phase greater
22 flexibility for nucleation and growth by introducing a buffer layer of Cu-rich atomic columns between the Q'
23 unit cells and the Al matrix.

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30 **Data availability**

1 The TEM images presented in this work are given in raw/processed form. The raw/processed scanning
2 diffraction data required to reproduce these findings cannot be shared at this time as the data also forms part of
3 ongoing study.

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Table 1 Chemical composition of experimental alloys (wt.%).

Alloy	Si	Cu	Mg	Fe	Al
0.2 Fe	4.44	1.09	0.39	0.15	Bal.
1 Fe	4.47	0.98	0.35	1.02	Bal.

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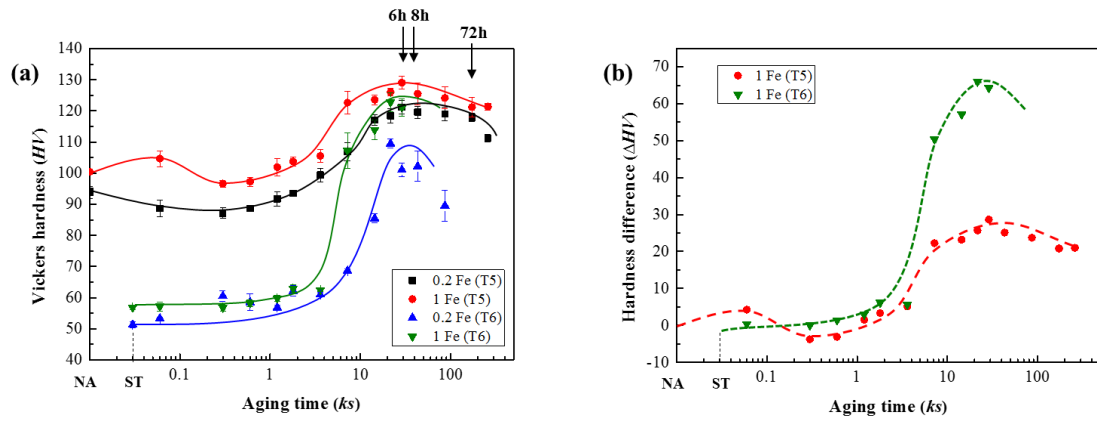
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Table 2 Statistic results of precipitates in T6 treated alloys.

Material	Cross section [nm ²]	Needle length [nm]	Number density [#/um ³]	Volume fraction [%]
0.2 Fe 6 h	2.9 ± 0.2	16.2 ± 10	2303.2 ± 523.8	(0.1, 0.13)
0.2 Fe 72 h	3.5 ± 0.4	30.8 ± 5.5	1236.6 ± 184.3	(0.18, 0.33)
1 Fe 6 h	3.3 ± 0.3	13.3 ± 2.2	1744.0 ± 265.5	(0.07, 0.12)
1 Fe 72 h	4.4 ± 0.8	40 ± 4.1	687.0 ± 240.9	(0.29, 0.38)

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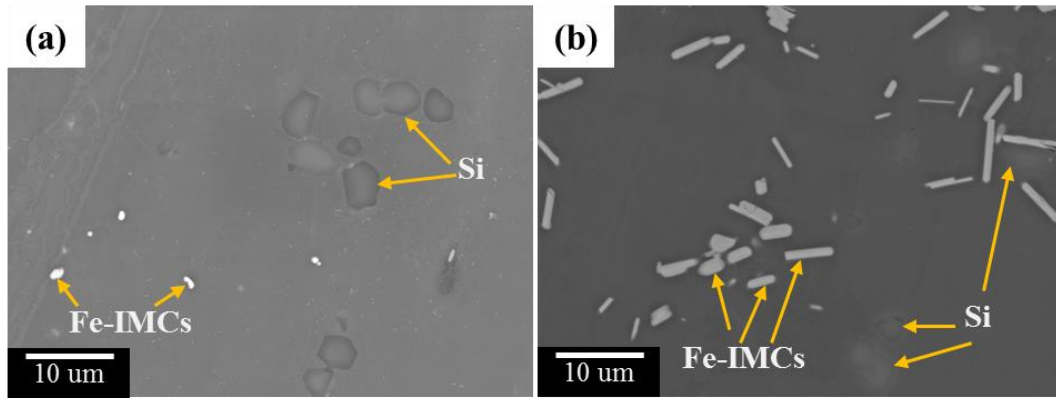
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2 Fig. 1 (a) Age-hardening curves of T5 and T6 treated 0.2 Fe and 1 Fe alloys and (b) hardness difference of T5
 3 and T6 treated 1 Fe alloys.

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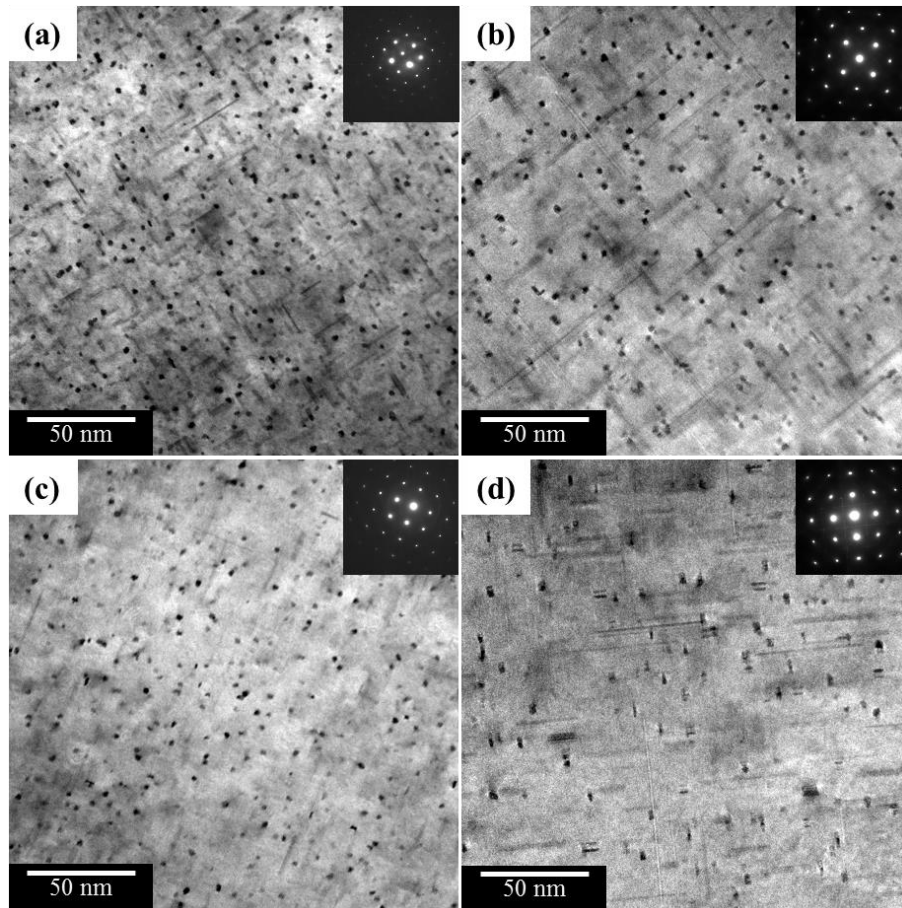


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2 Fig. 2 SEM images of as-extruded samples (before T5/T6 aging); (a) as-extruded 0.2 Fe and (b) as-extruded 1

3 Fe.

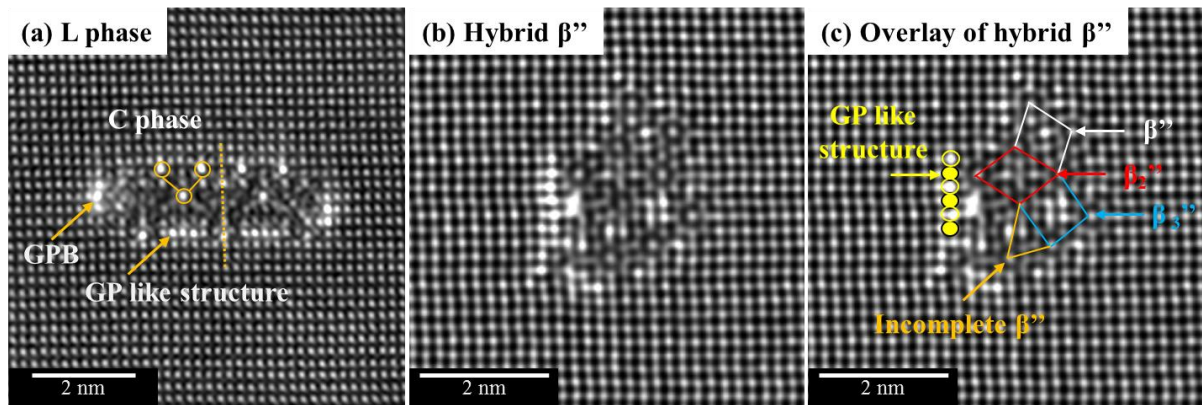
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2 Fig. 3 Overview bright-field TEM images of T6 treated alloys for different times; (a) 0.2 Fe for 6 h,(b) for 72
3 h ,(c) 1 Fe for 6 h, (d) for 72 h used to determine the precipitate statistics.

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2 Fig. 4 STEM images of T6 treated 0.2 Fe alloy aged for 6 h; (a) L and (b) β'' phase mixed with GP like structure.

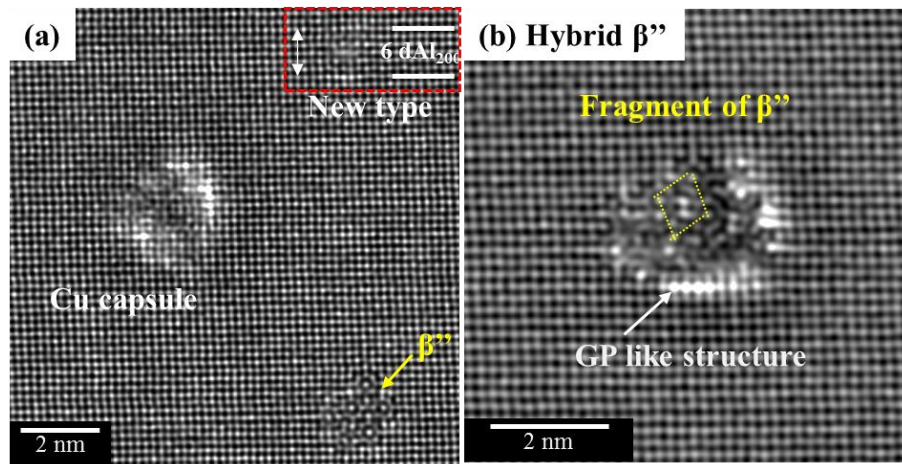
3 The location of the GPB unit and GP-like structure (Cu wall) are indicated by arrows in a). Three columns

4 delimiting a local C-phase configuration are connected by full line, and a mirror plane is indicated by a dotted

5 line. c) is an overlay of the image in b) showing the location of a GP-like structure and the different types of β''

6 that comprise this precipitate.

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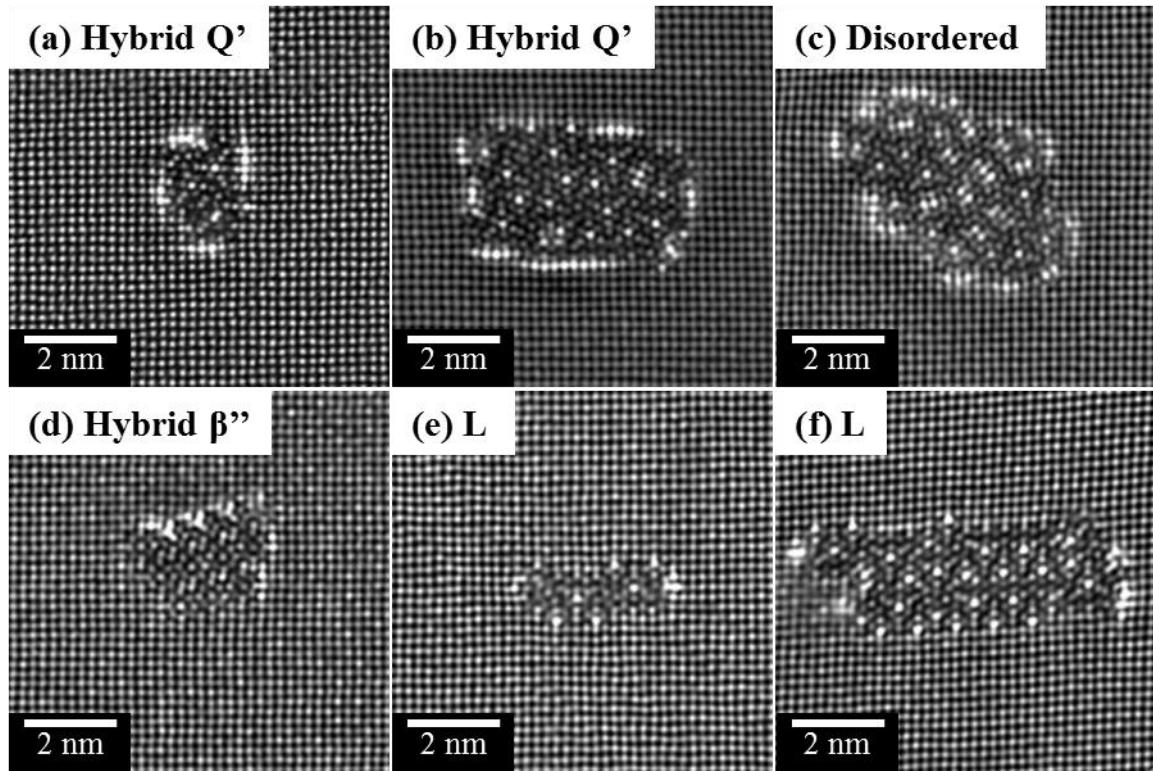
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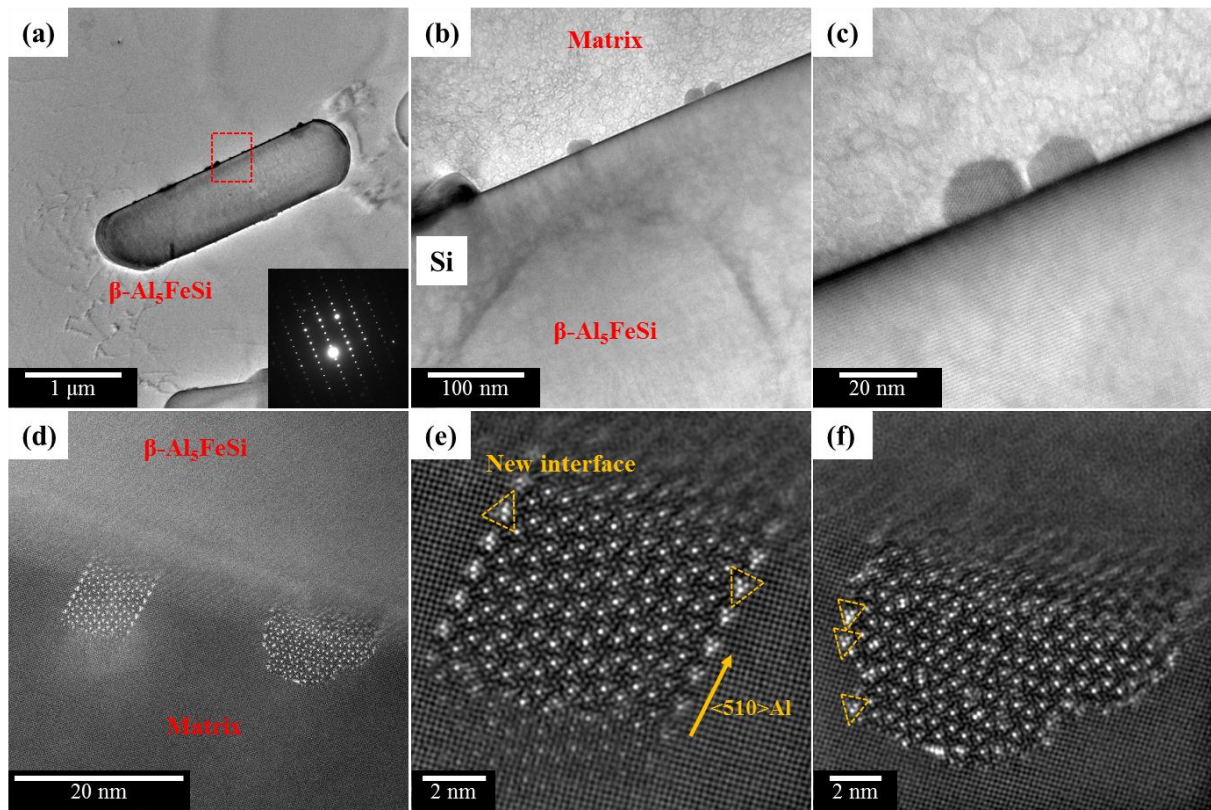
Fig. 5 STEM images of T6 treated 1 Fe alloy aged for 6 h; (a) Cu capsule, new type precipitate with Cu walls separated by 6d Al₂₀₀ and β'' phase, (b) hybrid β'' phase.



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2 Fig 6 HAADF-STEM images of T6 treated 1 Fe alloy aged for 72 h; (a-b) hybrid Q' phases, (c) disordered with
3 Cu walls, (d) hybrid β'' , (e-f) L phases.

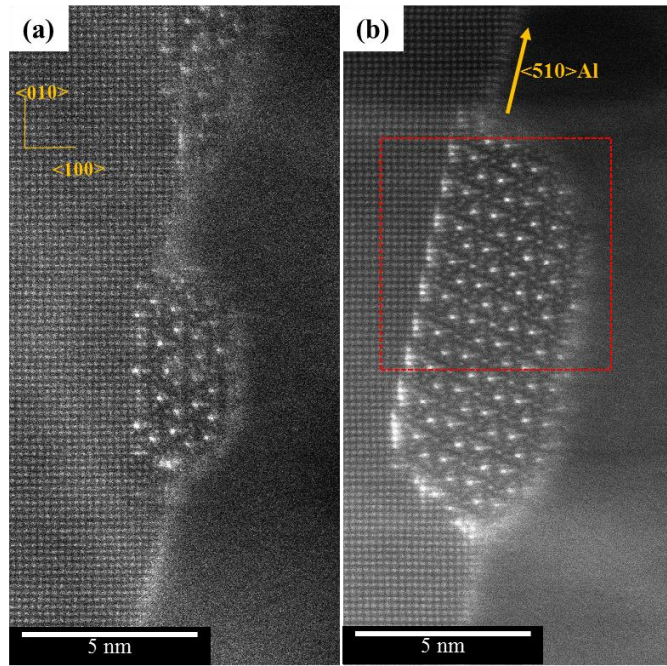
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2 Fig. 7 (S)TEM images of the T6 treated 1 Fe alloy aged for 8 h; (a) BF-TEM, β -Al₅FeSi particle, (b-c) BF-TEM,
 3 enlarged interface of (a) image, (d-f) HAADF-STEM images of Q' phases obtained at different locations on the
 4 boundary of β -Al₅FeSi. A new coherent Q' interface with the Al matrix along $\langle 510 \rangle$ is found, and its distinctive
 5 triangular appearance is marked in e). Structural units of this interface (triangles) are sporadically observed in
 6 the case of a less coherent Q' interface in f).

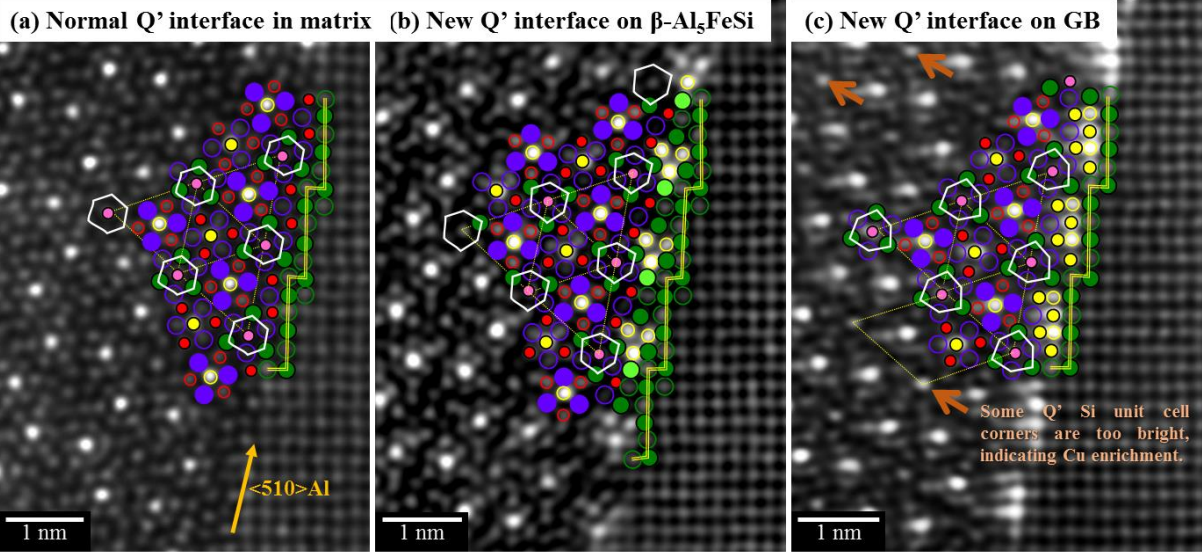
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2 Fig. 8 (a-b) HAADF-STEM images of grain boundary precipitates in the T6 treated 1 Fe alloy aged for 6 h. For
3 the Q' precipitates in b), a previously unreported coherent interface is identified, consisting of local GP-like
4 structures with 1.04 nm periodicity along $\langle 510 \rangle_{Al}$, marked by arrows.

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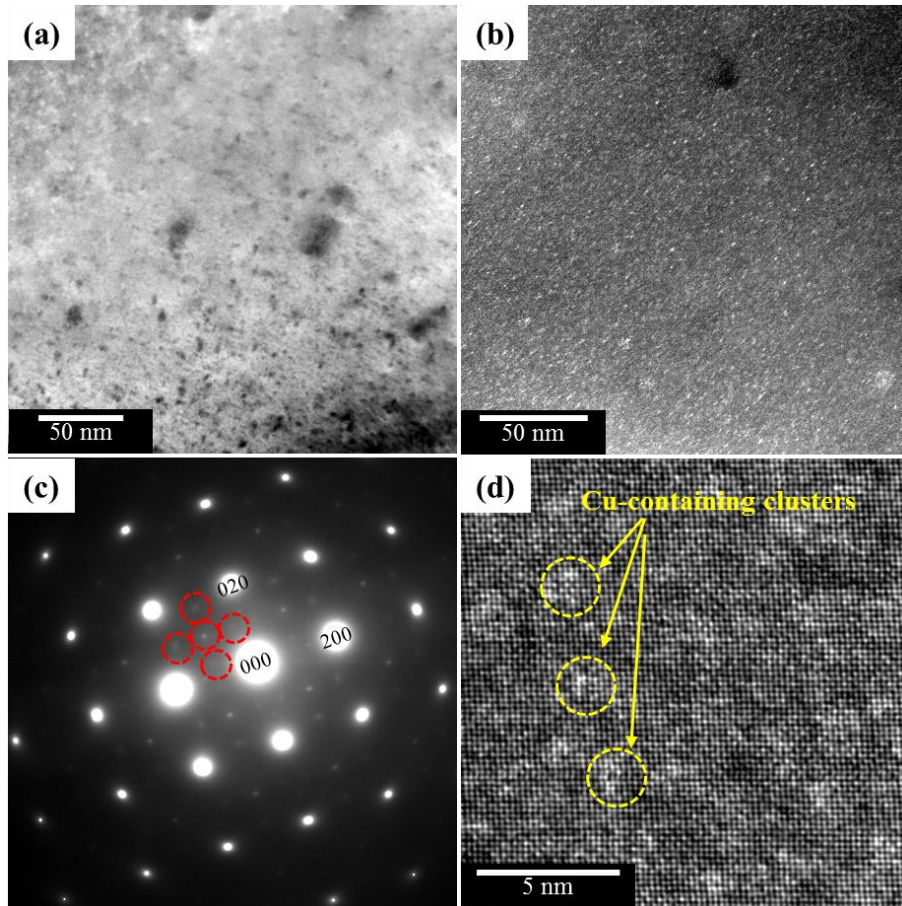


Elements/ height	Al	Cu	Mg	Si	Q' Si unit cell corner	Al column enriched with Cu
0						
1/2						

1

2 Fig. 9 Atomic overlay of Q' phases with different coherent interfaces with the Al matrix along $\langle 510 \rangle_{Al}$; (a)
 3 normal Q' interface [12], (b) new interface of Q' nucleated on a $\beta-Al_5FeSi$ particle from Fig. 7 (e) and (c) new
 4 interface of Q' nucleated on a GB from Fig. 8 (b). The flipped and rotated Fig. 9 (b) is presented in the
 5 supplementary material. To facilitate the comparison between the images, the rings of near neighbor atomic
 6 columns with the unit cell corners are connected by white lines, and the unit cell corners by dotted yellow lines.
 7 The matrix Al in contact with the Q' phases are indicated by yellow double lines.

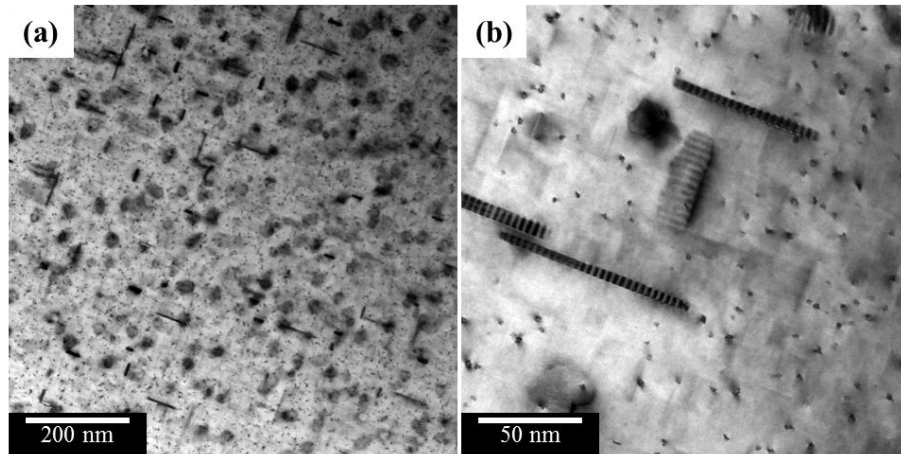
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2 Fig. 10 TEM images of the T5 treated 1 Fe alloy aged for 8 h; (a) BF-TEM, (b) DF-TEM, (c) selected area
3 diffraction pattern, (d) Gaussian blur filtered HAADF-STEM image. Bright spots indicate the presence of Cu-
4 containing atomic clusters.

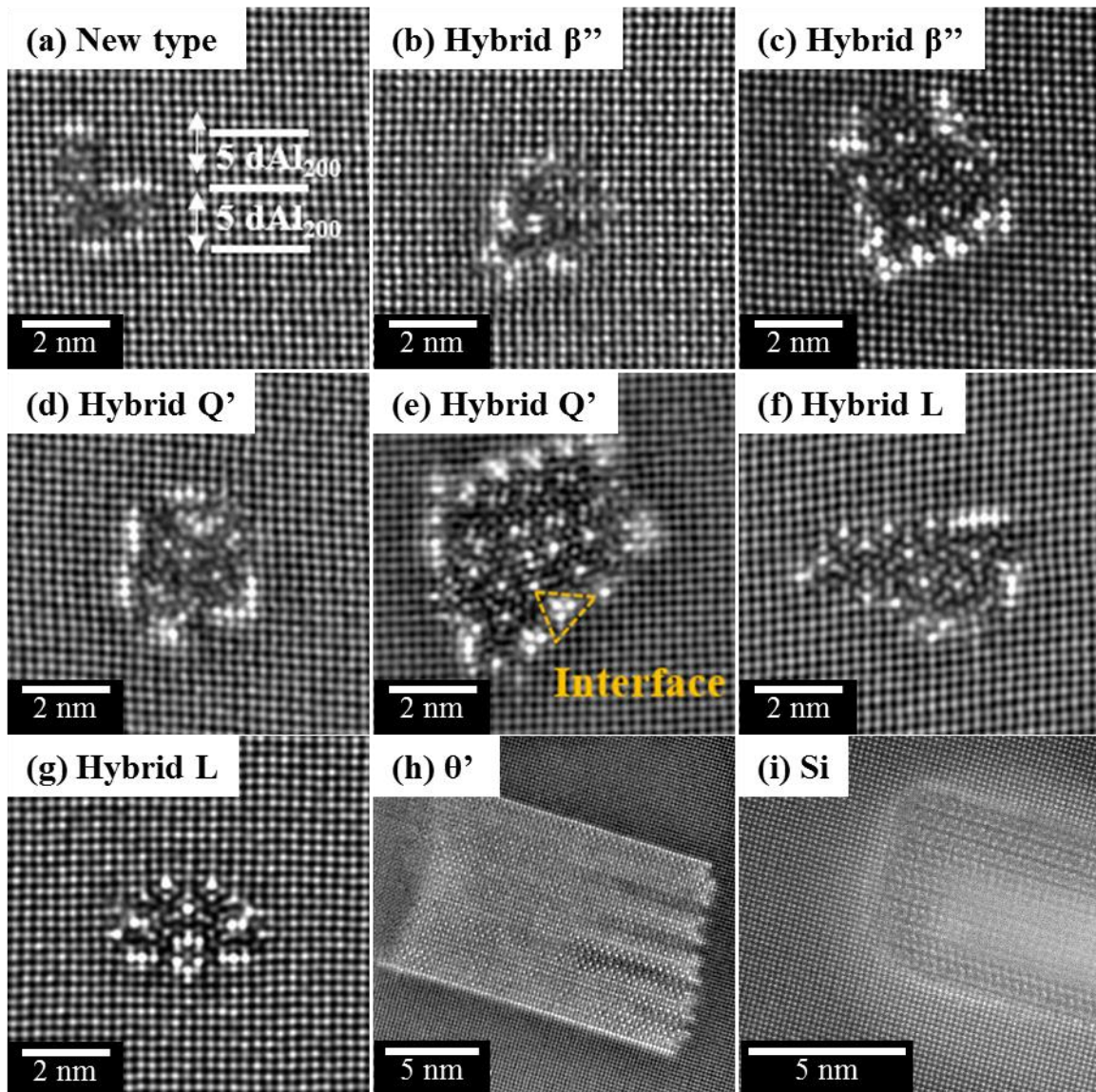
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2 Fig. 11 BF-TEM images of the T5 treated 1 Fe alloy aged for 72 h; (a) low and (b) high magnification.

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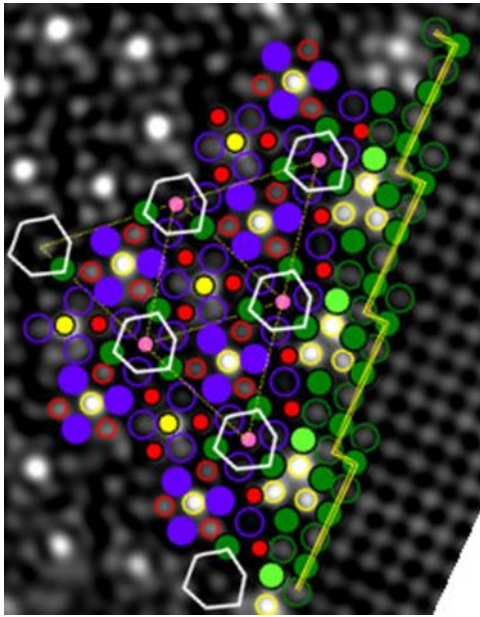


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2 Fig. 12 HAADF-STEM images of the T5 treated 1 Fe alloy aged for 72 h; (a) new type of precipitate, (b-c)
 3 hybrid β'' , (d-e) disordered/hybrid Q' with Cu walls (d) or with the new periodic Cu-containing interface along
 4 $\langle 510 \rangle_{Al}$ (e), (f-g) hybrid L, (h), θ' , (i) Si plate.

5

1 **Supplementary material**



2

3 Fig. The Fig 9 (b) image flipped and rotated is provided to have the same precipitate orientation as in the normal
4 interface image. The new interface allows the Q' precipitate to connect to the Al matrix in a different orientation.