

Reactivity tests for supplementary cementitious materials: RILEM TC 267-TRM phase 1

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10 **Abstract:**

11 A primary aim of RILEM TC 267-TRM: “Tests for Reactivity of Supplementary Cementitious Materials
12 (SCMs)” is to compare and evaluate the performance of conventional and novel SCM reactivity test methods
13 across a wide range of SCMs. To this purpose, a round robin campaign was organized to investigate 10 different
14 tests for reactivity and 11 SCMs covering the main classes of materials in use, such as granulated blast furnace
15 slag, fly ash, natural pozzolan and calcined clays. The methods were evaluated based on the correlation to the
16 28 days relative compressive strength of standard mortar bars containing 30% of SCM as cement replacement
17 and the interlaboratory reproducibility of the test results.

18 It was found that only a few test methods showed acceptable correlation to the 28 days relative strength over
19 the whole range of SCMs. The methods that showed the best reproducibility and gave good correlations used
20 the R^3 model system of the SCM and $\text{Ca}(\text{OH})_2$, supplemented with alkali sulfate/carbonate. The use of this
21 simplified model system isolates the reaction of the SCM and the reactivity can be easily quantified from the
22 heat release or bound water content. Later age (90 d) strength results also correlated well with the results of the
23 IS 1727 (Indian standard) reactivity test, an accelerated strength test using an SCM/ $\text{Ca}(\text{OH})_2$ -based model
24 system. The currently standardized tests didn't show acceptable correlations across all SCMs, although they
25 performed better when latently hydraulic (slag) materials were excluded. However, the Frattini test, Chapelle
26 and modified Chapelle test showed poor interlaboratory reproducibility, demonstrating experimental difficulties.
27 The TC 267-TRM will pursue the development of test protocols based on the R^3 model systems. Acceleration
28 and improvement of the reproducibility of the IS 1727 test will be attempted as well.

29 **Keywords:** supplementary cementitious materials, reactivity test, heat release, bound water, compressive
30 strength

31 **1 Introduction**

32 The use of supplementary cementitious materials (SCMs) as a partial replacement for clinker in blended cements
33 or concrete is becoming increasingly widespread. In addition, the availability of traditionally used SCMs (e.g.
34 blast furnace slag and fly ash) is decreasing and a wider range of materials and combination are being considered
35 as SCMs. The first criterion for such replacements is the contribution they make to the development of
36 mechanical properties so there is great interest both in testing this directly and in the development of tests which
37 give a rapid assessment of this reactivity. RILEM TC 267-TRM (Tests for Reactivity of Supplementary
38 Cementitious Materials) was established to evaluate the existing reactivity tests and develop a pre-normative
39 recommendation for rapid SCM reactivity tests that can be adopted as standard testing methods. Ideally test
40 methods should supply results more rapidly than the standard compressive strength testing regimes, they should

41 be straightforward and robust to execute and should not require expensive equipment or advanced training of
42 practitioners.

43 The current standardized methods for SCM or pozzolanic reactivity test are 1) the Chapelle test [1] or a modified
44 version of it (NF P18-513) [2], 2) the Frattini pozzolanicity test (EN 196-5) [3], and 3) the determination of
45 reactive silica (EN 196-2:2013). An Indian standard (IS 1727-1967) – locally known as the lime reactivity test
46 - is also in use. Both the (modified) Chapelle [4,2,1] and Frattini test methods [4,3] measure the reactivity of
47 the SCM with $\text{Ca}(\text{OH})_2$, either by titrating the amount of $\text{Ca}(\text{OH})_2$ remaining in a dilute suspension or by
48 evaluating the saturation degree of solution towards $\text{Ca}(\text{OH})_2$, respectively. Both tests intend to test pozzolanic
49 reactivity (with portlandite) and were not intended to work for latent-hydraulic, Ca-rich SCM such as blast
50 furnace slags. The Chapelle test takes less than 1 day to carry out, the Frattini test at least 8 days, and for less
51 reactive SCMs up to 15 days. The IS 1727 test measures the compressive strength of a portlandite ($\text{Ca}(\text{OH})_2$)
52 and SCM binary mix cured initially at 27°C and then at 50 °C until 10 days after casting. Previous work,
53 indicated that these standard reactivity testing methods for supplementary cementitious materials have
54 shortcomings [4], particularly in terms of correlation to strength development of cements, test duration and
55 reproducibility.

56 There has been much research on the mechanism of reaction of SCMs in blended systems, which has benefited
57 from advances in analytical methods and thermodynamic modelling [5,6]. In contrast, few advances have been
58 made regarding reactivity test methods.

59 More recently, the spread of new or improved experimental techniques such as isothermal conduction
60 calorimetry has inspired new research into the topic [7,8]. The so called “R³” test was developed initially to test
61 the pozzolanic activity of calcined clays. R³ stands for rapid, reproducible and relevant: The aim is to have a
62 method which can give results correlating to strength in standard mortars (*relevant*) in a much shorter time
63 (*rapid*) and which is relatively simple to carry out giving *reproducible* results. This method tries to better
64 simulate the conditions occurring in a blended cement by the addition of small amounts of sulfate and alkali to
65 an SCM portlandite mixture [8]. This test was accelerated by measuring the reaction at 40 °C either by the heat
66 release in isothermal calorimetry continuously up to 7 days, or bound water between 110 °C and 40 °C after 7
67 days of curing [8]. For calcined clay a very good correlation was found between the amount of reaction at 1 day
68 at 40 °C and the 28 day strength in standard mortar bars.

69 This paper reports on the round robin study which was phase 1 of the committee work. The objective was to
70 look at the performance of a range of methods proposed to measure reactivity across a wide range of SCMs.
71 The tests methods were selected according to the experience of the committee members and an overview of
72 SCM reactivity tests [4]. Two categories of test methods were defined: the existing standard methods and the
73 R³ model system tests (non-standard). For the R³ system, measurements of portlandite consumption using
74 thermogravimetric analysis (TG) [9] and the chemical shrinkage [10] were included in addition to the
75 calorimetry and bound water methods.

76 The reactivity test results on a selected range of SCMs were compared to a benchmark - the compressive strength
 77 results of cement mortar bars (EN 196-1) - in which 30 wt.% of the Portland cement was replaced by SCM. The
 78 interlaboratory reproducibility of the test methods was assessed. A selection of test methods which seem to be
 79 giving best results for further testing and optimization was made for the phase 2 work of this committee.

80

81 2 Experimental

82 2.1 Participants and work plan

83 In total, there were 21 participants (see Table 1), who were free to choose which methods to test. The summary
 84 of the number of participants for each test is shown in Table 2.

85 *Table 1 Summary of the participants*

Continent	Europe		North America	Asia
Participants	Empa	RWTH Aachen	U. Laval	IIT Delhi
	EPFL	TU Delft	U. Texas Austin	IIT Madras
	LafargeHolcim	Università degli Studi della Basilicata	U. Toronto	Sinoma
	HeidelbergCement Technology Center	U. Gent		
	INSA Rennes	U. Sheffield		
	INSA Toulouse	VITO		
	KU Leuven	ZAG		
	NTNU			
Subtotal	15		3	3
Total	21			

86 Notes: U. = University,

87

Table 2 Summary of the test planning.

	Test	Total participants
Standard method	Mortar test: EN 196-1	6
	Fratini test: EN 196-5	5
	Chapelle test: standard	4
	Modified Chapelle test: NF P18-513	5
	IS 1727 (Indian standard)	2
	Reactive silica: EN 197- 1/EN 196-2	1
R ³ model	Calorimetry	13
	Bound water	13
	Chemical shrinkage	5
	Portlandite consumption	7

88

89 2.2 SCMs

90 In this phase 1, the aim was to look at a wide range of SCMs, including those most commonly used. Eleven
91 materials, were selected:

- 92 ○ 2 calcined clays (labelled as *CC1* and *CC2*)
- 93 ○ 2 ground granulated blast-furnace slags (labelled as *S1* and *S8*)
- 94 ○ 2 calcareous fly ashes from coal combustion (labelled as *CFA_P* and *CFA_S*)
- 95 ○ 3 siliceous fly ashes from coal combustion (labelled as *SFA_E*, *SFA_I* and *SFA_R*)
- 96 ○ 1 natural pozzolan (labelled as *Po*)
- 97 ○ quartz (labelled as *Q*) as a reference for an inert material

98 The Supplementary Material gives the chemical composition (measured by X-ray fluorescence analysis); origin
99 and the physical properties of the SCMs (Blaine fineness, density measured according to ASTM C188-09 using
100 isopropanol instead of kerosene [11]; particle size distribution (PSD) measured using Malvern laser diffraction
101 using isopropanol); mineralogical compositions of the materials obtained by X-ray powder diffraction (XRD)
102 with Rietveld analysis. For SCMs, the external standard method was used to determine the amorphous content
103 (details on the XRD experiments and Rietveld analysis are given in supplementary material).

104 2.3 Benchmark testing

105 It was decided to use a classic strength test as a benchmark for the reactivity tests. The level of replacement of
106 the SCMs was chosen as 30% to give good sensitivity to the contribution of the SCMs.

107 Six participants carried out the mortar strength tests according to EN 196-1 using local Portland cements (in
108 total 6 different cements were used) of type CEM I 42.5 N/R or similar. The characteristics of the cements used
109 for the mortar tests are given in the Supplementary Material.

110 The mortar tests were carried out according to EN 196-1. 30% by mass replacement of cement by SCMs was
111 used, an adjustment of gypsum content (similar to Antoni et. al. [12]) was applied and superplasticizer (PCE
112 type) was introduced for calcined clays to control the reaction of the Al_2O_3 in the SCMs and the workability of
113 the mortar, respectively. The compressive strength was measured at 2, 7, 28 and 90 days.

114 It was not possible to average the absolute strengths for the different cements as 6 local CEM I 42.5N/R cements
115 were used. So, the relative compressive strength $R_{SCM,relative}$ (%) was used for the correlation analysis:

$$116 R_{SCM,relative} = \frac{R_{SCM} - R_{PC}}{R_{PC}} \times 100 \quad \text{Eq. (1),}$$

117 where R_{SCM} and R_{PC} are the absolute strength in MPa for the SCM blended cement and the pure PC from the
118 same source, respectively. The $R_{SCM,relative}$ was calculated for each cement and then averaged. The strengths
119 relative to the quartz references were also calculated.

120

121 2.4 Methods

122 Detailed protocols for each of the methods are given in the Supplementary Material.

123 2.5 Standard SCM reactivity tests

124 The standardised methods used were:

125 1) Chappelle test or a modified version of it (NF P18-513)

126 The Chappelle test [1] assesses the consumption of calcium hydroxide by a test material in a dilute heated
127 suspension as a measure of pozzolanic activity. 1 g of SCM is reacted with 1 or 2 g of Ca(OH)_2 (Chappelle or
128 modified Chappelle test, resp.) in 200 ml of water at 90-100 °C for 16 h. The non-reacted lime is then analyzed
129 and the result expressed in mg Ca(OH)_2 fixed by the SCM.

130 2) Frattini or pozzolanicity test (EN 196-5)

131 The Frattini test evaluates portlandite saturation in a supernatant solution of a hydrated slurry of Portland cement
132 and a pozzolan test material by measuring the OH^- and Ca^{2+} concentrations. The test consists of mixing a blend
133 of Portland cement (CEM I) and SCM with distilled water at a water to solid ratio of 5. The interpretation of
134 the Frattini test results was made according to Donatello et. al.[13] and Snellings et al.[4], which calculates the
135 vertical distance of data points from the lime solubility curve.

136 3) The determination of reactive silica (EN 196-2 and EN 197-1)

137 Reactive silica is defined according to EN 197-1 as that fraction of the SiO_2 which is soluble after treatment
138 with HCl and a boiling KOH solution. The measurement procedure is established in EN 196-2.

139 4) The Indian test method for pozzolanic materials (IS 1727 - 1967)

140 In this method, a volume based mix design is used to keep the same volume of the binder in each mix. A 1:2:6
141 portlandite : pozzolan : sand ratio is used and the w/b ratio is adapted to keep the mortar flow fixed. The mortars
142 are cast and kept in RH saturated conditions and at 27 °C until 2 days, after which the samples are demoulded
143 and further cured at 90-100% RH and 50 °C. The compressive strength of the mortar cubes is measured after
144 10 days of curing. The strength data are taken as indication of the reactivity of the pozzolan.

145

146 2.6 R^3 test

147 The basic principle of the R^3 test is to use a simplified model system to separately measure the reaction of an
148 SCM. This is to avoid interference and overlap with the clinker hydration reactions that occur in a blended
149 cement system. Moreover, the use of lab-grade chemicals instead of local Portland cements avoids much
150 material related variability. The two main components of the R^3 model system are the SCM and Ca(OH)_2 . The
151 mix design of the R^3 model paste, shown in Table 3, was based on Avet et. al. [8].

152

Table 3 R^3 model mix design

Components	SCM	Portlandite ^(a)	Deionized Water	KOH ^(b)	K ₂ SO ₄ ^(b)	Calcite ^(c)
Mass (g)	11.11	33.33	60.00	0.24	1.20	5.56

153 Notes: (a) Lab-grade, less than 5 wt.% CaCO₃ present

154 (b) Lab-grade

155 (c) Lab-grade, d₅₀ 5-15 μm.

156 The R^3 pastes were used for the bound water, isothermal calorimetry, portlandite consumption using TG and
 157 the chemical shrinkage tests.

158

159 *R^3 bound water*

160 The R^3 pastes were cured in sealed plastic containers at 40 °C for 7 days. The hydrated samples were crushed
 161 and dried in an oven at 105 °C until reaching constant weight. The dried samples were heated at 350 °C for 2
 162 hours and the bound water (for hydrates, excluding portlandite) was calculated from the weight difference.

163 *R^3 portlandite consumption*

164 The R^3 pastes were cured in a sealed container at 40 °C for 7 days. The hydration of the samples was stopped
 165 by solvent exchange according to [14,15]. The dried samples were analyzed by thermogravimetry. 50 mg of
 166 sample was introduced in the crucible which was heated from 30 °C to 950 °C at 10 °C/min. A protective
 167 nitrogen atmosphere at a flow rate of 50 mL/min was used. The portlandite content was determined using the
 168 tangent method described by Lothenbach et al. [16] and the portlandite consumption calculated by difference to
 169 the initial content.

170 *R^3 calorimetry test*

171 Isothermal conduction calorimetry at 40 °C was carried out to measure the heat release during hydration of the
 172 R^3 systems. The heat release was recorded until 7 days.

173 *R^3 chemical shrinkage*

174 Chemical shrinkage was measured using a modified protocol based on the ASTM C1608-12 and Geiker [17].
 175 4-6 replicate samples were used for all measurements. The fresh R^3 paste was added into the test vial (weight
 176 m_{vial}) up to ~3 cm (half to two thirds of the container's capacity). De-aerated water at 40°C was carefully added
 177 on top avoiding mixing with the paste to completely fill the vial. The sealed samples were placed in a water
 178 bath at 40 °C and the volume changes were recorded for 14 days to calculate the chemical shrinkage.

179 2.7 Data treatment

180 The inputs from different participants for the same test were averaged, and the standard deviation (σ) on the
 181 average of the test results was calculated.

182 The coefficient of variation (CV, in %) was used to estimate the reproducibility of a test between laboratories:
183 the smaller the CV, the higher the reproducibility. For the calculation of the CV, the difference between the
184 averages of the SCM and the quartz results were used in the denominator. This way the quartz acts as the
185 reference and comparison of the CV amongst samples and techniques is possible.

$$186 \quad CV_i = \frac{\sigma_i}{\bar{x}_i - \bar{x}_Q} \times 100 \quad \text{Eq. (2),}$$

187 where the σ_i and \bar{x}_i are the standard deviation and the mean of the input of a test method from all the laboratories
188 for a specific SCM, respectively. \bar{x}_Q is the mean of the input from all laboratories for quartz for the test method.

189 The mean CV of all the SCMs for a specific testing method was used to assess reproducibility of the method.

190 Linear fitting of the data from the test methods to relative strength was used for all the SCMs tested. The
191 regression coefficient, R^2 , of the linear fitting was taken as the indicator of quality of correlation between the
192 relative strength and the respective test method.

193 3 Results

194 The original data are reported in the Supplementary Material. The following sections present an overview of
195 the processed results.

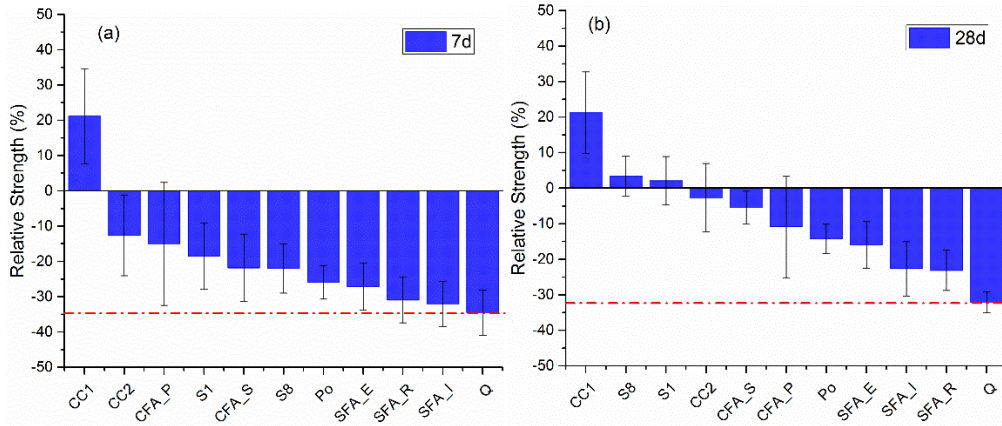
196 3.1 Compressive strength benchmark test

197 The strength development of the cement reference samples showed significant differences both at early (2 and
198 7 days) and late ages (28 and 90 days), even though the cements used for the mortar test were all CEM I 42.5N/R
199 (see supplementary material). These differences were enlarged when the cements were blended with the SCMs.
200 Even when the results were expressed relative to the reference cement, there were still large differences for the
201 results from different laboratories.

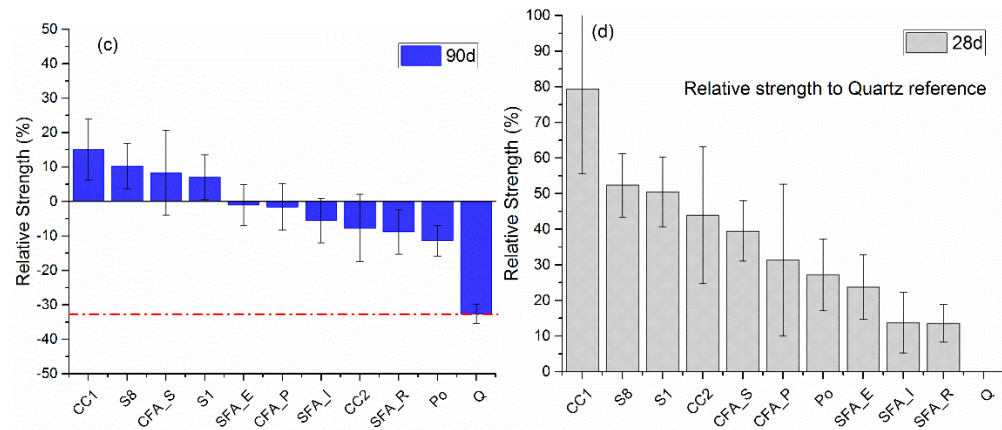
202 The average of the relative strength at 28 days was retained as the critical measure for comparison. For early
203 and later strength, the average of the relative strength at 7 days and 90 days were regarded as the indicators,
204 respectively. The relative strength based rankings of the 10 SCMs and the quartz are shown in Figure 1. The
205 strength relative to the quartz reference were also calculated Figure 1 (d). These show the same ranking (Figure
206 1 (b) and (d)).

207

208



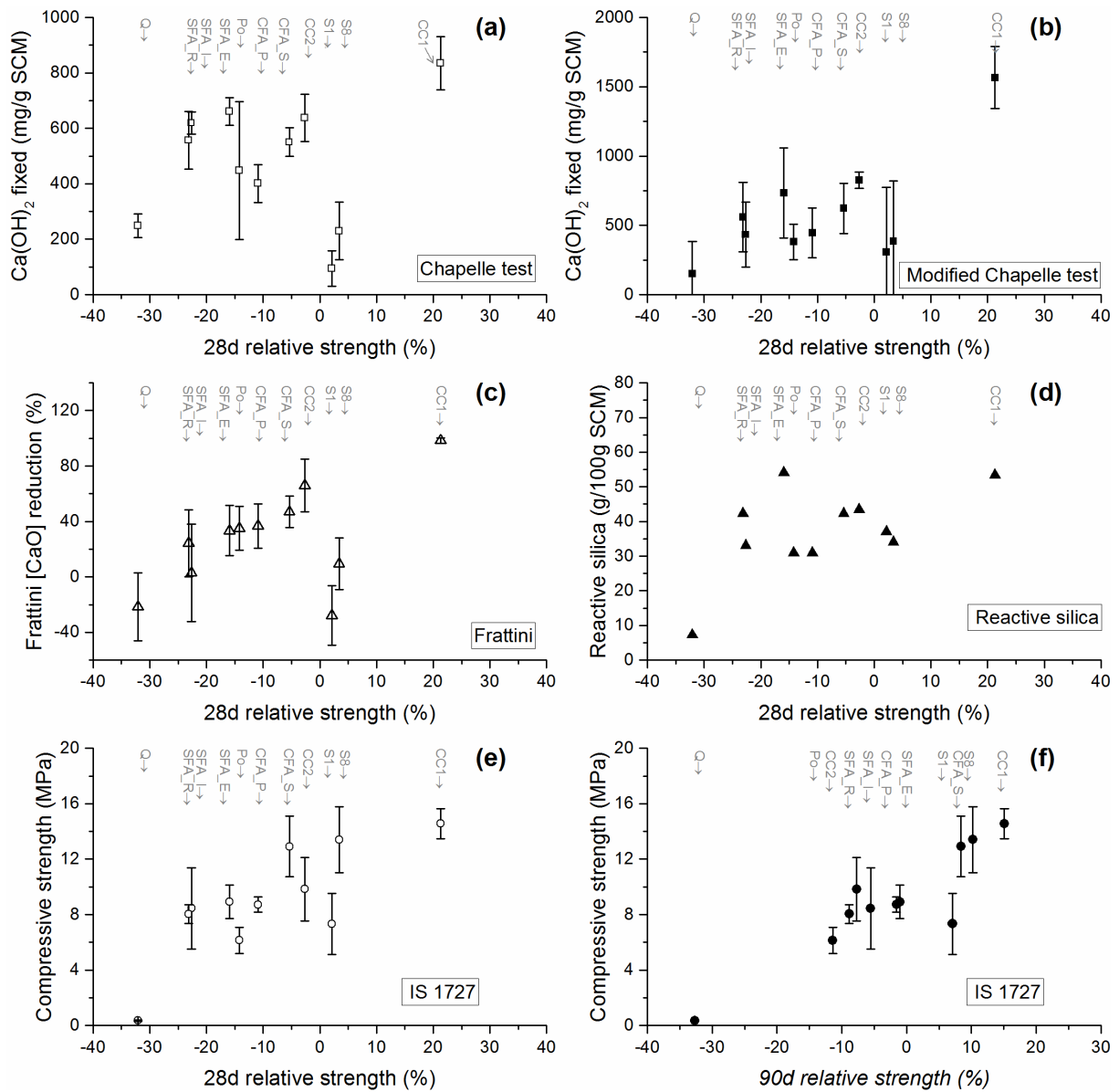
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210 *Figure 1 Relative strengths of the SCM blended cement mortar bars, (a), (b) and (c) are relative strengths compared to the PC*
 211 *reference, (d) shows relative strength compared to the quartz (Q) as inert reference*

212 3.2 Correlation analysis of reactivity test results

213 The global average and the standard deviation of the output for each reactivity test are shown in Figure 2 and
 214 Figure 3. Characteristic heat release and chemical shrinkage values at 0.5, 1, 3 and 7 days (and 14 days only for
 215 chemical shrinkage) were used for the correlation analysis (the 3 and 7 days values are shown in the
 216 Supplementary Material) for continuous measurements such as R³ calorimetry and chemical shrinkage.



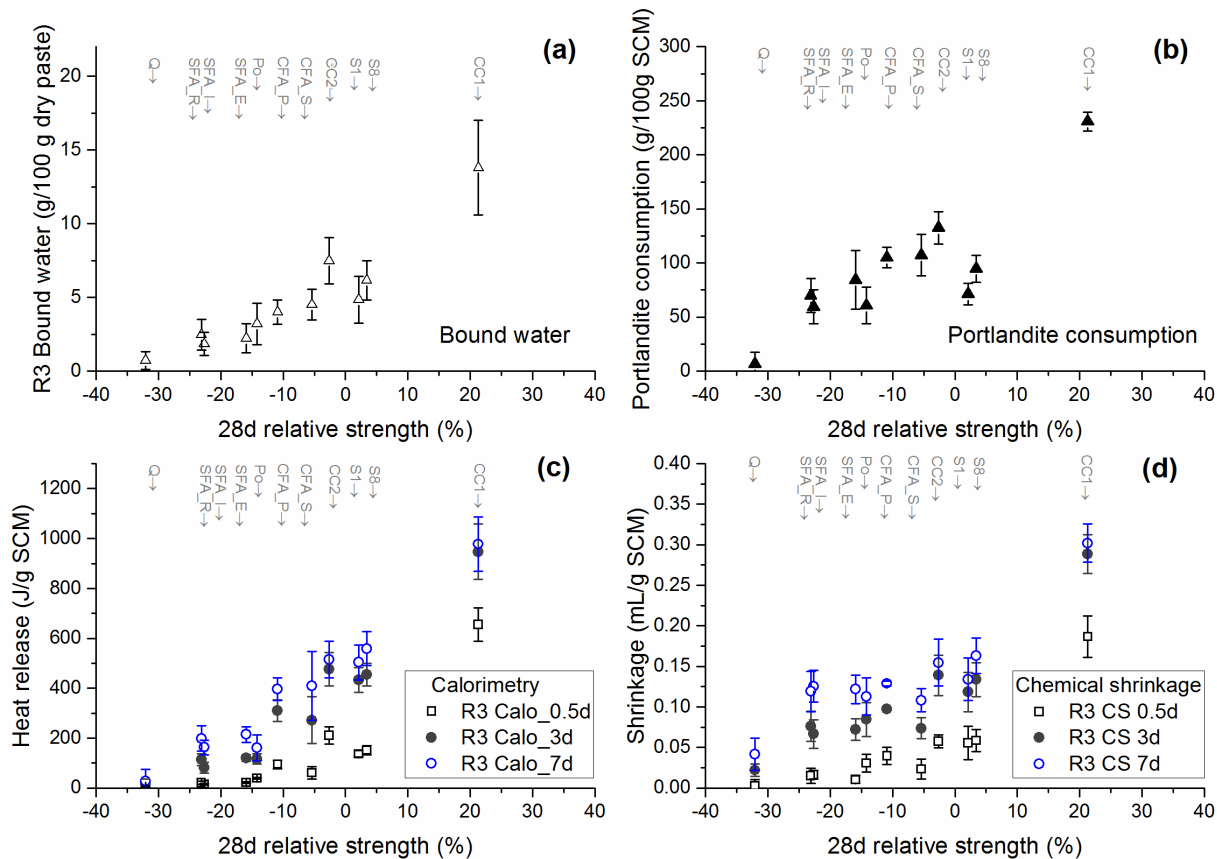
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Figure 2 Plots of the standard testing methods against relative strength; the SCMs corresponding to the points are labelled on top of the plot. (a) Chapelle test, (b) modified Chapelle test, (c) Frattini test, (d) Reactive silica, no error bar because there is only 1 input, (e) IS 1727 and (f) IS 1727 (vs. 90 days relative strength). Average values are shown by symbols, the error bars represent 1σ .



221

222 Figure 3. Plots of the R^3 model test methods to 28 days relative strength, the SCMs corresponding to the points are labelled on top of
 223 the plot. (a) Bound water test, (b) Portlandite consumption, (c) Cumulative heat release for 0.5, 3 and 7 days and (d) Chemical
 224 shrinkage at 0.5, 3 and 7 days. Average values are shown by symbols, the error bars represent 1σ .

225

226 The R^2 values of the linear fitting of the reactivity test to the relative strength using all the SCMs (including the
 227 quartz) are summarized in Table 4². Here we considered an R^2 of more than 0.85 as the criterion for acceptance
 228 in terms of correlation.

229 Table 4. R^2 index of linear correlation of the reactivity test results to the relative strength at 7, 28 and 90 days for all SCMs tested.

Relative strength at	Standard method					R3 model										
	Chapelle	Modified Chapelle	IS 1727	Frattini	Reactive silica	Bound water	CH consum.	Calorimetry (heat released)				Chemical shrinkage				
				[CaO] reduction				0.5d	1d	3d	7d	0.5d	1d	3d	7d	14d
7 days	0.20	0.74	0.39	0.54	0.27	0.93	0.89	0.95	0.95	0.90	0.86	0.93	0.94	0.92	0.87	0.72
28 days	0.03	0.46	0.62	0.31	0.33	0.86	0.74	0.72	0.80	0.91	0.94	0.77	0.76	0.80	0.75	0.55
90 days	0.04	0.29	0.82	0.17	0.52	0.43	0.51	0.31	0.36	0.50	0.63	0.36	0.33	0.41	0.49	0.47

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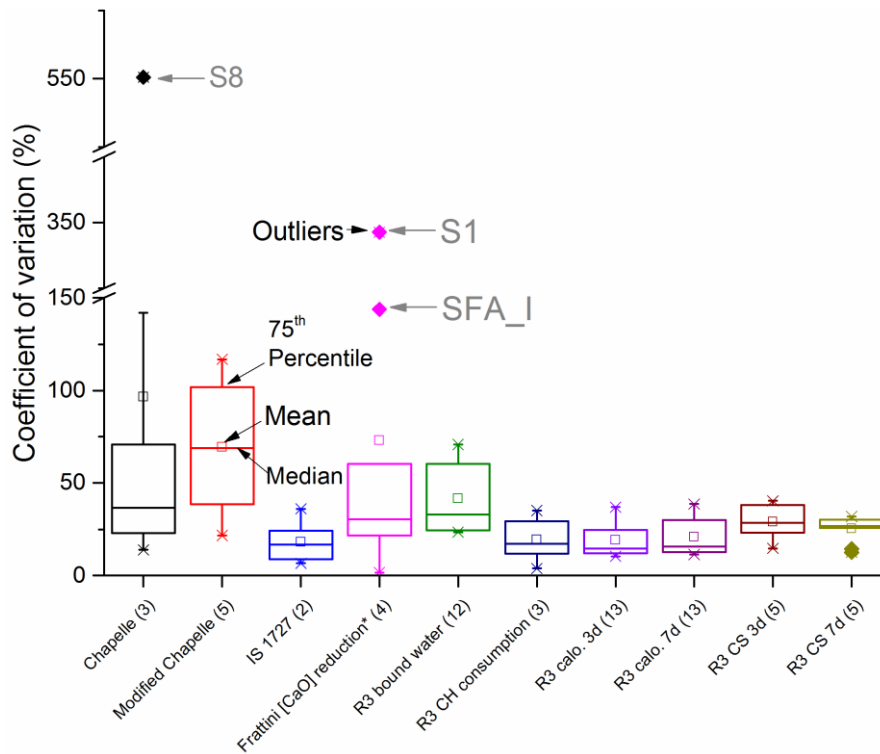
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232 3.3 Interlaboratory reproducibility

233 The coefficient of variation (CV defined in Eq. (8)) was used to indicate the reproducibility of the reactivity test
 234 methods (see Figure 4). As there was only one participant for the reactive silica test, the CV was not available

² The R^2 values of the linear correlation of the relative strength to the quartz reference strength are given in the Supplementary Material

235 for this test. The heat release and chemical shrinkage values at 3 and 7 days were used to evaluate the
 236 reproducibility for these continuous tests.



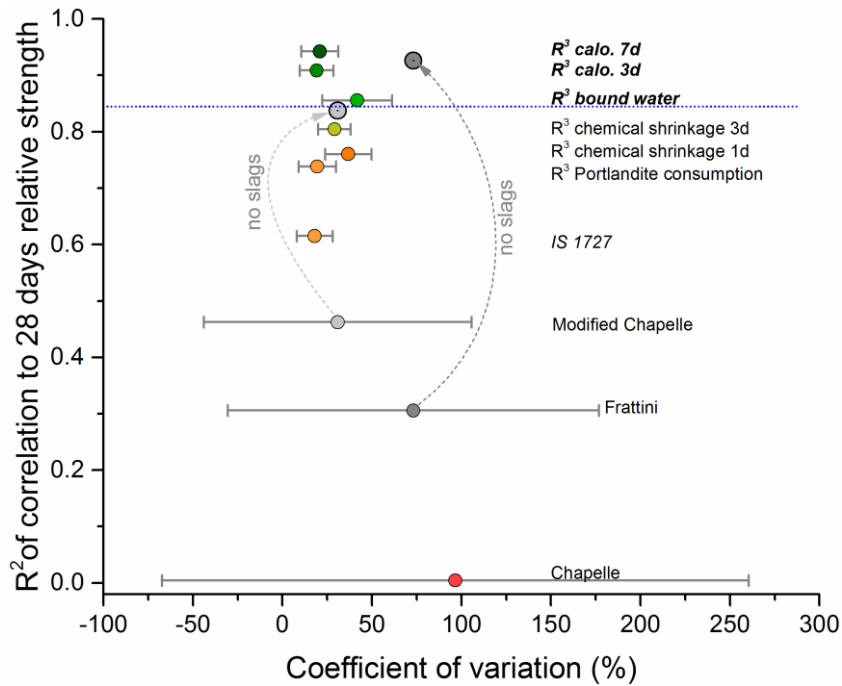
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238 *Figure 4* Box chart for coefficient of variation (CV) for different methods, numbers in the bracket along x-axis refer to the number of
 239 participants. R³ CH consumption refers to portlandite consumption for R³ model test; R³ calo. 3d and R³ CS 3d refers to calorimetry
 240 heat release and chemical shrinkage for R³ model.

241

242 4 Discussion: Evaluation of the methods

243 The test methods were evaluated based on the correlation to the benchmark (relevance) and the interlaboratory
 244 reproducibility (reliability). Other factors such as test duration, complexity and cost of equipment also need to
 245 be taken into consideration. Figure 5 shows the CV against the R^2 value for the correlation to the 28 days relative
 246 strength. An ideal test should be located as close as possible to 1.0 on the R^2 scale while showing the lowest CV
 247 in Figure 5. The dotted blue line corresponds to $R^2 = 0.85$. The results are summarized and compared to the
 248 other factors for each reactivity method in Table 5. In the following sections the results for the reactivity test
 249 methods are discussed one by one.



250

251 Figure 5 Correlation to 28 days relative strength vs. coefficient of variation (CV) plot, dotted blue line corresponds to R^2 value equal
 252 to 0.85, dotted grey arrows indicate the improvement of the correlation for Frattini and Modified Chapelle tests without slags.

253

254 Table 5 Summary of the methods, ranked based on the correlation to 28 days relative compressive strength.

Methods	Correlation to 28d relative strength ^(a)	Coefficient of variation	Time		Equipment investment	Key equipment
			Operating	Test duration		
<i>Units</i>	--	%	<i>Hours</i>	<i>days</i>	relative ^(c)	--
R ³ calorimetry 7 days	0.94	20.9	1	7	20	Calorimeter
R ³ calorimetry 3 days	0.91	19.1	1	3	20	Calorimeter
R ³ bound water	0.86	41.7	2	8	2	Oven
R ³ chemical shrinkage 3 days	0.80	29.1	4	3	2	Water bath
R ³ portlandite consumption	0.74	19.5	2	8	10	TG
IS 1727 (Indian standard)	0.62	18.1	1	10	2	Compression testing machine
Modified Chapelle	0.46	30.9	2	1	1	Reflux condenser
Frattini ([CaO] reduction)	0.31	73.1	2	8	1	Glass, pipettes
Reactive silica	0.31	-- ^(b)	2	1	2	Glass, oven
Chapelle test	0.00	96.6	2	1	1	Reflux condenser

255 Notes: (a), R^2 of the linear fitting; (b) no data as there is only one input; (c) relative cost.

256

257 4.1 Standard reactivity test methods

258

259 *Chapelle and modified Chapelle test*

260 The Chapelle test showed no correlation to the 28 days relative strength. The Modified Chapelle test showed
261 poor correlation ($R^2 = 0.46$) because that the results for slag fell out of the linear trend. When leaving out the
262 slags the R^2 correlation coefficient is improved from 0.46 to 0.84 for the modified Chapelle test (see Figure 2
263 (b), Figure 5, Table 4 and Supplementary Material). The Chapelle test showed the worst reproducibility (mean
264 CV = 96%) of all tests. The improved protocols of the modified Chapelle test resulted in significantly less
265 dispersion of results with a mean CV of 31%. However, the committee noted that the experimental set-up is
266 rather complex and much care is required to control the experiment and avoid carbonation.

267 *Frattini test*

268 The Frattini test also showed poor correlation to the 28 days relative strength. The results of slags fell out of the
269 trend for the Frattini test (see Figure 2 (c)). When leaving out the slags the R^2 correlation coefficient is much
270 improved from 0.31 to 0.93 (see Figure 5, Table 4 and Supplementary Material). This indicates that the Frattini
271 method does perform well for purely pozzolanic materials, but cannot cover SCMs that show a (latent) hydraulic
272 nature. On the other hand the Frattini test results showed a rather high CV (mean CV = 73%), which reflects
273 the use of different local Portland cements with different alkali content.

274 *Indian lime reactivity test (IS 1727)*

275 The Indian standard lime reactivity test (IS 1727) showed only moderate correlation to the 28 days relative
276 strength benchmark, but better than any other standard method when all SCMs are taken into account. For 90
277 days strength however the IS 1727 test performed best in terms of correlation (see Figure 2 (f) and Table 4).
278 This may be related to the higher curing temperature of 50 °C and the longer test duration of 10 days as also for
279 the R^3 test methods an increase in correlation was found for increased test durations (e.g. compare the R^2 for 3
280 and 7 days R^3 test heat release). The CV for IS 1727 is relatively good but less representative because only two
281 laboratories used this technique at this stage, more testing is required to better constrain the reproducibility of
282 the test.

283 *Reactive silica test*

284 Reactive silica test did not give acceptable correlation to the compressive strength results. The reproducibility
285 could not be assessed as the test was only carried out by one participant.

286

287 4.2 R^3 model tests

288 Both the R^3 bound water and calorimetry tests gave good correlations passing the acceptance criterion. For
289 methods compared to 28 days relative strength with R^2 higher than 0.85 (R^3 calorimetry at 7 d and 3 d, and
290 bound water test), the linear fitting to the 28 days relative strength is shown in the plots in the Supplementary
291 Material.

292 With respect early strength (7 days relative strength), the R^3 model systems perform better than the standard
293 methods. The correlation coefficients were greater than 0.85 for all measurement methods, with the exception
294 of the 14 days chemical shrinkage measurements. The CV for the R^3 model tests were relatively low, better than
295 those of the standard tests.

296

297 *R^3 calorimetry test*

298 The R^3 calorimetry test showed the best correlation to 28 days relative strength with an R^2 of 0.94 for the heat
299 release results taken at 7 days (as shown in Table 5), also the correlation to the 3 days cumulative heat was
300 acceptable. The cumulative heat at shorter ages (0.5 and 1 day) gave the best correlation to the 7 days strength
301 measurements. This indicates that different time intervals in the continuous measurements may be selected for
302 correlation to the compressive strength at different ages. It can be observed in Table 4 that the R^3 heat release
303 correlates better with the 90 days strength as the total heat is calculated at longer times. The relatively low CV
304 indicates good reproducibility of the results. As a drawback, the equipment cost of an isothermal conduction
305 calorimeter is relatively high. This is partially mitigated by the relatively low staff effort required compared to
306 more laborious standard tests (see Table 5).

307

308 *R^3 bound water test*

309 The R^3 bound water test showed acceptable correlations to the 28 days relative strength with an R^2 of 0.86 for
310 the linear correlation. Even though the linear correlation is not as good as that for the calorimetry at 7 days, the
311 simplicity and the relatively low cost of the equipment needed (see Table 5) would enable widespread use of
312 this test. Between the different methods for the R^3 system, the bound water test has the highest CV (42%). While
313 the equipment used in this test is inexpensive and widely available in basic cement laboratories, the
314 measurement protocol requires more staff effort (see Table 5). However, the technique is straightforward and
315 does not require advanced training.

316

317 *R^3 portlandite consumption test*

318 The R^3 portlandite consumption test showed a rather weak correlation to the strength benchmark as the results
319 of slags biased the linear trend (see Figure 3 (b)). Similarly to the Frattini and modified Chappelle tests, the
320 correlation is much improved when the results for the slags are removed from the analysis. The relatively low
321 CV indicates good reproducibility of the results. The current protocol requires thermogravimetric equipment
322 which is costly, in addition the need for hydration stoppage (here by solvent exchange) makes the test rather
323 laborious and introduces an additional source of variation.

324

325 *R³ chemical shrinkage test*

326 The R³ chemical shrinkage test did not give acceptable correlations to the 28 days relative strength (see Figure
327 3 (d)). The relationships between early age chemical shrinkage and strength appear to be non-linear, moreover
328 later age chemical shrinkage results did not show an improvement of the correlation to the 28 days relative
329 strength. The rather low CV indicates fair reproducibility of the results. The chemical shrinkage measurement
330 apparatus is inexpensive, however correct execution of the measurement requires experience. Notably the
331 loading of the containers with the paste is difficult and may strongly affect the results.

332

333 Only tests based on the R³ system gave good performance across the whole range of SCMs investigated.
334 Standardized methods conceived for pozzolans perform poorly when slag is included (Frattini, modified
335 Chapelle test). Some standardized methods, e.g. reactive silica, did not show correlation to the benchmark
336 strength development.

337

338 5 Conclusions and perspectives

339 This paper reports on an extensive multi-laboratory evaluation of SCM reactivity test methods carried out as
340 part of the work of RILEM TC-267 TRM.

341 When taking all SCMs into consideration, all standardized methods showed poor correlation to the benchmark
342 of 28 days relative strength. In contrast, the R³ model calorimetry and bound water tests were able to give
343 acceptable correlations, i.e. $R^2 > 0.85$. When slags are excluded, the correlation of the Frattini test results
344 becomes acceptable as well. The IS 1727 test is the only method that gave reasonable correlation to the later
345 age 90 days relative strength, possibly because of its longer duration than most other tested methods.

346 The Chapelle showed the worst interlaboratory reproducibility while Frattini test and modified Chapelle test
347 had better reproducibility. The reproducibility of the R³ model tests was the best of all the methods investigated,
348 and can probably be improved by specifying in more detail some critical aspects in the execution of the tests.

349 In the phase 2 work of RILEM TC 267-TRM, the R³ model bound water and calorimetry will be further studied
350 due to their very promising correlations to the relative 28 days compressive strength. Further work will focus
351 on improving the reproducibility of these methods by optimizing the test protocols. Possibilities to reduce the
352 duration and improve correlations with early age strength development for the IS1727 test will also be included
353 in the work for phase 2.

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367 8 References

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- 369 1. Chapelle J (1958) Attaque sulfocalcique des laitiers et pouzzolanes. *Rev Matér Constr* 512:136-145
- 370 2. Raverdy M, Brivot F, Paillère A, Bron R (1980) Appréciation de l’activité pouzzolanique de constituents
371 secondaires. *Proceedings of 7e congrés international de la chimie des ciments, Paris, France*:6-41
- 372 3. Frattini N (1949) Ricerche sulla calce di idrolisi nelle paste di cemento. *Ann di Chim Appl* 39:616-620
- 373 4. Snellings R, Scrivener K (2015) Rapid screening tests for supplementary cementitious materials: past and
374 future. *Mater Struct*:1-15. doi:10.1617/s11527-015-0718-z
- 375 5. Lothenbach B, Scrivener K, Hooton RD (2011) Supplementary cementitious materials. *Cem Concr Res* 41
376 (12):1244-1256. doi:<http://dx.doi.org/10.1016/j.cemconres.2010.12.001>
- 377 6. Scrivener KL, Lothenbach B, De Belie N, Gruyaert E, Skibsted J, Snellings R, Vollpracht A (2015) TC 238-SCM:
378 hydration and microstructure of concrete with SCMs State of the art on methods to determine degree of
379 reaction of SCMs. *Mater Struct* 48 (4):835-862. doi:10.1617/s11527-015-0527-4
- 380 7. Suraneni P, Weiss J (2017) Examining the pozzolanicity of supplementary cementitious materials using
381 isothermal calorimetry and thermogravimetric analysis. *Cem Concr Comp* 83:273-278.
382 doi:<https://doi.org/10.1016/j.cemconcomp.2017.07.009>
- 383 8. Avet F, Snellings R, Alujas Diaz A, Ben Haha M, Scrivener K (2016) Development of a new rapid, relevant and
384 reliable (R3) test method to evaluate the pozzolanic reactivity of calcined kaolinitic clays. *Cem Concr Res* 85:1-
385 11. doi:<http://dx.doi.org/10.1016/j.cemconres.2016.02.015>
- 386 9. Durdziński PT, Ben Haha M, Bernal SA, De Belie N, Gruyaert E, Lothenbach B, Menéndez Méndez E, Provis
387 JL, Schöler A, Stabler C, Tan Z, Villagrán Zaccardi Y, Vollpracht A, Winnefeld F, Zajac M, Scrivener KL (2017)
388 Outcomes of the RILEM round robin on degree of reaction of slag and fly ash in blended cements. *Mater Struct*
389 50 (2):135. doi:10.1617/s11527-017-1002-1
- 390 10. Kocaba V, Gallucci E, Scrivener KL (2012) Methods for determination of degree of reaction of slag in
391 blended cement pastes. *Cem Concr Res* 42 (3):511-525.
392 doi:<http://dx.doi.org/10.1016/j.cemconres.2011.11.010>
- 393 11. Helsel MA, Ferraris CF, Bentz D (2015) Comparative study of methods to measure the density of
394 Cementitious powders. *J Test Eval* 44 (6):2147-2154

- 395 12. Antoni M, Rossen J, Martirena F, Scrivener K (2012) Cement substitution by a combination of metakaolin
396 and limestone. *Cem Concr Res* 42 (12):1579-1589. doi:<http://dx.doi.org/10.1016/j.cemconres.2012.09.006>
397 13. Donatello S, Tyrer M, Cheeseman CR (2010) Comparison of test methods to assess pozzolanic activity. *Cem*
398 *Concr Comp* 32 (2):121-127. doi:<https://doi.org/10.1016/j.cemconcomp.2009.10.008>
399 14. Snellings R, et.al. (2018) TC 238-SCM: Hydration stoppage methods for phase assemblage studies of
400 blended cements – results of a round robin test. *Mater Struct* under review
401 15. Snellings R, et.al. (2018) TC 238-SCM: RILEM TC-238 SCM recommendation on hydration stoppage by
402 solvent exchange for the study of hydrate assemblages. *Mater Struct* under review
403 16. Lothenbach B, De Weerd K (2016) Thermogravimetric analysis. In: *A Practical Guide to Microstructural*
404 *Analysis of Cementitious Materials*. CRC Press Oxford, UK, pp 177-212
405 17. Geiker M (2016) Characterisation of development of cement hydration using chemical shrinkage. *A*
406 *Practical Guide to Microstructural Analysis of Cementitious Materials*:75

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