

- 1 Mini-review
- 2 The use of hydrocolloids in physical modelling of complex biological matrices
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10 1 introduction

11 1.1 Scope

12 Life on earth is water-based, and nature has evolved complex water-based matrix materials,
13 biological hydrogels, that are highly adapted to support cellular functions(Baranova, Attili, Wolny, &
14 Richter, 2011). Found within cells and tissues(Chirila & Hong, 1998; Frantz, Stewart, & Weaver,
15 2010; Lieleg & Ribbeck, 2011; Oyen, 2014), between cells and the external environment(Corfield,
16 Carroll, Myerscough, & Probert, 2001; Flemming & Wingender, 2010; Fudge, Levy, Chiu, & Gosline,
17 2005; Lillehoj & Kim, 2002) or at cell surfaces(Authimoolam & Dziubla, 2016; Baranova, Attili, Wolny,
18 & Richter, 2011; Button, et al., 2012; Hattrup & Gendler, 2008) (see figure 1), these matrices are
19 highly diverse both in composition and functions but both their highly hydrated nature, and the
20 functional importance of their physical properties are fundamental shared features. Historically,
21 these materials were considered primarily as inert filler materials but the complexity of their
22 functions is becoming ever clearer(Lutolf, 2009; Tatko, 2008), and with this the challenge of
23 understanding these materials and their molecular components grows. As with many complex
24 systems the use of models to aid their study is widespread, and hydrocolloids, with physical
25 properties similar to natural tissue, are frequent components of such models(Lutolf, 2009).

26 This mini review does not aim to provide a comprehensive overview of the structure and function
27 biological hydrogels or hydrocolloid based models of these materials, but rather to illustrate the
28 challenges they pose to scientific study and, through the use of specific examples, demonstrate how
29 hydrocolloids can contribute to the modelling, and hence better understanding, of these systems.

30 1.2 Examples of biological hydrogels

31 A number of biological hydrogels, their central molecular components and functions are presented
32 in table 1. Many classes of biomolecules are represented in these hydrogels but typically the long
33 range order in the hydrogel matrix is primarily provided by polysaccharides, fibrous proteins,
34 proteoglycans and high molecular weight glycoproteins or a combination of these, with other
35 smaller biomolecules contributing both physically and biochemically to the total functionality of the
36 material. Some biological hydrogels, such as the cumulus cell-oocyte complex matrix(Camaioni,
37 Salustri, Yanagishita, & Hascall, 1996) or the membrane bound mucin layer(Hattrup & Gendler,
38 2008), are surface bound and in this case the cell membrane has a central structural role to play in
39 maintaining matrix architecture. Cell surface structures and membrane bound molecules also
40 interact with secreted hydrogel matrices with both anchoring and signalling functions(Flemming &
41 Wingender, 2010; Frantz, Stewart, & Weaver, 2010; Hattrup & Gendler, 2008; Migliorini, et al., 2014)
42 and thus may contribute significantly to the overall properties of the material. This intimate
43 relationship between the cell and the environment adds further complexity to already complex
44 systems. Within these cell plus matrix systems the cells may exist as a structured part of an
45 organised tissue(Frantz, Stewart, & Weaver, 2010), or as individual cells acting in a cooperative way
46 as for example in a bacterial biofilm(Flemming & Wingender, 2010; Parsek & Fuqua, 2004). Figure 1
47 presents a graphic representation of A, the secreted mucus matrix and membrane bound mucins of
48 the glycocalyx that protect mucosal epithelia from the external environment, B, extracellular matrix
49 between cell types within a tissue and C, the cells and extracellular polymeric substance of a
50 bacterial biofilm, illustrating these various situations.

51 1.3 studying biological hydrogels

52 There are many motivations to study biological hydrogels; to investigate how they function and what
53 molecular elements are important for their function, to study how we can control, improve or
54 disrupt their function (eg medical applications), to understand how they contribute to a 'bigger
55 picture' in nature (eg biofilms in ecosystems and whole organ function), or to copy their functionality
56 in synthetic or semi synthetic systems (biomimetics).

57 However there are multiple challenges in studying biological hydrogels. They are highly dynamic
58 systems both in terms of molecular turnover and in terms of environmental
59 responsiveness(Baranova, Attili, Wolny, & Richter, 2011), and are highly susceptible to both enzyme
60 and bacterial induced degradation, so for example the quality of *ex vivo* matrices tends to decline
61 quite rapidly limiting the opportunities for study. It is often challenging to understand structure
62 function relationships in biological hydrogels. Firstly, a single matrix often both contains multiple
63 components and performs several functions, making it difficult to clearly assign a functional role to a
64 single component. Secondly, structurally differing matrices may perform very similar functions,
65 suggesting the structural requirements for such functions are broad(Lieleg & Ribbeck, 2011),. Finally,
66 for classes of materials (such as extracellular matrix, mucus or bacterial biofilms) that are produced
67 by multiple cell types or tissues the specific physical, topological and biochemical composition of the
68 matrix is both source specific and heterogeneous,(Authimoolam & Dziubla, 2016; Flemming &
69 Wingender, 2010; Frantz, Stewart, & Weaver, 2010) further complicating the elucidation of structure
70 function relationships. Indeed, simply identifying and quantifying to biomolecules present in these
71 complex matrices may be extremely challenging with analysis tools performing less well in mixed,
72 non-dilute systems and extraction procedures resulting in damage, selective loss or concentration of
73 component molecules even in well understood systems such as the alginate matrix of
74 seaweeds(Hernandez-Carmona, McHugh, Arvizu-Higuera, & Rodriguez-Montesinos, 1998; Vauchel,
75 Kaas, Arhaliass, Baron, & Legrand, 2008).

76 Given the inherent complexity of the native materials, there is a clear role for model matrices to
77 enable and assist research efforts. When the hydrogel is formed from surface anchored biopolymers
78 it can be modelled by grafting the appropriate constituents onto solid supports(Authimoolam &
79 Dziubla, 2016; Migliorini, Thakar, Sadir, Pleiner, Baleux, Lortat-Jacob, Coche-Guerente, & Richter,
80 2014) but this approach is not sufficient for reproducing bulk hydrogel materials. As previously
81 stated, hydrocolloids are frequently used in models of biological hydrogels due to their physical
82 similarities to native tissue(Lutolf, 2009). In some cases the hydrocolloids used in models are native
83 constituents of the matrix being modelled, such as alginate based models of bacterial
84 biofilms(Schmid, Messmer, Yeo, Zhang, & Zenobi, 2008), in other cases hydrocolloids may be added
85 as additional support to other functional matrix molecules(Boegh, Baldursdottir, Mullertz, & Nielsen,
86 2014; Taylor, Pearson, Draget, Dettmar, & Smidsrod, 2005), alternatively, studies may exploit the
87 similarities between hydrocolloid gels and the native biological hydrogel to replicate a single
88 properties such as viscosity(Hasan, Lange, & King, 2010) or water binding(Covington, Gardner,
89 Hamilton, Pearce, & Tan, 2007) to investigate the influence of biological hydrogels on other
90 functions. Model hydrogels can be highly controlled in terms of composition and behaviour and
91 therefore be used to provide highly reproducible material for high throughput screening
92 studies(Groo & Lagarce, 2014), or to investigate the influence of individual components on bulk
93 properties(Stewart, Ganesan, Younger, & Solomon, 2015). This level of control also makes model

94 hydrogels useful for validating tools that may be of use for studying native biological
95 hydrogels(Ntarlagiannis & Ferguson, 2009). The use of hydrocolloids in the modelling of biological
96 hydrogels will now be explored for two biological hydrogel matrices, mucus and bacterial biofilms,
97 by considering the nature of the native hydrogel, the specific challenges associated with their study
98 and exploring examples of hydrocolloid use in modelling these systems.

99 2 Use of hydrocolloids in mucus models

100 2.1 Mucus

101 Mucus is the collective name given to viscoelastic secretions composed primarily of highly hydrated
102 networks of polymeric mucin molecules(Corfield, Carroll, Myerscough, & Probert, 2001; Lillehoj &
103 Kim, 2002). Mucins are high molecular weight glycoproteins with a protein backbone that is heavily
104 O-glycosylated forming a bottlebrush structure where up to 80% of the molecules weight may be
105 carbohydrate. Mucins can be membrane bound or secreted, and the secreted mucins can be further
106 divided into those which are polymeric and those which are secreted as monomers(Corfield, Carroll,
107 Myerscough, & Probert, 2001; Hatstrup & Gendler, 2008; Lillehoj & Kim, 2002). Mucus secretions are
108 widespread in nature being found in organisms as diverse as mammals, snails and corals, but in a
109 mammalian context mucus forms a protective barrier at epithelial surfaces of the body exposed to
110 the external environment, such as the gastrointestinal, respiratory and genitourinary
111 tracts(Authimoolam & Dziubla, 2016; Corfield, Carroll, Myerscough, & Probert, 2001; Lieleg &
112 Ribbeck, 2011; Lillehoj & Kim, 2002). The secreted mucus has diverse roles including
113 lubrication(Authimoolam & Dziubla, 2016; Corfield, Carroll, Myerscough, & Probert, 2001; Taylor,
114 Draget, Pearson, & Smidsrod, 2005) (for example in the GI and female reproductive tracts), an
115 integral role in mucociliary transport systems(Button, Cai, Ehre, Kesimer, Hill, Sheehan, Boucher, &
116 Rubinstein, 2012; Lillehoj & Kim, 2002) (for example in the lungs), and providing a robust unstirred
117 layer to allow the maintenance of a pH gradient at the gastric mucosal surface. The barrier function
118 of mucus is ubiquitous, helping to reduce the contact of damaging agents, including bacteria and
119 viruses, with the epithelial cell surface. In this context it is worth noting that the membrane bound
120 mucins of the cell glycocalyx provide an even tighter steric barrier than the overlying secreted mucus
121 layer(Button, Cai, Ehre, Kesimer, Hill, Sheehan, Boucher, & Rubinstein, 2012; Hatstrup & Gendler,
122 2008). When we consider the functions performed by secreted mucus it is clear that the physical
123 (viscoelastic) properties are critical to effective function, and that the ideal viscoelastic properties for
124 maintaining the pH gradient in the stomach differ from those needed as part of the mucocilliary
125 clearance system in the airways. Unfortunately there is no clear universal link between mucin
126 structure, either in terms of gene product or glycosylation and the viscoelastic properties of the
127 mucus matrix.

128 2.2 Modelling mucus in drug delivery research

129 Interest in mucus as a barrier in the context of drug delivery has increased significantly over the last
130 decade(Groo & Lagarce, 2014). This increased interest has been driven primarily by an increased
131 interest in nanomedicine and nanoscale drug delivery systems. These drugs have length scales
132 similar to those associated with the bacteria or viruses, entities that mucus has evolved to act as a
133 barrier to. As a result of this, whilst small molecule drugs typically experience a few fold reduction in
134 diffusion in mucus as opposed to water(Larhed, Artursson, & Bjork, 1998) nanoscale drug delivery
135 systems may be effectively immobilised(Groo & Lagarce, 2014) so they are prevented from contact

136 with mucosal cells and drug uptake is not achieved. It is therefore highly important to be able to
137 study the fate of nanomedicines in mucus, in terms of nanoparticle stability and mobility in mucus
138 matrices and nanoparticle uptake in *in vitro* models that include a mucus component(Boegh,
139 Baldursdottir, Mullertz, & Nielsen, 2014), in order to engineer the most functional drug delivery
140 systems. Such comparative studies require mucus matrices which are sufficiently homogeneous and
141 reproducible in their functionality, and available in sufficient quantity to allow conclusions about the
142 relative effectiveness of different drug delivery systems to be drawn. Although small intestinal
143 mucus is available in reasonable quantities from pigs at slaughter, there is significant inter and intra
144 individual variation in the material properties(Boegh, Baldursdottir, Mullertz, & Nielsen, 2014; Groo
145 & Lagarce, 2014) and *ex vivo* small intestinal mucus samples have a tendency to degrade rather
146 rapidly at room temperature due to the high load of digestive enzymes and bacteria within the
147 samples further complicating the picture. For these reasons mucus models have the potential to
148 make a significant contribution in this research area, however such a contribution is dependent on
149 both the quality of the model in its ability to replicate the functions of the native mucus and a clear
150 understanding of the limitations of the model so that the appropriate controls and comparisons to
151 the native mucus are undertaken. In 2014 Groo and Lagarce published a review of mucus models to
152 evaluate nanomedicines for diffusion. In this article they describe a number of different mucus
153 models that have been developed and highlight some of the challenges, with perhaps the most
154 significant being the limited number of mucin preparations available commercially (porcine gastric
155 or bovine submaxillary) and the fact that the commercial extraction process damages the polymeric
156 structure of the mucins reducing their ability to form a viscoelastic network and thereby their ability
157 to replicate the mucus barrier. In an attempt to overcome this problem Boegh and co-workers have
158 developed a cell compatible mucus model, which is like many others based on commercially
159 available porcine gastric mucin with the addition of other mucus components but uniquely includes
160 the addition of the hydrocolloid polyacrylic acid to add long range support to the matrix and better
161 replicate the viscoelastic properties of the native secretion(Boegh, Baldursdottir, Mullertz, &
162 Nielsen, 2014). Whilst polyacrylic acid is not a component of native mucus, diverse mucous systems
163 have be shown to be capable of incorporating other polymers within their matrix, to provide this
164 same kind of long range support, without dramatically altering other functional properties of the
165 material (Bocker, Ruhs, Boni, Fischer, & Kuster, 2016; Taylor, Draget, Pearson, & Smidsrod, 2005;
166 Taylor, Pearson, Draget, Dettmar, & Smidsrod, 2005) so this strategy certainly has a sound
167 background. However, at the time of writing there is no universally accepted mucus model for use
168 in intestinal drug delivery research, and a similar situation applies to mucus models for other
169 mucosal surface.

170 2.3 Modelling mucus in lung clearance systems

171 Mucus has a protective function in the airways as part of the mucociliary clearance system.
172 Turbulent airflow promotes the deposition of inhaled particular matter, bacteria and viruses on to
173 the mucus surfaces of the conducting airways, where they become entrapped in the sticky mucus
174 blanket that covers the epithelium and is propelled upwards and out of the lungs by the beating cilia.
175 In this manner the air reaching the delicate gas exchange surfaces of the alveoli is relatively cleaner
176 than the inhaled air(Lillehoj & Kim, 2002). The lungs also have a secondary clearance mechanism,
177 cough, which comes into play when the baseline mucociliary clearance is overwhelmed either
178 through increased mucus production or viscosity as a result of infection or underlying disease, or as

179 a first line response to inhalation of a high concentration of particulate matter such as dust (Hasan,
180 Lange, & King, 2010; Lillehoj & Kim, 2002).

181 Baseline mucociliary clearance is often compromised in patients undergoing artificial ventilation but
182 physical interventions to improve mucus clearance often have a poor evidence base to establish
183 how, and under what circumstances they improve mucus clearance. Tatkov and Pack (Tatkov & Pack,
184 2011) have taken one such intervention, symmetrical waveform high frequency oscillation (HFO), the
185 effectiveness of which is debated in the literature, and designed a study which combines *ex vivo*
186 mammalian tracheal tissue with its endogenous mucus and the addition of larger volumes of a
187 hydrocolloid gel comprising of polyethylene oxide in phosphate buffer to simulate pathological
188 mucus secretions found in lung diseases such as COPD. By subjecting their *ex vivo* tissue to HFO they
189 were able generate data suggesting that whilst the clearance of the endogenous mucus in the
190 trachea was unchanged by this intervention the clearance of the simulated pathological mucus
191 model improved. In this case the use of a mucus model which was rheologically matched to
192 pathological sputum but homogeneous and stable in its properties allowed conclusions to be drawn
193 using far fewer *ex vivo* trachea than would be required to produce statistically significant data using
194 highly variable and inhomogeneous *ex vivo* sputum samples.

195 Looking at mucus and cough clearance, Hasan and co-workers (Hasan, Lange, & King, 2010) have
196 considered the disease transmission angle and investigated the influence of artificial mucus on the
197 characteristics of airborne bioaerosol droplets generated during simulated coughing. They used
198 locus bean gum, sodium tetraborate and sodium dodecyl sulphate to produce mucus preparations
199 where the viscoelasticity and the surface tension could be controlled by altering the hydrocolloid and
200 surfactant components respectively. They conclude that viscoelastic properties of the mucus but
201 not surface tension significantly influence droplet production, and propose, rather interestingly, that
202 pharmacological interventions to alter mucus rheology (and thus reduce droplet formation) could be
203 given simultaneously with treatment for infectious respiratory diseases to limit transmission of such
204 diseases within the population.

205 2.4 Modelling mucus as a functional element of organ systems

206 The role of mucus in the effective functioning of major organs such as the stomach is well
207 understood but this is not the case for all structures within the body. Dollinger and co-workers have
208 considered the case of the larynx (Dollinger, Grohn, Berry, Eysholdt, & Luegmair, 2014) where the
209 influence of mucus on phonation, that is sound production, is poorly understood despite widespread
210 consensus that mucus plays an important role in voice performance. In their preliminary study the
211 authors used *ex vivo* human larynges and investigated the influence of two hydrocolloid mucus
212 models, linear polystyrene sulfonate and the same polystyrene sulfonate crosslinked with the ionic
213 porphyrin TAPP to form nanoscale networks. A central element of phonation is the mechanical
214 vibration of the vocal folds, which is defined by both the mechanical properties of the tissue and the
215 transfer of energy from the airstream to the tissue. It is reasonable to assume that the boundary
216 layer of mucins/mucus on the vocal folds will affect this transfer, and this preliminary study
217 supported that conclusion suggesting that not only the presence of mucus, but also the structure of
218 the mucus can influence the mechanics of vocal fold vibrations and thus phonation.

219 Gardner and co-workers (Gardner, Covington, Tan, & Pearce, 2007) have taken a biomimetic
220 approach and made use of a polymer systems mimicking mucus, not to understand an *in vivo* organ

221 system but rather to improve the function of an electronic ‘nose’. Taking inspiration from studies
222 demonstrating that the mucus layer that coats the nasal epithelium has partitioning properties
223 similar to gas chromatography contributing to the coding of olfactory information, they have
224 engineered an artificial nose that incorporates a retentive polymer coating in addition to chemical
225 sensors, improving its functionality. In this case the ‘model mucus’ used
226 (Polymonochloroparaxylene C) bears little similarity to the native material and rather falls outside
227 the scope of this article, however it is included here to illustrate the sheer breadth of applications
228 where models for biological hydrogels can potentially be utilised.

229 3 Use of hydrocolloids in bacterial biofilm models

230 3.1 Bacterial biofilms

231 Bacteria have historically studied as pure cultures of dispersed planktonic single cells, but in the
232 environment they are most often found accumulated at surfaces as polymicrobial aggregates with
233 bacterial cells embedded in a matrix of different types of biopolymers collectively referred to as
234 extracellular polymeric substances (EPS)(Battin, et al., 2007; Flemming & Wingender, 2010; Parsek &
235 Fuqua, 2004), and as such it is important to gain an understanding of bacteria within these matrix
236 colonies. These bacterial biofilms pose the same kind of challenges to research as the hydrogel
237 matrices originating from eukaryote cells; they are highly variable and inhomogeneous, isolation of
238 matrix components is difficult, and understanding of the precise role and interactions of biofilm
239 matrix polymers is limited. In fact EPS have been called “the dark matter of biofilms” as a result of
240 the difficulty in analysing them (Flemming & Wingender, 2010). Again, the case for using hydrogel
241 models to increase our understanding of these materials is strong, aided by the fact than some of
242 the EPS polysaccharides such as alginate, gellan and xanthan can be generated by bacteria in
243 bioreactors providing a better source of molecular constituents than is available for many of the
244 eukaryote matrix polymers. Indeed Hellriegel and co-workers have performed a detailed study of
245 gellan, obtained from *Sphingomonas elodea*, gelled under differing polymer and mono and divalent
246 cation concentrations and evaluated it as a physiochemical biofilm model(Hellriegel, et al., 2014).

247 3.2Hydrocolloids models for studying diffusion in biofilms

248 Diffusion in biofilms is of significant interest as the bacteria in biofilms are less susceptible to anti-
249 microbial treatment. Biofilm formation can provide a significant challenge to controlling
250 microorganisms in diverse situations from implanted medical devices to microbial induced clogging
251 and corrosion in industrial processes(Hu, Miyanaga, & Tanji, 2012). By utilizing an agarose gel
252 biofilm model with and without the addition of *Pseudomonas aeruginosa* and a chlorine
253 microelectrode to monitor chlorine concentration Chen and Stewart have elegantly demonstrated
254 that the reaction rate of chlorine with cellular biomass is sufficiently fast that diffusion of chlorine
255 into the biofilm is rate-limiting, providing an explanation for the poor efficacy of chlorine as a
256 disinfectant when used against microorganisms in biofilms(Chen & Stewart, 1996). Hu and co-
257 workers have also utilized agar as a model biofilm to study the diffusion of a bacteriophage. Using
258 *Escherichia coli* as host cells they found that dead host cells significantly slowed phage diffusion, but
259 the addition of the phage to the biofilm model when the host cells were in an exponential growth
260 phase significantly increased phage diffusion(Hu, Miyanaga, & Tanji, 2012). Thus model biofilms
261 show promise in increasing our understanding of the efficacy (or lack of efficacy) of antimicrobial
262 strategies against bacteria in biofilms.

263 3.3 Hydrocolloid models of biofilms for understanding matrix interactions

264 Another strategy for biofilm control is to gain an understanding of cell matrix interactions in biofilm
265 assembly and disassembly. Stewart and co-workers have used native *Staphylococcus epidermidis*
266 biofilms and biofilm models of *S. epidermidis* with chitosan to study the assembly, disassembly and
267 viscoelastic properties of biofilms(Stewart, Ganesan, Younger, & Solomon, 2015). They conclude
268 that thermodynamic phase instability of the extracellular polymeric substances (EPS) drives colloidal
269 self-assembly of biofilms and that pH induced solubilisation of the EPS matrix drives disassembly of
270 the biofilm.

271 3.4 hydrocolloid models for validating use of characterisation techniques in biofilms

272 There is clearly interest in *in situ* characterising the matrix of biofilms and the content and structures
273 of the molecular components, and confocal laser scanning microscopy (CLSM) combined with
274 fluorescent dyes or fluorescently labelled lectins and antibodies is a widespread tool(Flemming &
275 Wingender, 2010). However labelling techniques carry an associated risk of introducing artefacts,
276 making label-free chemical characterisation of nanostructures in biological systems, including
277 biofilms, particularly attractive. Schmid and co-workers have used well defined alginate gels to test
278 the feasibility of tip enhanced Raman spectroscopy as a tool for identifying nanostructures in
279 biofilms and concluded that this method shows some promise(Schmid, Messmer, Yeo, Zhang, &
280 Zenobi, 2008). At a rather different length scale Ntarlagiannis and Ferguson have addressed the
281 problem of identifying environmental biofilms in the subsurface by geophysical methods. They used
282 a controlled alginate gel within a packed column system to mimic a biofilm in the subsurface and
283 evaluate the ability of the spectral induced polarization method to detect the biofilm, concluding the
284 method has potential for use in the field(Ntarlagiannis & Ferguson, 2009).

285 4 Summary

286 Hydrocolloid gels have shown their usefulness in modelling biological hydrogels in a wide range of
287 contexts. However, it is important to emphasise that the limitations of each individual systems must
288 be considered when drawing general conclusions, and these model systems should be seen as a
289 supplement to rather than a replacement for studies on native matrices.

290 5Future prospects

291 Whilst our understanding of biological hydrogels is improving, aided by the use of model systems, it
292 is generally difficult to model the spacial complexity and variability of matrix structures. However,
293 the rise in 3D printing and tissue engineering is giving rise to exciting developments in terms of
294 producing complex biological structures and microenvironments (Hinton, et al., 2015; Zhang, et al.,
295 2016), which could lead to a paradigm shift in the way we model and study biological hydrogels.

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297 This research did not receive any specific grant from funding agencies in the public, commercial, or
298 not-for-profit sectors.

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Biological hydrogel	Typical constituents	Functions	ref
Extracellular matrix	Fibrous proteins (collagens, elastin, fibronectin, tenascin, laminin), proteoglycans and glycosaminoglycans (hyaluronan, aggrecan, perlecan)	Physical scaffolding of tissues, biochemical and biomechanical signalling, selective diffusion barrier	(Frantz, Stewart, & Weaver, 2010)
Respiratory mucus	Secreted mucin (MUC5AC, MUC5B), shed membrane bound mucins (MUC1, MUC4, MUC16), lipids (e.g. cholesterol, phospholipids), proteins (e.g. IgA, anti-microbial peptides), cell debris (DNA etc)	Mucocilliary clearance, epithelial protection, selective diffusion barrier	(Hattrup & Gendler, 2008; Lillehoj & Kim, 2002)
Intestinal mucus	Secreted mucin (MUC2), shed membrane bound mucins (MUC1, MUC3, MUC4), lipids (e.g. phospholipids, ceramides), proteins (e.g. serum proteins, IgA, defensins, trefoil peptides), cell debris (DNA etc)	Epithelial protection, lubrication, selective diffusion barrier	(Corfield, Carroll, Myerscough, & Probert, 2001; Larhed, Artursson, & Bjork, 1998)
Cumulus cell-oocyte complex matrix	Hyaluronan, inter- α -trypsin inhibitor, dermatan sulphate proteoglycans	Cell adhesion and fertilization	(Baranova, Attili, Wolny, & Richter, 2011; Camaioni, Salustri, Yanagishita, & Hascall, 1996)
Vitreous humour	Hyaluronan, versican, collagens, albumin, IgG	Light penetration, mechanical support, diffusion of metabolic solutes	(Chirila & Hong, 1998)
Bacterial biofilm	EPS (extracellular polymeric substances) including neutral and charged polysaccharides, proteins, nucleic acids and lipids	Scaffold for biofilm architecture, surface adhesion, cohesion, signalling	(Flemming & Wingender, 2010)
Hagfish slime	Mucins, protein slime threads	Defence (produced when stressed or provoked)	(Bocker, Ruhs, Boni, Fischer, & Kuster, 2016; Fudge, Levy, Chiu, & Gosline, 2005)

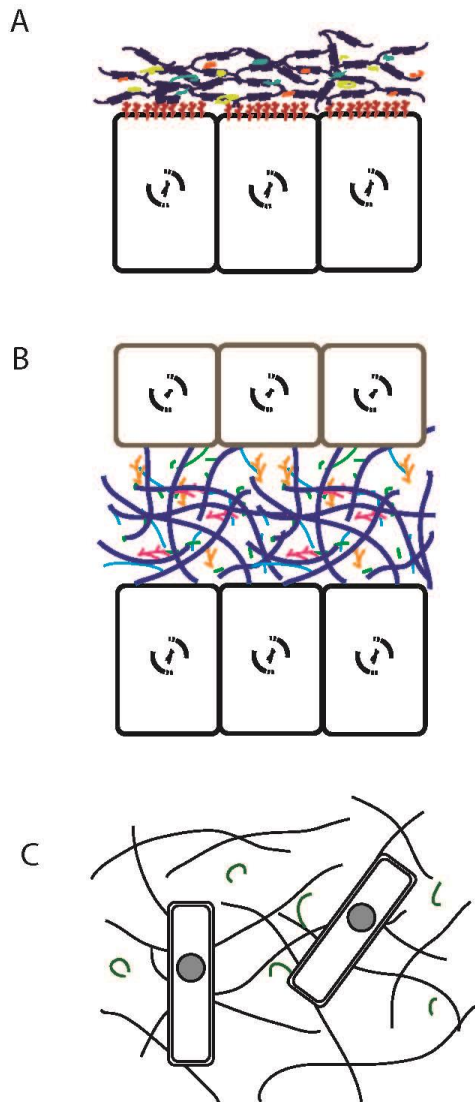
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303 Table 1 examples of biological hydrogels, their components and functions

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307 Figure 1 graphic representation of A, the secreted mucus matrix and membrane bound mucins of
308 the glycocalyx that protect mucosal epithelia from the external environment, B, extracellular matrix
309 between cell types within a tissue containing fibrous proteins and proteoglycans, and C, the cells and
310 extracellular polymeric substance (polysaccharides and proteins) of a bacterial biofilm

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