

# 1 Impact of Production Parameters on 2 Physiochemical Characteristics of Wood Ash for 3 Possible Utilisation in Cement-based Materials

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## 8 **Abstract**

9 Energy production is reorganised to mitigate the pressure on the global environment. This  
10 reorganisation leads to an increase in the production of wood ash (WA). Multivariate modelling  
11 was used to identify the link between production parameters and the physicochemical  
12 characteristics of different WAs and to determine which production parameters result in the WAs  
13 most suitable for utilisation in cement-based materials. Based on the multivariate model partial  
14 least square, WA originating from circulating fluidised bed combustion of wood chips made  
15 from whole trees is the optimal type of WA when utilised as a supplementary cementing  
16 material with pozzolanic activity. WA originating from the combustion of wood chips made  
17 from whole trees is the optimal type of WA when utilised as a supplementary cementing  
18 material with hydraulic activity. Furthermore, the combustion method and type of ash were  
19 seen to have the largest influence on the physiochemical characteristics of WAs compared to  
20 the other production parameters included in this study.

21 *Keywords:* wood ash, supplementary cementitious material, multivariate modelling

## 22 **1. Introduction**

23 The pressure on the global environment has led to an increase in the demand for renewable  
24 energy. In April 2017, 26 of the 28 EU nations, stated they would not invest in new coal-fired  
25 power plants after 2020, in close accordance to the Paris Agreement and the goal to provide

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26 100 % carbon neutral fuel by 2050 (Neslen, 2017). The agreement also includes a demand to  
27 reorganise the existing energy sector. Since coal combustion plants already exist, a prevalent  
28 option is to transform the existing plants to use alternative, sustainable fuels. One abundant  
29 fuel is biomass, defined as an organic material, e.g. plant material such as wood, straw and  
30 bagasse (McKendry, 2002).

### 31 *1.1. Utilisation of WA*

32 Incineration of different types of wood products, e.g. wood pellets or chips, for energy  
33 production causes a significant increase in the production of wood ash (WA). 48,000 tonnes  
34 WA is produced annually in Denmark (Emineral A/S, 2019). A major portion of the produced  
35 WA is land-filled and thus not utilised (Etiégni and Campbell, 1991), which could lead to a  
36 range of potential problems. WA consists of fine particulate matter, which is easily airborne by  
37 winds. Thus landfilling of WA can be a potential health risk to nearby residents or a source of  
38 pollution of the surrounding nature (Cheah and Ramli, 2011a). Further, landfilling of WA can  
39 lead to leaching of chemical elements from the WA and thus contamination of the groundwater  
40 (Udoeyo et al., 2006). To avoid contamination a properly engineered landfill is required,  
41 making landfilling of WA uneconomical (Cheah and Ramli, 2011a). Thus, utilisation of WA  
42 is of interest as an alternative to landfilling.

43 Application of WA in cement-based materials as a supplementary cementitious material (SCM)  
44 or filler has been studied in the literature (Berra et al., 2015; Chowdhury et al., 2015; Elinwa  
45 and Mahmood, 2002; Rajamma et al., 2009a; Ramos et al., 2013; Siddique, 2012; Udoeyo et  
46 al., 2006). The findings of these studies (Berra et al., 2015; Chowdhury et al., 2015; Elinwa  
47 and Mahmood, 2002; Rajamma et al., 2009a; Ramos et al., 2013; Siddique, 2012; Udoeyo et  
48 al., 2006) vary markedly e.g. in regards to the obtained compressive strength for mortar  
49 samples containing WA. Udoeyo et al., (2006), Berra et al., (2015), and Elinwa and Mahmood,  
50 (2002) all concluded a decrease in the compressive strength when using WA as a partial cement  
51 replacement. Rajamma et al., (2009a) saw no decrease in the compressive strength for a 10 %  
52 replacement of cement with WA and Siddique, (2012), Ramos et al., (2013), and Chowdhury  
53 et al., (2015) all concluded an initial decrease in the compressive strength, but an increase in  
54 the compressive strength over time due to pozzolanic activity. These different results are  
55 depending on which type of WA is used. WA physiochemical characteristics vary depending  
56 both on the wood products, e.g. wood chips or pellet, and on the combustion process, e.g.  
57 temperature, and technology (Cheah and Ramli, 2011a; Siddique, 2008).

58 Utilisation of WA in cement-based materials is not approved according to EN 450-1, (2012),  
59 which is the current European standard for utilisation of fly ashes as mineral admixtures in  
60 cement-based materials. Nevertheless, EN 450-1, (2012) will be used as a reference for  
61 evaluation of the WAs in this work, as in several previous studies (Berra et al., 2015; Rajamma  
62 et al., 2009a; Ramos et al., 2013).

### 63 *1.2. Environmental perspectives on utilisation of WA in cement-based materials*

64 About 5 % of the global CO<sub>2</sub>-emission originates from the production of Portland cement  
65 (International Energy Agency (IEA), 2007). Utilisation of WA as a partial cement replacement  
66 could potentially contribute to a reduction of the global CO<sub>2</sub>-emission and conserve raw  
67 materials (Schneider et al., 2011; Teixeira et al., 2016). Utilisation of WA as a partial filler  
68 replacement could also conserve raw materials.

69 Teixeira et al., (2016) conducted a life cycle assessment comparing the potential environmental  
70 impact from cradle-to-gate of 1 m<sup>3</sup> of concrete with Portland cement and 0, 20, 40 and 60 %  
71 substitutions of cement with different types of by-products (coal fly ash, biomass fly ash, and  
72 co-combusted coal and biomass fly ash). They studied cradle-to-gate, as the use and disposal  
73 of concrete were assumed to result in the same environmental impacts regardless the type of  
74 concrete. Teixeira et al., (2016) included six environmental impact categories: global warming,  
75 ozone layer depletion, acidification potential, eutrophication potential, formation potential of  
76 tropospheric ozone and abiotic depletion potential of fossil resources. In general, the  
77 incorporation of fly ash, regardless of type, reduced the environmental impacts, primarily due  
78 to the decrease in the cement consumption, thus the CO<sub>2</sub>-emission. They showed that the  
79 biomass fly ash had the best environmental performance, increasing with replacement rate.

80 The life cycle assessment presented by Teixeira et al., (2016) only takes into account  
81 environmental impacts and does not take into account the mechanical performance of concretes  
82 with biomass fly ash, both parameters that could limit the use. Further, the requirements for  
83 binders in cement-based materials, according to EN 450-1, (2012) are comprehensive and  
84 conservative. These requirements ensures the necessary durability of a concrete structure, which  
85 is another important aspect, not taken into account in the life cycle assessment (Teixeira et al.,  
86 2016).

87 The environmental performance of concrete with WA highly encourages utilisation of biomass  
88 fly ash in concrete; however, further studies are required, regarding, e.g. mechanical response,

89 durability, reactivity of WA, influence of WA on the hydration phases and possibilities and  
90 influence of pre-treatment of WA, in order to utilise biomass fly ashes in cement-based  
91 materials.

### 92 *1.3. Multivariate modelling*

93 The use of multivariate statistical methods has been applied for several purposes in the  
94 literature, e.g. by Pedersen et al., (2015) for evaluation of the efficiency of electro-dialytic  
95 removal of heavy metals from polluted harbour sediments and by Christensen et al., (2005) for  
96 evaluation of biodegradation of mineral oil. Voshell et al., (2018) used multivariate statistical  
97 methods to get a better understanding of the origin of the trace elements As, Cd, Cr, Cu, Ni,  
98 Pb, Zn, K, Mg, Ca, and Al in biomass ashes.

99 This work identifies the link between production parameters and the physicochemical  
100 characteristics of WA by the use of multivariate statistical analysis. The production parameters  
101 were selected by the authors and cover both material and process parameters (see table 1.). The  
102 optimal production parameters for WA to be applicable in cement-based materials were  
103 identified based on the assumption that requirements described in EN 450-1, (2012) and EN  
104 197-1, (2011) are valid for WA. The production parameters used in this work for multivariate  
105 statistical analysis is; the initial water content (of the biofuel), the mean combustion  
106 temperature, type of biofuel (wood chips, wood pellets or wood chips and powder), origin of  
107 biofuel (whole trees (including logs, bark, buds and pine needles) or logs), and combustion  
108 method (circulating fluidized bed (CFB) or grate combustion). The production parameters were  
109 supplied by the individual biomass combustion facility. The type of ash is further included as  
110 a production parameter (bottom (B), mixed (M) or fly ash (F)). The authors selected these  
111 production parameters, based on knowledge available at the individual biomass combustion  
112 facility. However, the inclusion of more production parameters, if possible, would result in a  
113 more detailed model.

## 114 **2. Materials and methods**

### 115 *2.1. Investigated WAs*

116 Eleven different types of WA were used in this study, nine different wood ashes from five  
117 different biomass combustions facilities in Denmark and two different wood ashes from one  
118 biomass combustion facility in Sweden. An overview of the eleven WAs and their production

119 parameters can be found in table 1. Collectively, the eleven WAs used in this study will be  
120 referred to as the investigated ashes.

121 The WAs were sampled from end of March to the beginning of April 2017, except WMFA2,  
122 which was collected at the beginning of July 2017. The individual biomass combustion  
123 facilities supplied samples of the ashes from deposit storage. All ashes were subsequently  
124 stored in closed plastic buckets protected from heat and light sources. The plastic bucket was  
125 mixed before sample collection from the bucket in order to ensure a representative ash sample.

## 126 *2.2. Characterisation methods*

127 Characterisation was made on dried WA (105 °C, 24 hours) in order to assure sample without  
128 free water. Scanning Electron Microscope (SEM) was used to analyse the morphology. The  
129 content of minerals was analysed by X-ray diffractometry (XRD) measured with a  
130 PanAnalytical X-ray diffractometer, sat at the PW3064 Spinner stage, with Cu-K $\alpha$  radiation  
131 measuring between 4 °2 $\theta$  and 100 5 °2 $\theta$  with a step size of 0.002 °2 $\theta$  and a sampling time per  
132 step of 24.8 s. The XRD plots were qualitatively evaluated using X'Pert HighScore Plus  
133 software, with data from the International Centre for Diffraction Data (ICDD). The chemical  
134 composition of the investigated WAs was determined by X-ray fluorescence (XRF). The  
135 particle size distribution and specific surface area were determined by laser diffraction using a  
136 Mastersizer 2000 instrument. The pH and conductivity of the WAs were measured in a 1:2.5  
137 solid to liquid ratio suspension in distilled water with the respective electrodes after 1-hour of  
138 stirring with a magnetic stirrer. The suspension was filtered followed by measurement of Cl<sup>-</sup>  
139 and SO<sub>4</sub><sup>2-</sup> concentrations by Ion Chromatography. Loss on ignition (LoI) was measured in  
140 accordance with CEN (European Committee for Standardization), (2009), i.e. at 550 °C and  
141 EN 196-2, (2005), i.e. at 950 °C. The solubility in water was determined: ash and distilled  
142 water were mixed to an L/S (liquid-to-solid) ratio 5 and shaken for 1 min. After settling, the  
143 water was decanted. This procedure was repeated three times (Kirkelund et al., 2016). Finally,  
144 the suspension was filtered and the ash dried and weighed. The carbonate volume was  
145 determined by reaction with hydrochloric acid measured by the use of Schreiber equipment  
146 (Hamid, 2009). All tests conducted for the characterisation of the WAs were repeated three  
147 times, except the XRF analysis. A ternary diagram was plotted for the components CaO, SiO<sub>2</sub>  
148 and Al<sub>2</sub>O<sub>3</sub>. All excess components were subtracted from the total bulk composition before  
149 plotting. Thus the ternary diagram displays the relationship between the relative quantities of

150 the three components. SIMCA 14.1 Software is used for conducting the multivariate statistical  
151 analysis.

152

### 153 **3. Results and discussion**

#### 154 *3.1. Physiochemical characteristics of WA*

155 The possibility of using the investigated WAs in cement-based materials has been evaluated  
156 based on the physiochemical characteristics measured for the investigated WAs. The  
157 characterisation comprises physicochemical characteristics (tables 2 and 3), percentage  
158 complying with the filler limit and compliance with category N (table 2), mineralogical  
159 composition (table 2 and figure 3) and morphology (figure 4).

##### 160 *3.1.1. Pozzolanic activity*

161 Pozzolanic activity is facilitated by a high amount of SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and Fe<sub>2</sub>O<sub>3</sub>, referred to as  
162 primary oxides, and according to (EN 450, 2012) the  $\sum$ primary oxides > 70% for a pozzolan.  
163 A pozzolan has, in itself, little or no cementitious value but will, in the presence of moisture,  
164 react with calcium hydroxide and form compounds possessing cementitious properties (ASTM  
165 International C125-15a, 2003).

166 Pozzolanic activity was reported for WAs in several studies (Chowdhury et al., 2015; Elinwa  
167 and Mahmood, 2002; Ramos et al., 2013; S. V. Vassilev et al., 2010), concluded due to a  
168 content of primary oxides above 70 % (EN 450, 2012). None of the investigated WAs in this  
169 study complies with this limit. The highest content of primary oxides was found for WCBA  
170 with a content of primary oxides of 64 %. The rest of the WAs had at a content  $\leq$  25 %. The  
171 primary oxides for the WAs in this study are in the following order SiO<sub>2</sub> > Al<sub>2</sub>O<sub>3</sub> > Fe<sub>2</sub>O<sub>3</sub> for  
172 bottom and mixed ashes, except for WPMA, and SiO<sub>2</sub> > Fe<sub>2</sub>O<sub>3</sub> > Al<sub>2</sub>O<sub>3</sub> for fly ashes, except  
173 for WCFA3. The WA originating from CFB combustion (WCBA and WCFA3) arrived at the  
174 highest amount of primary oxides, mainly due to a high content of SiO<sub>2</sub>. A high content of SiO<sub>2</sub>  
175 is due to sand particles, which make the suspension bed, and which are carried with the flue  
176 gas during combustion with the CFB technology (van Loo and Koppejan, 2010). Sand is  
177 considered inert (Wig, 1913), thus a part of the SiO<sub>2</sub> content in WCBA and WCFA3 could be  
178 inert bed sand.

179 The content of primary oxides was above 70 % in the WAs in Chowdhury et al., (2015), Elinwa  
180 and Mahmood, (2002), Ramos et al., (2013) and S. V. Vassilev et al., (2010) and in the order  
181  $\text{SiO}_2 > \text{Fe}_2\text{O}_3 > \text{Al}_2\text{O}_3$ , an order which corresponds to the findings in this study. Rajamma et  
182 al., (2009a) investigated a WA originating from a biomass thermal power plant using forest  
183 residues as fuel. The content of primary oxides was 52 % ( $\text{SiO}_2 = 41\%$ ,  $\text{Al}_2\text{O}_3 = 9\%$ ,  $\text{Fe}_2\text{O}_3 =$   
184  $3\%$ ), but still argued to contribute to the pozzolanic activity, due to the CaO and OH  
185 concentrations based on a direct pozzolanic activity test showing a saturation curve well below  
186 that of cement, which indicates pozzolanic activity of the WA (Rajamma et al., 2009a). The  
187 oxides of the WA investigated by Rajamma et al., (2009a) was similar to WCBA investigated  
188 in this study. Thus, WCBA might similarly show pozzolanic behaviour, even though it contains  
189 only 64 % primary oxides. The remaining WAs of this study are considered to have very little  
190 or no pozzolanic activity, due to the low amount of primary oxides ( $< 25\%$ ).

### 191 *3.1.2. Hydraulic activity*

192 Hydraulic activity describes the ability of a material to set and harden, while submerged in  
193 water, by forming cementitious products in a hydration reaction (Snellings et al., 2012).  
194 Hydraulic activity is governed by the content of  $\text{SiO}_2$  and CaO (EN 197-1, 2011) and  
195 requirements from (EN 197-1, 2011) is  $\text{CaO}/\text{SiO}_2 > 2$ .

196 Hydraulic activity for WAs was investigated in Berra et al., (2015), Cheah and Ramli, (2011b),  
197 and Rajamma et al., (2009a). Berra et al., (2015) investigated the hydraulic index  $K_3$ , defined  
198 as  $(\text{CaO} + \text{MgO} + \text{Al}_2\text{O}_3)/\text{SiO}_2$ . Values of  $K_3 > 1$  are an indication of good hydraulic properties  
199 (Berra et al., 2015). All WAs was investigated by Berra et al., (2015) and all WAs in the present  
200 study, except WCBA, have a hydraulic index above 1. Rajamma et al., (2009a) expected  
201 hydraulic reactions of a WA due to a content of CaO above 25%, substantiated by Cheah and  
202 Ramli, (2011b) determining WA to be an active hydraulic binder as it is rich in  $\text{CaCO}_3$  and  
203 CaO. However, none of the WAs in Berra et al., (2015), Cheah and Ramli, (2011b) and  
204 Rajamma et al., (2009a) complies with the normative compositional requirements set by EN  
205 197-1, (2011).

206 A CaO-SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> ternary diagram with the WAs of the present investigation is plotted in  
207 figure 2. All mixed and fly WA are located in the area for SCM with potential hydraulic  
208 properties (high content of CaO, low content of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub>) and WCBA is in the latent  
209 hydraulic area. None of the WAs are located in the sketched areas for fly ash, natural pozzolans

210 or slag. The composition of WCFA3 and WMFA2 are located just outside the sketched area  
211 for Portland cement, which is hydraulic, thus properties alike Portland cement could be seen  
212 for WCFA3 and WMFA2. The ternary diagram shows that all mixed and fly WA comply with  
213 the requirements set by EN 197-1, (2011), i.e. they potential have hydraulic activity. Thus there  
214 is a possibility for using mixed and fly WA as SCM partially replacing cement in cement-based  
215 materials.

### 216 *3.1.3. Filler*

217 Fillers are in principle inert and do not react themselves. However, the addition of WA as a  
218 filler could still positively influence the suitability of WA in cement-based materials. An inert  
219 filler can contribute to the properties of a cement-based material by filling the intergranular  
220 voids between the cement grains in the mixture (Deschner et al., 2012; Moosberg-Bustnes et  
221 al., 2004). An inert filler can also contribute by having a heterogeneous nucleation effect, acting  
222 as nucleation sites for the hydrates in cement, accelerating the hydration reaction and thus  
223 improving the compressive strength development (Lawrence et al., 2005; Moosberg-Bustnes  
224 et al., 2004; Ye et al., 2007).

225 The compliance of WA with the filler limit (250 $\mu$ m) (Herholdt et al., 1985; Moosberg-Bustnes  
226 et al., 2004) and category N described in EN 450-1, (2012) are highly depending on the type  
227 of ash. Only the fly ashes WPFA, WCFA2 and WCFA3, comply with category N (EN 450,  
228 2012), and the same three fly ashes have a > 95 % compliance with the filler limit (table 2).  
229 Thus, three fly ashes WPFA, WCFA2 and WCFA3 can be expected to contribute to a decrease  
230 in porosity and an increase in compressive strength through the filler effect when utilised in  
231 cement-based materials.

232 The ability of fillers to act as nucleation sites depends on the fineness of the particles, the  
233 amount of mineral admixture and the affinity of the filler to cement hydrates related to the  
234 origin of the mineral admixture (Lawrence et al., 2005). Particles > 215 $\mu$ m can be assumed to  
235 be large enough to exclude any heterogeneous nucleation effects (Lawrence et al., 2003;  
236 Neville, 1996). This facilitates the fly ashes, and particularly the three fly ashes WPFA,  
237 WCFA2 and WCFA3, to have a potential larger contribution to the compressive strength  
238 through the heterogeneous nucleation effect, compared to the bottom and mixed WA. Grinding  
239 the WAs may lead to an increase in the WAs possibility for utilisation as filler (Berra et al.,  
240 2015).



241 *3.1.4 Mineralogy and morphology*

242 The mineralogical composition was determined by XRD (table 2 and figure 3). Quartz, arcanite  
243 and calcite were seen to be the predominant mineral components for seven out of the eleven  
244 WAs, which is in accordance with Yeboah et al., (2014). Other detected mineral components  
245 were portlandite, sylvite, periclase and lime. WCBA differs from the other WAs by having  
246 only quartz and wollastonite as identified mineral components.

247 From the XRD analysis (coal fly ash included for comparison) in figure 3, the amorphous rise  
248 was observed from approximately  $15^{\circ}2\theta$  to  $35^{\circ}2\theta$ . A broad amorphous peak is common in  
249 XRD studies of coal fly ash due to the poorly ordered atomic structure of the amorphous glass  
250 content of the coal fly ash (Bellotto et al., 1989; van Roode et al., 1987; Yeboah et al., 2014),  
251 as seen in figure 3. For all the investigated WA no such peak was observed, thus they contain  
252 a very limited (if any) amount of amorphous glass.

253 The lack of amorphous glass is supported by the SEM images, figure 4, for the investigated  
254 WA. The SEM images, display all the investigated WA to contain none of the typical glassy  
255 aluminosilicate spherical particles, found in commercial coal fly ash (Yeboah et al., 2014). The  
256 SEM images, display the investigated WA to consist of large and fibrous wood particles,  
257 originating from the biomass (Yeboah et al., 2014). Low content of aluminosilicate leads to a  
258 low pozzolanic activity (Shearer, 2014), which is in accordance with the previous findings,  
259 stated that pozzolanic activity is expected for the investigates WAs, except for WCBA, based  
260 on the content of oxides.

261 *3.2 Multivariate analysis*

262 A statistical analysis of experimental data sets can establish trends and correlations in a system,  
263 e.g. assessing the possible relations between the inputs and outputs. In a traditional statistical  
264 analysis, e.g. multiple regression analysis, the input variables are assumed independent, which  
265 can result in biased results of an analysis, if the input variables are correlated (Pedersen et al.,  
266 2015). Biased results is, e.g. the case for the chloride content and solubility in water of WAs,  
267 which are co-dependent (Wang et al., 2001). Statistical analyses coping with possible  
268 collinearity between the variables are the multivariate methods principal component analysis  
269 (PCA) and partial least squares (PLS) regression, which in addition provides plots of the data  
270 compressed to fewer dimensions than the original dataset (Pedersen et al., 2015).

271 The production parameters for each of the collected WA are given in table 1, defined as the X-  
272 matrix and the measured, individual physicochemical characteristics of the WA are given in  
273 tables 2, 3 and 4, defined as the Y-matrix. Combined these tables represent the data set used  
274 for the multivariate analysis.

### 275 *3.2.1. Evaluation of physicochemical characteristics of WA by PCA modelling*

276 PCA is a statistical procedure for identifying differences and similarities in multivariate data.  
277 PCA modelling reduces the dimensions of multivariate data by an orthogonal transformation  
278 of the variables into a set of linearly uncorrelated variables, referred to as principal components  
279 (Voshell et al., 2018). This transformation entails the first principal component to account for  
280 as much of the variability from the original data set as possible, and each of the following  
281 principal components has the highest variance possible under the constraint that the principal  
282 component is orthogonal to the preceding principal component. A detailed description can be  
283 found in Jackson, (1991). A loading plot has been obtained by projection of the original  
284 variables onto the principal components. The Loading Scatter Plot is used to interpret the  
285 relationships between the original variables. In this work, the original variables used for PCA  
286 modelling is the physicochemical characteristics of the WAs (Y-matrix), and the obtained  
287 Loading Scatter Plot can be seen in figure 1. The influence of each of the original variables on  
288 the principal components is reflected by the location of the original variable in the obtained  
289 Loading Scatter Plot. Variables with a strong contribution to the variation are projected far  
290 from the axis centre (e.g.  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$ , figure 1), and variables with minor influenced are  
291 projected close to the axis centre (e.g. carbonate, figure 1). Variables, which are positively  
292 correlated, are found close to each other (e.g.  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$ , figure 1) and variables, which  
293 are negatively correlated, are projected opposite each other with respect to the origin of the plot  
294 (e.g. high compliance with the filler limit and a high mean particle size, figure 1) (Pedersen et  
295 al., 2015). As the physicochemical characteristics of the WA, in this work, are defined as the  
296 Y-matrix, the PCA model is referred to as a PCA-Y model.

297 Expected correlations between the measured physicochemical characteristics of WA can  
298 contribute to the validation of the composed Loading Scatter Plot for the PCA-Y model (figure  
299 1), and subsequently the PLS model. The Loading Scatter Plot (figure 1) displays the principal  
300 components 1 and 2. The PCA-Y model is based in total on five principal components, resulting  
301 in a model with 25 Loading Scatter Plots and 85% of the correlations between the  
302 physicochemical characteristics of WA explained. An example of an expected correlation seen

303 in the Loading Scatter Plot (figure 1) is a high amount of particles, complying with the filler  
304 limit (particle size  $< 250\mu\text{m}$  (Herholdt et al., 1985)) and category N (maximum 40.0% by mass  
305 of the particles retained on a  $45\mu\text{m}$  sieve (EN 450, 2012)). These again correlates with a high  
306 specific surface area (SSA) and a low mean size (D50) of the particles, supported by the  
307 literature.

308 Further, the following is displayed by the PCA-Y model:

- 309 • The composed PCA-Y model (figure 1) displayed the possible pozzolanic activity (defined  
310 as  $\Sigma$ primary oxides (EN 450, 2012)) of the investigated WAs to originate only from a high  
311 content of  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$ , as the WAs have a low content of  $\text{Fe}_2\text{O}_3$ . Further, a high  
312 pozzolanic activity is seen in the PCA-Y model to entails a low content of  $\text{CaO}$  and a low  
313 LoI measured at both  $550^\circ\text{C}$  and  $950^\circ\text{C}$ .
- 314 • From the composed PCA-Y model, the hydraulic activity (defined as the  $\text{CaO}/\text{SiO}_2$  ratio  
315 (EN 197-1, 2011)) of the WAs is seen to increase with a high content of particles complying  
316 with the filler limit (Herholdt et al., 1985; Moosberg-Bustnes et al., 2004), category N (EN  
317 450, 2012), and a high SSA and a low LoI at both  $550^\circ\text{C}$  and  $950^\circ\text{C}$ , and relatively low pH.
- 318 • WA with a high content of particles complying with the filler limit (Herholdt et al., 1985;  
319 Moosberg-Bustnes et al., 2004) is correlated to a high content of  $\text{Fe}_2\text{O}_3$  and  $\text{SO}_4^{2-}$  and a  
320 low LoI measured at both  $550^\circ\text{C}$  and  $950^\circ\text{C}$ .

### 321 *3.2.2. Relationship between production parameters and physicochemical characteristics of WA* 322 *evaluated by PLS modelling*

323 PLS is a multivariate method used for modelling the quantitative relationships between two  
324 data matrices, the descriptor matrix (X-matrix) and the response matrix (Y-matrix) (Wold et  
325 al., 2001). In this work, a PLS model was conducted in order to see how the production  
326 parameters (X-matrix, table 1) influences on the physicochemical characteristics of WA (Y-  
327 matrix, tables 2, 3 and 4). The PLS model is based on four principal components, resulting in  
328 a model with 16 Loading Scatter Plots and 90% of the correlations between the production  
329 parameters and the physicochemical characteristics of WA explained.

330 Table 3 summarises the assessment of whether the production parameters and ash type (X-  
331 matrix) included in the model have positive (+) or negative (-) influence response, thus leading  
332 to an increase or decrease, respectively, on the measurements for the physicochemical  
333 characteristics of WA (Y-matrix).

334 The relative importance of each of the production parameters included in the PLS model can  
335 be described by the variable importance in the projection (VIP) plot. The VIP-plot is  
336 constructed with respect to all the responses (Y-matrix) and the projections (X-matrix).  
337 Production parameters with VIP-values above one are considered to be most relevant for  
338 explaining the physicochemical characteristics of WA (Pedersen et al., 2015). VIP-plot for the  
339 production parameters, included in this work, is shown in table 4. The production parameters  
340 with VIP-values above 1 are the ash types 'B' and 'F', the combustion methods 'CFB' and  
341 'grate', and the biofuel 'chips'. Thus, these production parameters have, according to the  
342 conducted model, the largest influence on the physicochemical characteristics of WA  
343 (response). The production parameters ash type 'M', the types of biofuel 'chips+powder' and  
344 'pellets', origin of biofuel 'log' and 'tree', combustion temperature and initial water content of  
345 the biofuel had VIP-values between 0.5 and 1, indicating a moderate influence on the  
346 physicochemical characteristics of WA (response).

347 The presented PLS model and VIP-values corresponds to the findings of S. V. Vassilev et al.,  
348 (2010) of the combustion technology to have the most significant impact on the properties of  
349 biomass ash (ash types excluded). S. V. Vassilev et al., (2010) further concluded that WA  
350 originating from land clearing wood, pine chips and wood residue results in WAs complying  
351 with the limit for the primary oxides set by EN 450-1, (2012) and WA, originating from elm  
352 bark, olive wood, poplar bark, spruce bark and willow results in the WA complying with the  
353 limit for hydraulic activity set by EN 197-1, (2011). These conclusions corresponds to the  
354 findings in the presented PLS model, where the origin of biofuel is found to have the largest  
355 influence on the composition of oxides.

356 The particle size, governing the amount of WA particles to comply with the filler limit, (table  
357 2), is highly dependent on the type of ash, where fly ash has the smallest particles (Cheah and  
358 Ramli, 2011a). In addition, the type of biofuel and the combustion temperature also influences  
359 the particle size. Wood pellets and high combustion temperature generally facilitates smaller  
360 particles, substantiated by Cheah and Ramli, (2011a) and Lecuyer et al., (1996). This  
361 correlation is due to the link between the combustion technology and combustion temperature  
362 (S. V. Vassilev et al., 2010).

363 *3.3. Evaluation of the relationship between production parameters and physicochemical*  
364 *characteristics of WA for utilisation in cement-based materials by PLS modelling*

365 To identify which production parameters results in a WA suitable in cement-based materials,  
366 the PLS model is compared with the chemical requirements set by (EN 450, 2012) and (EN  
367 197-1, 2011), see tables 2 and 4. The standard (EN 450, 2012) only covers fly ashes, thus the  
368 ash types in the PLS model are neglected in the following.

### 369 *3.3.1 Pozzolanic activity*

370 The following points relate the experimental results to the chemical requirements for fly ash  
371 with pozzolanic properties set by EN 450-1, (2012):

- 372 • CFB combustion and biofuel originating from whole trees processed into wood chips as  
373 biofuel contributes to a high content of SiO<sub>2</sub> and thus a high content of primary oxides.
- 374 • CFB combustion with a low combustion temperature and biofuel originating from whole  
375 trees processed into a combination of wood chips and wood powder as biofuel contributes  
376 to a low content of CaO.
- 377 • Wood chips as biofuel contribute to a low content of MgO.
- 378 • CFB combustion and biofuel originating from wood logs contribute to a low content of  
379 P<sub>2</sub>O<sub>5</sub>.
- 380 • A low combustion temperature facilitates a low content of SO<sub>3</sub>.
- 381 • Biofuel originating from whole trees processed into wood pellets as biofuel contributes to  
382 a low content of Cl<sup>-</sup>.
- 383 • CFB combustion with a low combustion temperature and biofuel originating from wood  
384 logs contribute to a low LoI measured at 950 °C.

### 385 *3.3.2 Hydraulic activity*

386 The following points relate to the chemical requirements for fly ash with hydraulic properties  
387 set by EN 197-1, (2011):

- 388 • Biofuel originating from whole trees processed into wood chips as biofuel contributes to a  
389 high content of SiO<sub>2</sub> and CaO.
- 390 • Wood pellets as biofuel contribute to a high content of MgO. However, the content of MgO  
391 complies with the limit set by EN 197-1, (2011) for both wood pellets, chips and  
392 chips+powder as biofuel.

### 393 *3.3.3. Filler*

394 The following points relate to the chemical requirements for fly ash complying with the filler  
395 limit and category N set by EN 450-1, (2012):

- 396 • A high combustion temperature and wood pellets used as biofuel contribute to a high  
397 amount of WA particles complying with the filler limit.
- 398 • A low initial water content facilitates a WA complying with category N.

399

### 400 *3.4 General discussion on utilisation of WA*

401 Based on the physical and chemical characterization of the wood ashes this study shows that WAs have  
402 potential for utilisation as a SCM in cement-based materials with hydraulic properties. However,  
403 castings of concrete are needed to explore the full potential. To meet industrial use it is necessary to  
404 either reconsider the current standard EN 450-1, (2012) or establish a new standard taking into  
405 account, among others, the hydraulic properties and the alkali content.

406 Besides utilisation as a SCM, WA can be utilised as fertiliser. Spreading of WA and recycling of  
407 nutrients from wood combustion in the forest counteracts the export of nutrients from the forest when  
408 harvesting for energy production, thus creating a closed loop of the nutrients returning to the forest  
409 promoting forest growth (Ingerslev et al., 2011; Pitman, 2006). The suitability of WA as fertiliser  
410 depends on the retained nutrients in the ash e.g. potassium, phosphorous, calcium and magnesium,  
411 which must be relatively high and on the content of heavy metals e.g. arsenic, cadmium, lead, chromium  
412 and nickel, which must meet limiting values (Danish Environmental Protection Agency, 2017;  
413 Ingerslev et al., 2011). The Danish Environmental Protection Agency, 2017; Ingerslev et al.,  
414 2011 also sets limit for the conductivity of the WA of maximum 3600mS/m. The retained nutrients  
415 were not included in this study; however, based on the presented model and the measured conductivity,  
416 bottom or mixed ash and fly ash only originating from CFB combustion are suitable for recirculation to  
417 the forests.

418 It is not an either-or whether WAs can be used in concrete or as fertilizer. Some of the WAs may find  
419 use in concrete or as fertilizer, dependent on their characteristics and local conditions. When considering  
420 utilisation of the raw material WA, the utilization with the largest environmental and economic value  
421 as possible should be chosen.

## 422 **4. Conclusion**

423 In this study, WAs from eleven different plants were characterised and PCA and PLS modelling  
424 were performed in order to assess the linkage between the production parameters and the  
425 physiochemical characteristics of WA. The most important conclusion from the PLS model  
426 was the combustion method and type of ash (B or F) to have the largest influence on the WA  
427 characteristics. Based on the PLS model, a WA originating from CFB combustion of wood  
428 chips made from whole trees (logs, bark, buds and pine needles) at low temperatures are the  
429 most optimal type of WA when utilised as an SCM with pozzolanic activity, as these  
430 combustion parameters facilitate a higher content of primary oxides. A wood fly ash originating  
431 from the combustion of wood chips made from whole trees is the most optimal type of WA  
432 when utilised as an SCM with hydraulic activity. When used as filler, WA originating from  
433 high-temperature combustion with wood pellets used as a biofuel has the best characteristics.  
434 Low initial water content of the biofuel content facilitates the WA to comply with category N,  
435 which is consistent with wood pellets facilitating a low particle size, as wood pellets contain  
436 significantly less water than wood chips and wood chips + powder.

437 None of the investigated fly and mixed ashes complied with the normative compositional  
438 requirements for indications of pozzolanic activity of  $\sum$ primary oxides > 70 % (EN 450, 2012).  
439 Thus all of the investigated WAs have little or no pozzolanic potential, latter, which is primarily  
440 seen for, WAs from grate combustion. The opposite tendency is seen for the hydraulic activity,  
441 were all mixed and fly ashes complied with the normative compositional requirements, thus  
442 being able to set and harden, while submerged in water, by forming cementitious products in a  
443 hydration reaction. Only three of the investigated WAs are found to comply with the filler limit,  
444 thus expected to contribute to the compressive strength through the filler effect.

445

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455

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589 **Table 1.** Production parameters and ash type (X-variables for multivariate statistical analysis)

	Ash type	Combustion method	Initial water content [%]	Type of biofuel	Origin of biofuel	Combustion temperature [°C]
<b>WCBA</b>	Bottom	CFB	24-40	Wood Chips	Whole trees	760-930
<b>WPMA</b>	Mixed	Grate	5-8	Wood Pellets	Logs	800
<b>WCMA1</b>	Mixed	Grate	35-45	Wood Chips	Logs	800-900
<b>WCMA2</b>	Mixed	Grate	35-50	Wood Chips	Whole trees	800-900
<b>WCMA3</b>	Mixed	Grate	40-45	Wood Chips	Logs	800-900
<b>WPFA</b>	Fly	Grate	5-8	Wood Pellets	Logs	1,000-1,100
<b>WCFA1</b>	Fly	Grate	35-50	Wood Chips	Whole trees	800-900
<b>WCFA2</b>	Fly	Grate	40-45	Wood Chips	Logs	800-900
<b>WCFA3</b>	Fly	CFB	25-40	Wood Chips	Whole trees	760-930
<b>WMFA1</b>	Fly	Grate	25-50	Wood Chips and Powder	Whole trees	900-1,000
<b>WMFA2</b>	Fly	Grate	40-45	Wood Chips and Powder	Whole trees	1,000-1,100

590

591 **Table 2.** Chemical composition and physical characteristics of investigated WA (Y-variables for multivariate  
 592 statistical analysis) and mineralogical composition. \* For category C. ± defines the standard deviation.  
 593 Requirements for fly ash utilised in cement-based materials (EN 450, 2012), requirements for hydraulic activity  
 594 (EN 197-1, 2011) and values from literature (Berra et al., 2015; Cheah and Ramli, 2011b, 2011c; Chowdhury et  
 595 al., 2015; Dahl et al., 2009; Elinwa and Mahmood, 2002; Illikainen et al., 2014; Lanzerstorfer, 2015; Peyronnard  
 596 and Benzaazoua, 2011; Rajamma et al., 2009b; Ramos et al., 2013; Siddique, 2008; Udoeyo et al., 2006; Vassilev  
 597 et al., 2014; S. V Vassilev et al., 2010; Yeboah et al., 2014) are added for comparison.

598

	WCBA	WPMA	WCMA 1	WCMA 2	WCMA 3	WPFA	WCFA 1	WCFA 2	WCFA 3	WMFA1	WMFA2	EN 450-1, (2012)	EN 197-1, (2011)	Literary review
<b>Chemical composition (%)</b>														
SiO <sub>2</sub>	57.8	3.4	3.9	4.1	9.4	4.3	3.6	1.7	19.3	5.3	10.3	> 25.0		1.74 – 73.01
Al <sub>2</sub> O <sub>3</sub>	4.9	0.9	1.5	0.9	1.3	0.9	0.6	0.4	3.4	0.9	1.5			0.12 – 28
Fe <sub>2</sub> O <sub>3</sub>	1.1	1.6	0.6	0.7	1.0	2.1	0.9	1.0	2.1	1.1	1.6			0.09 – 27.9
Σ primary oxides	63.8	5.9	5.9	5.7	11.7	7.4	5.1	3.1	24.8	7.4	13.4	> 70.0		
CaO/SiO <sub>2</sub> (-)	3.6	11.1	8.6	19.5	5.7	15.0	13.6	36.2	2.2	6.3	3.1		> 2	
CaO	16.0	37.8	33.6	79.8	53.2	64.4	49.0	61.6	41.7	33.6	31.5	< 10.0		1.16 – 83.46
MgO	1.5	4.1	2.0	2.7	1.5	4.5	2.8	2.2	3.3	0.8	4.0	< 4.0	< 5.0	0.7 – 14.57
K <sub>2</sub> O	8.9	17.0	16.9	9.9	4.3	22.9	10.1	38.5	6.3	10.0	14.2			0.11 – 31.99
Na <sub>2</sub> O	0.0	1.1	1.1	0.7	0.7	0.9	0.8	2.7	1.2	2.6	1.9			0.08 – 29.82
P <sub>2</sub> O <sub>5</sub>	< 0.5	2.7	1.1	3.7	1.1	2.5	3.2	2.5	2.5	1.4	3.0	< 5.0		0.07 – 13.01
TiO <sub>2</sub>	0.2	0.3	0.0	0.0	0.1	0.0	0.0	0.0	0.2	0.1	0.2			0.01 – 1.2
SO <sub>3</sub>	0.2	3.0	0.7	1.2	0.7	6.0	2.5	19.2	4.2	3.5	4.5	< 3.0		0.05 – 12.5
Cl	0.001 ±	0.9 ±	1.0 ±	0.3 ±	0.05 ±	0.5 ±	0.6 ±	21.6 ±	0.5 ±		0.52 ±			
	0.0002	0.03	0.01	0.003	0.005	0.01	0.08	1.2	0.004	1.83 ± 0.2	0.008	< 0.1		0.1 – 1.74
SO <sub>4</sub> <sup>2-</sup>	0.02 ±	4.1 ±	0.43 ±		0.2 ±	5.6 ±	3.4 ±	10.2 ±	1.8 ±					
	0.0	0.1	0.0	1.3 ± 0.0	0.0	0.1	0.4	0.05	0.0	3.36 ± 0.0	4.2 ± 0.1			0.12 – 0.96
LoI, 550°C	0.0 ±	0.7 ±	9.8 ± 1.0	1.9 ± 1.2	1.7 ±	0.5 ±	8.4 ±	1.0 ±	0.8 ±					
	0.0	0.4			0.3	0.3	0.4	0.3	0.0	9.3 ± 1.5	7.2 ± 0.1			< 0.5
LoI, 950°C	0.4 ±	9.9 ±	23.2 ±	19.7 ±	7.6 ±	15.3 ±	25.0 ±	16.7 ±	10.2 ±					
	0.1	0.2	1.5	1.8	0.3	0.3	0.2	0.1	0.1	25.4 ± 1.1	18.3 ± 0.2	< 9.0*		0.66 – 95.9
<b>Physical characteristics</b>														
Mean particle size, D <sub>50</sub> (µm)	184.5 ±	94.5 ±	257.1 ±	65.1 ±	534.7 ±	18.3 ±	61.7 ±	10.5 ±	10.8 ±	273.9 ±	90.2 ±			3.2 – 1,440
	1.8	12.8	29.9	3.6	54.2	0.3	3.9	0.1	0.0	29.4	11.0			

Spread of particle size distribution (µm)	2.5 - 631.0	1.3 - 2,187.8	1.4 - 2,187.8	1.3 - 2,187.8	1.3 - 2,187.8	0.5 - 549.5	1.0 - 2,187.8	0.4 - 478.6	0.6 - 478.6	1.1 - 2,187.8	1.3 - 2,187.8	30 - 600
Specific Surface Area (m <sup>2</sup> /g)	0.04 ± 0.02	0.2 ± 0.02	0.1 ± 0.0	0.3 ± 0.04	0.1 ± 0.01	0.7 ± 0.01	0.4 ± 0.03	1.2 ± 0.01	1.0 ± 0.01	0.1 ± 0.02	0.2 ± 0.02	
pH	11.9 ± 0.0	12.3 ± 0.1	13.0 ± 0.1	12.8 ± 0.0	12.6 ± 0.0	13.1 ± 0.0	13.0 ± 0.1	12.7 ± 0.0	12.7 ± 0.0	12.8 ± 0.0	13.3 ± 0.01	10.10 - 12.5
Conductivity (mS m <sup>-1</sup> )	184.6 ± 7.8	1,894.3 ± 125.0	3,053.3 ± 100.7	2,253.3 ± 196.3	1,067.7 ± 103.8	4,760 ± 535.1	4,980.0 ± 1,496.4	3,976.7 ± 141.9	2,380 ± 10.0	3,950.0 ± 364.3	12,160 ± 317.6	
CaCO <sub>3</sub> (%)	21.4 ± 0.2	27.6 ± 0.5	19.3 ± 0.3	16.0 ± 0.9	12.1 ± 0.4	26.5 ± 3.4	16.9 ± 0.2	21.1 ± 0.3	1.5 ± 0.2	22.2 ± 0.4	16.1 ± 0.3	
Water solubility (%)	0.5 ± 0.4	14.1 ± 0.2	11.8 ± 0.6	9.0 ± 0.5	2.2 ± 0.2	17.9 ± 3.4	8.9 ± 0.5	43.6 ± 1.2	6.8 ± 0.4	14.8 ± 0.3	24.1 ± 1.2	

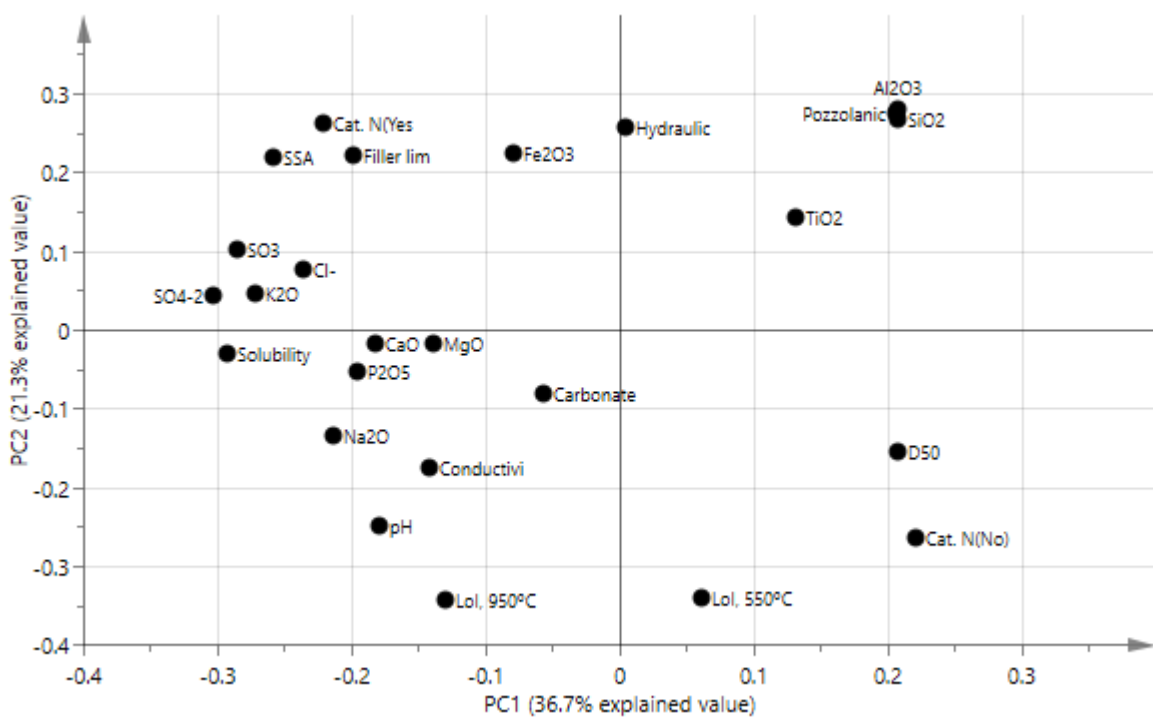
Percentage of WA complying with the filler limit and compliance with Category N (EN 450, 2012) (Yes = comply/No = does not comply)

Percentage complying with the filler limit (<250µm) (Herholdt et al., 1985; Moosberg-Bustnes et al., 2004)	72 ± 0.6	78 ± 4.6	49.7 ± 2.5	72.3 ± 1.9	31.3 ± 2.1	98.2 ± 0.7	65.7 ± 1.0	95.5 ± 0.4	98.6 ± 0.2	48.5 ± 1.9	76.6 ± 4.9	
Complying with category N (EN 450, 2012)	No	No	No	No	No	Yes	No	Yes	Yes	No	No	

Mineralogical compositions of investigated WA determined by XRD analysis. X marks detected minerals.

Quartz (SiO <sub>2</sub> )	X	X	X	X	X	X	X		X	X	X	
Portlandite (Ca(OH) <sub>2</sub> )			X	X	X			X				
Calcite (CaCO <sub>3</sub> )		X	X	X	X	X	X	X	X	X	X	
Sylvite (KCl)		X								X		
Arcintie (K <sub>2</sub> SO <sub>4</sub> )		X	X	X			X	X		X	X	
Periclase (MgO)		X				X	X					
Quick Lime (CaO)						X	X					
Wollastonite (CaSiO <sub>3</sub> )	X											

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603 **Table 3.** Overview of relationships between the production parameters and the wood ash characteristics based  
 604 on the PLS analysis. High values for the wood ash characteristics are achieved by high (+) or low values (-) of  
 605 the given production parameter.

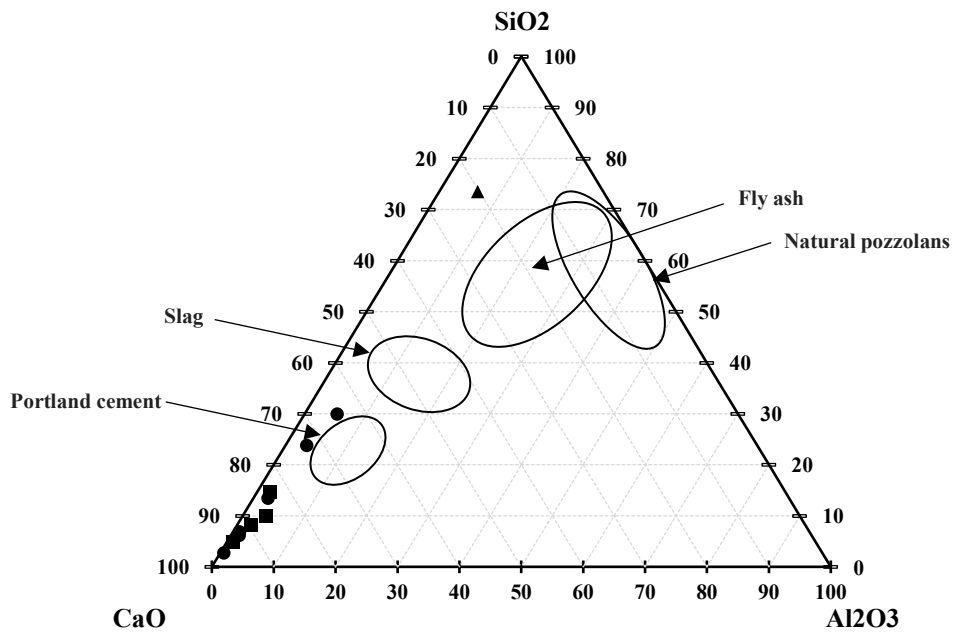
	Ash type: B	Ash type: F	Combustion temperature, middle	Combustion method: CFB	Combustion method: Grate	Type of biofuel: Chips+Powder	Ash type: M	Type of biofuel: Chips	Origin of biofuel: Log	Origin of biofuel: Tree	Type of biofuel: Pellets	Initial water content, middle
SiO <sub>2</sub> (%)	+			+	-			+	-	+		
Al <sub>2</sub> O <sub>3</sub> (%)	+			+	-				-	+		
Fe <sub>2</sub> O <sub>3</sub> (%)							-	-			+	-
Σ primary oxides (%)	+			+	-			+	-	+	-	
SiO <sub>2</sub> /CaO (-)								+	-	+		
CaO (%)	-		+	-	+	-			+	-		
MgO (%)								-			+	
K <sub>2</sub> O (%)	-							-	+	-	+	
Na <sub>2</sub> O (%)	-	+	+			+						
P <sub>2</sub> O <sub>5</sub> (%)	-	+		-	+				-	+		
TiO <sub>2</sub> (%)				+	-				-	+		
SO <sub>3</sub> (%)	-	+	+									
Cl (-)	-							+	+	-	-	
SO <sub>4</sub> <sup>2-</sup> (%)	-	+	+				-	-	+	-	+	
LoI, 550°C (%)						+			-	+	-	
LoI, 950°C (%)	-			-	+	+			-	+		
Mean particle size, D <sub>50</sub> (µm)	+	-	-			+	+	+			-	+
SSA (m <sup>2</sup> /g)	-	+	+			-	-					-
pH	-		+	-	+							
Conductivity (mS cm <sup>-1</sup> )	-	+	+			+						
Carbonate content (%)						-		-	+	-	+	-
Water solubility (%)	-	+	+	-	+	+		-	+	-	+	
Category N (Y = + / N = -)		+					-					-
<250 µm		+	+			-	-	-			+	-

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607 **Table 4.** VIP values of the variables in the X-matrix in descending order.

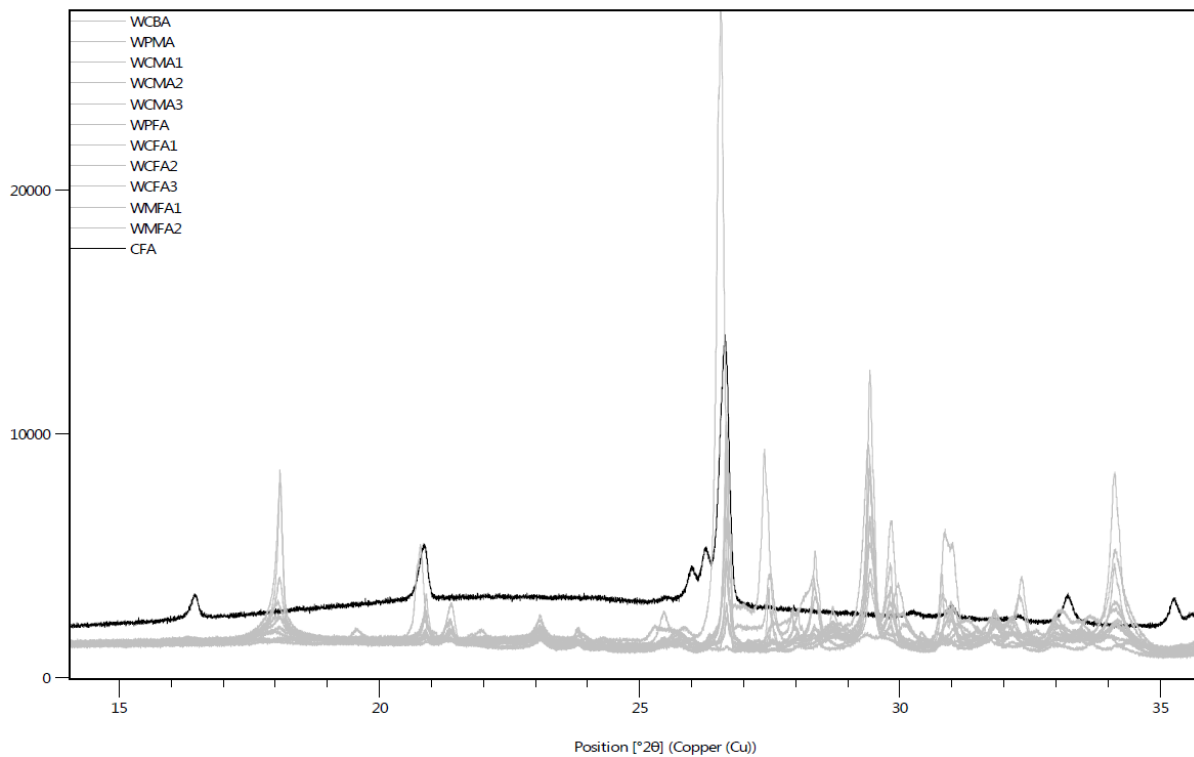
	Variable	VIP-value
608	Ash type: F	1.27
609	Combustion method: CFB	1.16
610	Combustion method: Grate	1.16
611	Ash type: B	1.13
612	Type of biofuel: Chips	1.09
613	Ash type: M	0.98
614	Type of biofuel: Chips+Powder	0.95
615	Initial water content, average	0.95
616	Type of biofuel: Pellets	0.90
	Combustion temperature, average	0.82
	Origin of biofuel: Log	0.70
	Origin of biofuel: Tree	0.70

**Figure 2.** CaO-SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> ternary diagram for the investigated wood ashes. Black triangle ▲: bottom ash, black square ■: mixed ash and black dot ●: fly ash. Ternary diagram adapted from Lothenbach et al., (2011).



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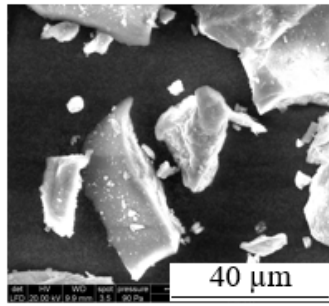
618



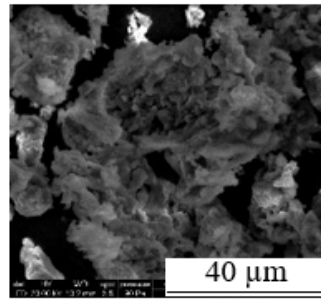
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**Figure 3.** Comparison between the XRD diffractograms from 15°2θ to 35°2θ (amorphous rise) for the investigated WA and a coal fly ash.

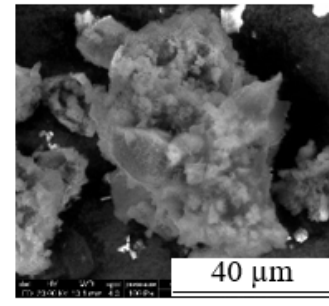




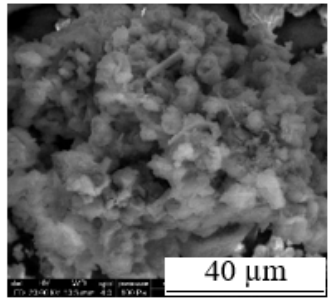
a) WCBA



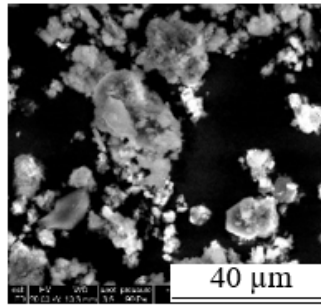
b) WPMA



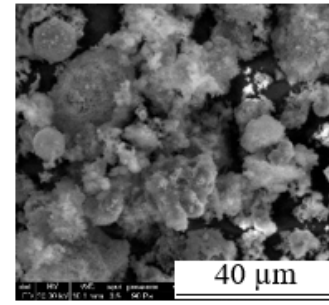
c) WCMA1



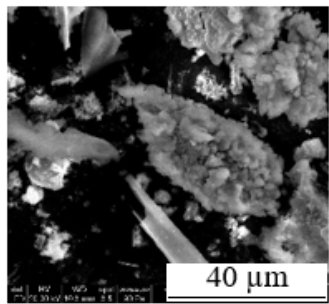
d) WCMA2



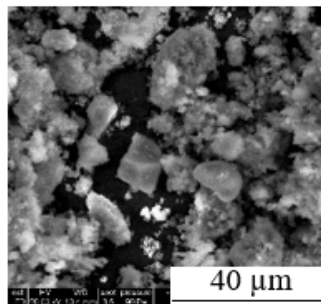
e) WCMA3



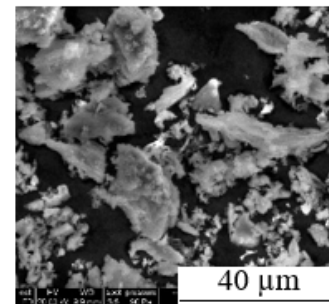
f) WPFA



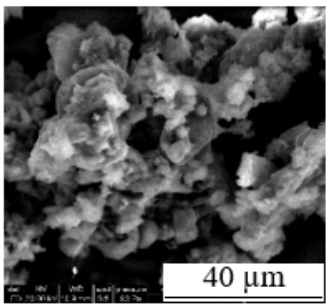
g) WCFA1



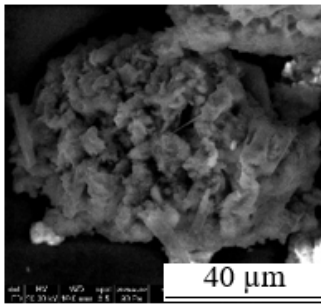
h) WCFA2



i) WCFA3



j) WMFA2



k) WMFA1

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623 **Figure 4.** SEM images displaying the 11 investigated WA.

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